





# STUC Project : **ST**udies of **U**ranium **C**ompounds

## Matthieu Lebois

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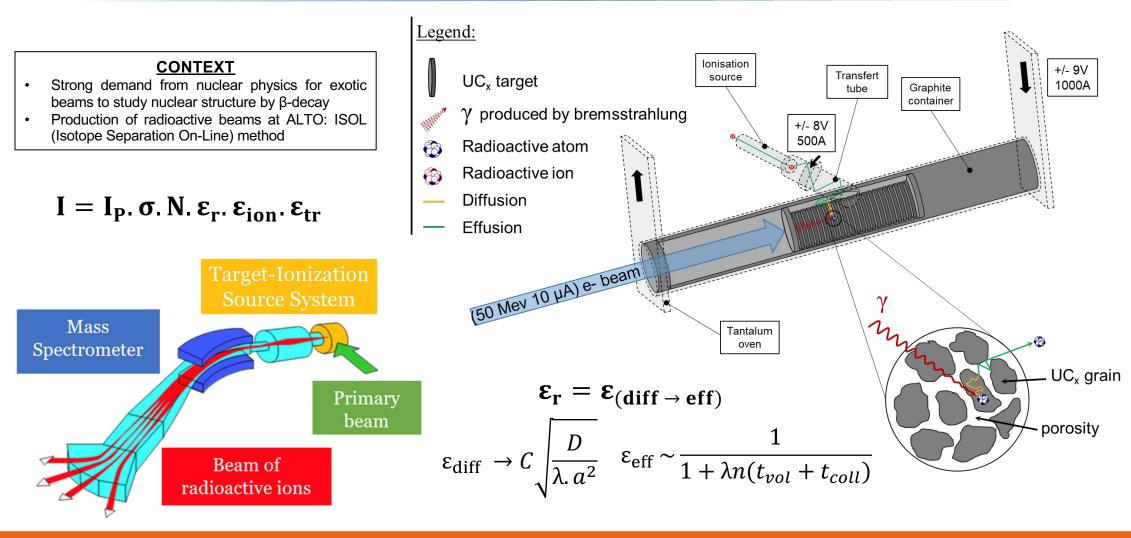
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workshop Targets - Ion Sources



#### Context

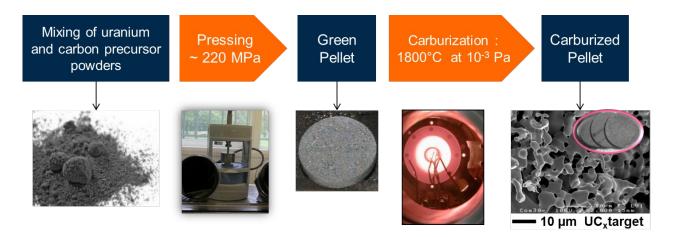




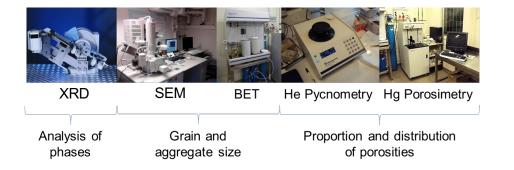
# How to build a target ? What caracterization?

• Synthesis of UC<sub>x</sub> target:  $UO_2 + 6C \rightarrow (1 - x)UC + xUC_2 + (3 - x)C + 2CO_{(g)}$ 

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Techniques used for the physico-chemical characterization of  $UC_x$  targets :



Equipment for measuring released fractions (off-line) and production (on-line) :



### Study on the influence of the microstructure

N° sample	Sample name	
1	UO <sub>2</sub> ground+CNT PM	
2	UO <sub>2</sub> ground+CNT UM	
3	UO <sub>2</sub> ground+graphene	
4	OXA+graphite PM	
5	OXA ground+CNT UM	
6	OXA+CNT UM	
7	PARRNe BP894	
8	PARRNe BP897 PM	
9	PARRNe BP897 PM 12d	
10	UO <sub>2</sub> ground+CNT PM 12d	
11	UO <sub>2</sub> ground+CNT UM 12d	
12	UO <sub>2</sub> ground+graphene 12d	
13	UO <sub>2</sub> ground+CNT-5mol UM	
14	UO2 ground+CNT-7mol UM	

#### Uranium precursor:

- Oxyde d'uranium
- Oxalate d'uranium

#### Carbone precursor:

- Graphite
- Graphene

• CNT

Molar ratio C/U 5 6 7

12-day heating after carburation :

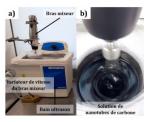
● Yes ○ No

#### Precursor powder mixing:

■ Robin<sup>™</sup> Powder Mixer



Ultrasonic liquid mixing

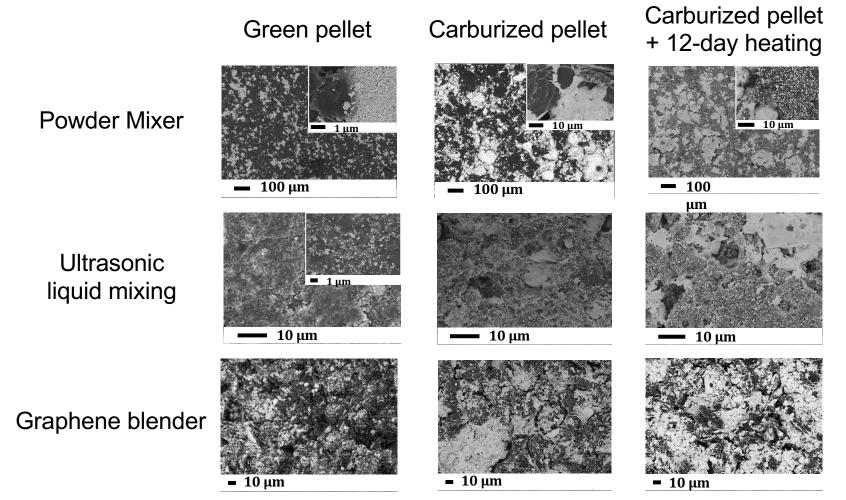


Graphene blender

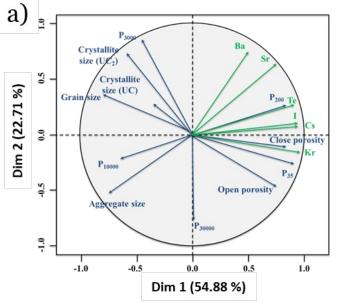




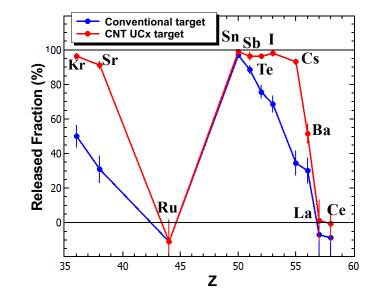
#### Study on the influence of the microstructure



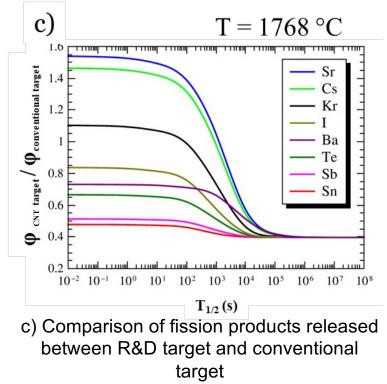




a) Correlations between release fractions (in green) and target properties (in blue)



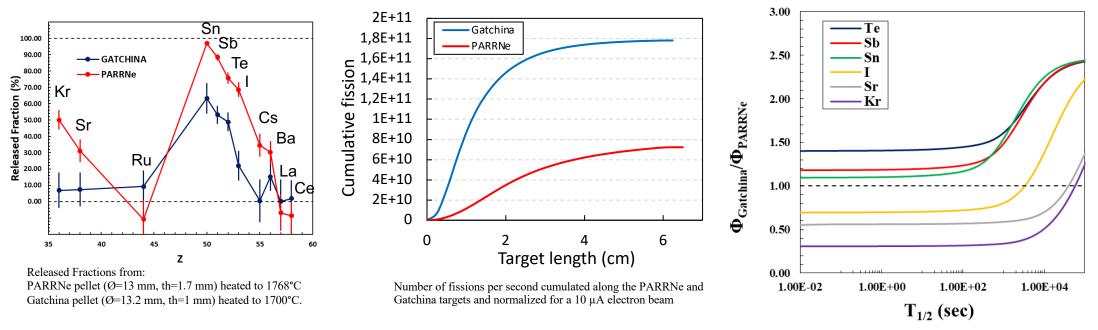
b) Released fraction comparison between conventional target (mostly UC<sub>2</sub>) and an R&D target made with CNT



Loss of production is compensated by improved release (Sr, Cs and Kr)



#### Study on the influence of the microstructure



Estimation of the production ratio between a Gatchina target and a PARRNe target

The loss in releases is compensated by improved production (Sn, Te and Sb)

Physico-chemical characteristics obtained by XRD and Helium pycnometry

		Gatchina	PARRNe
	UO <sub>2</sub>	4.5	-
Quantity of phases (%)	UC	86.9	5
	UC <sub>2</sub>	8.6	87
	С	-	8
Apparent density (g/cm <sup>3</sup> )		12.4	3.82
Porosity (%)	Open	5	51
	Close	2	5

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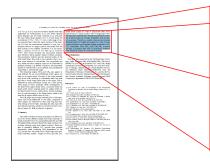
Elements	Release process	Target caracteristics
Sr	-	Low density
Cs	Diffusion <sup>1</sup>	Small grain
Kr	Diffusion <sup>2</sup>	Open pores
I	Diffusion/Effusion <sup>2</sup>	
Те	-	High density
Sb	-	Open pore (?)
Sn	Effusion <sup>2</sup>	

1: F. Hosni et al. NIM B 247 (2006) 205–209

2: B. Roussière et al. NIM B 246 (2006) 288-296

Table 1 List of the targets					
Target	UC2:C	UC2 particle size (µm)	Density (g/cm <sup>3</sup> )	Thickness (cm)	Production method
ANL- oxide	1:2	-	2.61	0.15	$UO_2 + C$
ANL 200	1:8	<250	5.65	0.076	CERAC UC <sub>x</sub>
ANL 325	1:3	<43	5.24	0.072	$U_{met} + C$
ANL 400	1:3	<37	5.49	0.077	$U_{met} + C$
ThO <sub>2</sub>	_	-	~7	_	Commercial
UB <sub>4</sub>	-	-	2.1	_	$UCl_4 + MgB_2$
Refrac	1:0.2	-	10.97	0.1138	Umet + C via UH





duced yields roughly an order of magnitude lower and a high-density UC<sub>2</sub> target another order of magnitude lower. The discussion indicates that the physical-chemical properties of these targets are strongly dependent upon production processes. Therefore, complete chemical and physical characterizations of targets are necessary in order to draw strong conclusions on the specific release properties of UC, UC<sub>2</sub>, oxycarbides, ThO<sub>2</sub>, ThC<sub>2</sub>, UB<sub>2</sub>, UB<sub>4</sub>, UB<sub>12</sub> as well as the type of graphite used. This is particularly necessary when designing targets for production facilities.

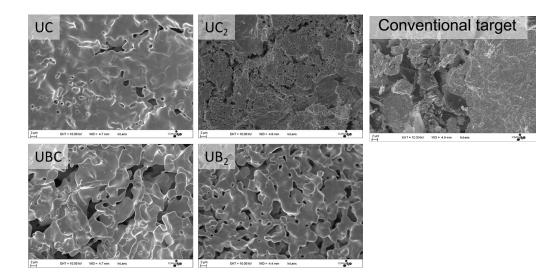
A. Kronenberg et al. Nucl. Instr. and Meth. in Phys. Res. B 266 (2008) 4267-4270

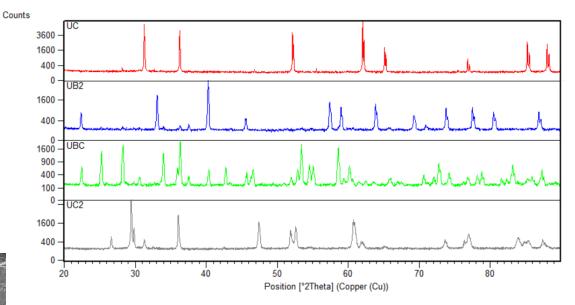
#### <u>Synthesis of various uranium compounds (UC, UBC, UB<sub>2</sub> and UC<sub>2</sub>):</u> (Aim: To study the influence of uranium alloy density on the release of fission products.)

- If we add porosity to a very dense target, does this improve release ?
- Is there a difference in release if we use uranium compounds with theoretical densities for UC, UBC, UB<sub>2</sub> and UC<sub>2</sub>? .
- Is there any influence of the chemical environment?



- $UO_{2(s)} + 4C_{(s)} \rightarrow UC_{2(s)} + 2CO_{(g)}$
- $UO_{2(s)} + 3C_{(s)} \rightarrow UC_{(s)} + 2CO_{(g)}$
- $UO_{2(s)} + 3C_{(s)} \rightarrow UC_{(s)} + 2CO_{(g)}$  $UC_{(s)} + BN_{(s)} \rightarrow UBC_{(s)} + N_{(g)}$
- $2UO_{2(s)} + 3C + B_4C_{(s)} \rightarrow 2UB_{2(s)} + 4CO_{(g)}$

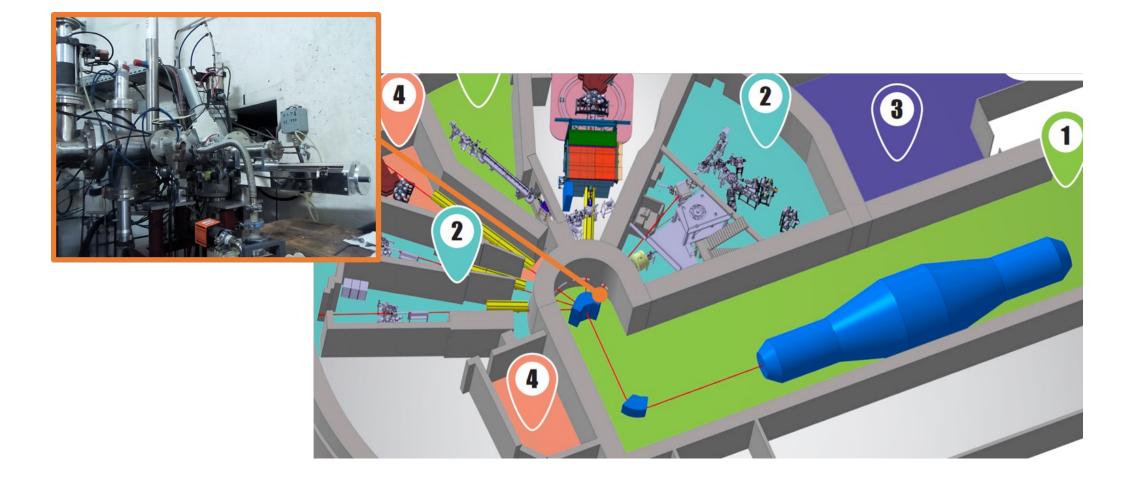




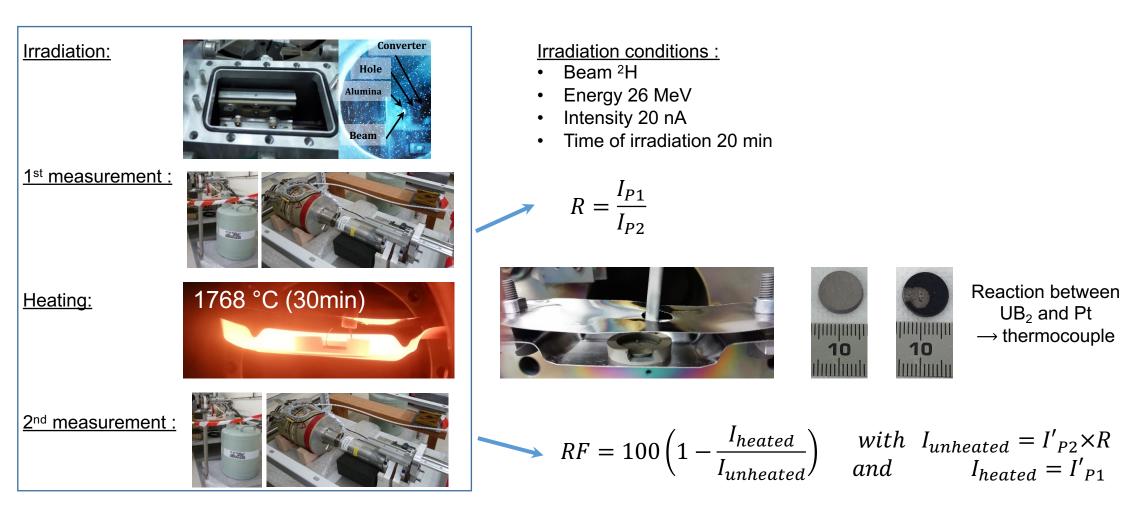
Target	Apparent density	Porosity (%)		SSA	Open pore size distribution (%)	
	(g/cm <sup>3</sup> )	open	close	(m²/g)	<10 µm	>150 µm
Conventional target $(UC_2 + 2C)$	4.5	44	1	0.4490	75	25
UC	8.16	39	1	0.0763	94	6
UB <sub>2</sub>	5.78	53	2	0.1032	80	20
UBC	6.93	42	1	0.0496	100	0
UC <sub>2</sub>	5.86	46	3	0.3965	88	12



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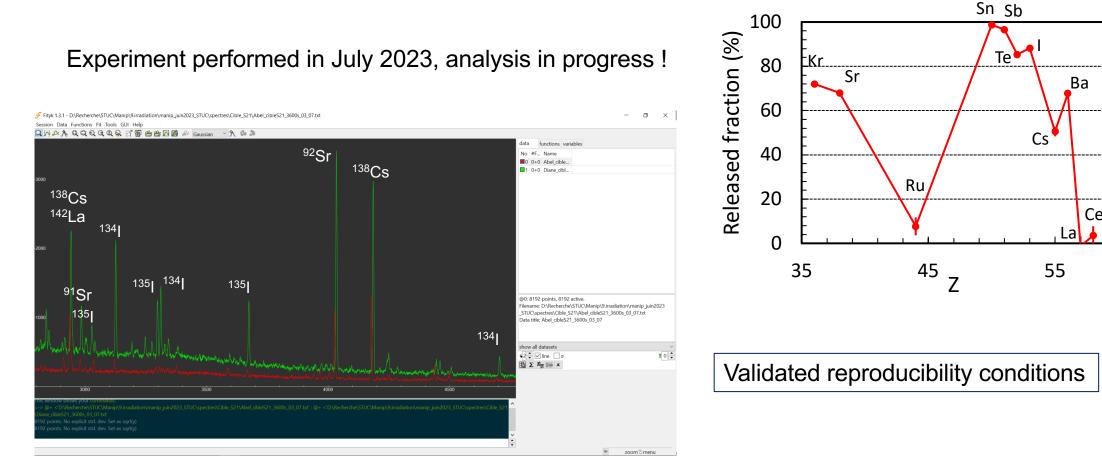








#### STUC PROJET: STudy of Uranium Compounds





Conclusion: there are no universal targets, but one target for each element.

The aim of this R&D project is to answer the following questions:

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- Is there any influence of the chemical environment ?



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# Thank you for your attention