

# Decay heat calculations and safety analyses for ARAMIS-A molten salt reactor

Jad HALWANI – PhD Hours: 03/04/2025  
PhD funded by ISAC project (France 2030)

## Supervisors:

- Lydie Giot (SUBATECH-Co-Director)
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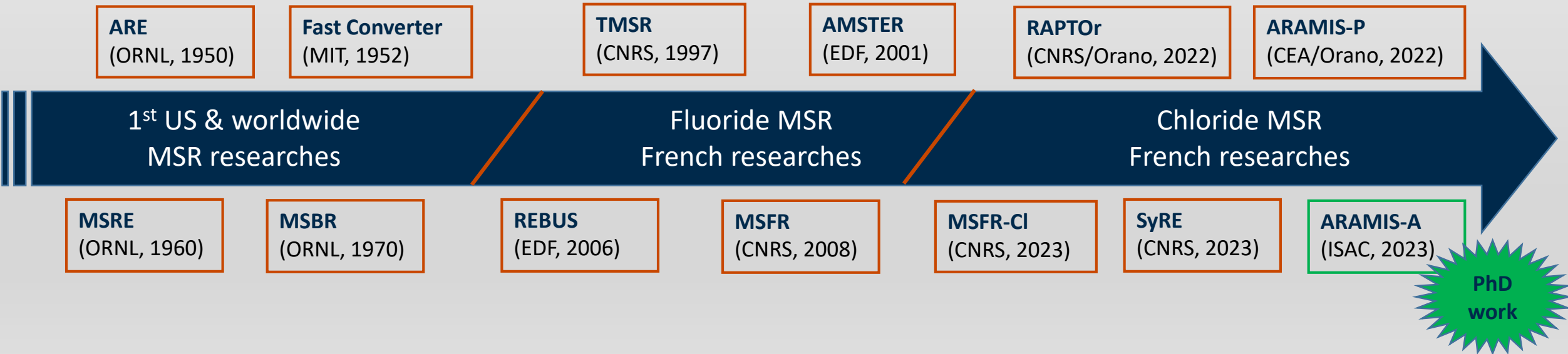
• **What is a Molten Salt Reactor (MSR)?**

- Gen IV fission concept
- Fuel is dissolved in a salt (fluoride / chloride)
- Burner (U, Pu, Actinides) or breeder (Th/U or U/Pu)
- Thermal or fast neutrons

- Fission products accumulation in fuel salt  
-> **Chemical treatments required**

Potential benefits	Key challenges
Actinides transmutation	Salt chemistry
Inherent safety behaviour	Materials and technologies qualification
Flexibility	Safety in operation (design & paradigm)

Summary of MSR potential benefits and key challenges [GIF4]

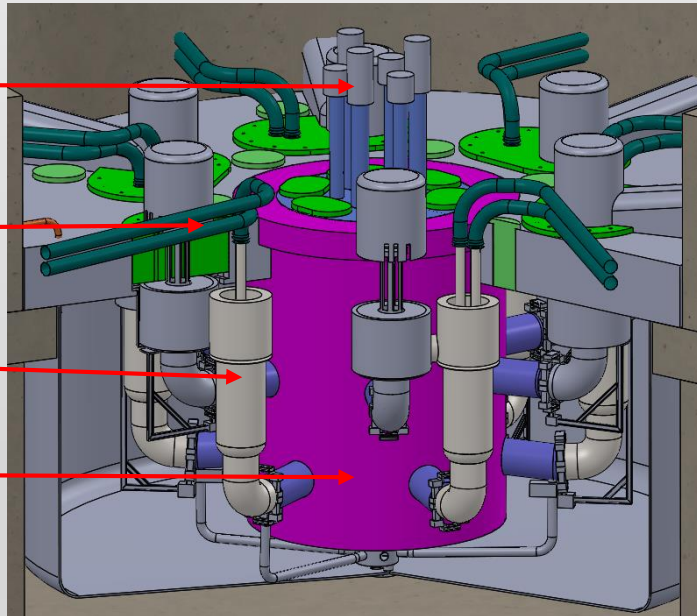


PhD work

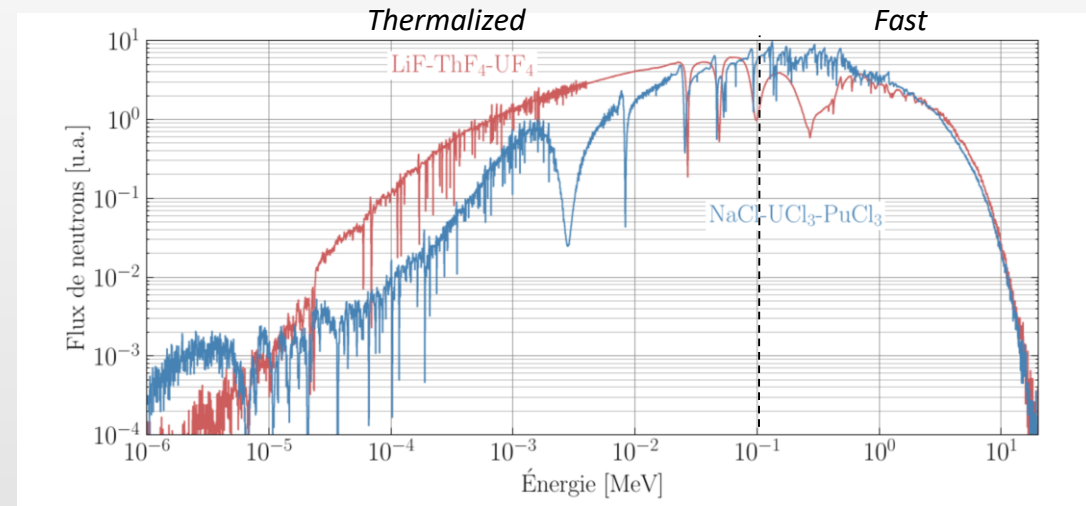
## • Why studying chloride MSR?

- Better Pu and minor actinides (Np, Am, Cm, ...) solubility in chlorides than fluorides salt
- Harder neutron spectrum in chlorides than fluorides salt
- Higher fission cross-sections for Pu and minor actinides

-> **Promising way to burn minor actinides like Am**  
(main radioactivity and decay heat contributor over a century)



*Illustrative schematic view*



*Neutron spectra in fluorides (red) and chlorides (blue) [LMes2023]*

## • ARAMIS-A concept

- 300 MW<sub>th</sub> fast chloride molten salt reactor
- Fuel salt: NaCl-MgCl<sub>2</sub>-(Pu-Am)Cl<sub>3</sub>
- Am burning target: 50 kg/TWh<sub>e</sub> (ADS maximum performance)
- Periodic fuel salt reprocessing to control reactivity
- Hypothesis: continuous gaseous and metallic fission products (FP) removal  
-> **3 source terms !**

## • ISAC project

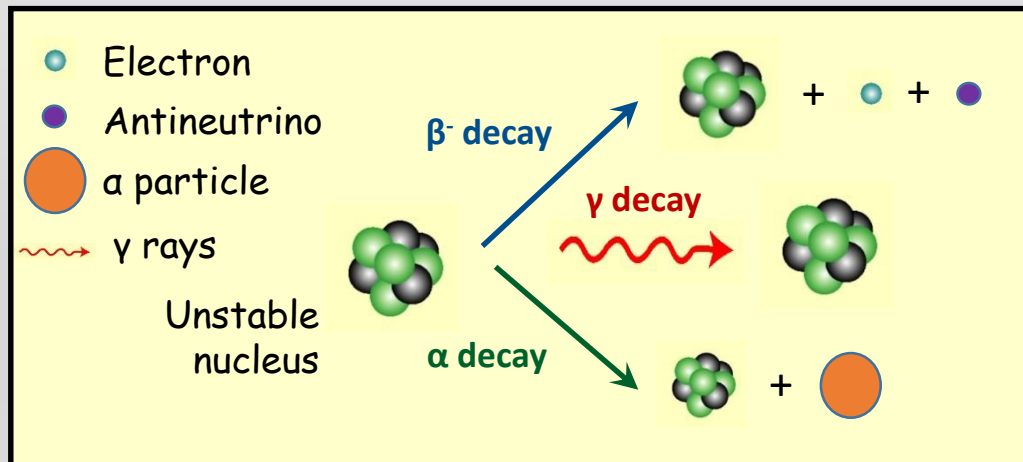
- French collaboration between CEA, CNRS, EDF, Framatome & Orano
- Topics studied: design, maintenance, salt chemistry, materials integrity, nuclear scenarios, **decay heat, safety, ...**

## • What is decay heat (DH)?

- Heat produced by nuclear power plant after shutdown
- **Calculated by summing all  $i$  nucleus contributions to DH**

$$P_{res}(t) = \sum_i (E_{i,LP} + E_{i,EM} + E_{i,HP}) \cdot \lambda_i \cdot N_i(t)$$

- $N_i(t)$ : atoms number
- $\lambda_i$ : decay constant
- $E_{i,LP}$ : light particles mean decay energy (mostly  $\beta^-$  decays)
- $E_{i,EM}$ : electromagnetic mean decay energy (mostly  $\gamma$  decays)
- $E_{i,HP}$ : heavy particles mean decay energy (mostly  $\alpha$  decays)



Concept	Maximum DH (% nominal power)
Pressurized Water Reactor (PWR) [MBro2013]	6%
Sodium-cooled Fast Reactor (SFR) [ACal2020]	7%
Molten Salt Fast Reactor (MSFR) [MBro2013]	5.5%

*DH orders of magnitude*

**-> Inventories have to be calculated during and after reactor operation**

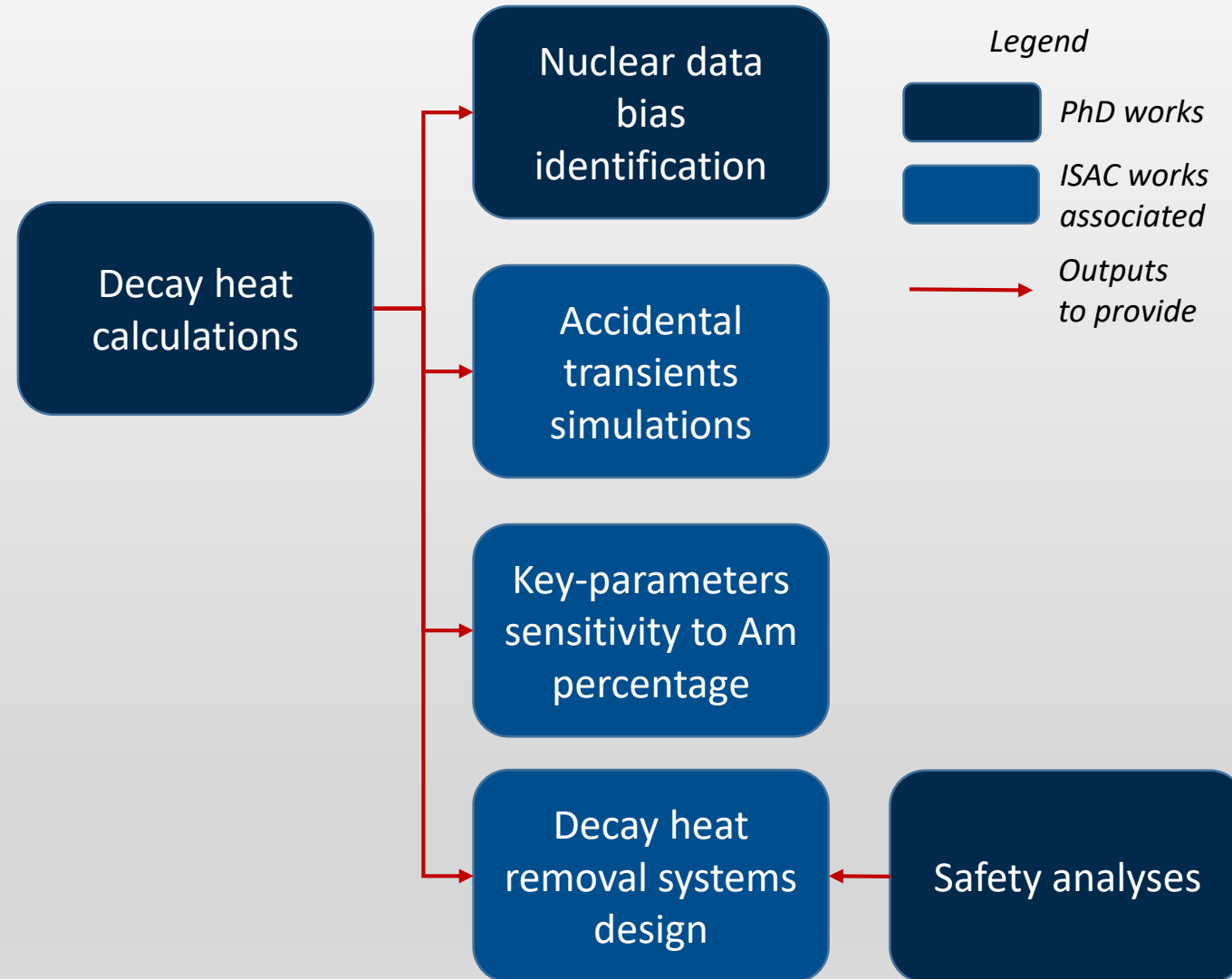
**Cross sections, decay data and mean decay energies uncertainties and bias can have a significant impact on DH**

**-> Biased mean decay energies (Germanium energy efficiency)**

- **What are the motivations of this PhD?**

- Decay heat calculations and safety analyses done for fluoride MSFR [MBro2013] [DGer2019]
- Decay heat calculations done for 3 GW<sub>th</sub> chloride MSFR reactivity-controlled by continuous reprocessing [HPit2023]

**-> Not any decay heat calculations and safety analyses performed for a small fast chloride MSR reactivity-controlled by discontinuous reprocessing**

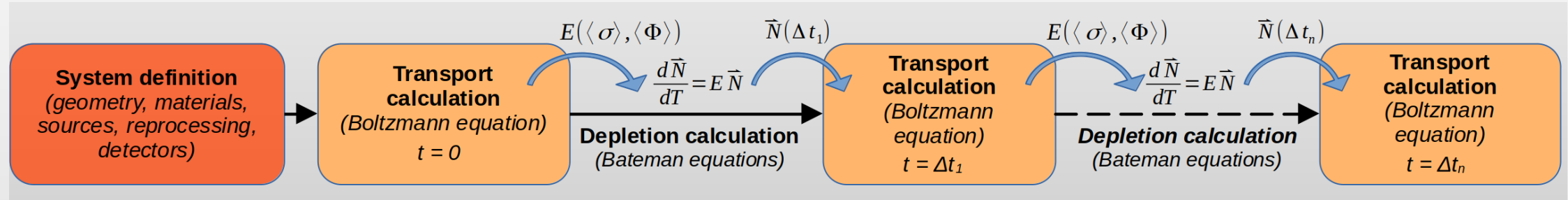


[MBro2013] M. Brovchenko, *Études préliminaires de sûreté du réacteur à sels fondus MSFR*, PhD Thesis, Grenoble-Alpes University, LPSC (2013)

[DGer2019] D. Gérardin, *Développement de méthodes et d'outils numériques pour l'étude de la sûreté du réacteur à sels fondus MSFR*, PhD Thesis, Grenoble-Alpes University, LPSC (2018)

[HPit2023] H. Pitois, *Design and optimisation of a Molten Salt Fast Reactor using chloride salts and uranium cycle*, PhD Thesis, Grenoble-Alpes University, LPSC (2023)

## • How inventories are calculated?



Transport and depletion equations coupling

- Transport (Boltzmann) and depletion (Bateman) equations are solved by SERPENT2 [Lep2015]
- Transport equations are solved with probabilistic (Monte-Carlo) approach
- Depletion equations are solved with CRAM method (order 16)

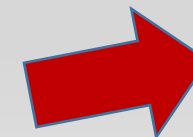
$$\frac{dN_i}{dt}(t) = \underbrace{-\lambda_i \cdot N_i(t) + \sum_{j \rightarrow i} \lambda_j \cdot N_j(t)}_{\text{Terms related to decays}} \underbrace{- N_i(t) \cdot \sigma_i \cdot \phi + \sum_l N_l(t) \cdot \sigma_{l \rightarrow i} \cdot \phi}_{\text{Terms related to reactions}} \underbrace{- \sum_k \lambda_{ext,k} \cdot N_i(t) + \sum_p \lambda_{inj,p} \cdot N_i(t)}_{\text{Terms related to reprocessors (MSR specificity)}}$$

Depletion equation for MSR

At CNRS MSR team, transport and depletion codes used: REM/MCNP (MSR reference code), OpenMC & SERPENT2

- **None of them can simulate reactivity control by periodic batch reprocessing**

**-> A new interface code, coupled with an existing one, is required to simulate new physical processes**



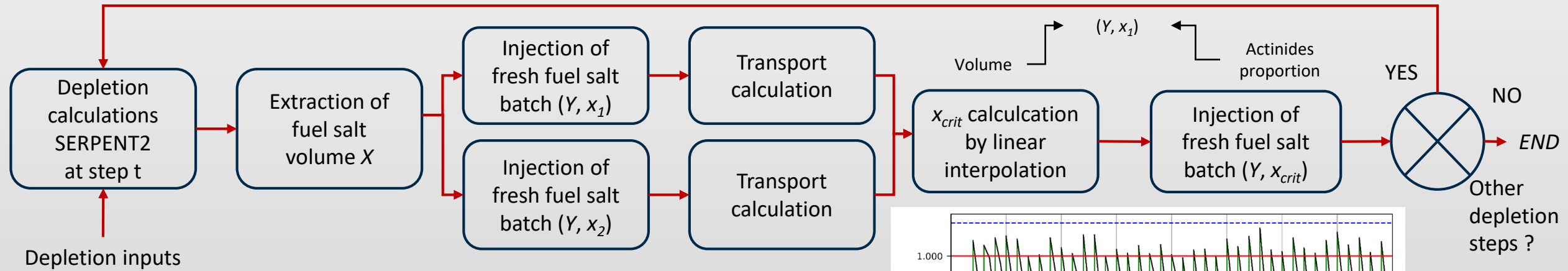
CERESIS logo

## • What is CEREIS (*Contrôleur en Evolution d'un Réacteur par Extraction et Injection de Sel*)?

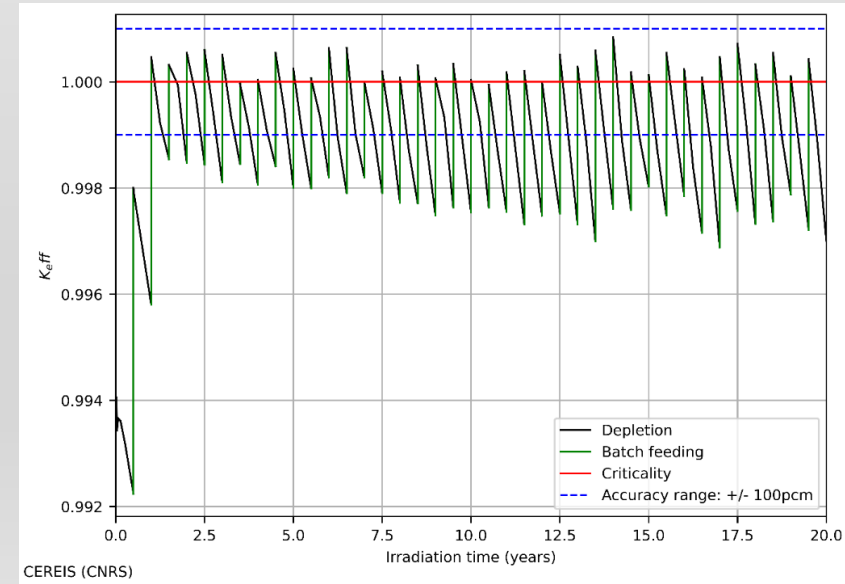
- Python interface code between transport and depletion code (here SERPENT2) and control modules (here of reactivity)
- Development started with Alexis Bodinier in thernship works
- **Hypothesis:  $k_{eff}$  increase is linear proportional to actinides proportion injected**

$$k_{eff} = \frac{\text{Neutrons produced by fission}}{\text{Neutrons lost by absorption and leakage}}$$

Target:  $k_{eff} = 1$  (criticality)

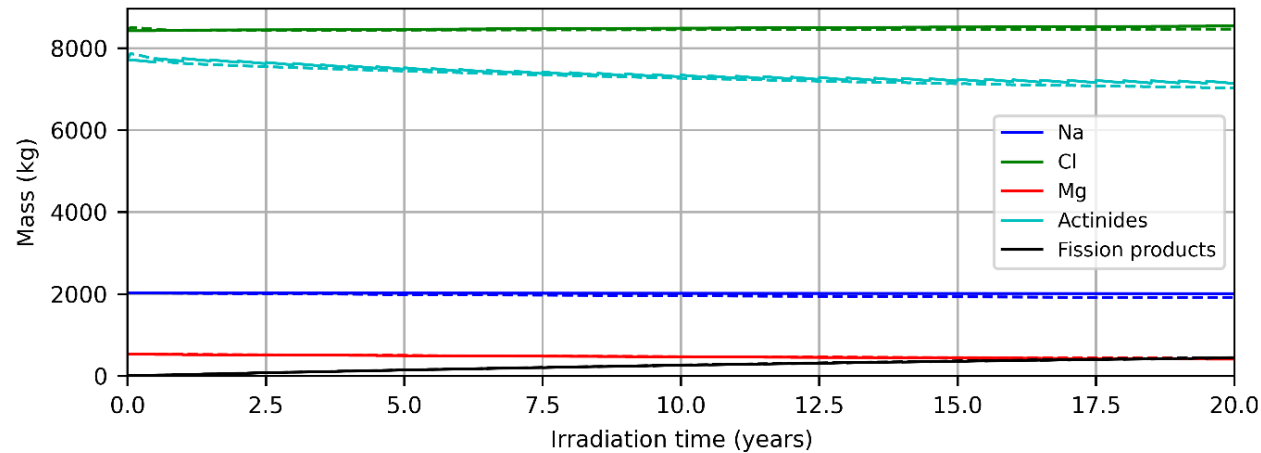


- **Approach precise (+/- 100 pcm) but time-consuming**  
-> **Not suited for operator controls**
- **Alternative approaches under-development**
  - $k_{eff}$  approximation (4 factor-formula + leakage estimation)
  - Sensitivity factor ( $\Delta\%(\text{Pu-Am})\text{Cl}_3/\Delta k_{eff}$ )



$k_{eff}$  evolution of ARAMIS-A

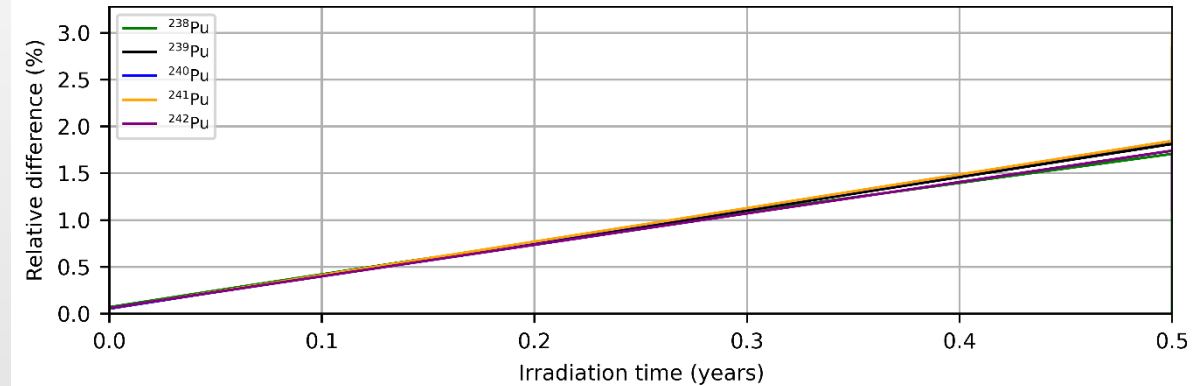
Poster at MIMOSA  
Summer School  
(July 2024)



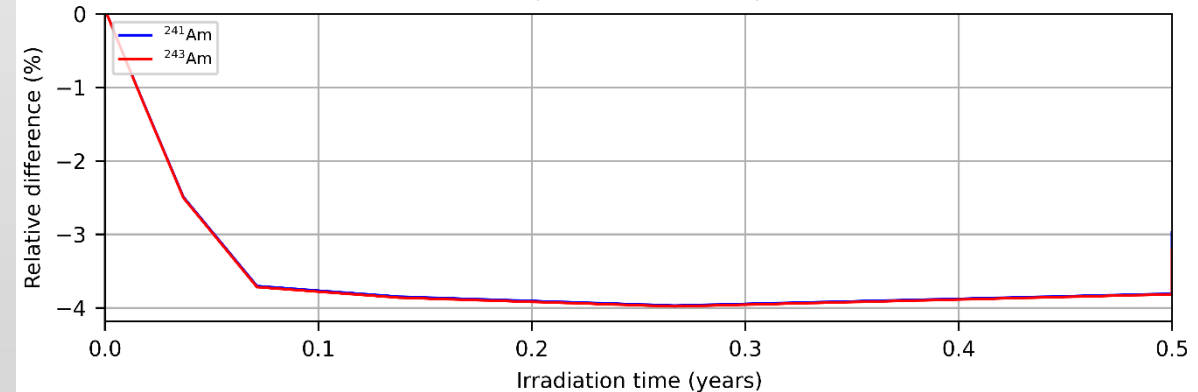
Global inventories evolution (CEREIS with continuous lines and REM with dashed lines)

### Impact of modelling differences under analysis

- Reactivity control process
- Multiplication factor estimator used for reactivity control
- Feeding volume and composition

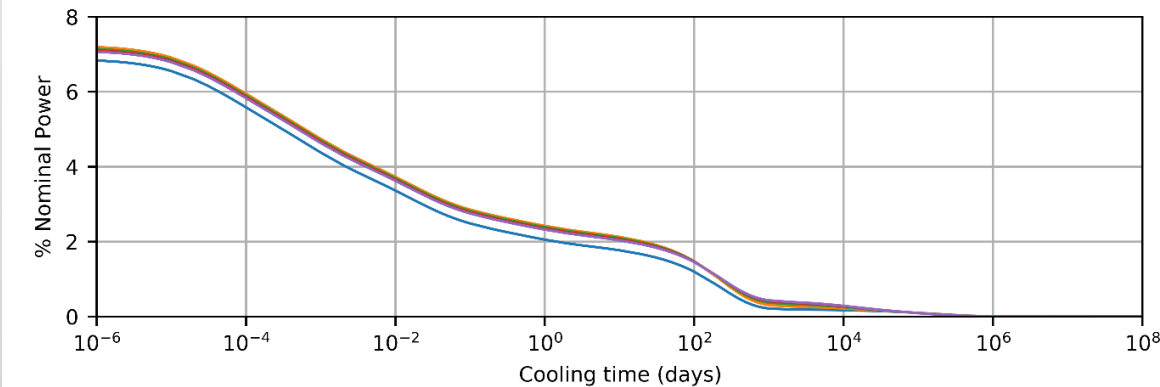
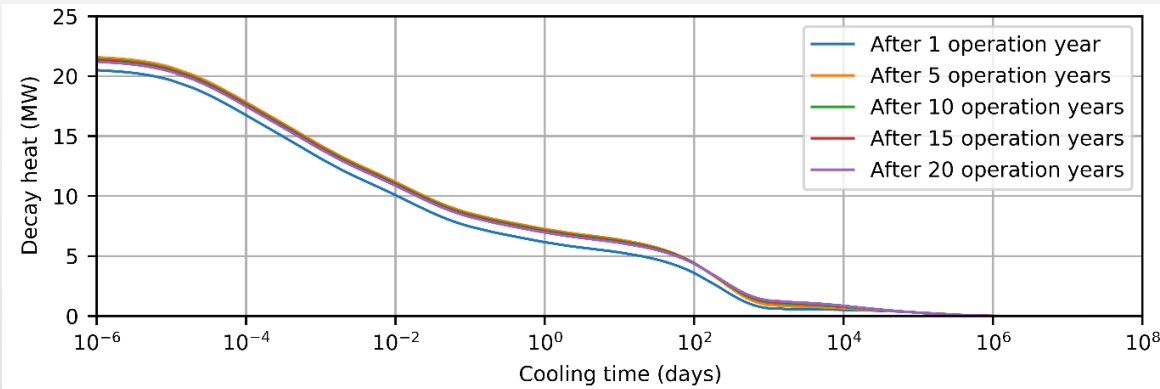


CEREIS/REM discrepancies on  $^{239}\text{Pu}$  and  $^{240}\text{Pu}$



CEREIS/REM discrepancies on  $^{241}\text{Am}$  and  $^{243}\text{Am}$





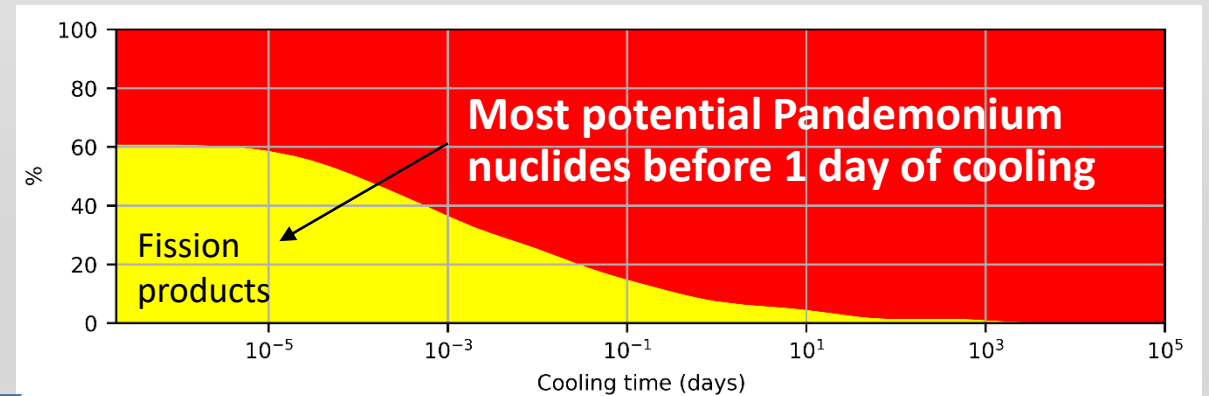
Decay heat from depleted fuel and extracted fission products

**After 20 operation years, decay heat maximum of 21.32 MW (7.11 % of nominal power)**

- 15.43 MW from depleted fuel
- 1.83 MW from extracted gaseous FP
- 4.06 MW from extracted metallic FP

**Significant decay heat increase between 1 and 5 operation years**

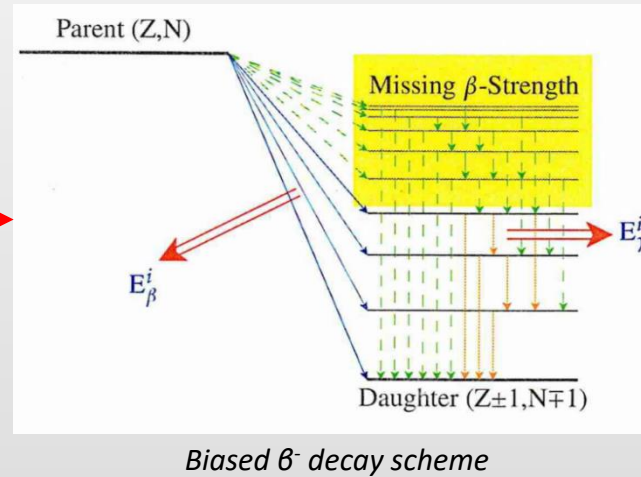
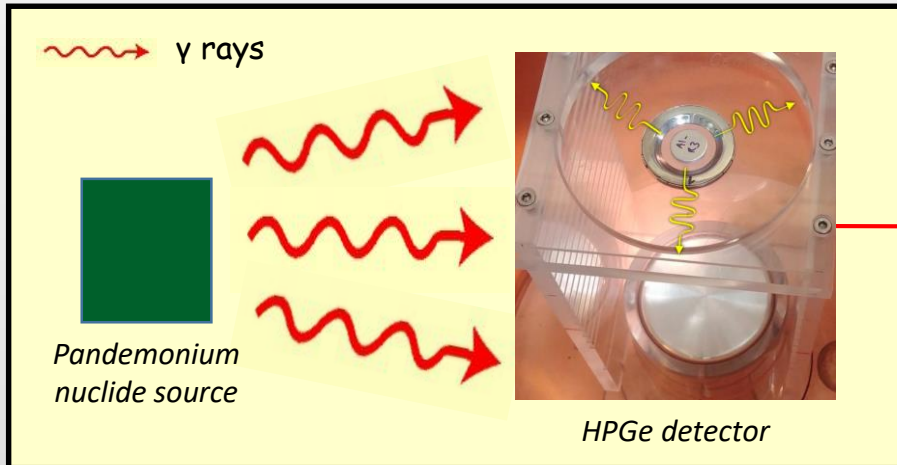
- **Consequence of mid-life actinides accumulation:  $^{238}\text{Pu}$ ,  $^{242}\text{Cm}$  and  $^{244}\text{Cm}$  -> Investigation in progress**



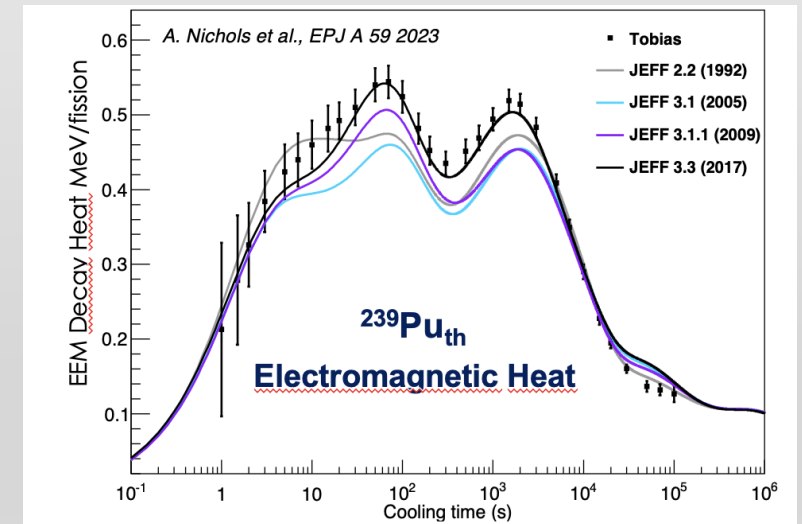
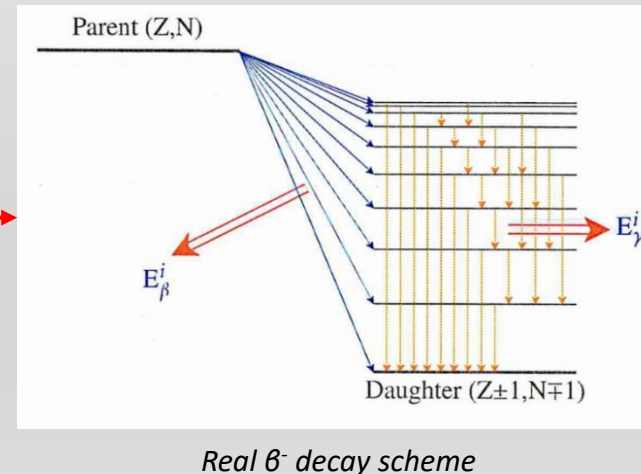
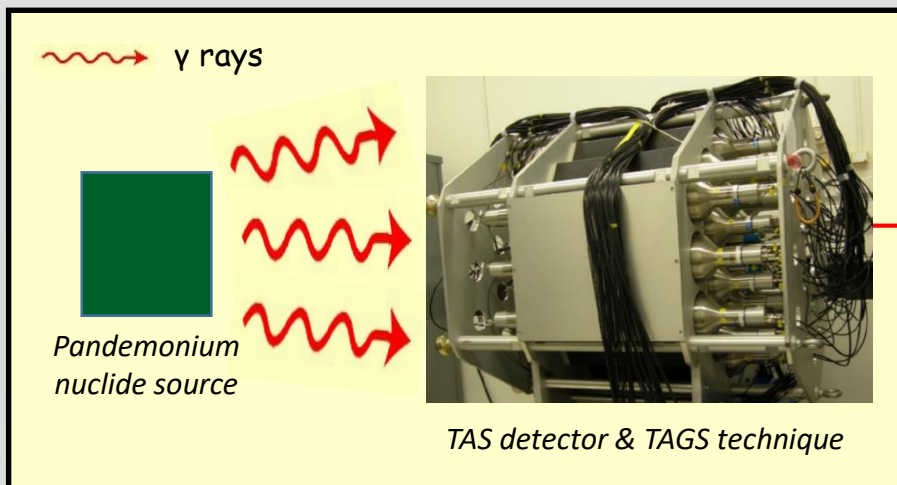
Decay heat repartition between actinides (in red) and fission products (in yellow) in depleted fuel

• **What is the Pandemonium effect?**

- To calculate  $E_{i,LP}$  and  $E_{i,EM}$ ,  $\beta^-$  decay scheme has to be determined
- $\beta^-$  decay scheme is deduced from  $\gamma$  intensities coming from excitation levels measured with high-resolution  $\gamma$  detector, like Hyper Pure Germanium (HPGe)



->  $\beta^-$  feeding at high excitation levels is not detected and is wrongly assigned to  $\beta^-$  feeding at lower excitation levels  
 ->  $E_{i,LP}$  ( $\beta^-$  decays) is overestimated &  $E_{i,EM}$  ( $\gamma$  decays) is underestimated



$E_{i,EM}$  comparison for  $^{239}\text{Pu}$  thermal fission pulse [ANic2023]

Before switching to safety, say « Hi » to Teddy



## • What is nuclear safety?

- Radionuclides are contained by physical **confinement barriers**
- Safety provisions to protect these barriers have to ensure **3 functions at any operation condition and at any time:**
  - Reactivity control (as much neutrons produced by fission than lost by absorption or leakage)
  - Heat removal (from fission and decay heat)
  - Radionuclides confinement
- Safety provisions are positionned according to **defence in depth principle**



Defence in depth schematic view

### Level 1: Normal operation

Monitoring and control provisions to stay there

### Level 2: Incidents

Small impact failure

Monitoring and control provisions to return to level 1

### Level 3: Accidents

High impact failure

Protection provisions to reach safety state

### Level 4: Severe accident

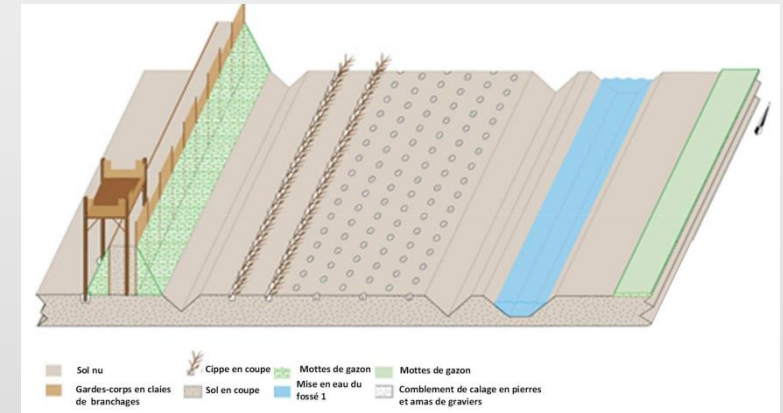
Several high impact failures (melting core for PWR, not identified yet for MSR)

Protection provisions to reach safety state and mitigate consequences

### Level 5: Crisis management

How to avoid to « RUN FOR YOUR LIFE »?

Public plans to protect people and environment



Defence in depth military application for Alesia battle [Vid2010]

**What provisions are required for ARAMIS-A?**

**-> Focus on decay heat removal function**

[ASN2017] ASN/IRSN, *Guide n°22 : Conception des réacteurs à eau sous pression*, 2017

[FBer2018] F. Bertrand, *Méthodologies d'études de sûreté et applications pour la pré-conception de différents types de réacteurs nucléaires de quatrième génération*, HDR, Grenoble Alpes University, CEA (2018)

[Vid2010] J. Vidal et C. Petit, *L'eau sur le site d'Alésia : la contrainte hydrogéologique lors du siège de 52 av. J.-C.*, Revue Archéologique de l'Est, Tome 59-1 | 2010 : Fasc. 1 - n° 181

- **How to conduct safety analyses?**

- Collaborative workshops involving conceptual design and safety specialists among CEA, CNRS, EDF, Framatome and Orano
- Through 4 workshops, **Lines of Defence method** has been applied to identify decay heat removal (DHR) systems requirements

- **What is the Lines of Defence method?**

- Ensure that every abnormal evolution of the reactor into a severe accident (contact between fuel salt and last confinement barrier) is always prevented by a minimum set of homogenous (in number and quality) safety provisions called *lines of defence* (LoD)

- **-> If not, make a feedback to design**

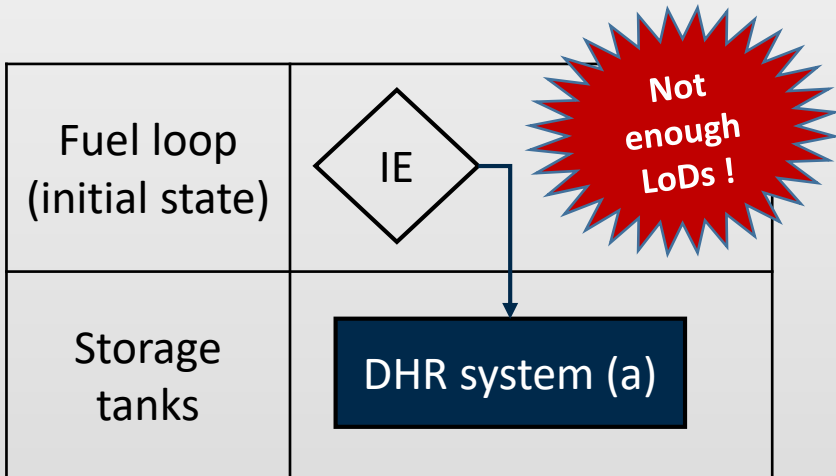
- 2 types of lines of defence:
  - Strong lines (a): Provisions with a failure rate between  $10^{-3}$  to  $10^{-4}$  by year and use (active and passive systems with redundancies)
  - Medium lines (b): Provisions with a failure rate between  $10^{-1}$  to  $10^{-2}$  by year and use (active systems without redundancies, inherent safety behavior, operator actions)
  - 2b lines are equal to 1a
- An incidental or accidental initiating event (IE) is the starting point of an abnormal evolution
- Lines of defence minimum number is dependant to IE frequency

Initiating event category	Initiating event frequency (per year)	Number of LoDs required
Incidental	$> 10^{-2}$	2a+b
Accidental	$< 10^{-2}$	2a
Hypothetical accident	$< 10^{-4}$	a+b

*LoDs number needed per initiating event category*

**What are LoD application conclusions?**

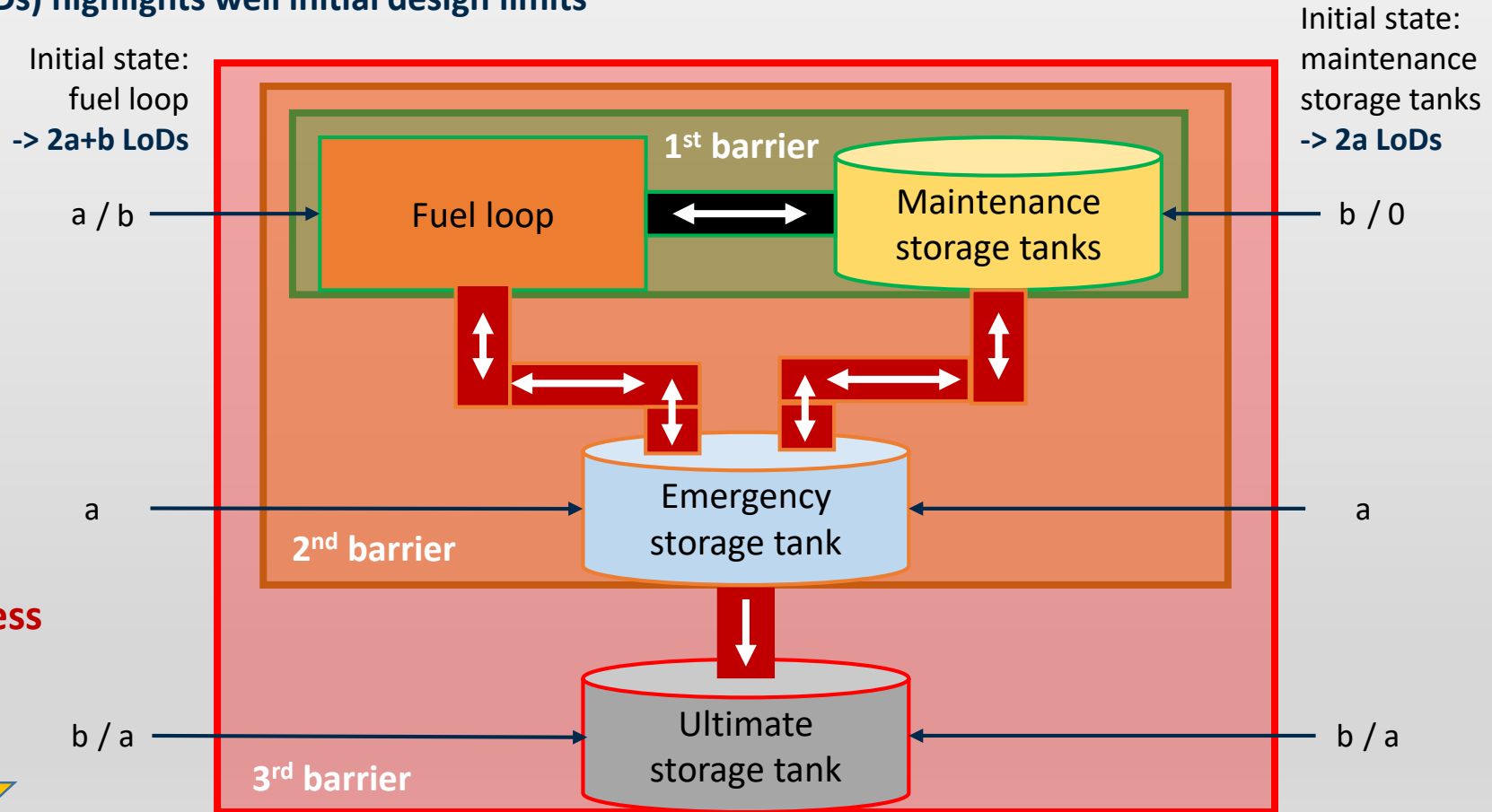
- LoD method applied to several IE: loss of main heat sink (LOHS), loss of liquid fuel (LOLF), reactivity anomalies accident (RAA), loss of fuel flow (LOFF), overcooling (OVC) and total loss of power (TLOP)
- Loss of liquid fuel (accidental IE = 2a LoDs) highlights well initial design limits**



*LoD application to fuel loop break with initial design*

- Confinement barrier definition in progress**
- What about radioactive gases ?**

Proceeding submitted to ICAPP 2025



*Proposed fuel salt locations configuration with DHR systems and preliminary confinement barrier definition*

## Decay heat calculations

- CERESIS interface code with reactivity control modules developed
- 1<sup>st</sup> inventories and decay heat calculations on ARAMIS-A
- Coming:
  - Discrepancies analysis with REM code
  - Decay heat parametric studies
  - Pandemonium nuclides identification
  - Internship: Implementation and comparison of valencies calculation methods (Sylvain Le Coupanec & Nathan Le Bourdonnec)

## Safety analyses

- LDD method applied to several initiating events
- Configuration proposed for fuel salt locations and decay heat removal systems
- Coming:
  - Deepening of confinement barriers definition
  - Analyse extension to radioactive gases confinement and decay heat removal



UNE PEPITE  
*Sens Critique*



**Coming in August 2025**

UN BANGER  
*Les Cahier du cinéma*

SURPRENANT  
*Télérama*

# Acknowledgments



Michel Allibert  
Louiliam Clot  
Axel Laureau  
Elsa Merle



Lydie Giot



Sylvie Delpech



Frédéric Bertrand  
Elena Martin-Lopez  
Vincent Pascal



Delphine Gérardin



Etienne Courtin  
Régis Marlier  
Herman-Wilfried Siaka



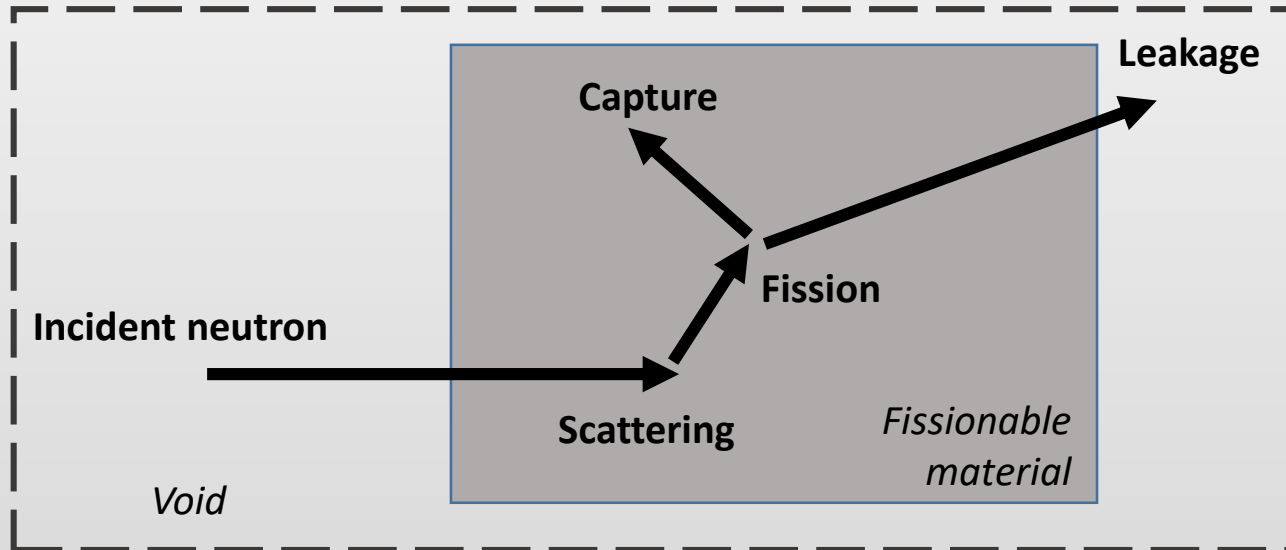
Rémi Hémery  
Frédéric Souvigné

Thank you for  
your attention !



# Appendix

- How does Monte-Carlo approach works?



*Monte-Carlo simulation example with 1 incident neutron*

- Reaction probabilities  $P_i$  (cross-sections) are computed by transport code before process
- N incidents neutrons are simulated multiples times
  - > **Reduce  $k_{\text{eff}}$  estimator uncertainty**
  - > **Stabilise Shannon entropy**

# Appendix

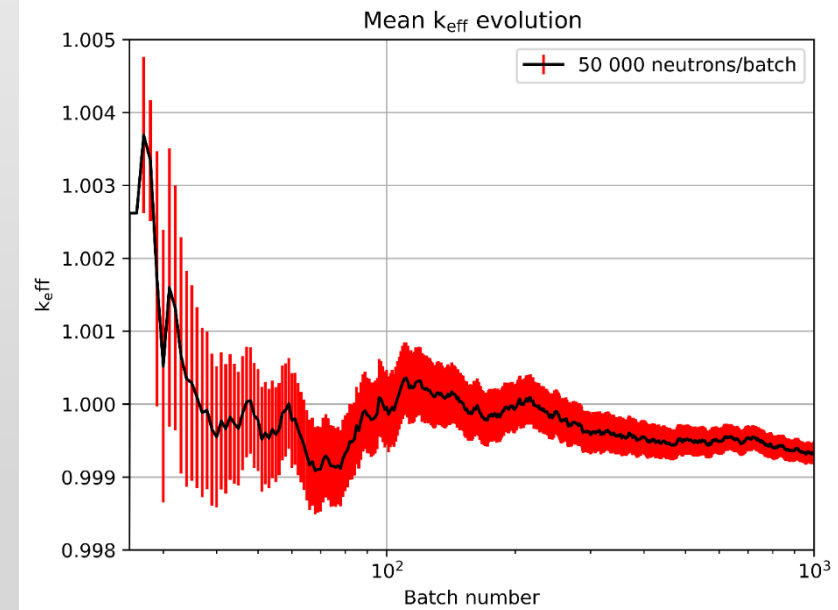
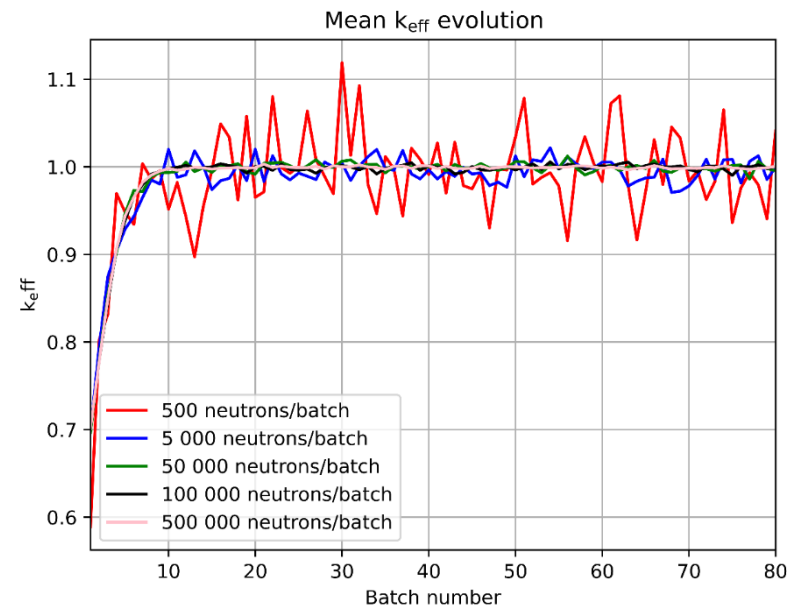
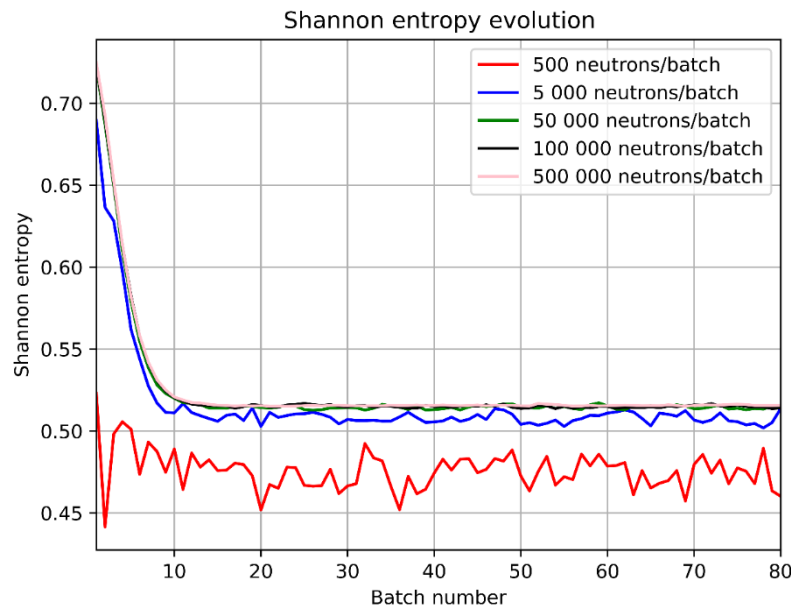
- What is the Shannon entropy?

$$H(X) = -E[\log_2 P(X)] = -\sum_{i=1}^N P_i \cdot \log_2(P_i)$$

with  $X$  the discontinuous random variable describing  $N$  neutrons

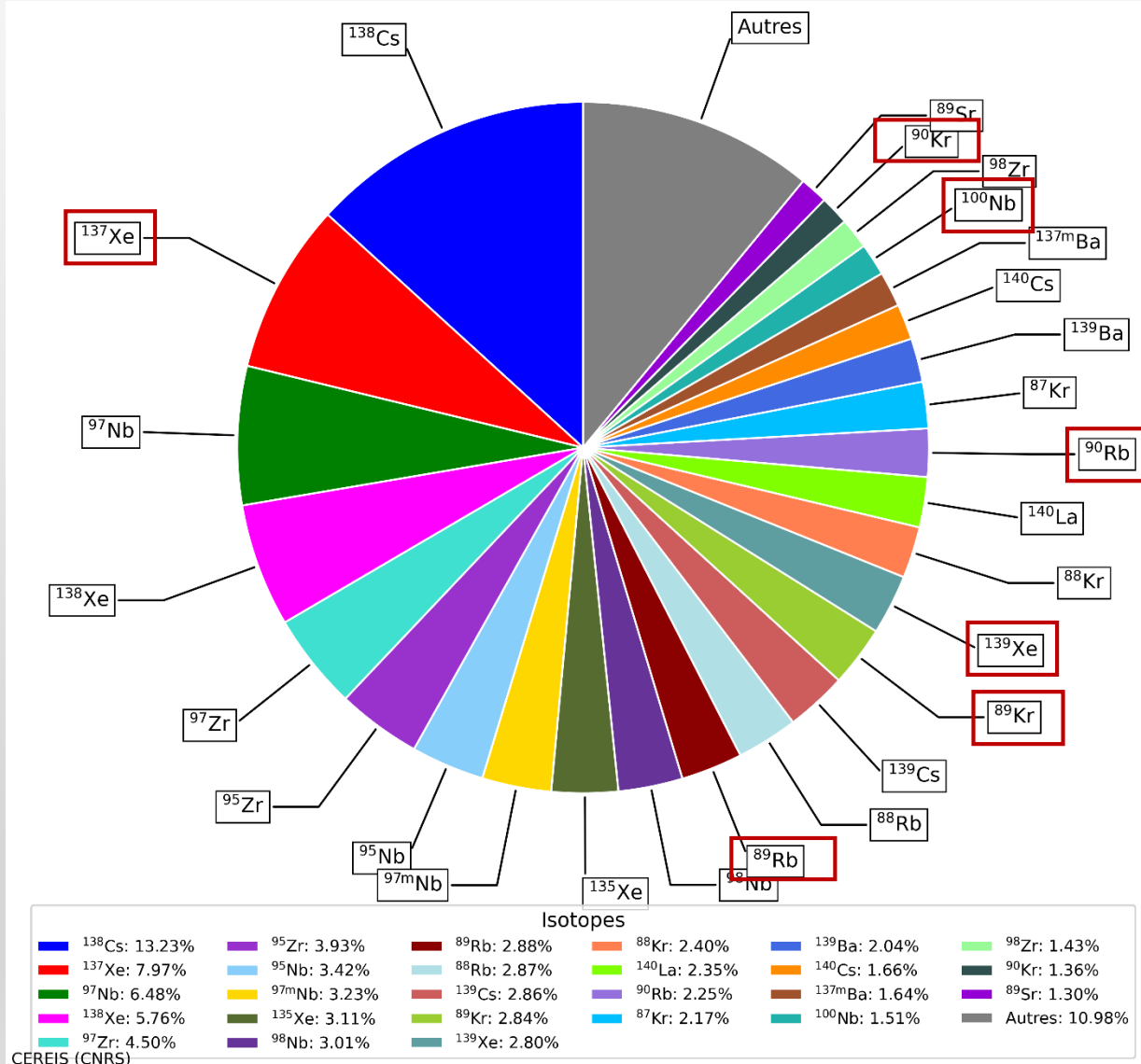
- To quantify uncertainty level of neutron source distribution
- In Monte-Carlo calculations, neutron flux and reaction cross-sections are averaged over multiples calculations (batches) involving each  $N$  neutrons

-> Every batch must have similar Shannon entropy



Parameters considered: 50 000 neutrons, 25 inactive cycles and 1000 active cycles

# Appendix



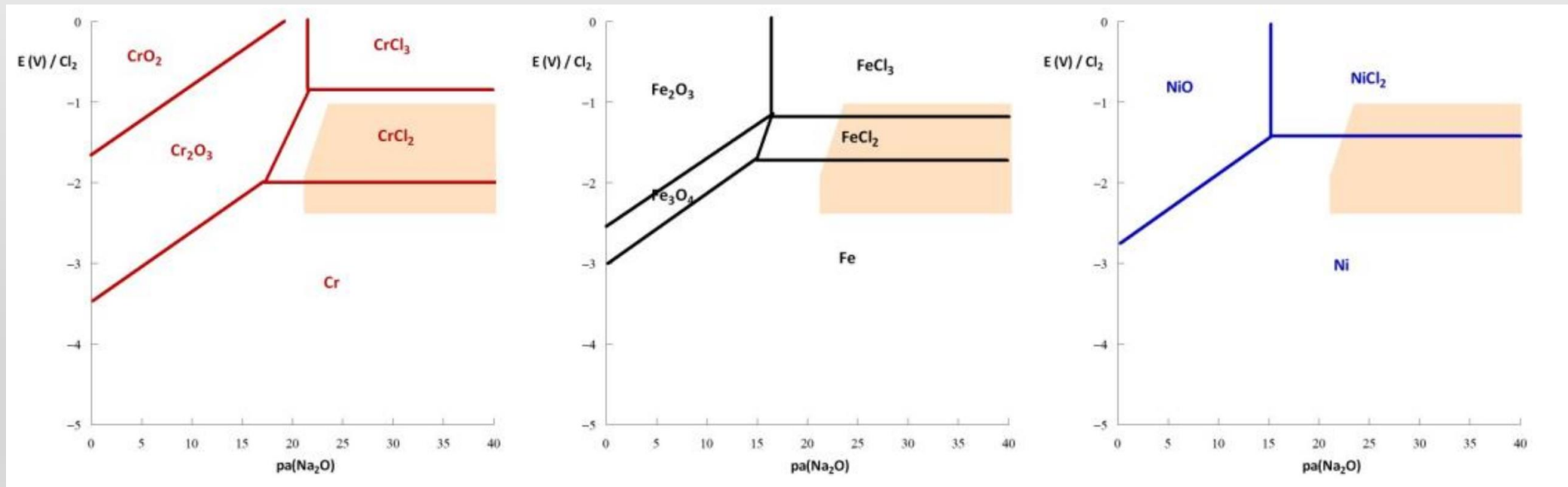
- At least 6 main contributors to DH are already identified and published Pandemonium nuclides
  - They contribute to 13.64% of total DH !
  - Are there other ones ?
- > Among main decay heat contributors, apply criteria to identify potential Pandemonium nodes
- > Identified nodes will be input for potential future TAGS measures

Decay heat main contributors (> 0.2% of total DH) in extracted gaseous PF after 20 operations years and 1 second of cooling

# Appendix

## Why calculating valencies?

- The valency of a chemical element is the maximal number of chemical bonds it can form
- Valency is dependant of fuel salt RedOx potential
  - > Valencies of some elements, from fuel salt or structural materials, could change
  - > New chemical species could be formed -> **safety issues (corrosion, precipitation, gaseous species)**
- In current depletion calculations, no prediction of the chemical species formed in the fuel salt
  - > **This internship is a “pioneer” work to implement chemical predictions modules into CEREIS**



Thermodynamic diagrams of Cr (left), and Fe (middle) and (right) in NaCl-based molten salt at salt at 600 °C.  
The orange domain corresponds to the stability domain of the NaCl-UCl<sub>3</sub>-PuCl<sub>3</sub> (80-10-10 mol%) mixture [SDel2024]