

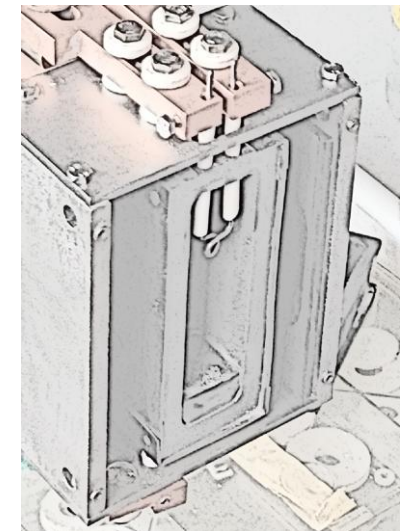
Nier-Bernas ion sources program at IJCLab: ISOL beams

Workshop on R&D for new ISOL beams (SPIRAL 1 and ALTO)



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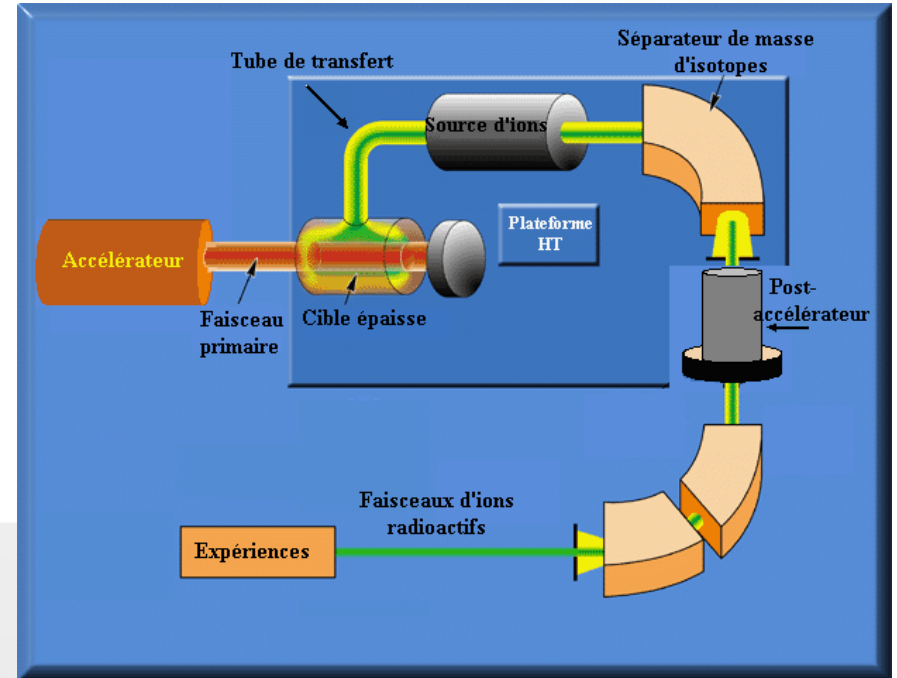




ISOL Technique

- ISOL method is an approved technique since ~70 years
- Allows to use thick targets
- Primary beam does not need to hit the ISOL target directly
- Nobel gases are very suitable for ISOL method

$$I = \Phi \cdot \sigma \cdot N \cdot \varepsilon_{\text{target}} \cdot \varepsilon_{\text{source}} \cdot \varepsilon_{\text{separ}} \cdot \varepsilon_{\text{det}}$$

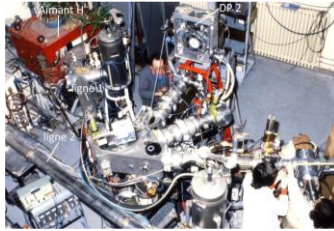


ISOL constraints: Selectivity + Rapidity



ISOL ion sources at Orsay (Legacy from the past and present)

1973



ISOCELE 1&2

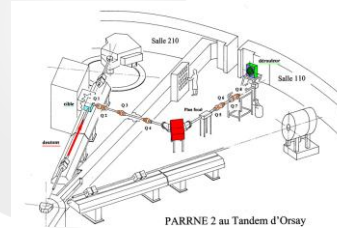
- Nielsen type ion source
- Nier-bernas ion source

1st Spectrometer closed to synchro-cyclotron

+

1st RIB experiment with online ISOL method in march 1974

1998



PARRNe 1&2

- Nier-bernas ion source
- FEBIAD MK5 ion source at Orsay
- Surface ionization ion source

2001



ISOL ALTO

Pioneering photofission experiment at CERN with the LEP injector

- FEBIAD MK5 ion source
- Surface ionization ion source
- Laser ion source

2020



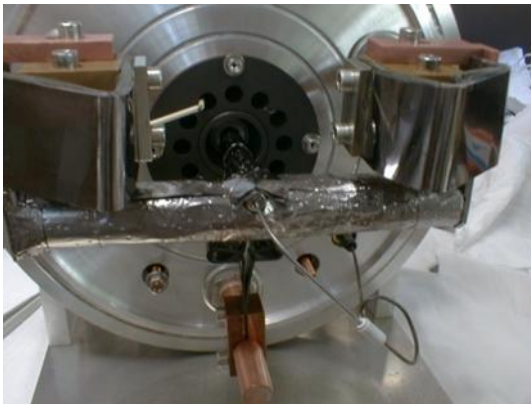
No universal ion source !



Hot plasma source :

(Isolde MK5 type)

Up to 30% for
gaseous & condensable
elements



Surface ionization

Source :

(Isolde MK1 type)

Suitable for $E_i \leq 6\text{eV}$

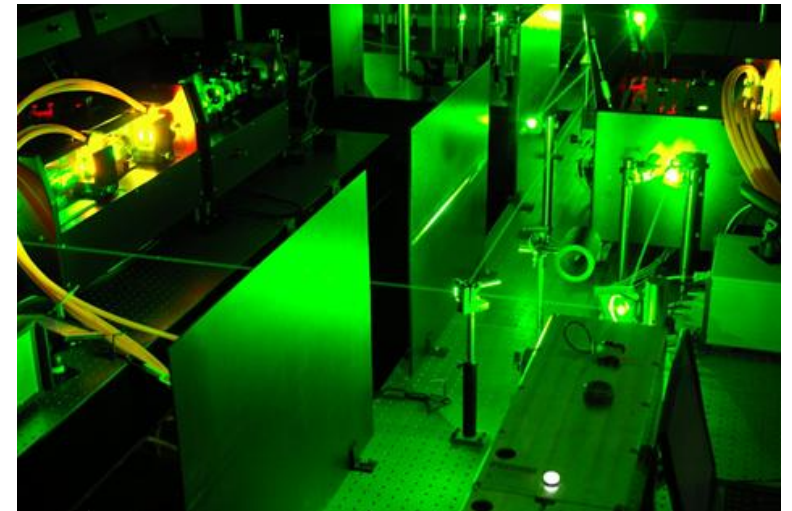


Laser Ion source:

Maximum selectivity

Good efficiency if

$F \geq 10\text{ kHz}$ (~10%)





Surface ionization

Source :

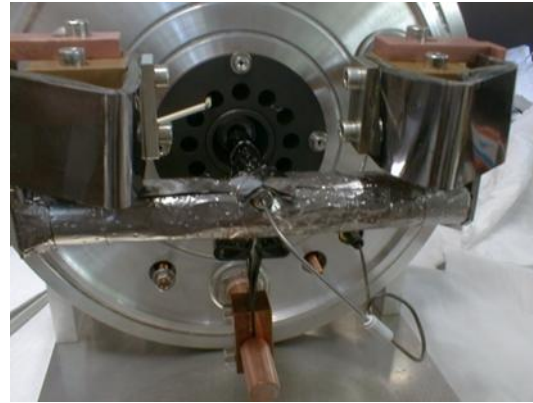


$< 1 \mu\text{A}/\text{mm}^2$

(10 – 20 π mm mrad)

(< 2 eV)

FEBIAD ion source:



$< 20 \mu\text{A}/\text{mm}^2$

(< 20 π mm mrad)

(few eV)

Nier-Bernas ion source:

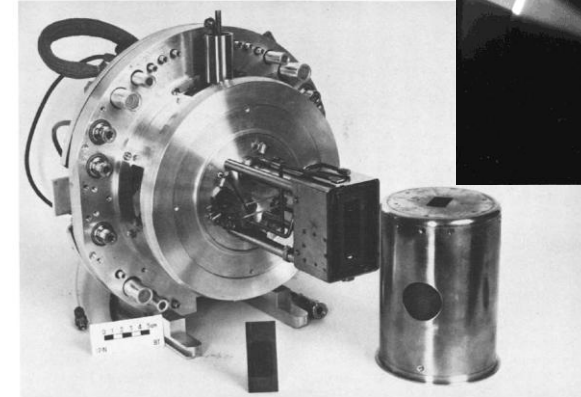


Fig 2 ISOCELE 2 ion-source

$< 5 \text{mA}/\text{mm}^2$

(< 20 π mm mrad)

(few eV)



- Efficiency : (20-70% for the elements above Ne)

Nier-Bernas (~mA range)

- Threshold pressure of 10^{-3} to 10^{-2} mbar
- Efficient ionization of gaseous and condensable elements
- Need of High intensity mass separator that handle space charge problems



Suitable

- When running molten targets or target materials with much intrinsic impurities at very high temperatures
- When using a molecular separation technique : Target consumption up to 1 g/hr + a high gas load from the target.

FEBIAD (~μA range)

- Threshold Pressure $5 \cdot 10^{-4}$ to $3 \cdot 10^{-5}$ mbar
- Very efficient for elements heavier than Ar
- ionization efficiencies for these sources are high and widely pressure-independent
- Simple and economic low-intensity mass separator needed
- Not able to cope with high gas load → ionization efficiency is quenched



FEBIAD and Nier-Bernas problem with Light elements

1. Lower ionisation cross section
2. Shorter transit times through the ionising volume → ionization efficiency ↘

Particularly troublesome elements : C, N, O → their volatile molecular compounds CO, CO₂, N₂, O₂ are very reactive in hot enclosures → cold-enclosure ECR sources

Successful proof of a concept : injection of suitable gases close to the exit slit.

TABLE I
Observed intensities of negative ion beams obtained from a Bernas-Nier type source. Because of the limitations noted in the text, these currents are probably not the maximum obtainable.

Ion	Gas	Current (μA)
C^-	CO	0.03
O^-	CO	1.05
N^-^{a}	N_2	0.3
NH^-	N_2	0.007
$^{35}\text{Cl}^-$	CCl_4	12
$^{37}\text{Cl}^-$	CCl_4	4

^a Attributed to the ^1D state^{2,3}.

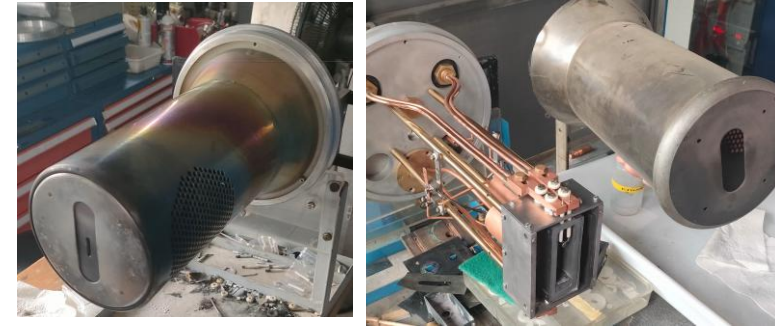


J. Denimal *and al.* (NIM B. 109 (1973) 409.



Actual uses and developments

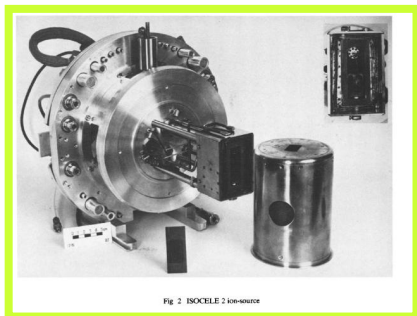
- ❑ Nier-Bernas ion source is already used at IJClab for ion implanter at the MOSAIC facility.
- ❑ Widely used in the field of ion implanters and the semiconductor doping
→ Many ion source developments are achieved last years in these field¹



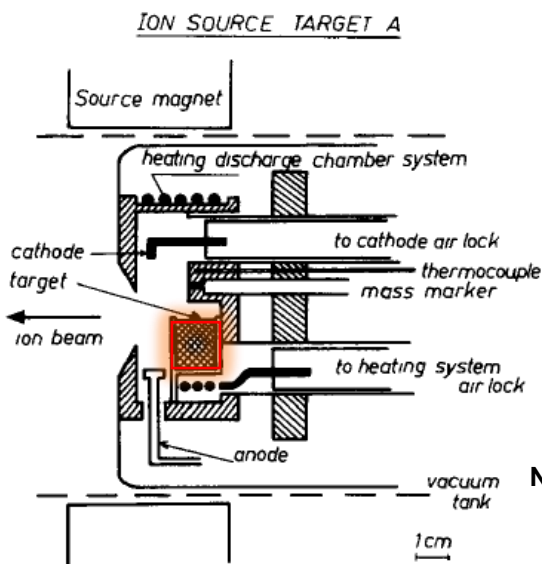
¹ <http://www.gursung.com/global/en/cont/2/2.php#n>



ISOCELE concept:



Nier-Bernas ion source @ ISOCELE



Nier-Bernas arc chamber

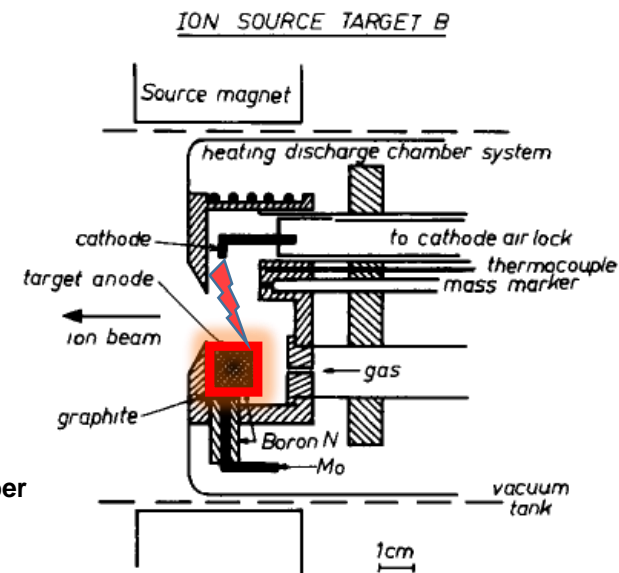


Fig 1 Schematic view of target-ion-source systems

A – target behind the arc, B – target under the arc

Main technical features:

Target

- 2 target positions (behind / under the arc)
- High temperature volume for the target
- Most of release issues from conventional ISOL targets will be ruled out

Ion source

- Versatile plasma ion source
- High intensity ion source (~mA)
- Low beam emittance

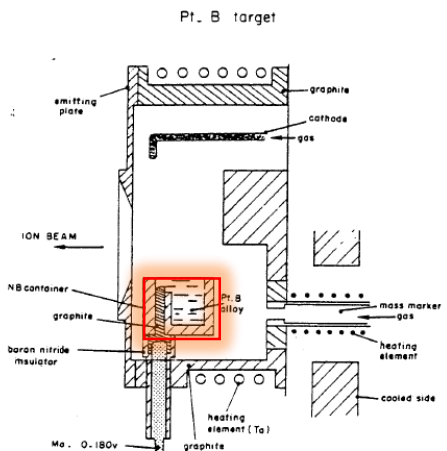


Fig. 5. Pt-B target ion-source arrangement.

- Target heated with arc discharge power ($\sim 100\text{ v} - \sim 2\text{ A}$)
- Beam intensities : $\sim 1 - 2\text{ mA}$

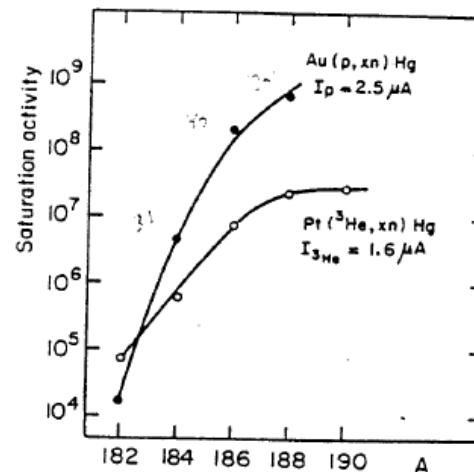


Fig. 4. Yields of Hg from Au and Pt-B targets.

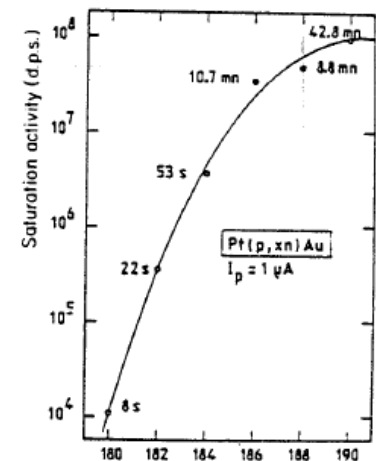


Fig. 6. Yields of Au from a Pt-B target.



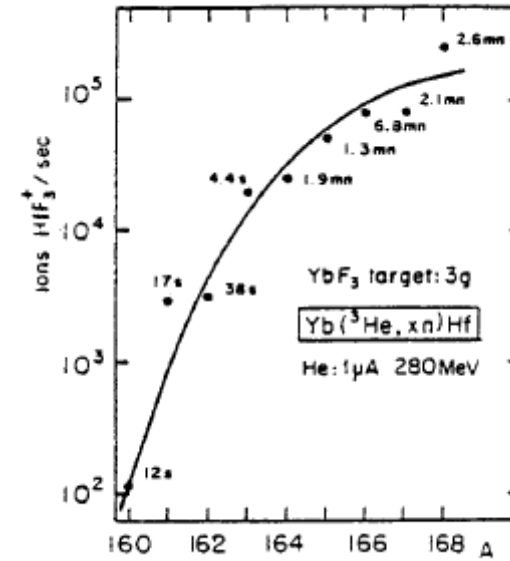
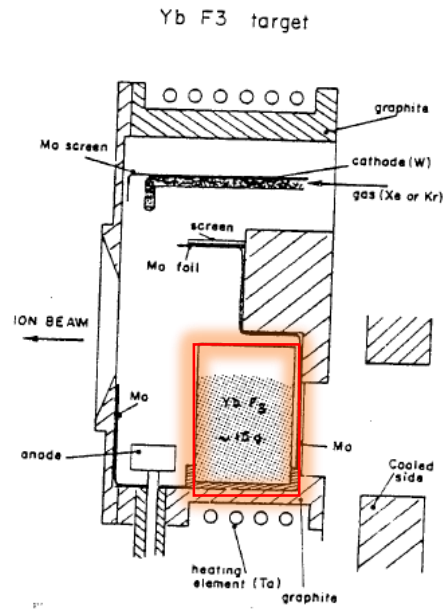
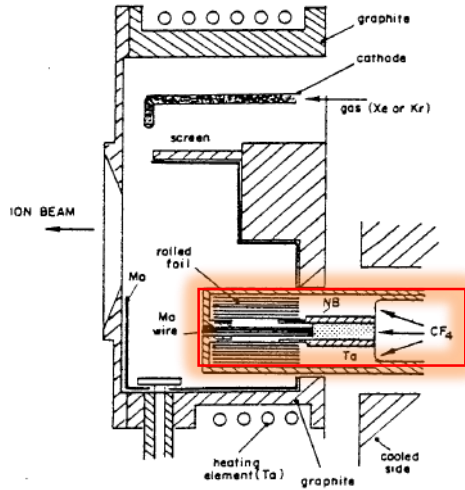


Fig. 12. Yields of Hf (HfF_3^+) from the YbF_3 target.

- Target slightly set back from the direction of the arc
- **Solid way fluorination**



- Mo target
- Target slightly set back from the direction of the arc
- Rolled sheets target
- **gaseous way fluorination**

Table II. Boiling points of elements ($36 < Z < 42$) and of their most volatile fluoride.

Element	Kr	Rb	Sr	Y	Zr	Nb	Mo
Oxidation number		1	2	3	4	5,3	6-2
Boiling point °C	-152	688	1384	3338	4377	4742	4612
Fluoride the most volatile		RbF	SrF ₂	YF ₃	ZrF ₄	NbF ₅	MoF ₆
Boiling point °C		1410	2489	1387	~600*	236	35

*Sublimation point.

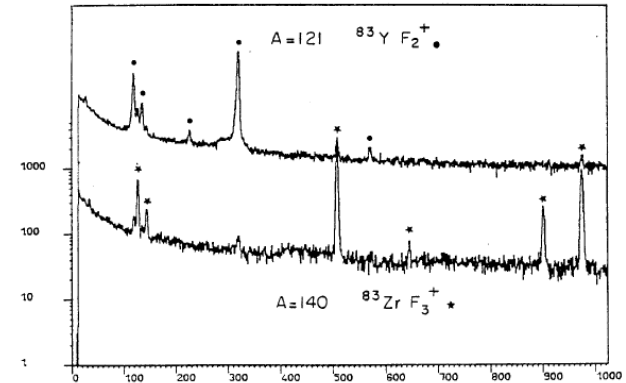
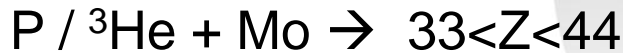


Fig. 10. Comparison of γ spectra for ZrF_3^+ ($A = 140$) and YF_2^+ ($A = 121$), separation of the isobars Zr and Y of the chain 83 can be seen.



Xe production with an outsided target (NaI or CsI)

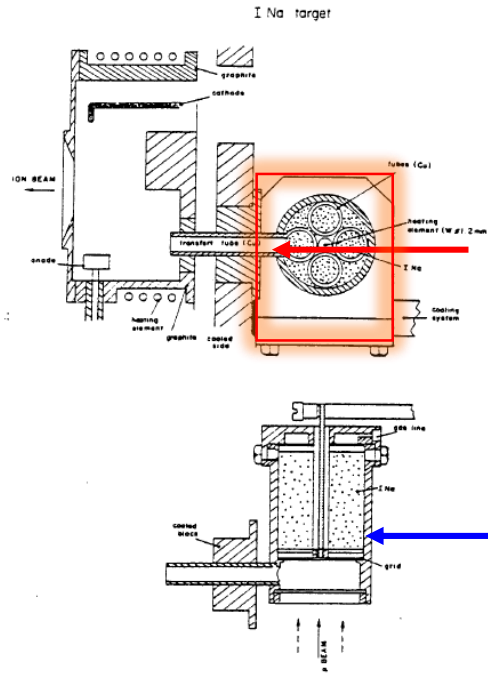


Fig. 7. NaI (or CsI) target ion-source arrangement.

Carrier gas used to accelerate the transfer of Xe to the ionization chamber

Outsided and cooled target

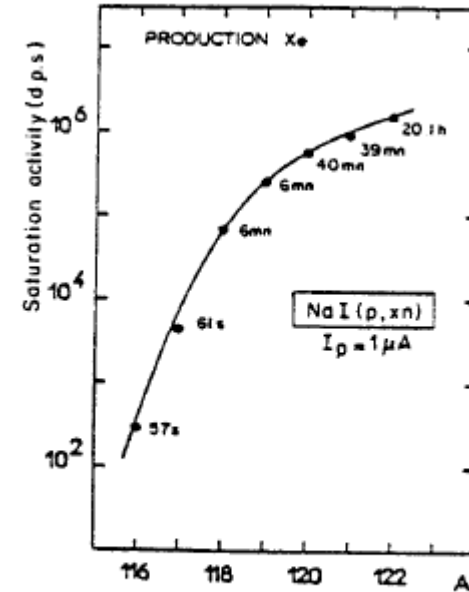
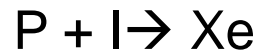


Fig. 8. Yields of Xe from a NaI target.



Achieved targets/beams

Target	SC beam	Extracted elements	
Ni.	^3He	Cu, Zn	
Ge	^3He	Zn, Ga	
Ag	^3He	In, Cd	
Sn	$\alpha, p, ^3\text{He}$	Te, Sb, In, Cd	
Pr	p	Nd	
Nd	^3He	Sm	
Gd-Cu	^3He	Dy, Eu, Sm, Pm	
Tb	p, ^3He	Ho, Dy, Eu, Sm	
Er-Cu	p	Tm, Ho, Dy	
Pt-B	p, ^3He	Au, Hg	
Au	p, ^3He	Hg, Tl	
La	p, ^3He	La, Ba, Cs	
Ce	p, ^3He	La, Ba, Cs	
YbF	^3He	Hf	powder
SrF ₂	^3He	Zr, Y, Rb	powder
YF ₃	^3He	Zr, Sr, Rb	powder
Y	^3He	Y, Sr, Rb	
LuF ₃	^3He	Ta, Hf	powder
Mo	^3He	Nb, Zr, Y, Sr, Rb	rolled foil
W	^3He	Ta, W, Hf	rolled foil
TeO ₂	^3He	Xe	powder
NaI	p	Xe	powder
CsI	p	Xe	powder
Bi	p	Po	

Not less than 31 combinations tested with ISOCELE Nier-Bernas Ion source!

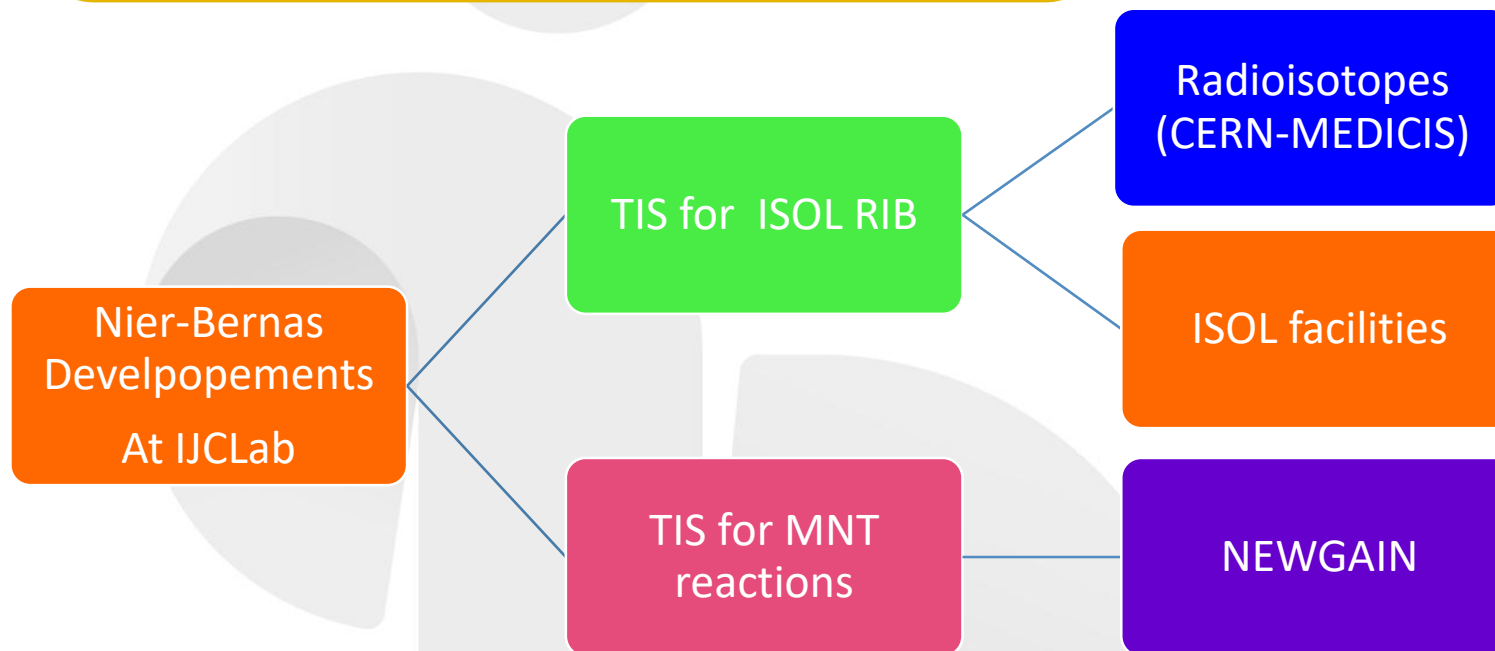
powder
powder
powder
powder
rolled foil
rolled foil
powder
powder
powder

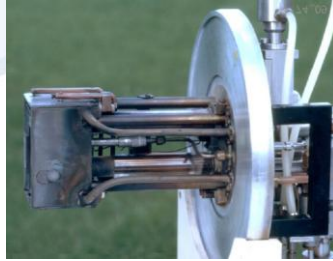
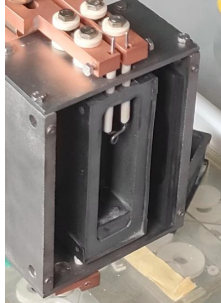
} isobar separation using CF₄



Main advantages :

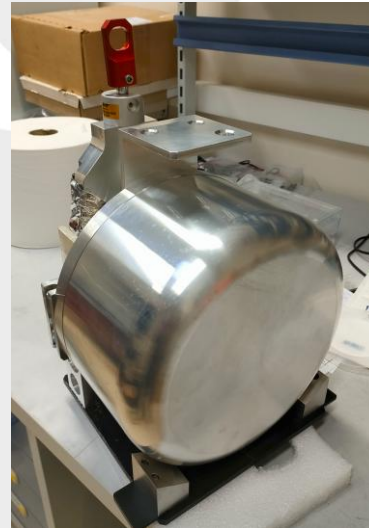
- High current intensity
- High pressure operating regime
- Good capability of handling molecular beams
- Target in the ionization volume...



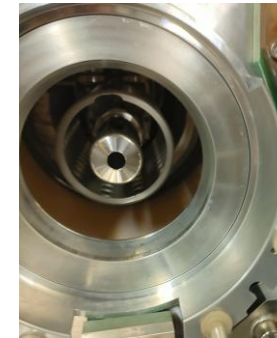


Extraction : slit shape
+ 4 directions

TIS for ISOL RIB



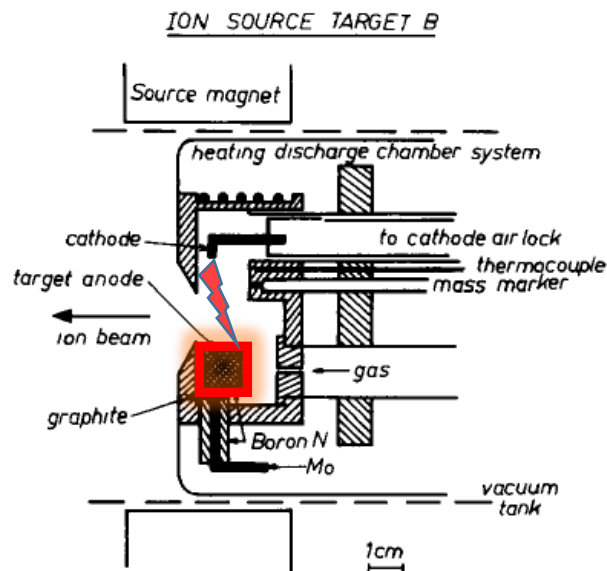
ISOLDE target
vacuum chamber



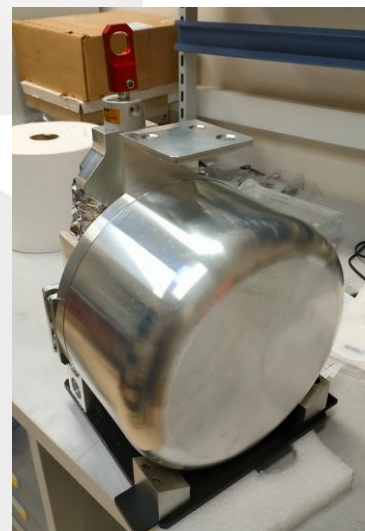
Extraction : hole
shape + 1 directions



TIS for MNT reactions



A – target behind the arc, B – target under the arc



Or other design...

ISOLDE target
vacuum chamber



New ISOL ion sources test bench @ IJCLab



Gray room: assembly of prototypes



Thermal test bench: thermal measurements (16 sensors + pyrometer) and filament developments.

→ Commissioning in progress