

Developments for low background experiments

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GDR DI2I



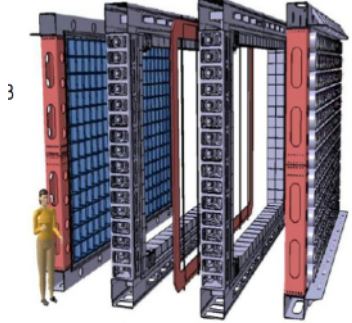
What means Low background experiments

- Looking for rare events:
 - Fundamental Physics
 - Proton decay
 - Nuclear structure
 - Neutrino double beta decay $\beta\beta\nu 0$
 - Dark matter
 - Other disciplines
 - Low radioactivity measurement
 - Logical test failure in nano/micro electronics
 - Qubits (Quantum computing)
 - Biology
- Protected from any irradiation
 - Cosmic rays
 - Natural irradiation
 - Radon
- Located in the Underground Laboratory (UL)
- Experiment components are radiopure

Example of the experiments

- Fondamental Physic

SuperNemo



- Other disciplines



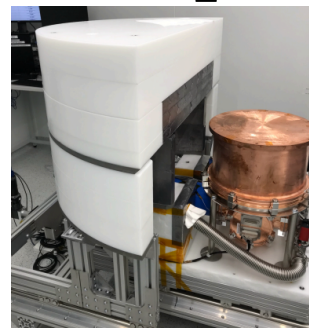
Gamma spectrometry



NEWS_G (SEDINE)



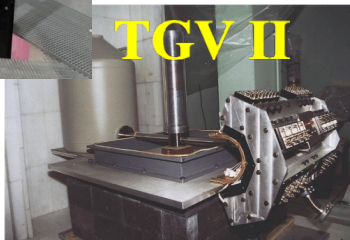
DAMIC_M



MIMAC



TGV II



- Then TESSERACT ; Qubits

Common part of LB experiments

- Material selection
- Low radioactive components such as:
 - Electronics components
 - Shielding
 - ==> Standard Lead
 - ==> Ancient Lead
- Detector structures/support
 - ==> **Copper mainly**
 - Commercial production \approx 5N (99.999%) \approx 0.1-10 mBq/kg
 - **Electroform Copper** \approx 6N (99,9999%) \approx 10-100 nBq/kg

Electrolytic Cell and Electroplating

- Electrolytic cell driven by **oxidation and reduction** reactions
- Current supplied to drive reactions
- Desired result is generally **the reduction of ions** from the electrolyte to form atoms on the cathode surface
- Mass deposited proportional to current supplied:

$$M = \frac{m_r \int I(t) dt}{zF}$$

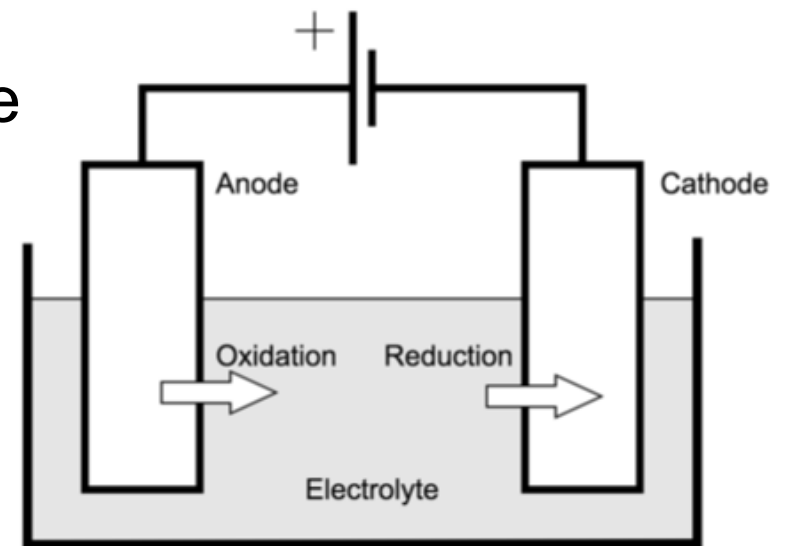
M – mass

m_r – molar mass

$I(t)$ – current as function of time

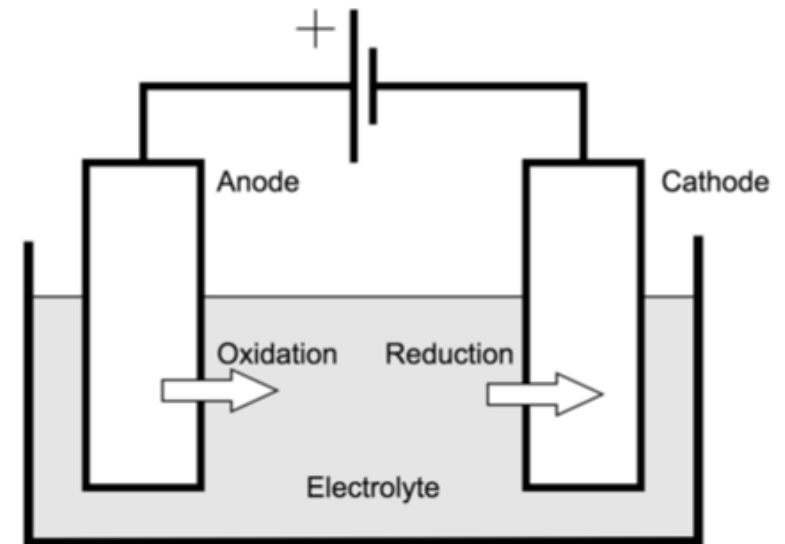
z – number of electrons transferred in reduction reaction

F – Faraday Constant ($= e N_A$)



Electroplating Copper

- Some ions reduce more readily than others – **reduction potentials**
- Voltage between anode and cathode limits **electroplating of some species**
- Copper benefits from ‘**electrowinning**’ – high reduction potential $+0.34\text{ V}$
- Reduction potential of:
 - Uranium: -1.80 V
 - Thorium: -1.90 V
 - Lead: -0.44 V
- **All lower than copper**; refined during electroplating



Why don't impurities plate too?

- Which reaction proceeds determined by standard **cell potential**:

$$E_{cell}^0 = E_C^0 - E_A^0$$

E_{cell}^0 - standard cell potential

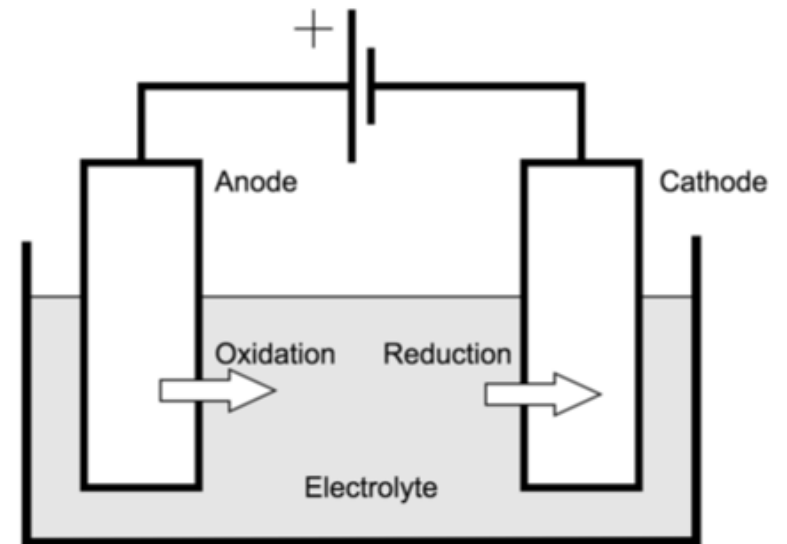
E_C^0 - standard cathode reduction potential

E_A^0 - standard anode reduction potential

- Related to Gibbs Free Energy:

$$\Delta G^0 = -z F E_{cell}^0$$

- If $\Delta G^0 < 0$ reaction is spontaneous
- If $\Delta G^0 > 0$ need extra energy input



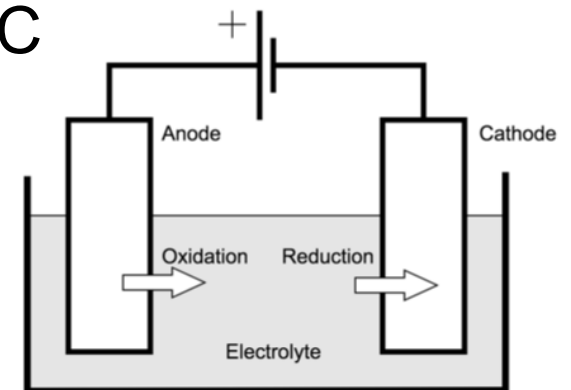
An Example

E_{cell}^0 - standard cell potential

E_c^0 - standard cathode reduction potential

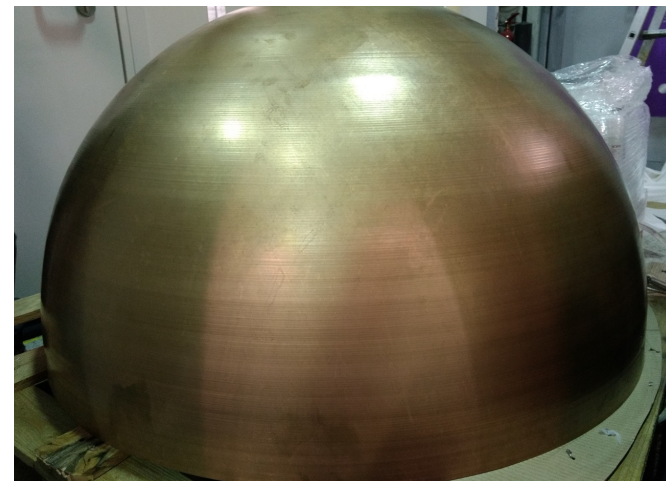
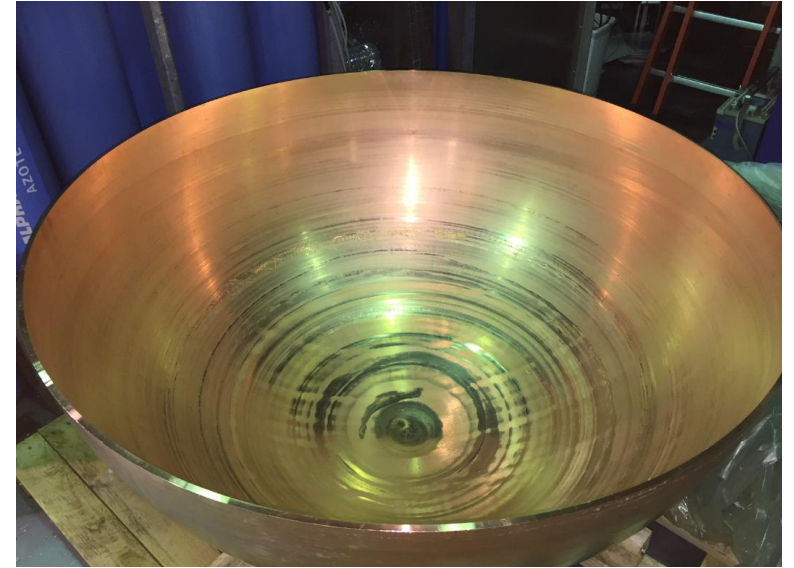
E_A^0 - standard anode reduction potential

- Example of solution containing U^{3+} and Cu^{2+} , with a Cu anode:
 - U^{3+} to U
 - $E_{Cell} = -2.138\text{ V} \rightarrow$ Requires energy Cu^{2+} to C
 - Cu^{2+} to Cu
 - $E_{Cell} = 0\text{ V} \rightarrow$ In equilibrium
- **Cu^{2+} reduction** will occur at lower potential
- Still require a potential difference between electrodes to **overcome energy losses**



Detector Hemispheres

- Two 140 cm diameter hemispheres
- 4N copper (99.99% pure)
- Plates purchased and then spun into hemispheres
- 110 cm diameter hemisphere also spun as the anode for electroplating



Preparation Procedure

- Cleaned with domestic detergent
- **Sanded** – different sanding for each hemisphere – but final stage with silicon carbide paper
- Estimated $\sim 10\text{-}100\ \mu\text{m}$ removed by sanding ($\sim 500\ \mu\text{m}$ second hemisphere)
- Cleaned with **Micro90** detergent (1%) in deionised water
- Inner hemisphere received same treatment for first use – just Micro90 second time

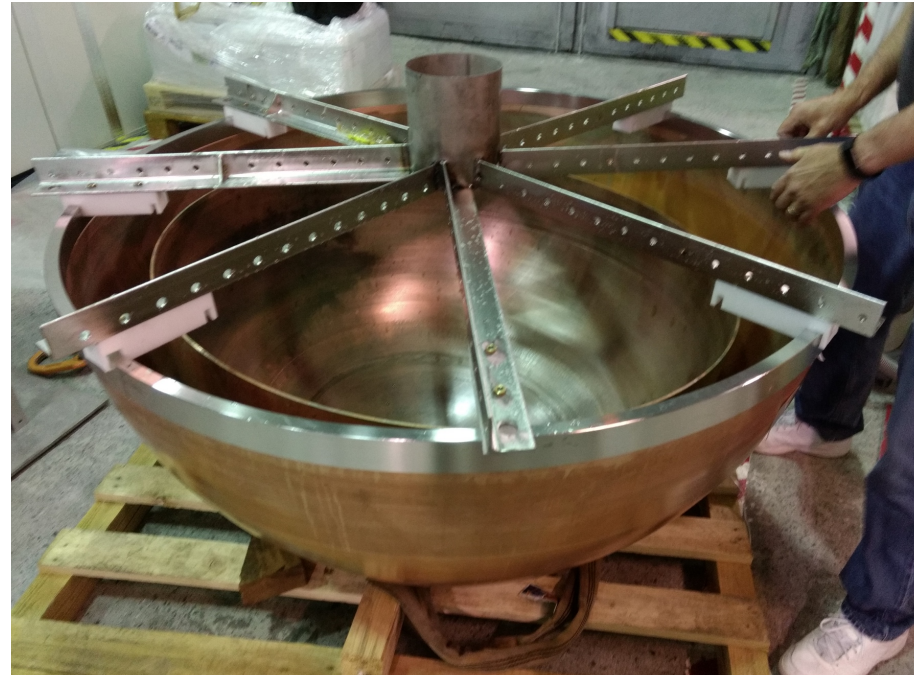
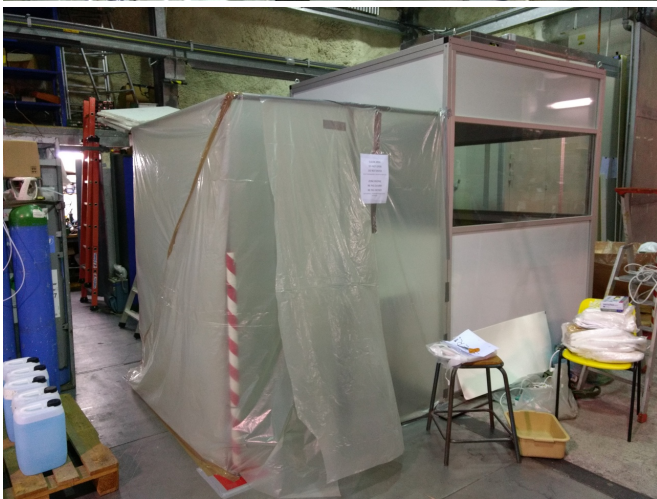


Chemical Etching

- Surface **chemically etched** with 3 % H₂O₂, 1 % H₂SO₄ in deionised (DI) water
 - Shown to be effective etchant while less aggressive than some alternatives
- To remove any trapped particulates from spinning process or sanding, and make surface more smooth
- Estimate **5-10 μm** removed during etching



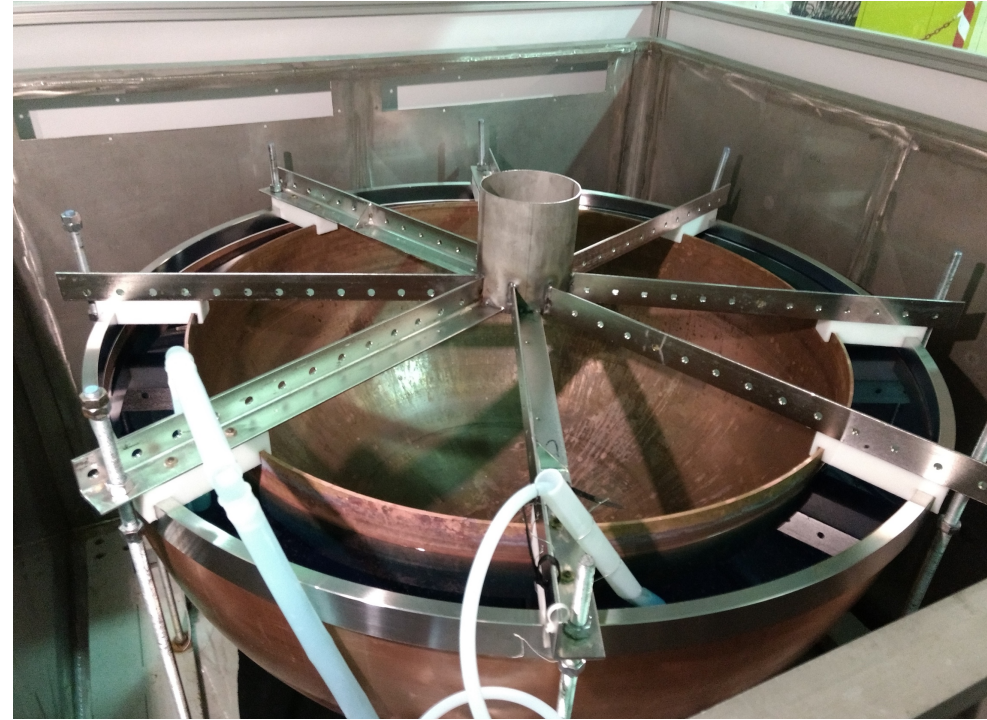
Cleanroom



Process carried out in cleanroom to minimise contamination and particulates

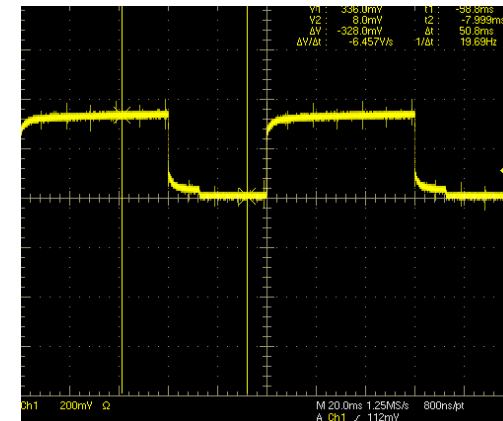
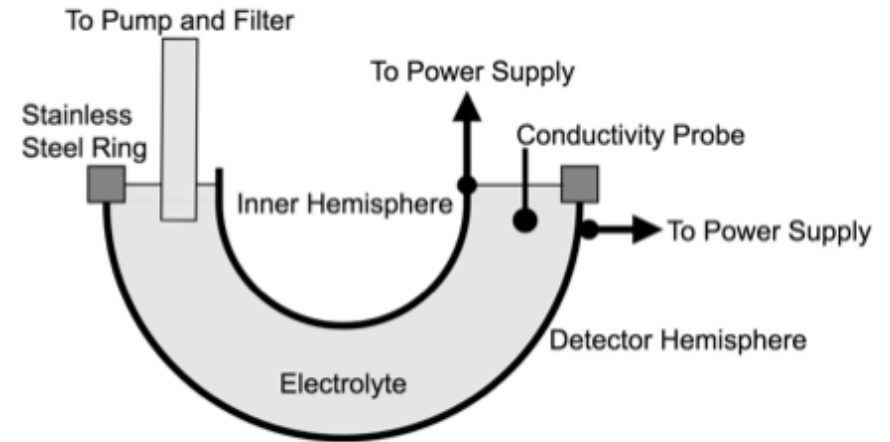
Electrolyte

- First hemisphere filled with 80 l electrolyte from SEDINE electroplating, 5.5 l H₂SO₄ and topped up to ~300 l with DI water
- Second hemisphere used ~275 l first-hemisphere electrolyte, 2 l H₂SO₄ and up to ~300 l with DI water
- Both have some CuSO₄ in solution already
- **Pump and filter** used to move electrolyte and remove particulates



Electropolishing

- Electropolishing serves several purposes
 - Removes layer from hemisphere without chemical or mechanical attack
 - Preferentially **removes high spots** from surface
 - **Increases concentration** of CuSO_4 in electrolyte
 - **Waveform** used to polish
 - **+ve = detector hemisphere**
- voltage > inner hemisphere**
- First (second) hemisphere $21.2 \pm 0.1 \mu\text{m}$ ($28.2 \pm 0.1 \mu\text{m}$) polished

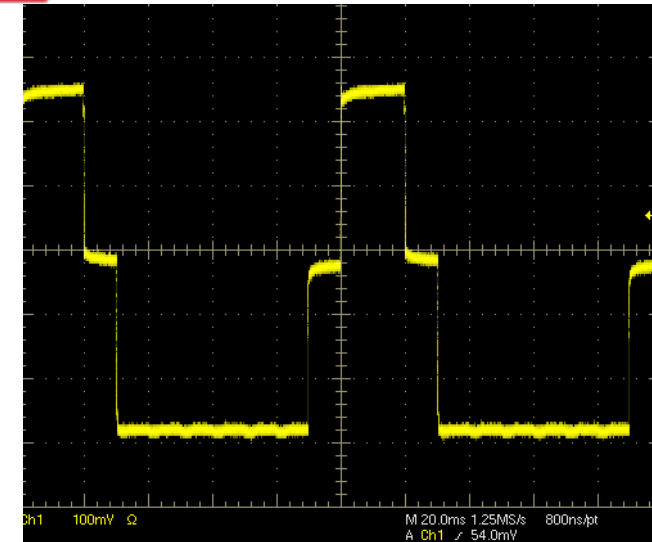
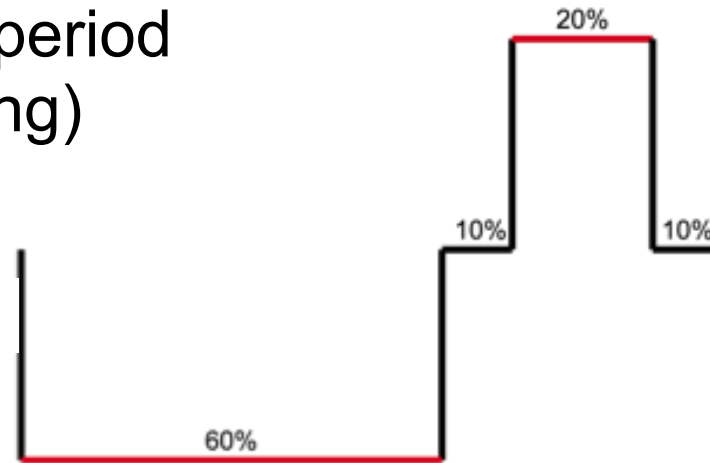


Why Pulse-Reverse Plating?

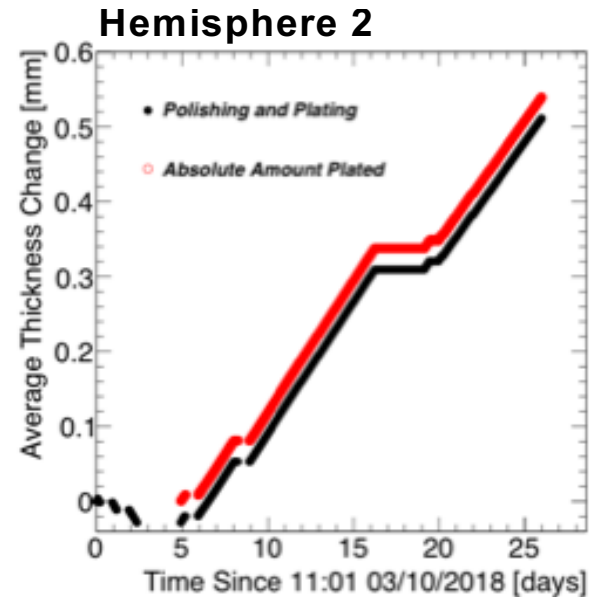
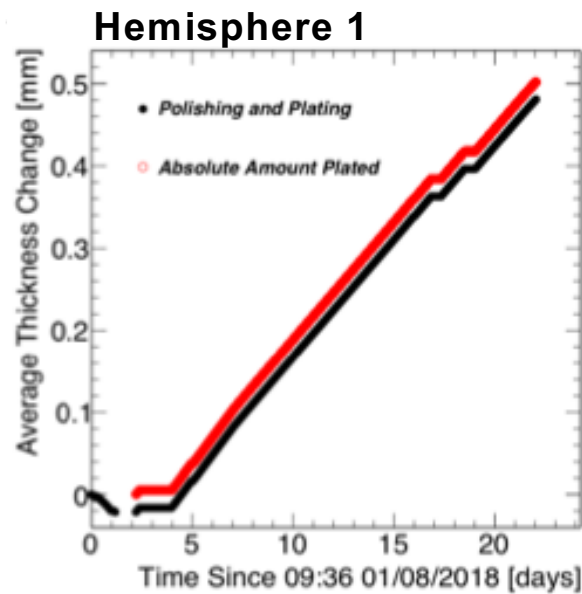
- Waveform for plating with relaxation period and/or reverse bias (effectively polishing)

- Benefits compared to DC plating:

- – **High points** are preferentially **removed** in reverse bias section
- – Greater **uniformity** of plate – relaxation period allows diffusion of ions
- – Higher density copper plate



Electroplating



- Plated for approximately 20 days for both hemispheres
- 0.27 to 0.3 V amplitude in forward and reverse directions of plating
 - Established value for copper plating
- In total first (second) hemisphere plated $502.1 \pm 0.2 \mu\text{m}$ ($539.5 \pm 0.2 \mu\text{m}$)

Results

- Drained electrolyte and rinsed surface with DI water
- Passivated with 1% $C_6H_8O_7$ in DI water for 30 min-1 hour
 - Prevents further oxidisation
- Rinsed again with DI water
- Stored in plastic ready for electron beam welding



- First hemisphere had many nodules – frequent and 0.5-1.5 mm in height
- Second hemisphere had very few nodules

Hemispheres 1st / 2nd

- Second hemisphere has less nodules. Main difference with first hemisphere:

- **Higher initial CuSO_4** concentration – already used for one hemisphere

- But also:

- **More sanding** – ensure uniform surface before beginning

- **Pump outlet moved** daily – improve mixing and spread excess-plating

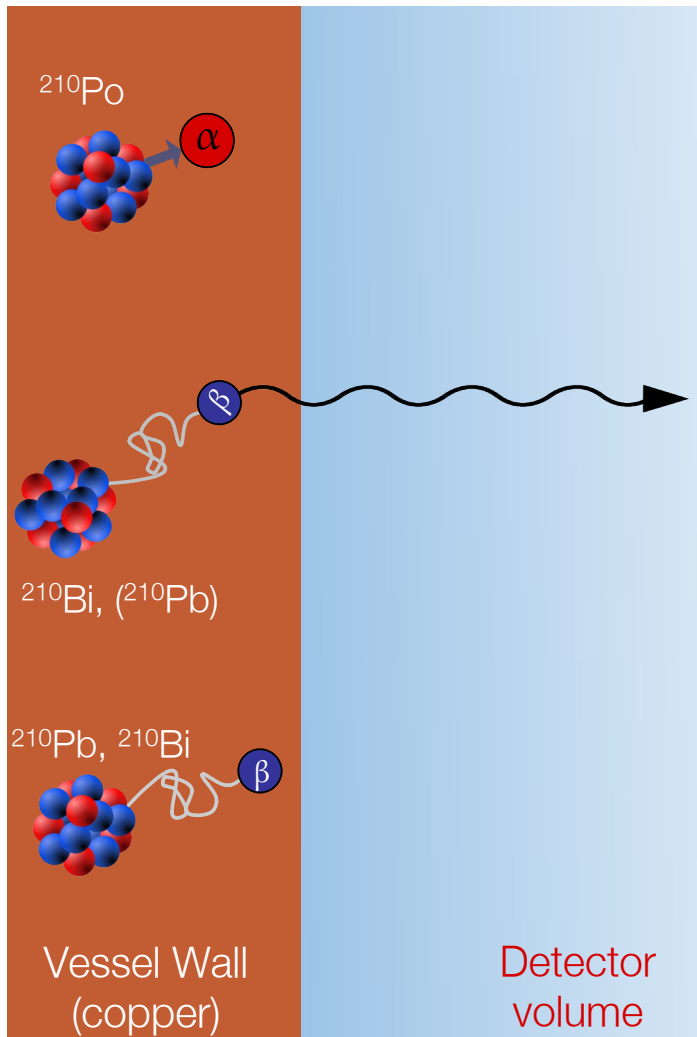
- Improved electrical isolation from container

- Maybe higher H_2SO_4 concentration (to be determined by analysis)

- Solution filtered for longer before beginning and between polishing and plating

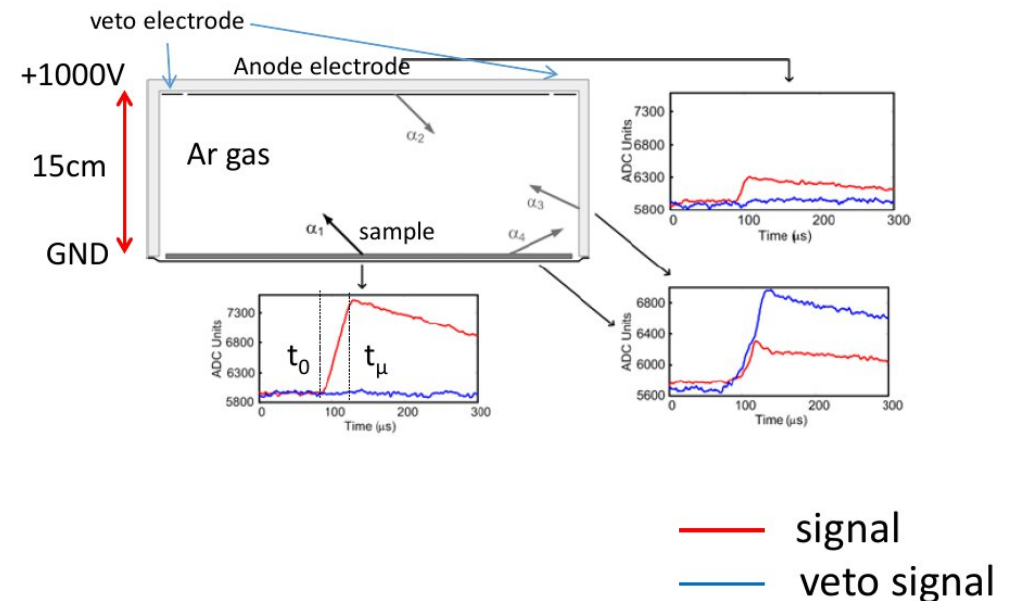
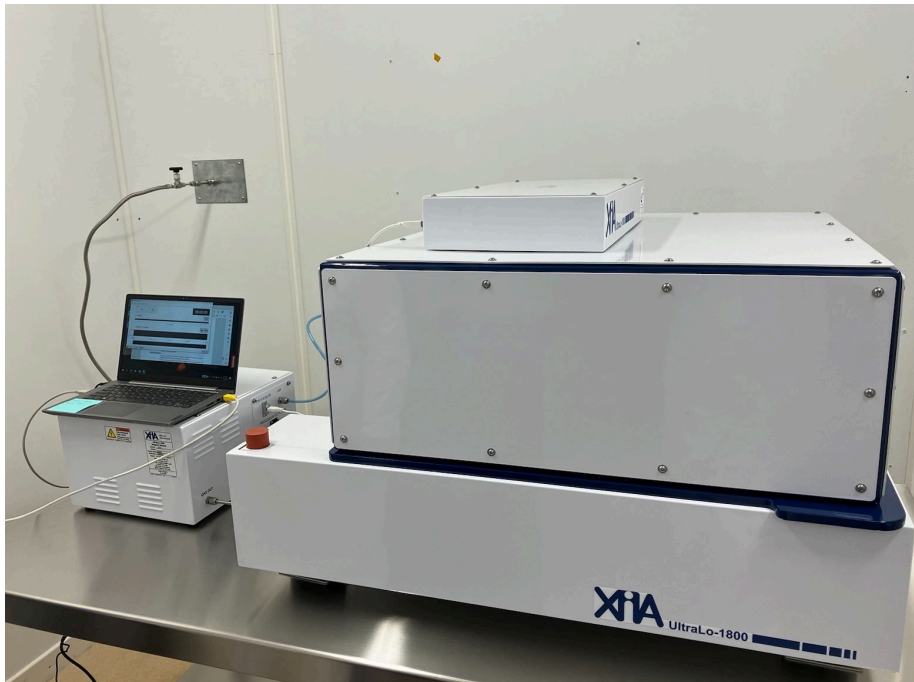
- Inner hemisphere had layer removed from polishing already

Then, surface contamination



- ^{210}Pb is a long-lived radio-impurity found in copper
- Most radiation is stopped inside the copper but...
 - Bremsstrahlung x-rays ($\sim\text{keV}$) from ^{210}Pb and ^{210}Bi β - decay in the copper escape, travel through whole volume

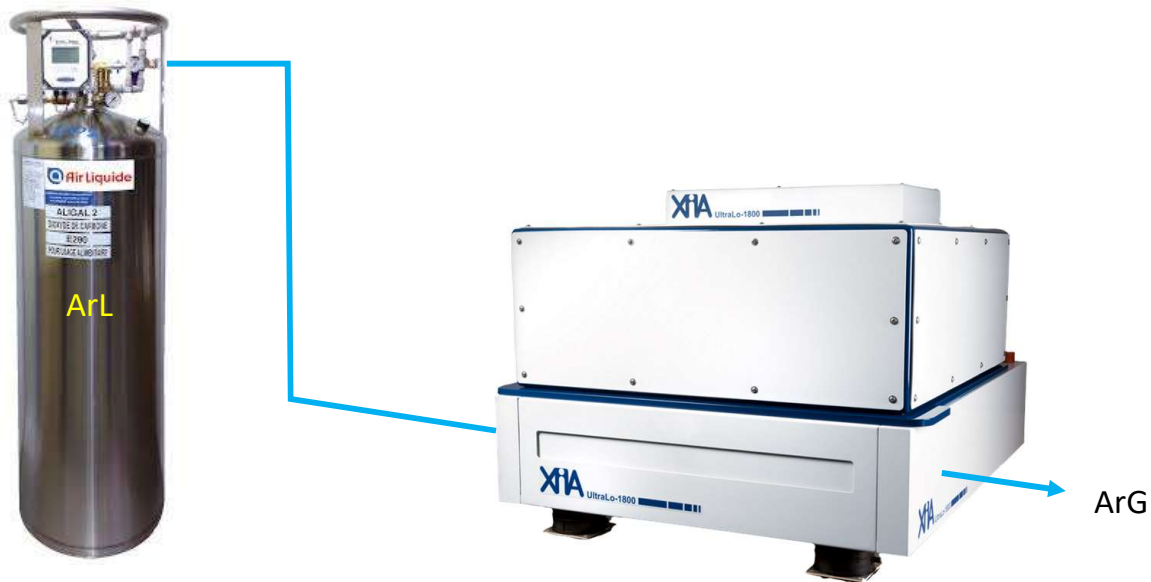
XIA, Ultralo-1800 Alpha counter



- Installation in a Cleanroom @ LPSC for commissioning run
- Final installation @ LSM

- By measuring rise time ($t_\mu - t_0$), surface event can be distinguished from event in Ar gas or ceiling

How XIA operate

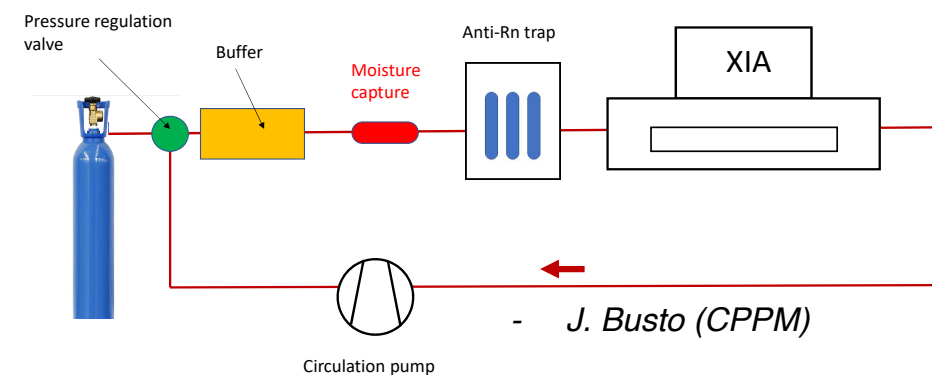


Flushing : ~ 700 L (45 min)
Measurement : ~300 L/ h

Proposition : replace Liquid-Ar by Gas-Ar and operate in a closed circuit with an anti-Rn purification system and very low moisture.

Using XIA in a closed circuit

Basic desing



- **Development of GPS: Gas Purification System**
- Manufacture is finished
- Pre-Installation @LPSC
- To be operational after confirming the XIA “normal” operating mode



Conclusion

- Radiopurity of off-the-shelf copper not sufficient – Solution: electroplate 0.5 mm ultra-pure copper layer to remove alpha particle component
- Electrode potential can be selected to plate only copper but not U, Th and Pb
- Surface preparation is critical – sanding, cleaning and etching
- Electropolished $\sim 20\ \mu\text{m}$ from surface and electroplated $\sim 500\ \mu\text{m}$ ultra-pure copper
- Difference in surface quality between the two hemisphere
- Measurement of the surface is necessary by an Alpha Counter
- For the cost and environment reason, gas purification and circulation system is welcome => if the XIA operation mode is respected

Merci

سپاس

(Persan)

Sépâsse

Thanks

Grazie

Gracias

धन्यवाद

28/11/2023