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The trinitites formation revealed by silicon and oxygen triple isotopes analyses.

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On July 16th, 1945, during the first nuclear explosion, glasses called trinitites formed as a nuclear by-product. They cover the surface of the explosion crater. The origins of the trinitites remain debated (Bonamici et al., 2017; Eby et al., 2010). Here, a scenario on the trinitites formation is proposed based on the chemical (realized with the Camparis Electron Microprobe) and the silicon and oxygen triple isotopes analysis (made with the CRPG-Nancy ion microprobe IMS1270) of three trinitites.

The trinitites are an assemblage of smaller glass beads and crystals from three chemical families: CaMgFe glasses, feldspar trinitites, and silica trinitites. Their silicon and oxygen isotopic compositions have wide variations between $12.4 \pm 0.6\text{‰}$ and $-15.6 \pm 0.6\text{‰}$ for $\delta^{30}\text{Si}$ and between $23.5 \pm 0.5\text{‰}$ and $1.5 \pm 0.5\text{‰}$ for $\delta^{18}\text{O}$. These wide variations are unusual for terrestrial solids. The silicon isotopic compositions can result from condensation and evaporation at high temperatures.

The trinitites chemical families show different origins: (i) the refractory rich CaMgFe trinitites are the result of a silicate vapor condensation ($\delta^{30}\text{Si} < 0\text{‰}$), and (ii) the silica trinitites are mainly a product of melting and evaporating sediments ($\delta^{30}\text{Si} > 0\text{‰}$). So, a scenario for the trinitites formation could be that the explosion vaporized a large amount of material and formed the crater. The surface of the crater had a high-temperature ($>1980\text{K}$) layer of melted sediment (quartz, feldspar...). In the nuclear fireball the silicate vapors condensed rapidly ($<10\text{s}$) into liquid droplets. These droplets fell into the melted sediment layer at the crater surface. Finally, the layer quenched and formed the trinitites.

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