Kilonovae and Charged Particle Thermalization

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Transient Universe 2023, Cargese

1 H	H ydrogen																	² He Helium
3	Li	⁴Be											⁵B	6 C	7 N	8 ∩	° F	Ne
	ithium	Beryllium											Boron	Carbon	IN NG AN	0	Fluorine	Neon
1		12											13	14	Nitrogen 15	0xygen 16	17	18
ľ	Na	Mg											Ä1	Si	Ρ	ຶS	ÜC1	År
	Sodium	Magnesium											Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon
19			21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
	Κ	Ca	Sc	Ti	V	Cr	Mn	Fe	Со	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
P	otassium	Calcium	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Gallium	Germanium	Arsenic	Selenium	Bromine	Krypton
3		38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
	Rb	Sr	Y	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Те	I	Xe
R	ubidium	Strontium	Yttrium	Zirconium	Niobium	Mo1ybdenum	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium	Tin	Antimony	Tellurium	Iodine	Xenon
5		56		72	73	74	75	76	77_	78	79	80	81	82	83	84	85	86
	Cs	Ba		Hf	Та	W	Re	0s	Ir	Pt	Au	Hg	T1	Pb	Bi	Po	At	Rn
(Cesium	Barium		Hafnium	Tantalum	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Polonium	Astatine	Radon
8	7_	88		104	105	106	107	108	109	110	111	112	113	114	115	116	117	118
	Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	F1	Mc	Lv	Ts	0g
F	rancium	Radium		Rutherfordium	Dubnium	Seaborgium	Bohrium	Hassium	Meitnerium	Darmstadtium	Roentgenium	Copernicium	Nihonium	Flerovium	Moscovium	Livermorium	Tennessine	Oganesson

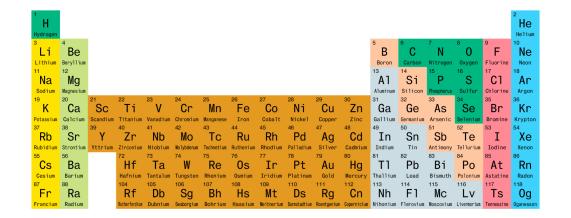
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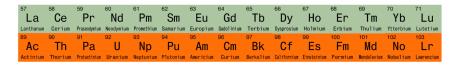
 Lanthau
 Cerium
 Prasodytiu
 Neodeliniu
 Promethiu
 Samarium
 Europiu
 Gdolinium
 Terbium
 Dysprostum
 Holenium
 Ebrium
 Thulium
 Ytterbium
 Lutium

 89
 90
 91
 92
 93
 94
 96
 96
 97
 98
 99
 100
 101
 102
 103

 Acc
 Th
 Pa
 U
 Np
 Pu
 Am
 Curium
 Earlewing
 Einsteinium
 Fermium
 Medelevium
 Nobelium
 Lamericum

1.Most H, He created during BB

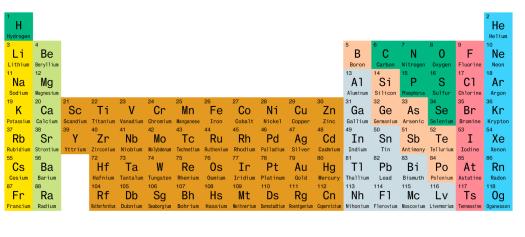




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2.Elements up to Fe created by nuclear fusion in stars or during

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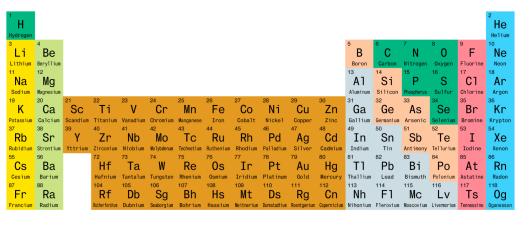
57 La	58 Ce	⁵⁹ Pr		⁶¹ Pm		⁶³ Eu		⁶⁵ Tb		⁶⁷ Ho			Yb	
Lanthanum	Cerium	Praseodymium										Thulium		
89	90	91	92			95			98	99	100	101	102	103
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
Actinium	Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	Lawrencium

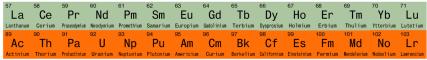
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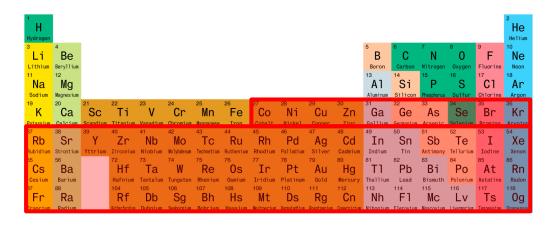
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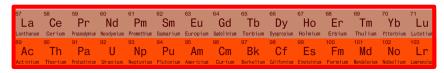
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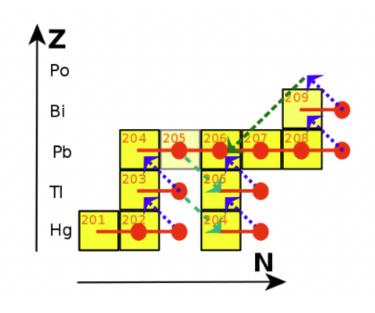




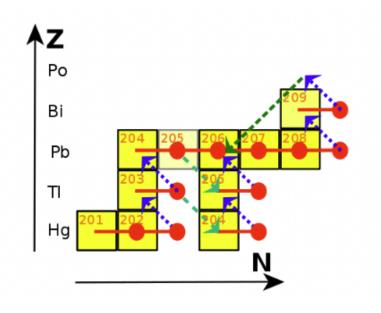
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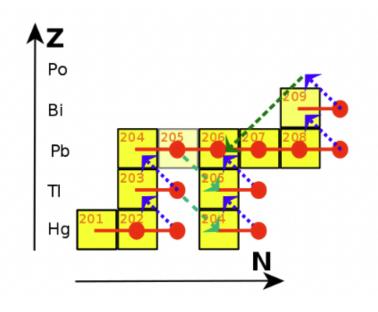


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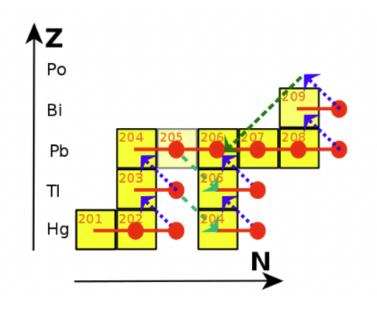


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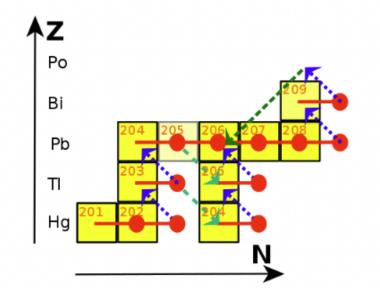
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4. Slow, inefficient, can't reach all elements.

1. Stops at lead, can't reach Uranium, thorium, etc.



2.Heavy elements, such as Uranium require an environment with a **large** neutron flux to allow a **rapid** capture of neutrons by nuclei - **before** beta-decay.

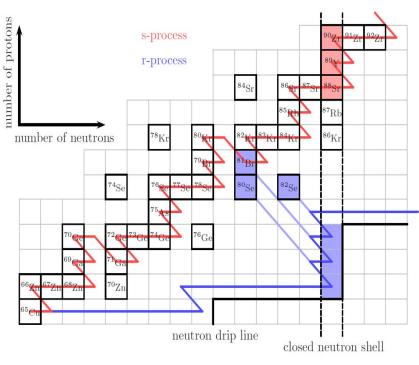
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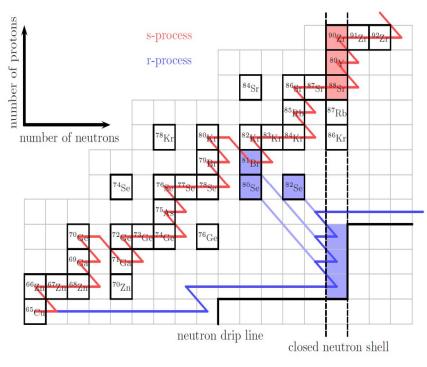
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5.We need neutron-rich environment

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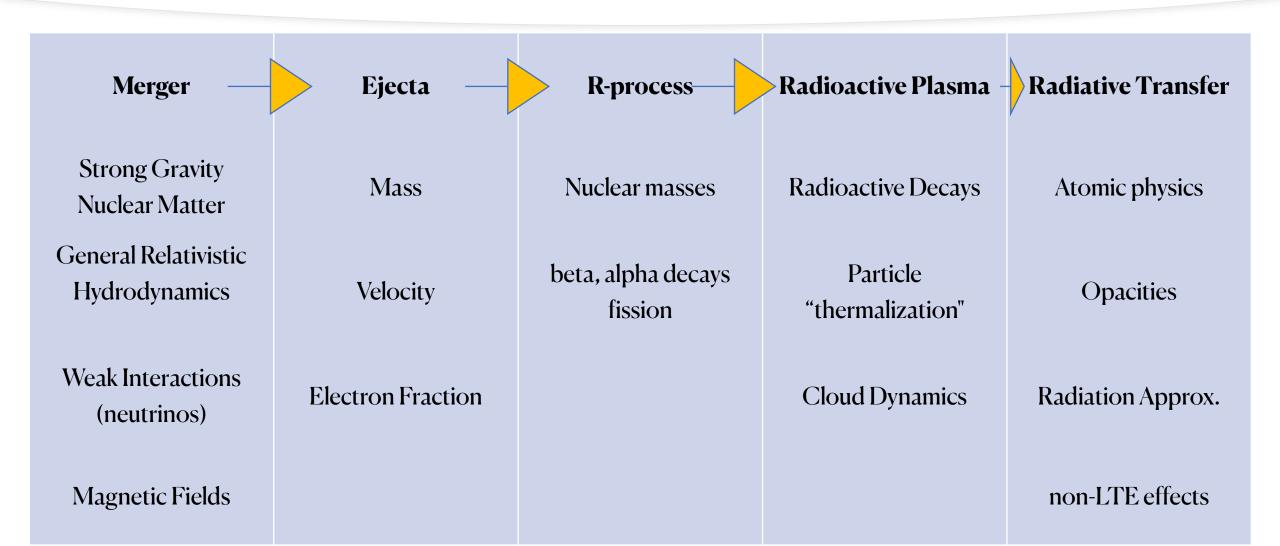
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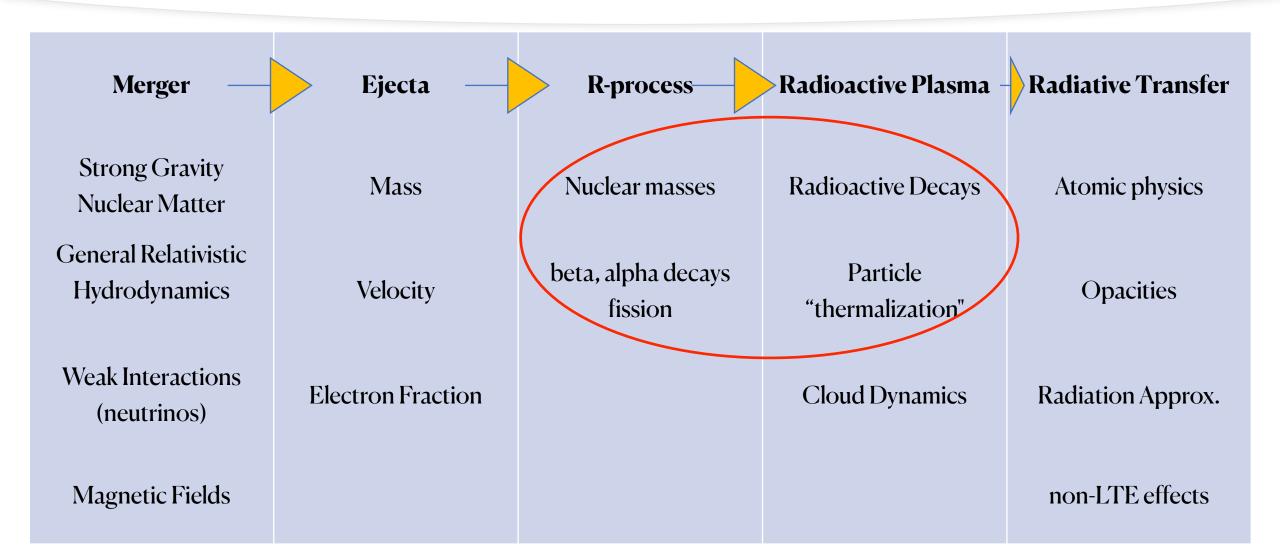
=> Can we tell which heavy elements, and what amount, were synthesized by analyzing kilonova's thermal emission?

* additional EM emission such as synchrotron, gamma-ray burst, etc.

Kilonovae Modeling Challenge



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GW170817 and Simulations

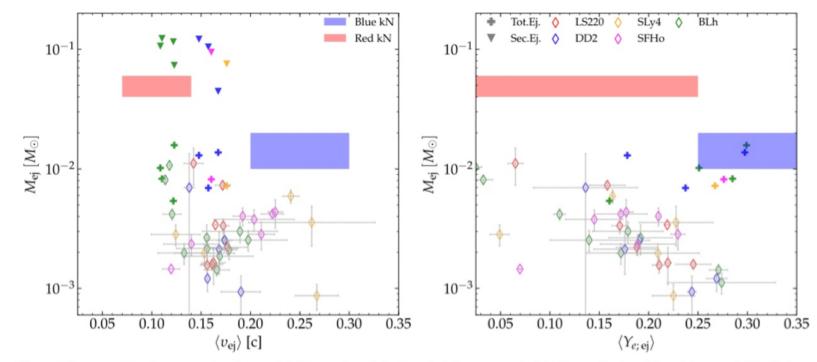


Figure 6. Summary of the ejecta properties of our models. Diamonds mark the dynamical ejecta, crosses include the contribution of the spiral-wave wind for the longlived models, and triangles are an estimate of the total ejecta mass on a secular timescale, assuming 40% of the disk mass is unbound on secular timescales. The ejecta mass is shown is terms of the mass-averaged velocity (left) and of the averaged electron fraction (right). The filled blue and red patches are the expected values of ejecta mass and velocity for blue and red components of AT2017gfo compiled by Siegel (2019), based on Villar et al. (2017).

Nedora et al., 2021

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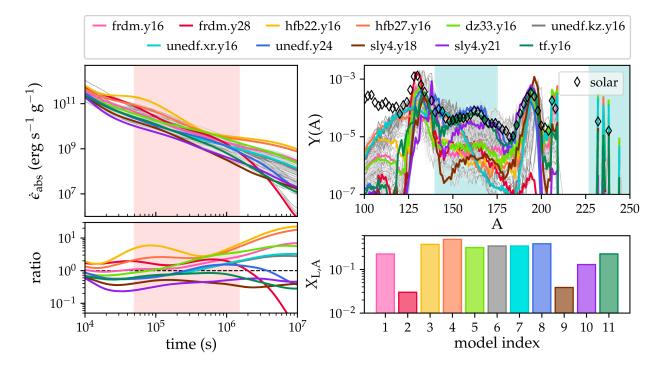
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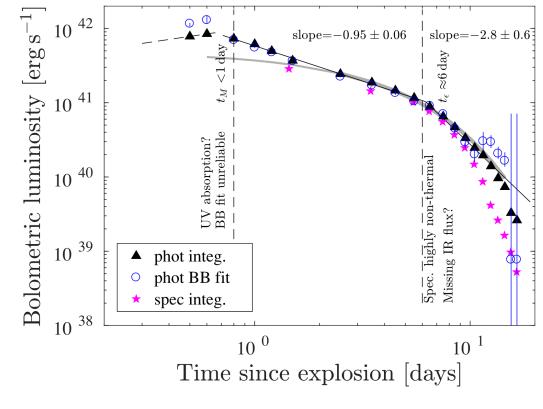
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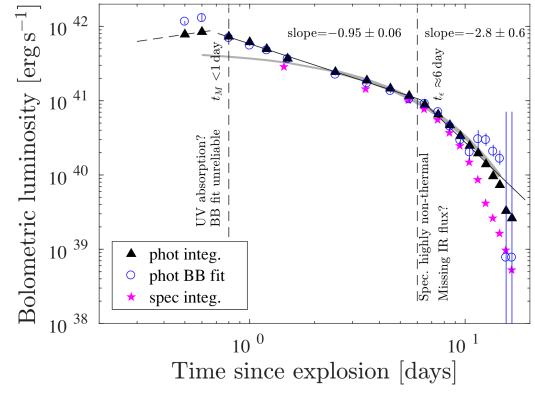


Barnes et al., 2020

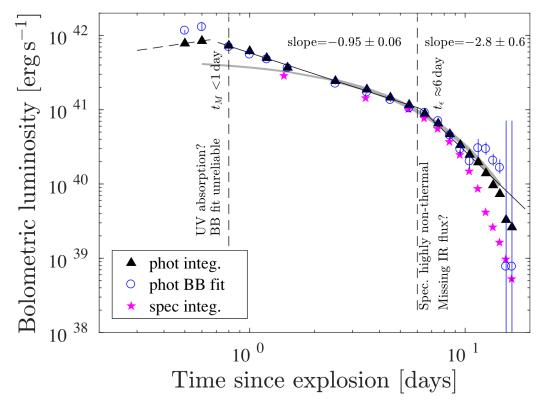


GW170817

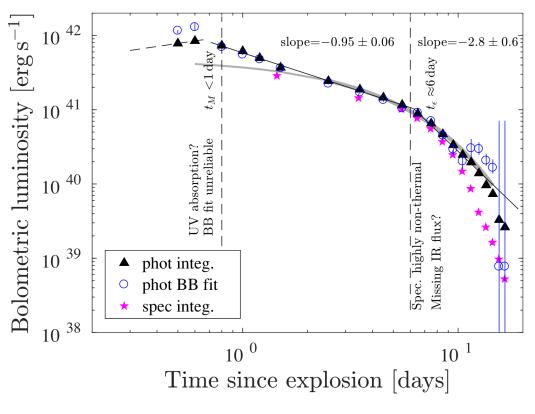
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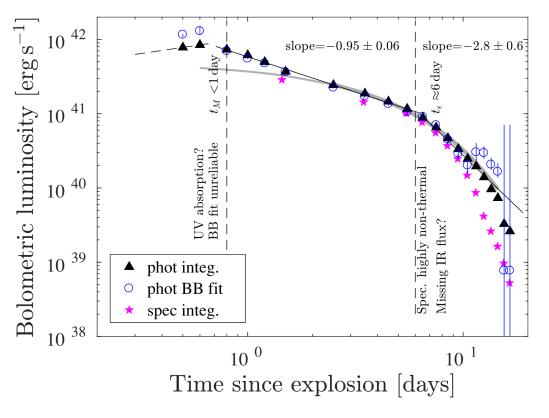
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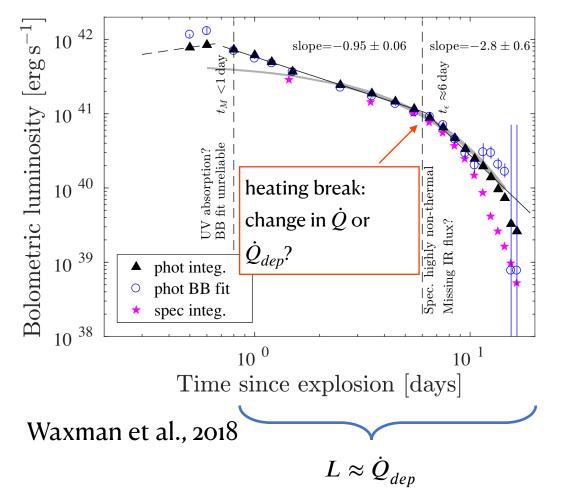
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- I examine decay particle thermalization for charged decay products (e, α -particles) based on extensive nucleosynthesis simulations.

How do Electrons Lose Energy?

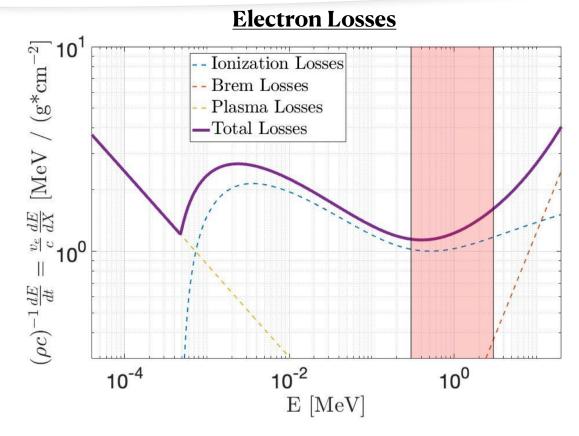


Figure 1: Energy loss rate of electrons propagating in a singly ionized $\chi_e = 1$ Xe plasma (Z = 54, A = 131). We take $\hbar\omega_p = 10^{-7} eV$. Shaded area shows typical average initial energies of β -decay electrons. For most relevant energies, ionization losses dominate.

Shenhar et al., in prep.

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- Three primary loss mechanisms:
 - plasma losses
 - ionization losses
 - Bremmstrahlung

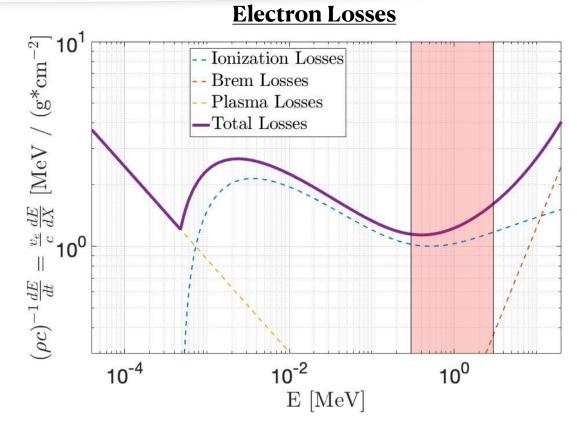


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adiabatic stopping power

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- We have some preliminary results... to be continued