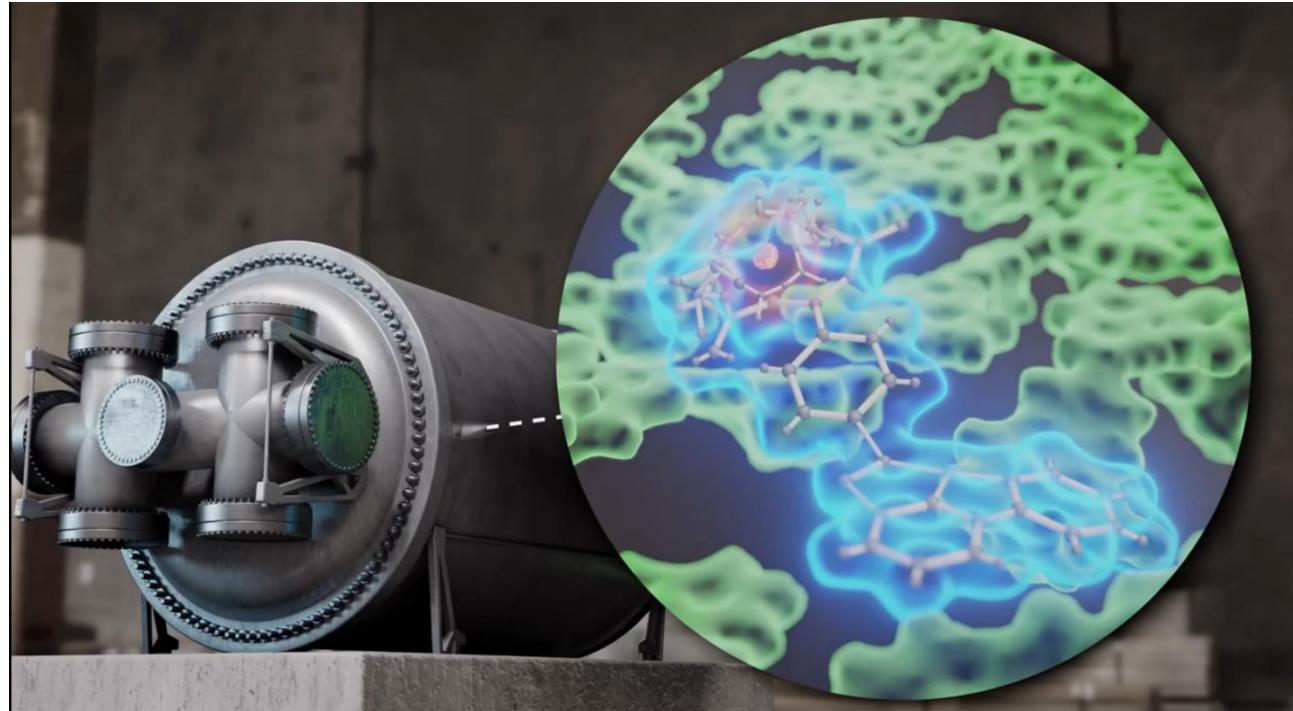


Searching the Grail: A background free $\beta\beta^0\nu$ experiment using Ba²⁺ tagging in a High-Pressure Xenon Chamber

Iván Rivilla and on behalf of

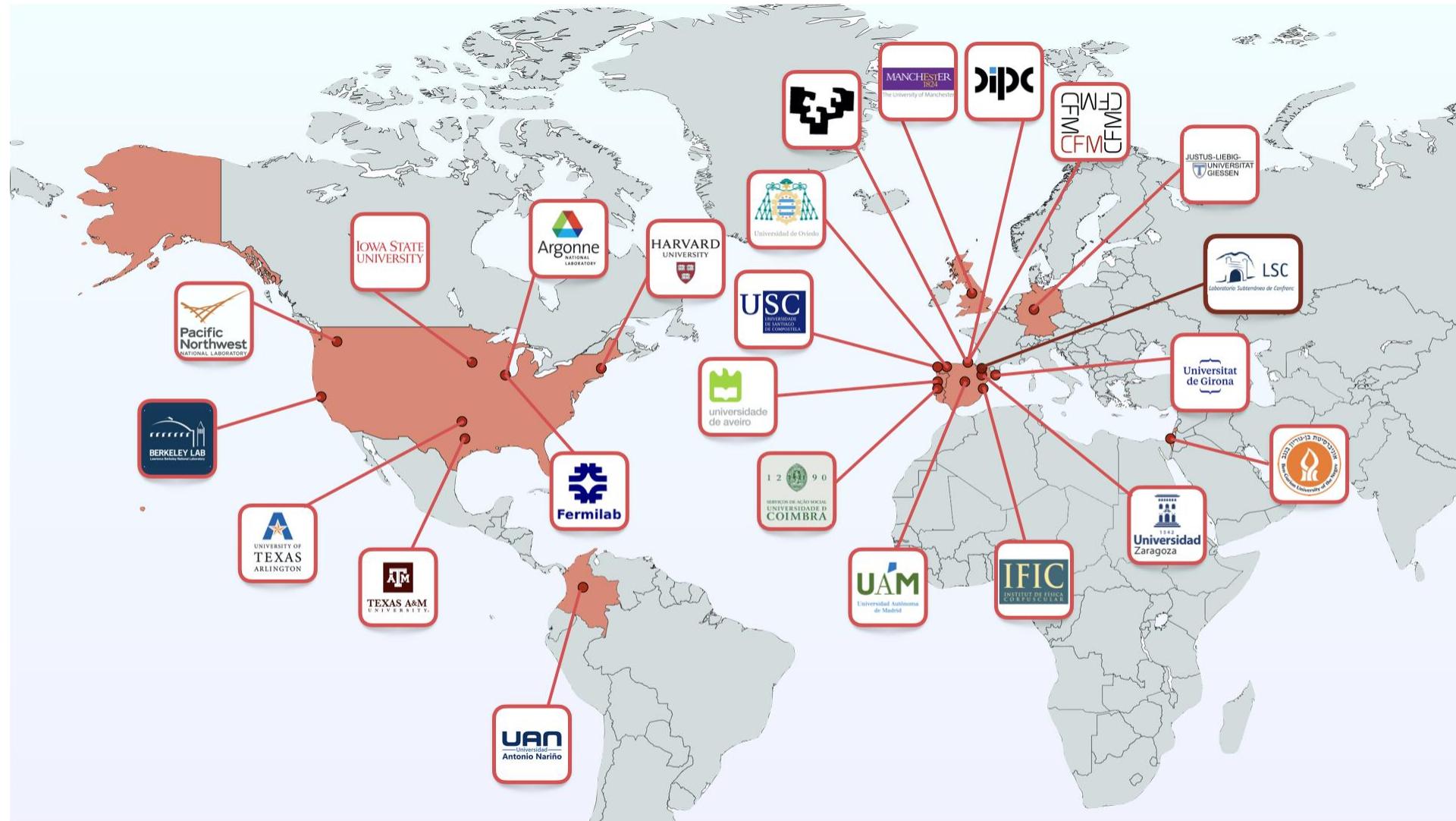


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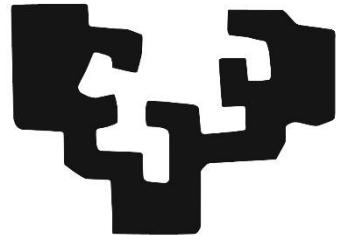
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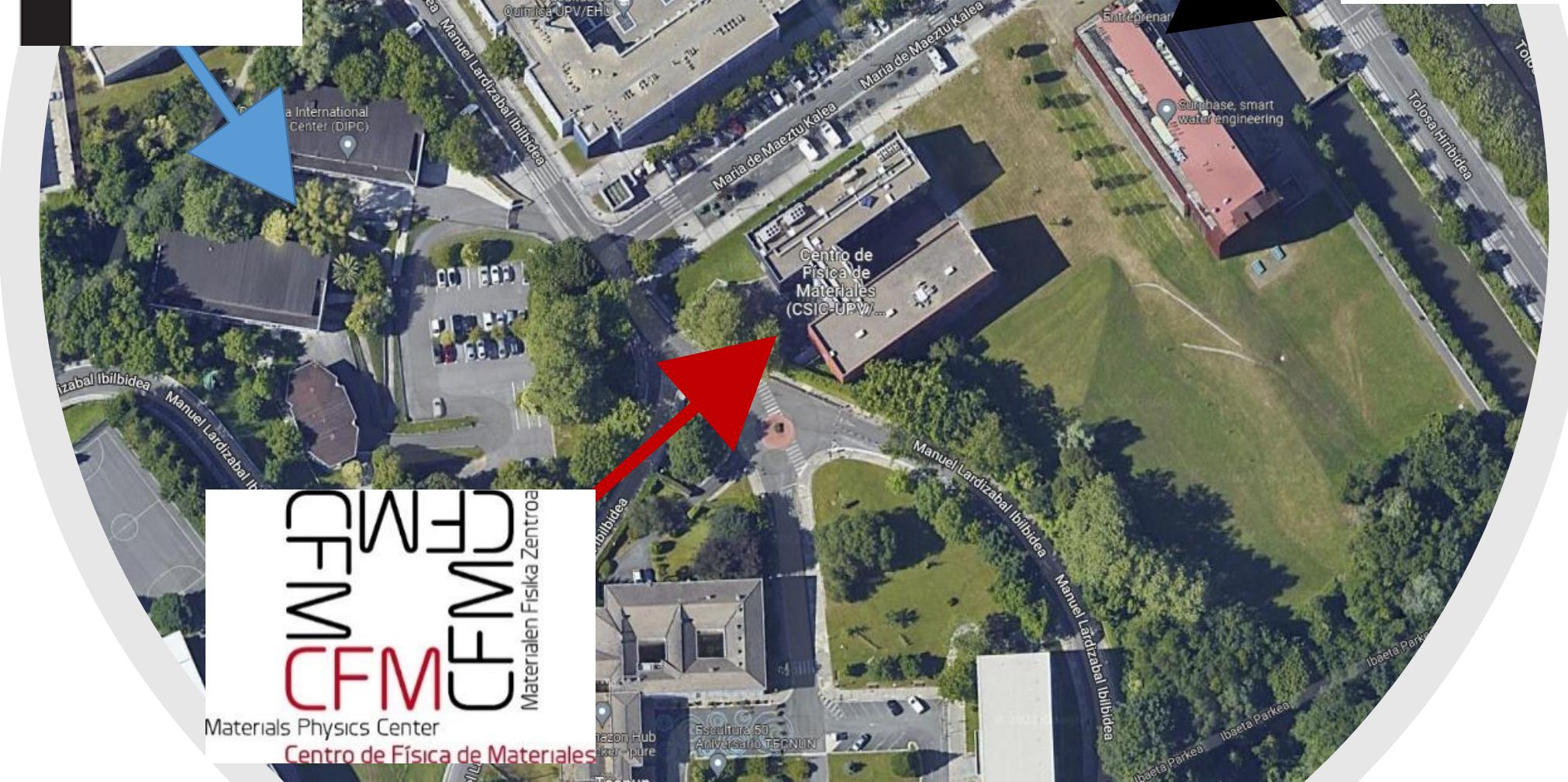






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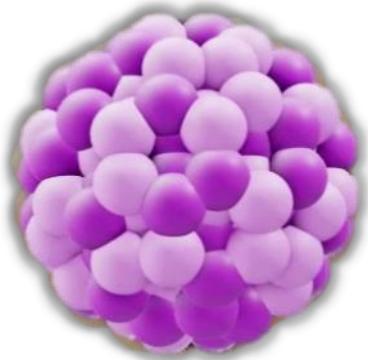
DIPC



CFM
Materials Physics Center
CFM Centro de Física de Materiales

How to confirm experimentally if neutrino is a Majorana particle?

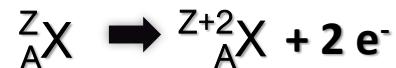
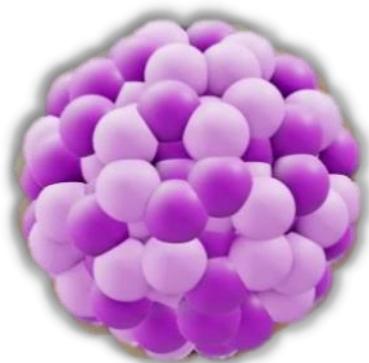
$\beta\beta 2\nu$



SM-allowed process
Measured in several nuclei

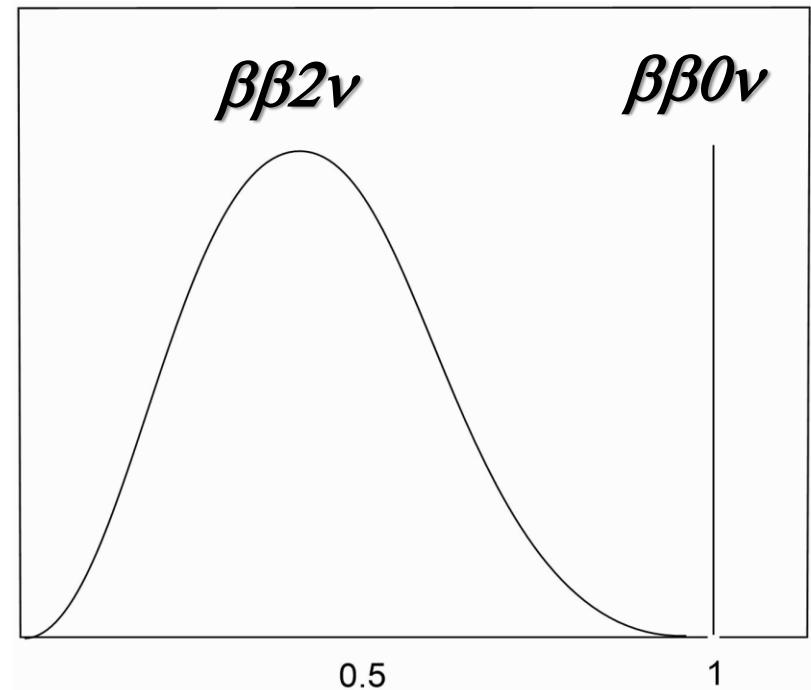
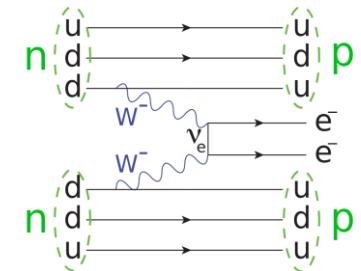
$$T_{1/2}^{2\nu} \sim 10^{19} - 10^{21} \text{ yr}$$

$\beta\beta 0\nu$



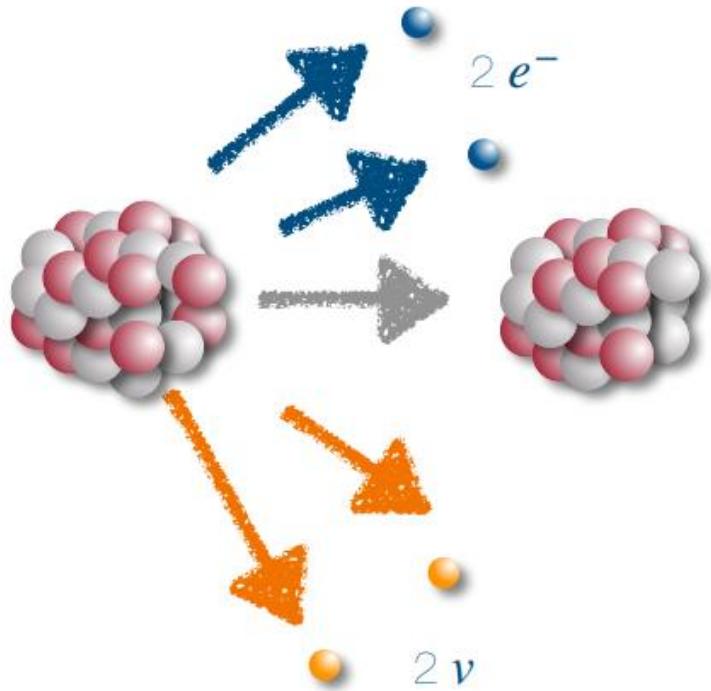
Lepton number violating process
Requires massive, Majorana neutrinos

$$T_{1/2}^{0\nu} > 10^{27} \text{ yr}$$

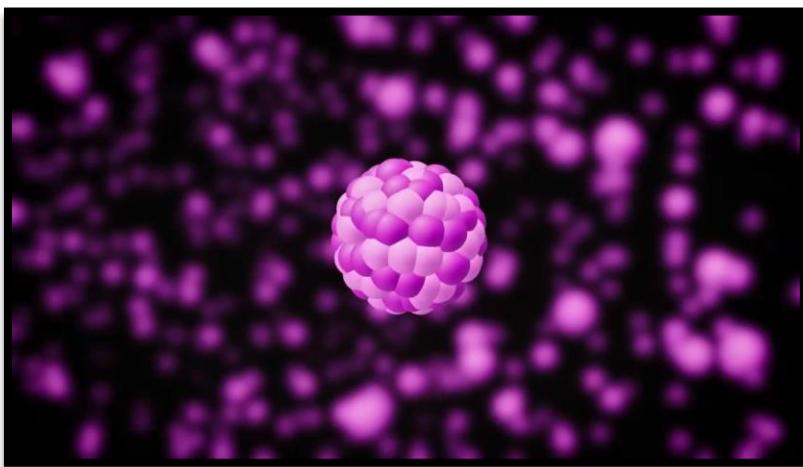


$$(T_{e1} + T_{e2}) / Q_{\beta\beta}$$

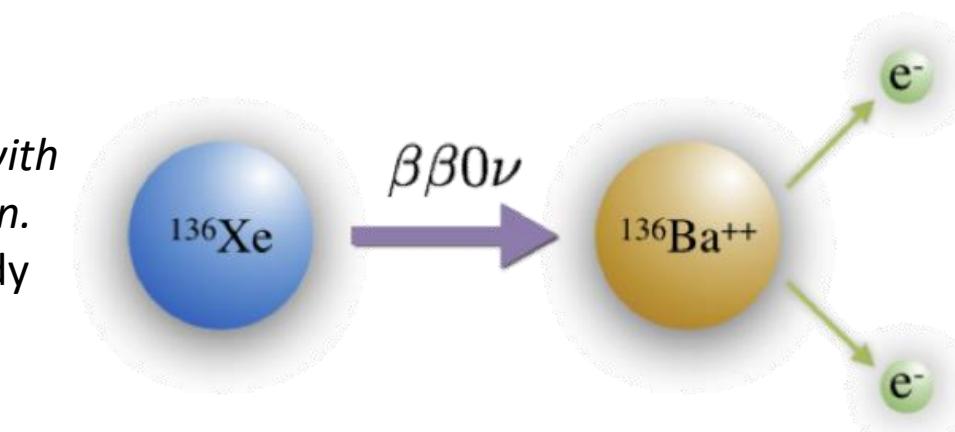
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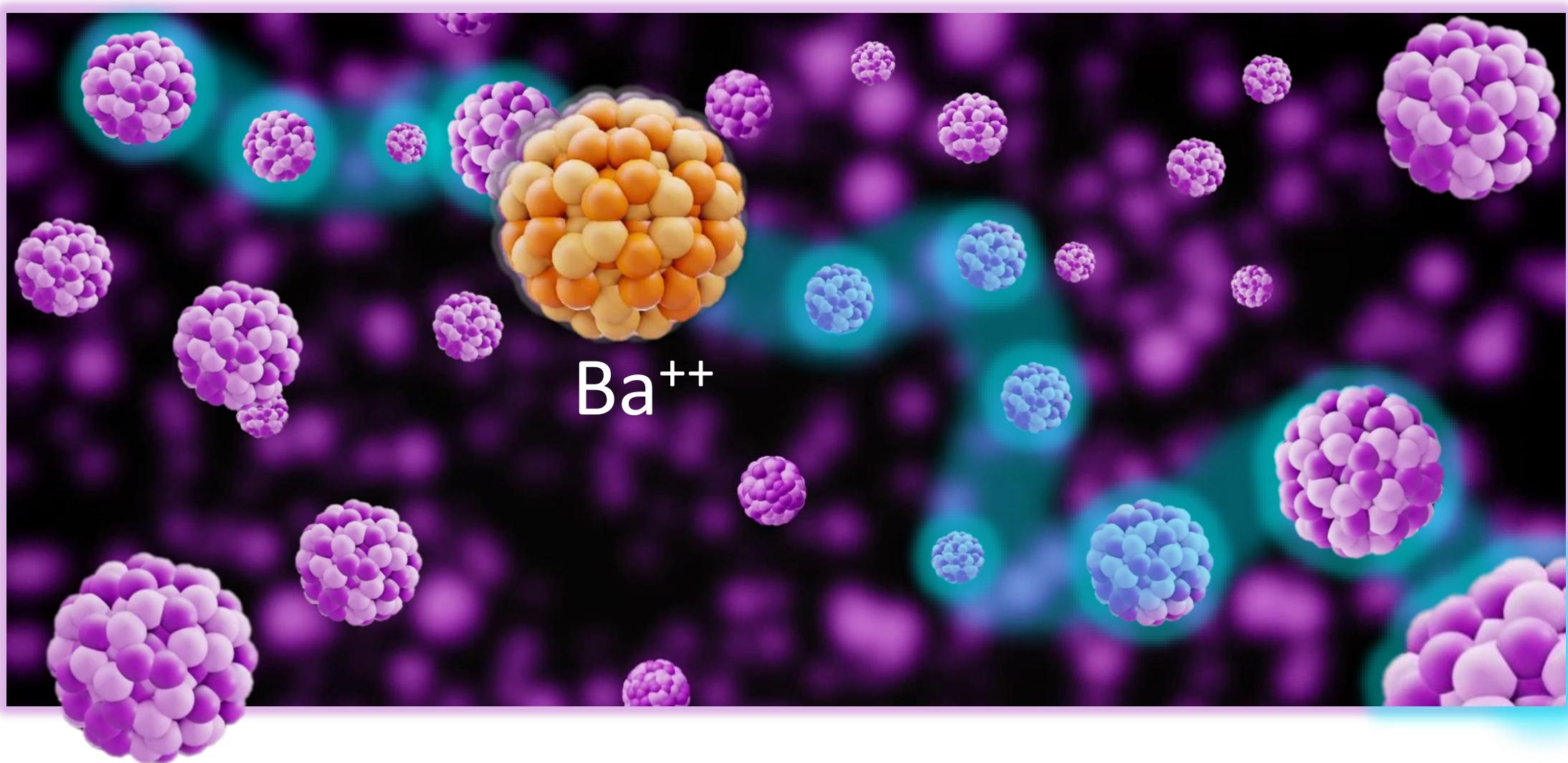
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1 H Hydrogen 1.008	2 Be Boron 6.94	3 Li Lithium 6.94	4 Be Boron 10.81	5 V	6 Cr Chromium 55.915	7 Mn Manganese 54.938044	8 Fe Iron 55.845	9 Co Cobalt 58.931794	10 Ni Nickel 58.6934	11 Cu Copper 63.546	12 Zn Zinc 65.40	13 Al Aluminum 26.9815395	14 Si Silicon 28.0855	15 P Phosphorus 30.973760988	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ne Neon 40.025902	
11 Na Sodium 22.989769	12 Mg Magnesium 24.305	19 K Potassium 39.0983	21 Sc Scandium 44.955908	22 Ti Titanium 47.867	23 V Vanadium 50.9415	24 Cr Chromium 51.976	25 Mn Manganese 54.938044	26 Fe Iron 55.845	27 Co Cobalt 58.931794	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.40	31 Ga Gallium 69.723	32 Ge Germanium 72.610606	33 As Arsenic 74.921605	34 Se Selenium 78.971	35 Br Bromine 79.904	36 Kr Krypton 83.800000
37 Rb Rubidium 85.4678	38 Sr Strontium 87.672	39 Y Yttrium 88.90584	40 Zr Zirconium 91.224	41 Nb Niobium 92.90637	42 Mo Molybdenum 95.94	43 Tc Technetium (98)	44 Ru Ruthenium 101.07	45 Rh Rhodium 101.927	46 Pd Palladium 106.42	47 Ag Silver 107.882	48 Cd Cadmium 112.464	49 In Indium 113.418	50 Sn Tin 118.739	51 Sb Antimony 121.760	52 Te Tellurium 127.88	53 I Iodine 126.447	54 At Astatine (221)	55 Xe Xenon (222)
55 Cs Caesium 132.90546	56 Ba Barium 133.227	57 - 71 Lanthanoids 130.94768	72 Hf Hafnium 178.49	73 Ta Tantalum 180.94768	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.23	78 Pt Platinum 191.024	79 Au Gold 196.967	80 Hg Mercury 204.38	81 Tl Thallium 204.272	82 Pb Lead 207.2	83 Bi Bismuth 208.98440	84 Po Polonium (209)	85 At Astatine (221)	86 Rn Radon (222)	
87 Fr Francium (223)	88 Ra Radium (226)	89 - 103 Actinoids (227)	104 Rf Rutherfordium (267)	105 Db Dubnium (266)	106 Sg Seaborgium (269)	107 Bh Bohrium (270)	108 Hs Hassium (269)	109 Mt Meitnerium (278)	110 Ds Darmstadtium (281)	111 Rg Roentgenium (282)	112 Cn Copernicium (285)	113 Nh Nihonium (286)	114 Fl Florium (288)	115 Mc Moscovium (288)	116 Lv Livermorium (293)	117 Ts Tennessee (294)	118 Og Oganesson (294)	



- ^{136}Xe ideal candidate
- Allows to build large detectors with Access to topological information.
- The NEXT experiment can already reconstruct this track



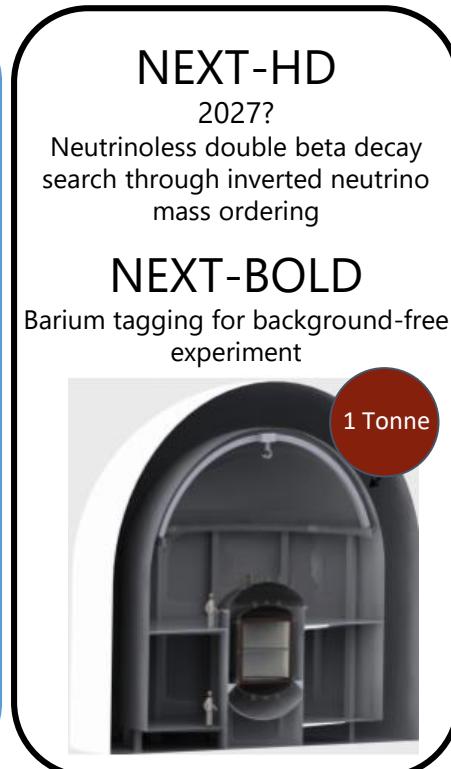
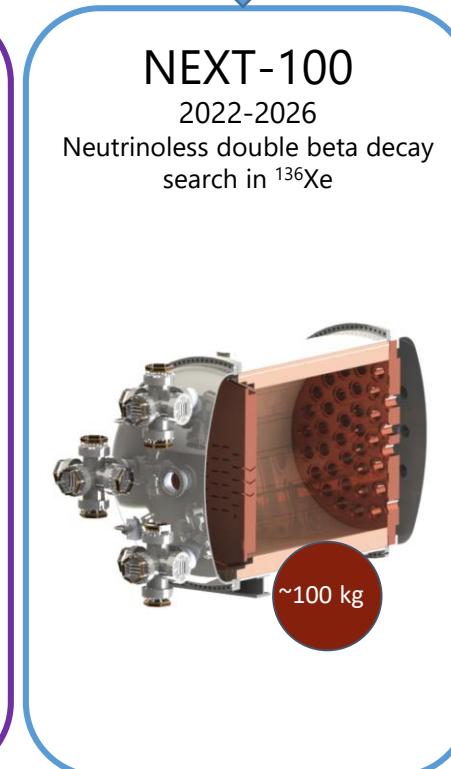
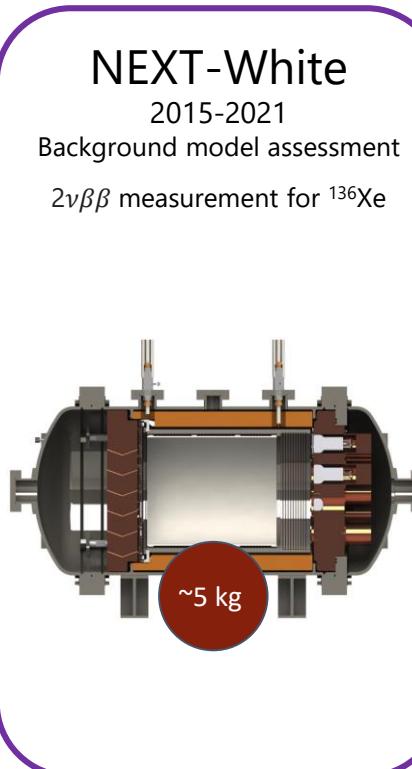
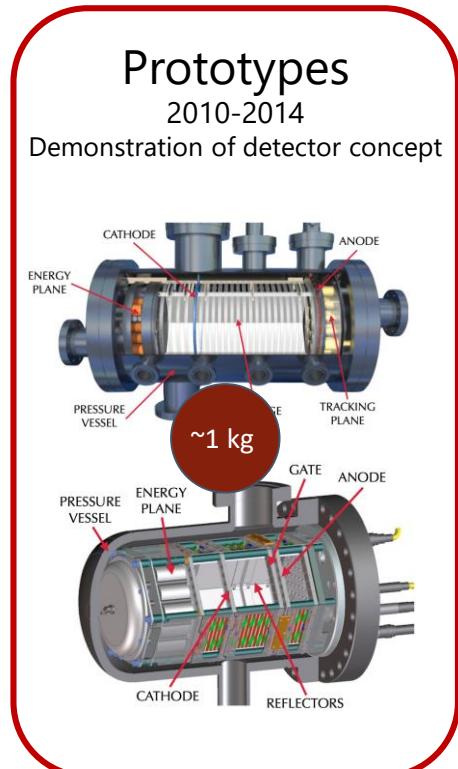
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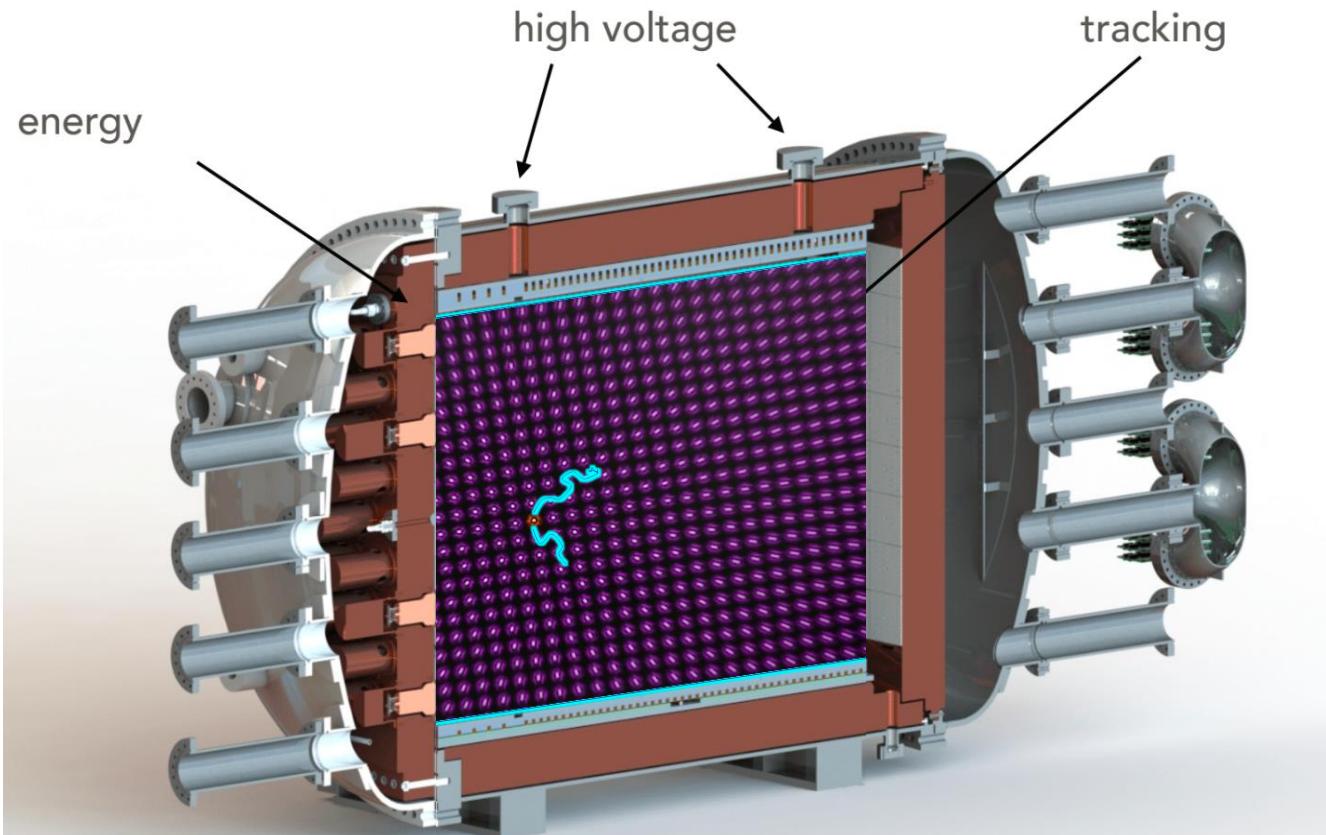
Detecting “tagging” the Ba^{++} signaling a $\beta\beta0\nu$ process has been a long sought holy grail of xenon chambers.

The NEXT Program

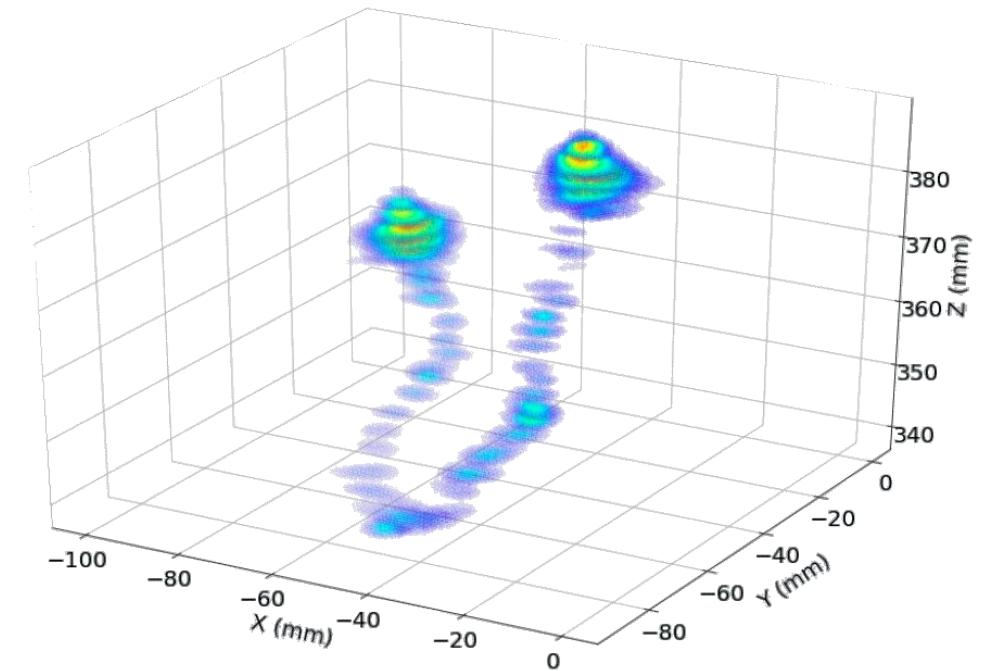
A. Usón
Talk



NEXT detectors capabilities



$\beta\beta0\nu$ in high-pressure ^{136}Xe gas

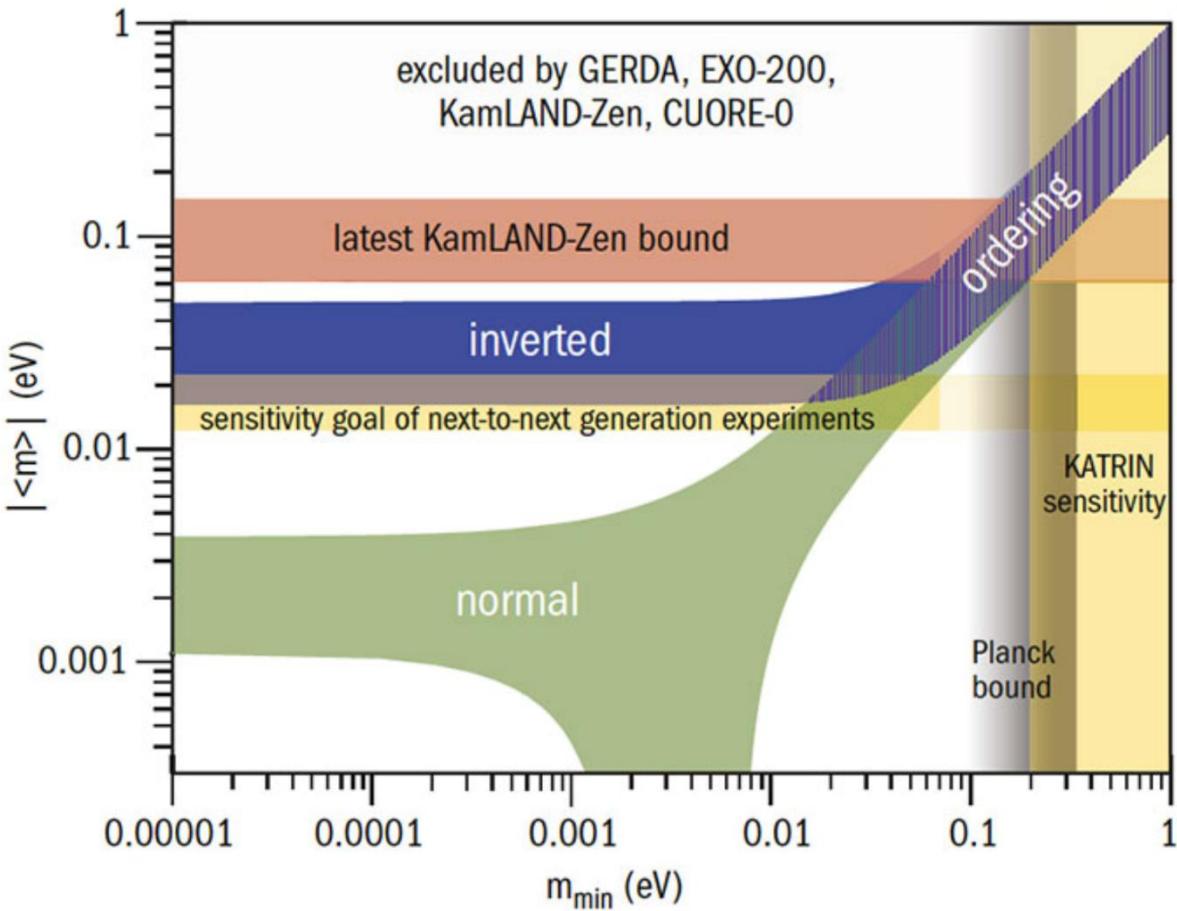


$\beta\beta2\nu$ candidate at 2 MeV, showing 2 energy blobs at the extremes.

Energy resolution better than 1% FWHM at $Q_{\beta\beta}$.

Detecting “tagging” the Ba^{++} signaling a $\beta\beta0\nu$ process has been a long sought holy grail of xenon chambers.

The NEXT thing:



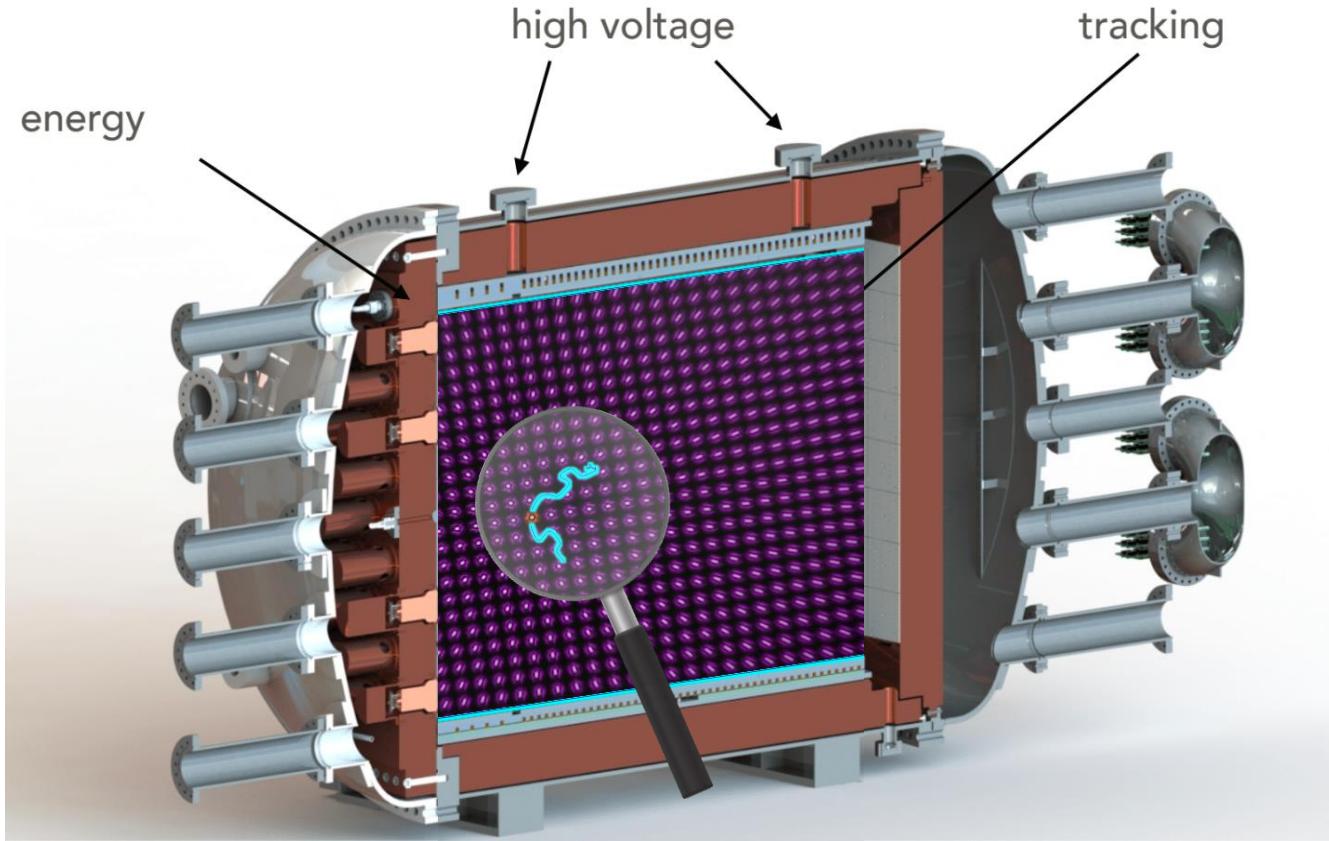
Current generation detectors “touching” inverse hierarchy régión.

Objective for the next generation is to cover this region.

Lifetimes $\sim 10^{27-28}$ years

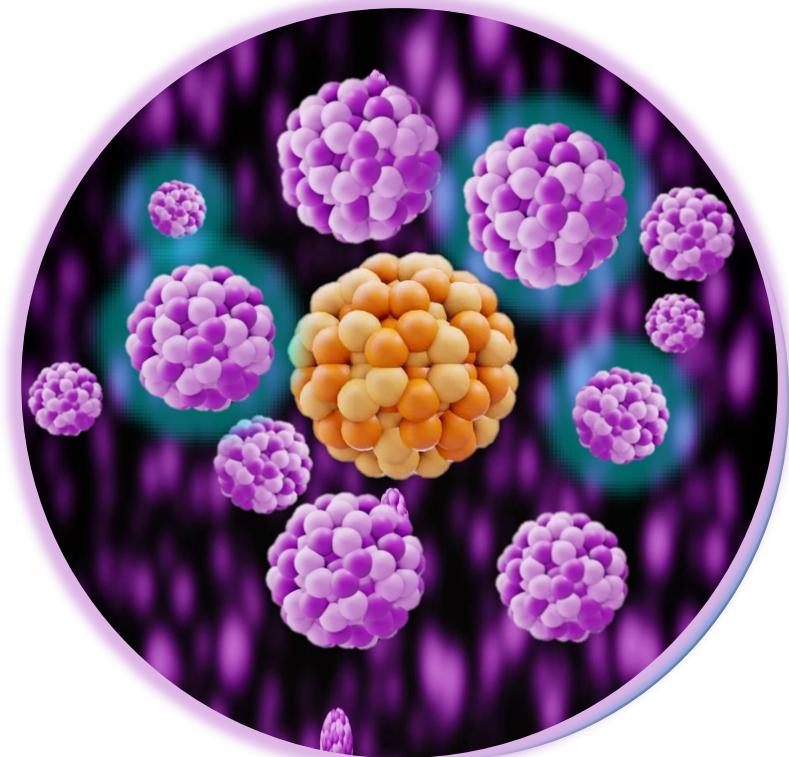
This implies almost background free detectors at the ca.Tonne-scale

How to get there with xenon gaseous detectors:



$\beta\beta0\nu$ in high-pressure ^{136}Xe gas

Ba^{++}



Detecting “tagging” the Ba^{++} signaling a $\beta\beta0\nu$ process has been a long sought holy grail of xenon chambers.

Detecting the barium daughter in ^{136}Xe $0-\nu\beta\beta$ decay using single-molecule fluorescence imaging techniques

David R. Nygren

Department of Physics, University of Texas at Arlington
Box 19059, Arlington, TX 76015

nygren@uta.edu

Abstract. Single-molecule fluorescent imaging may provide an avenue to efficiently detect the Ba^{++} daughter atom in the decay $^{136}\text{Xe} \rightarrow \text{Ba} + 2\text{e}^-$, and, unambiguously associate the birth point in space within the electron trajectories of the decay event. Chelation of doubly-charged alkaline earth elements such as calcium and barium by certain precursor molecules converts the resulting complex from a non-fluorescent to a fluorescent state. Repeated photo-excitation of a single fluorescent complex reveals both presence and location with high precision. This technique, widespread now in biochemistry, biophysics and biology, may permit a similar discriminating response in a large high-pressure xenon gas TPC for the Ba^{++} ion from xenon double-beta decay. The TPC measures the event time and energy of the two nascent electrons, as well as topology and position in 3-D from their trajectories in the gas. Measurement of the 2-D location of the molecular ion after arrival at the cathode plane permits an association of ion with the event. Demonstration of an efficient, highly specific detection of the barium daughter would provide a long-sought pathway to a background-free result in the search for this decay mode, of central importance for determining the nature of the neutrino.

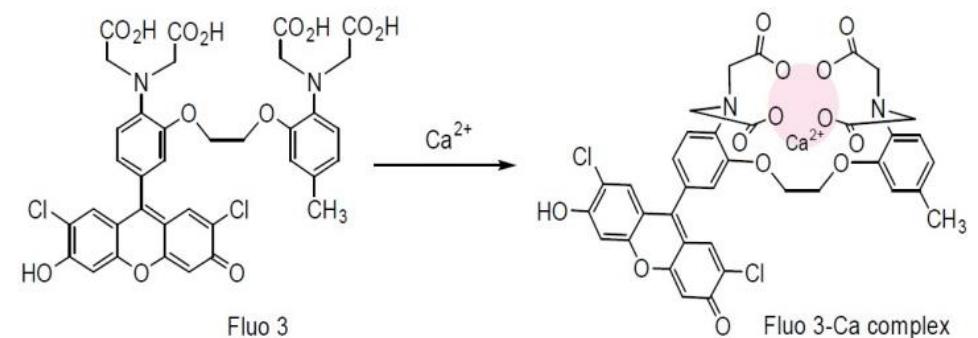
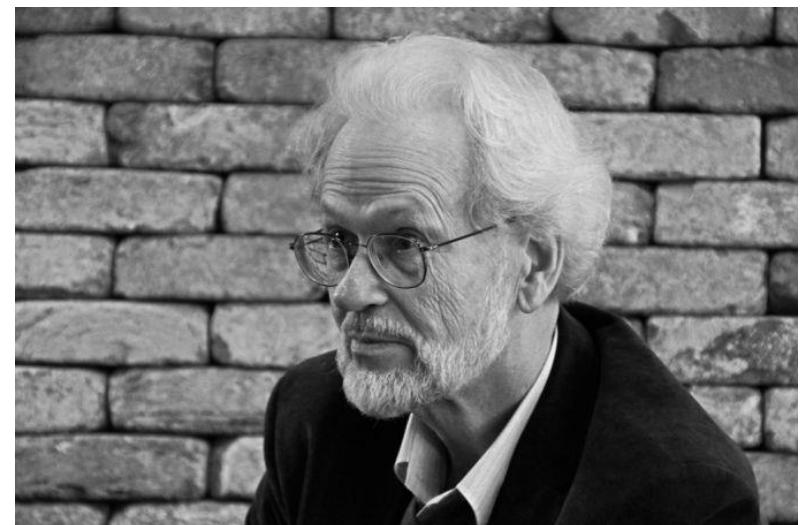


Figure 1. The skeletal formula for Fluo-3 is shown before and after complex formation. The conformal change that Fluo-3 undergoes in chelation with Ca^{2+} creates a fluorescent complex. In that state, the fluorescent response increases by a factor of 60 - >100 in the cellular environment. *Image source:* Dojindo Molecular Industries, Inc.

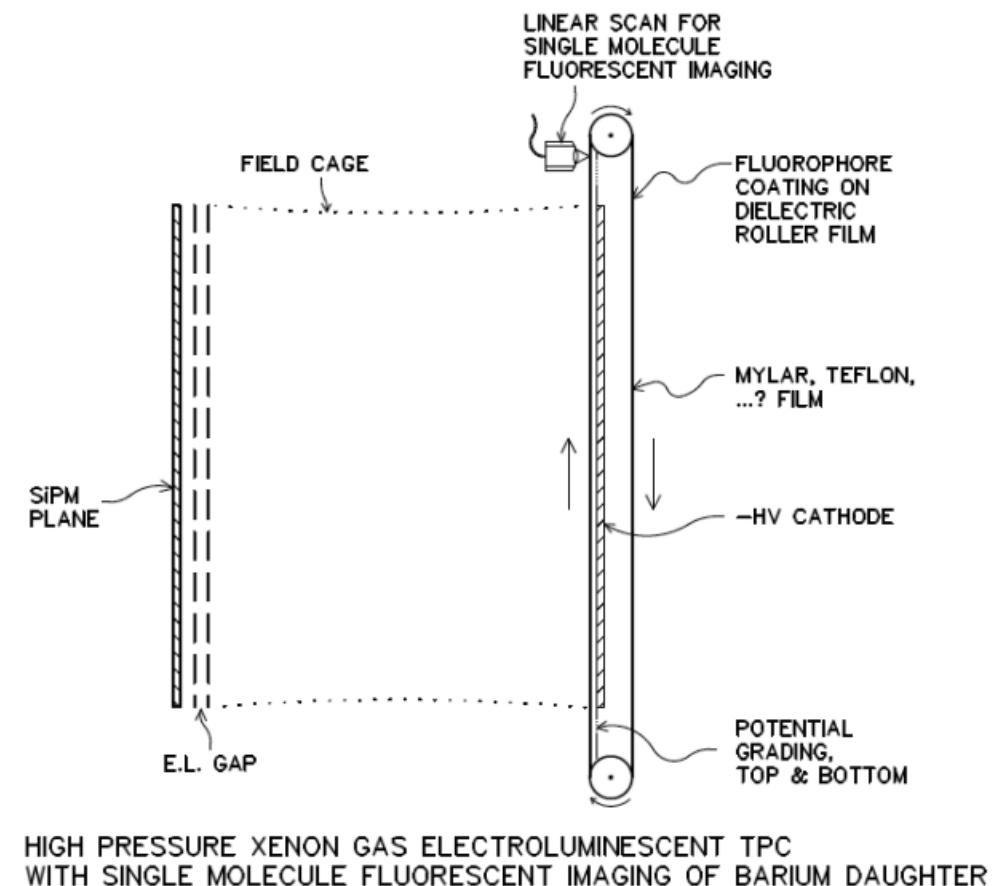
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Demonstration of Single-Barium-Ion Sensitivity for Neutrinoless Double-Beta Decay Using Single-Molecule Fluorescence Imaging

A. D. McDonald,^{1,†} B. J. P. Jones,^{1,‡} D. R. Nygren,^{1,§} C. Adams,² V. Álvarez,³ C. D. R. Azevedo,⁴ J. M. Benlloch-Rodríguez,³ F. I. G. M. Borges,⁵ A. Botas,³ S. Cárcel,³ J. V. Carrión,³ S. Cebrián,⁶ C. A. N. Conde,⁵ J. Díaz,³ M. Diesburg,⁷ J. Escada,⁵ R. Esteve,⁸ R. Felkai,³ L. M. P. Fernandes,⁹ P. Ferrario,³ A. L. Ferreira,⁴ E. D. C. Freitas,⁹ A. Goldschmidt,¹⁰ J. J. Gómez-Cadenas,^{3,§} D. González-Díaz,¹¹ R. M. Gutiérrez,¹² R. Guenette,² K. Hafidi,¹³ J. Hauptman,¹⁴ C. A. O. Henriques,⁹ A. I. Hernandez,¹² J. A. Hernando Morata,¹¹ V. Herrero,⁸ S. Johnston,¹⁵ L. Labarga,¹⁶ A. Laing,³ P. Lebrun,⁷ I. Liubarsky,³ N. López-March,^{1,3} M. Losada,¹² J. Martín-Albo,² G. Martínez-Lema,¹¹ A. Martínez,³ F. Monrabal,¹ C. M. B. Monteiro,⁹ F. J. Mora,⁸ L. M. Moutinho,⁴ J. Muñoz Vidal,³ M. Musti,³ M. Nebot-Guinot,³ P. Novella,³ B. Palmeiro,³ A. Para,⁷ J. Pérez,³ M. Querol,⁸ J. Repond,¹⁵ J. Renner,³ S. Riordan,¹⁵ L. Ripoll,¹⁷ J. Rodríguez,³ L. Rogers,¹ F. P. Santos,⁵ J. M. F. dos Santos,⁹ A. Simón,³ C. Sofka,^{3,||} M. Sorel,³ T. Stiegler,¹⁸ J. F. Toledo,⁸ J. Torrent,³ Z. Tsamalaidze,¹⁹ J. F. C. A. Veloso,⁴ R. Webb,¹⁸ J. T. White,^{18,*} and N. Yahlali³

(NEXT Collaboration)

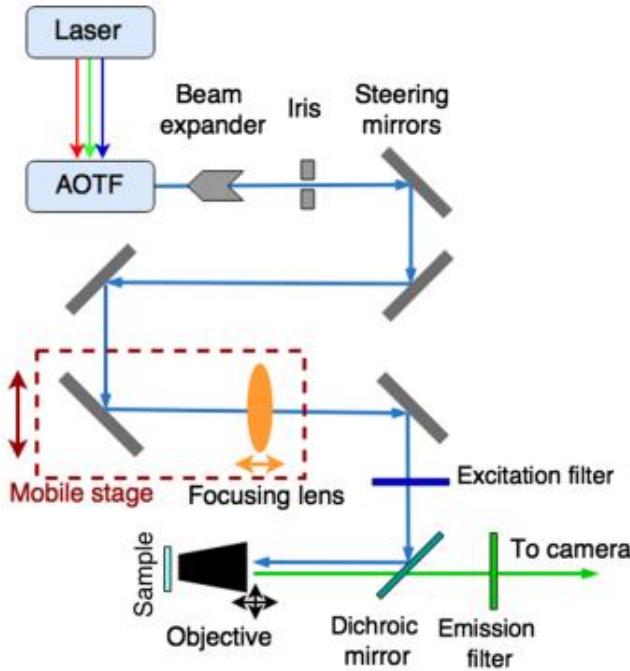


FIG. 2. A schematic view of the TIRF system.

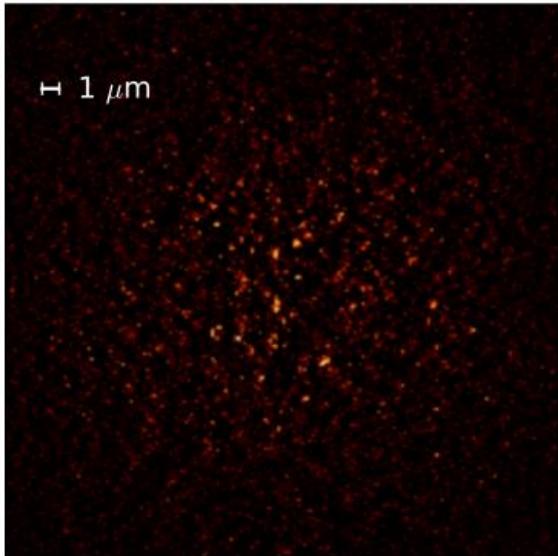


FIG. 3. A sample image from the EM-CCD in one of the barium-spiked samples showing both near-surface (bright) and deeper (dim) fluorescent molecules.

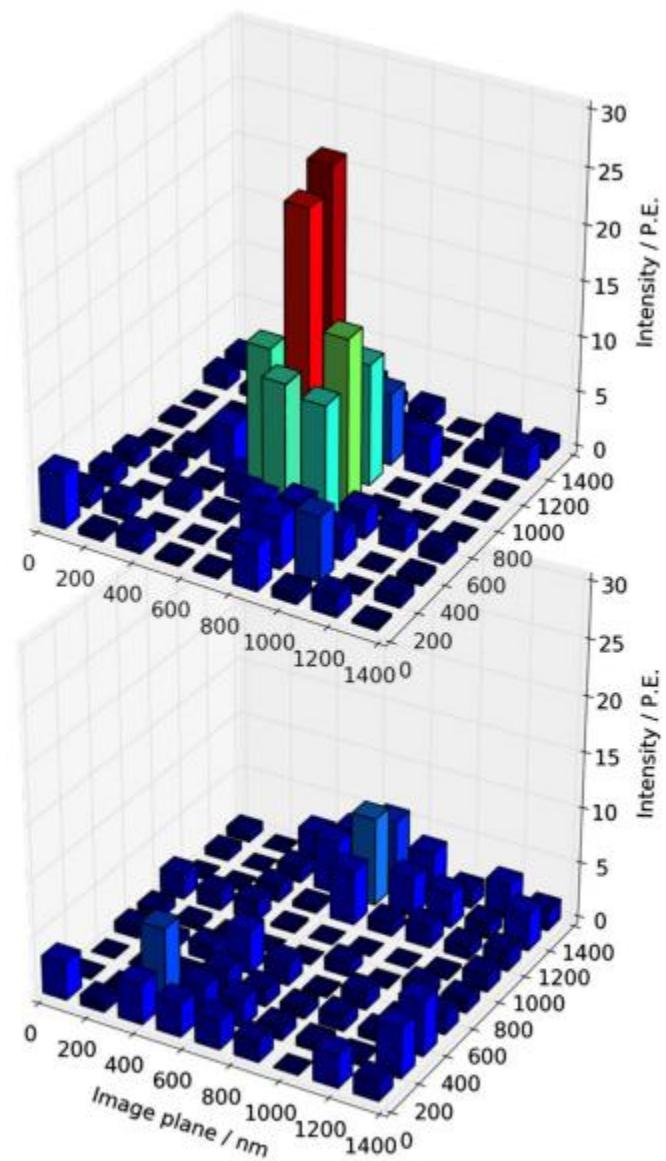
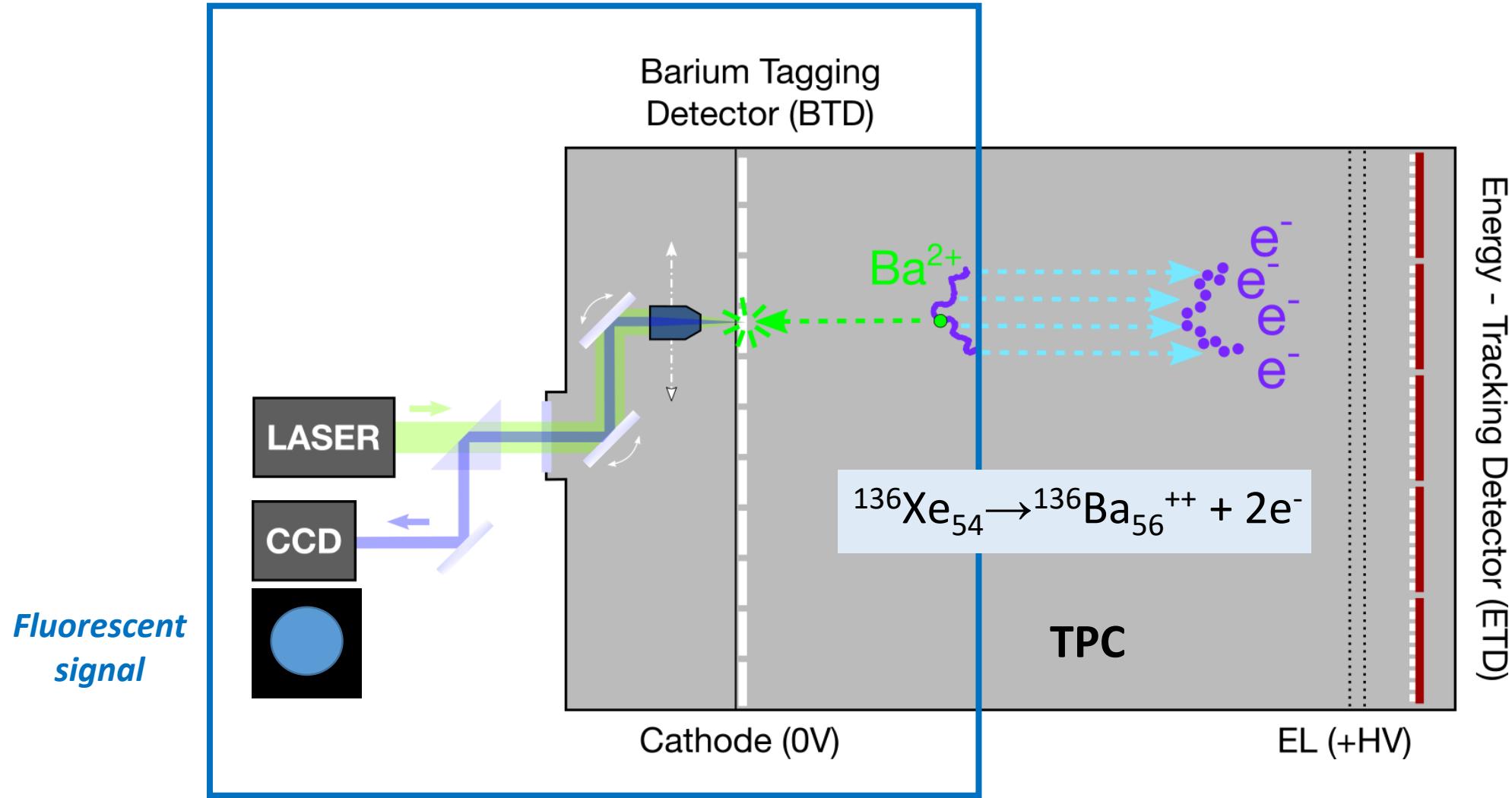


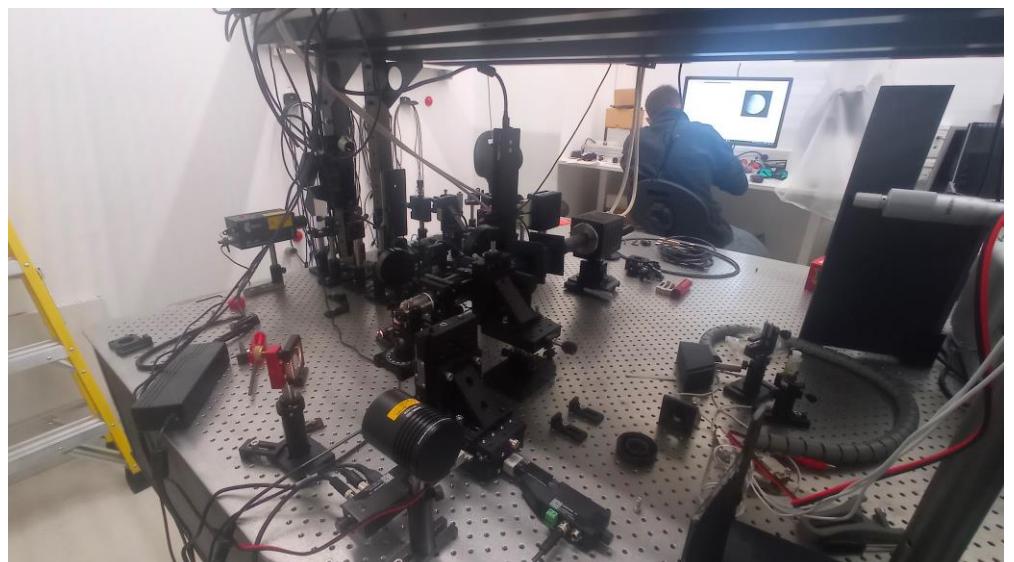
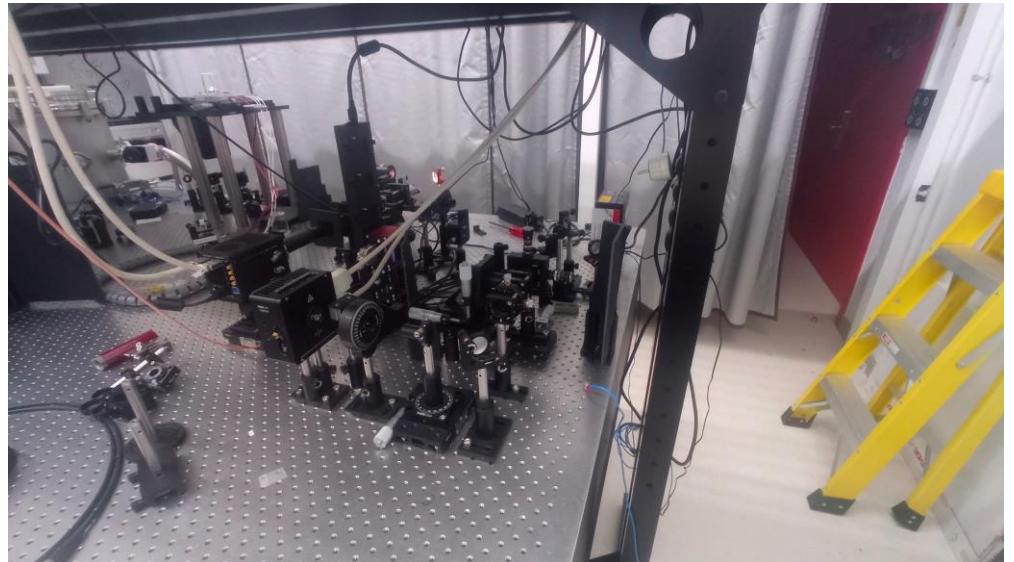
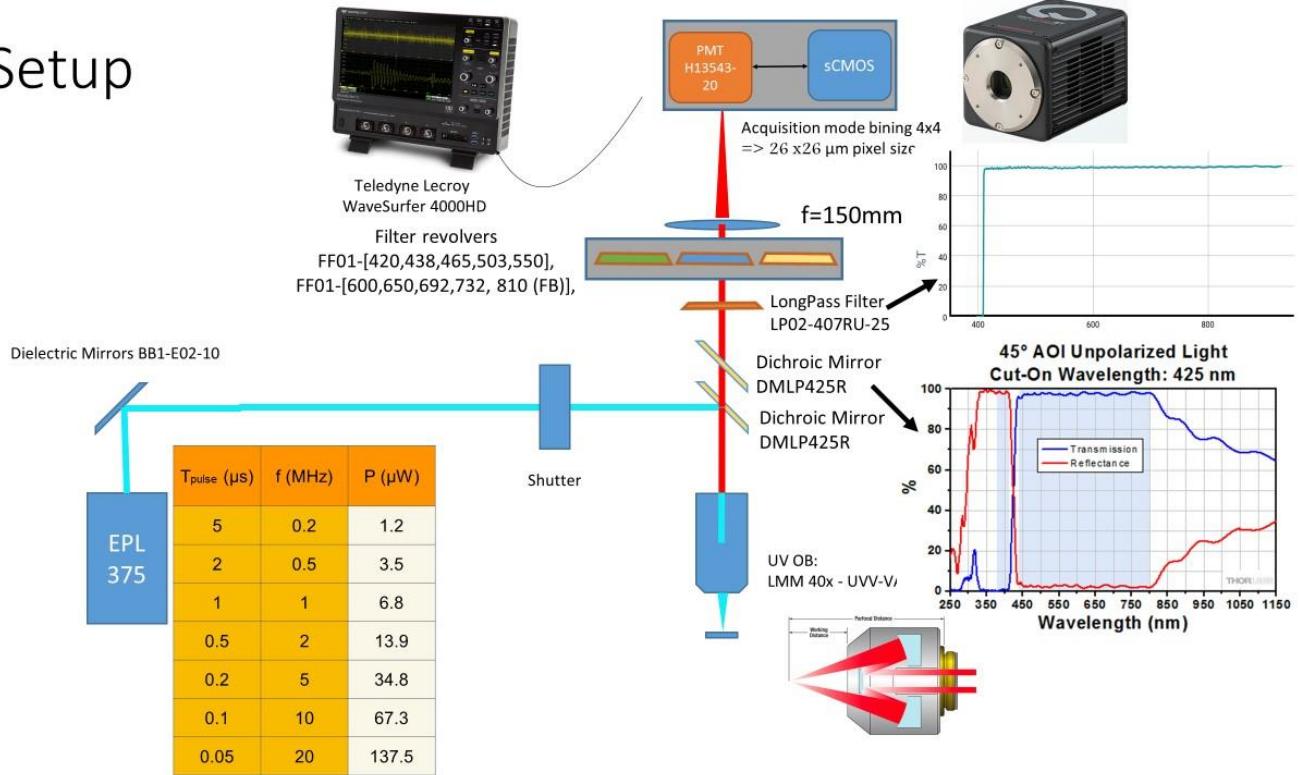
FIG. 1. A single Ba^{++} candidate. A fixed region of the CCD camera is shown with a 0.5 s exposure before (top) and after (bottom) the photobleaching transition.

BOLD/SABAT: The basic scheme

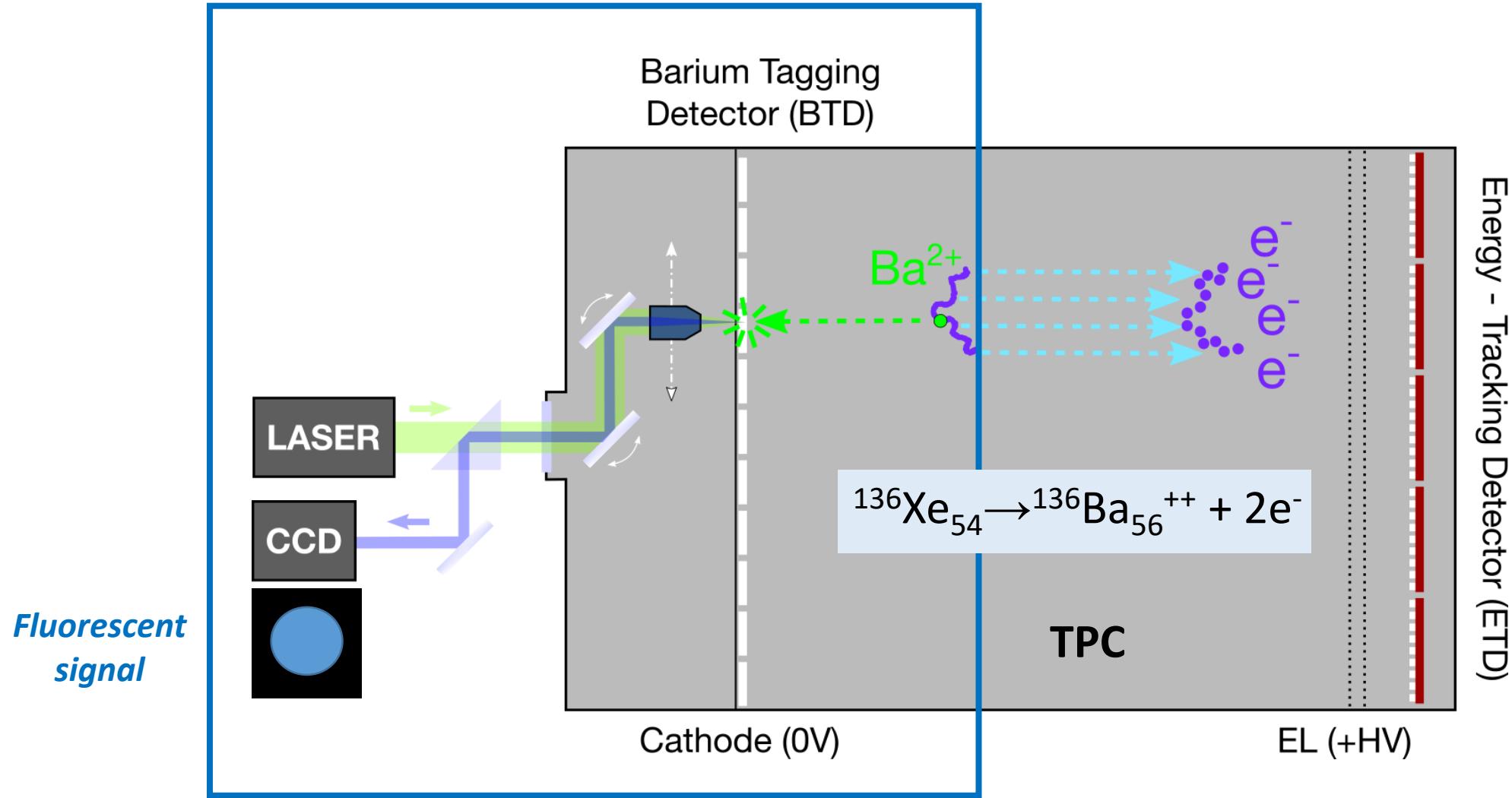


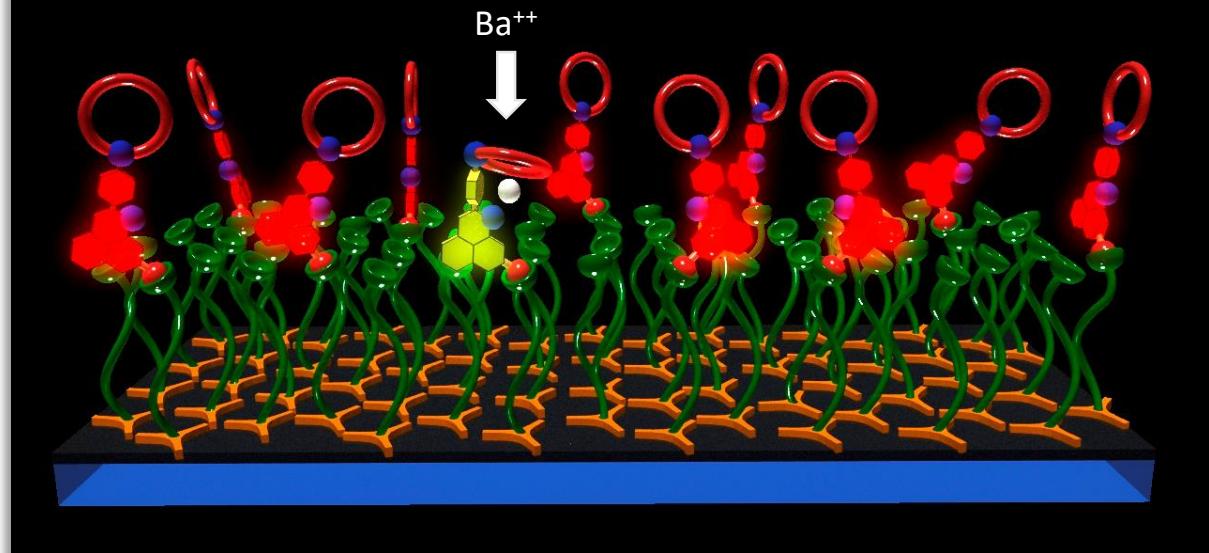
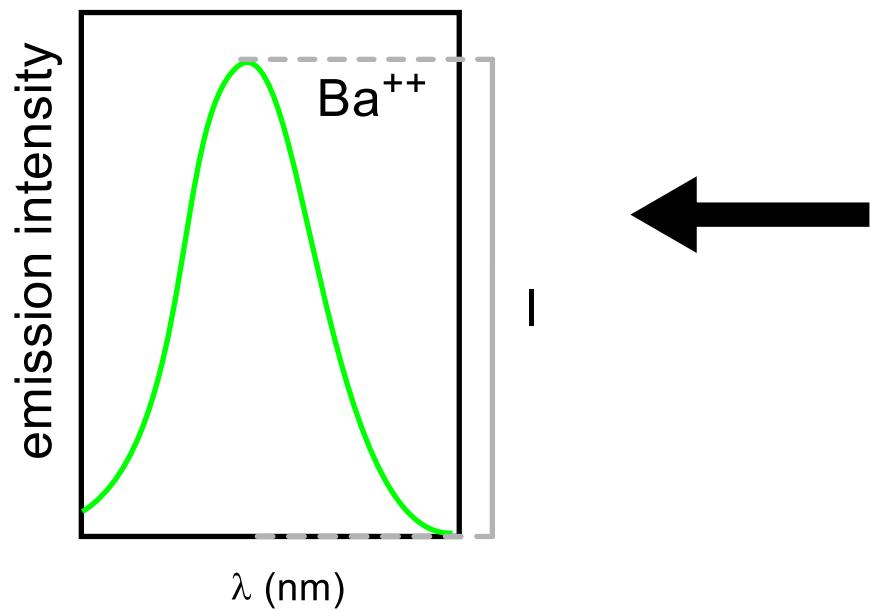
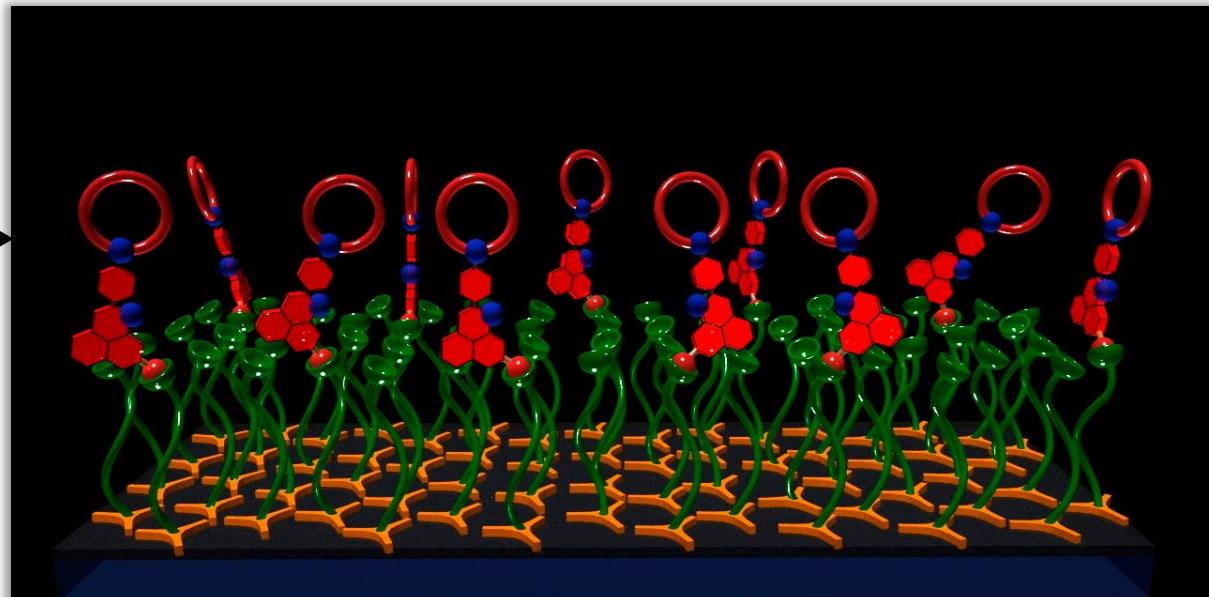
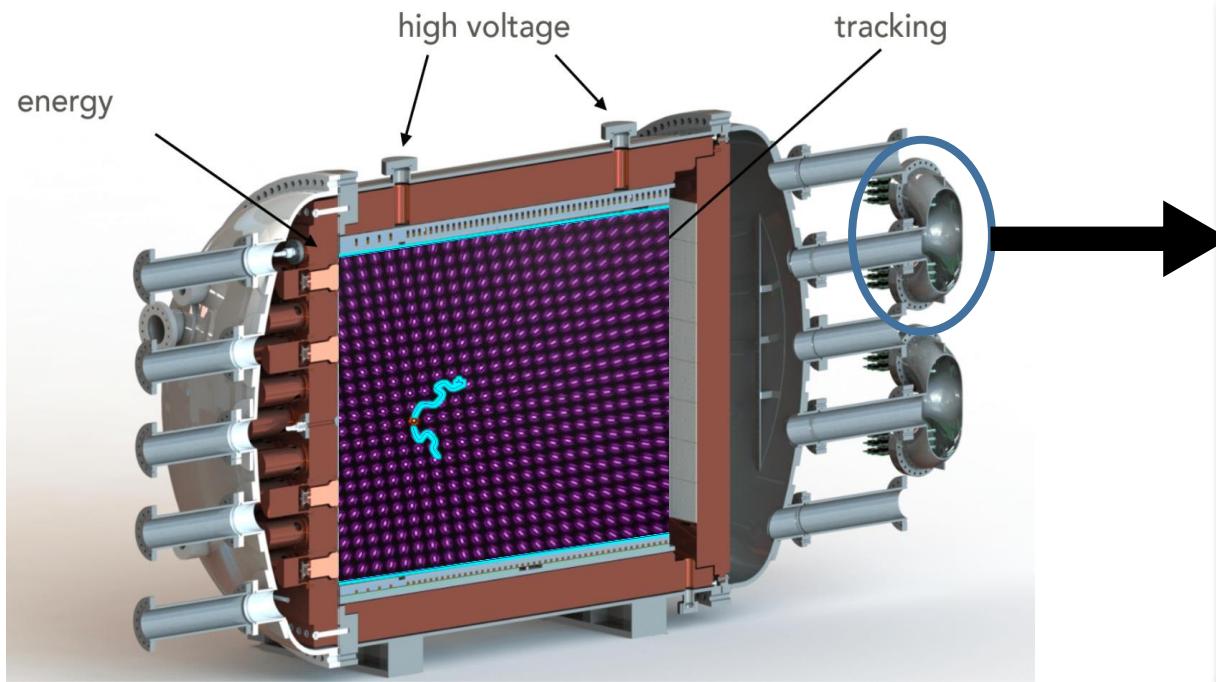
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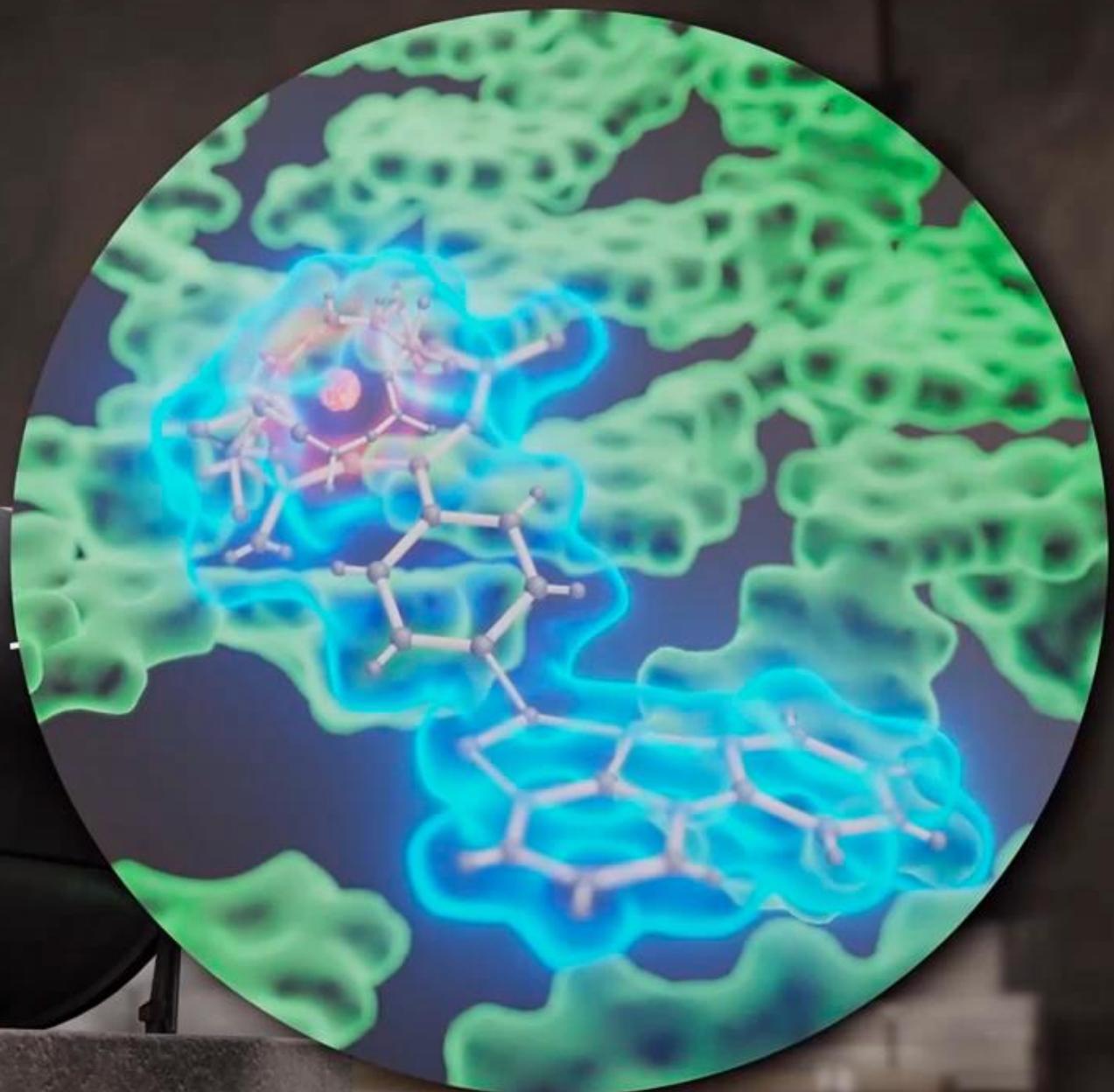
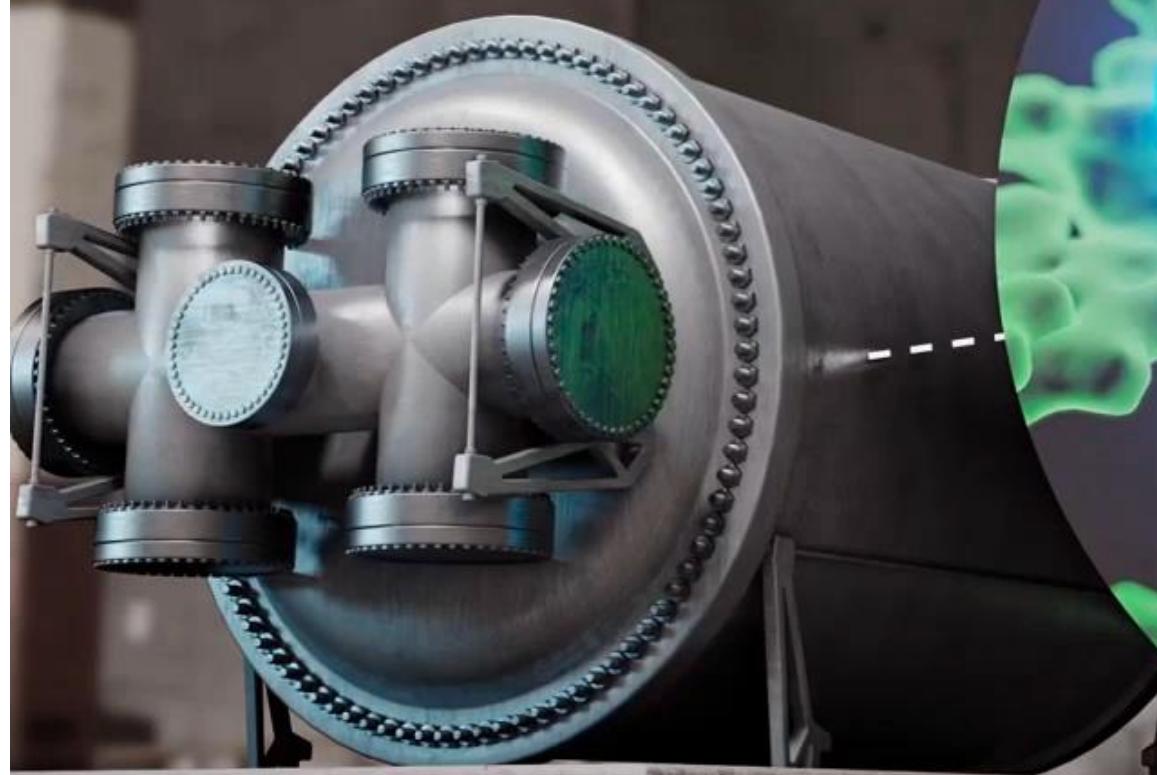
Setup



BOLD/SABAT: The basic scheme





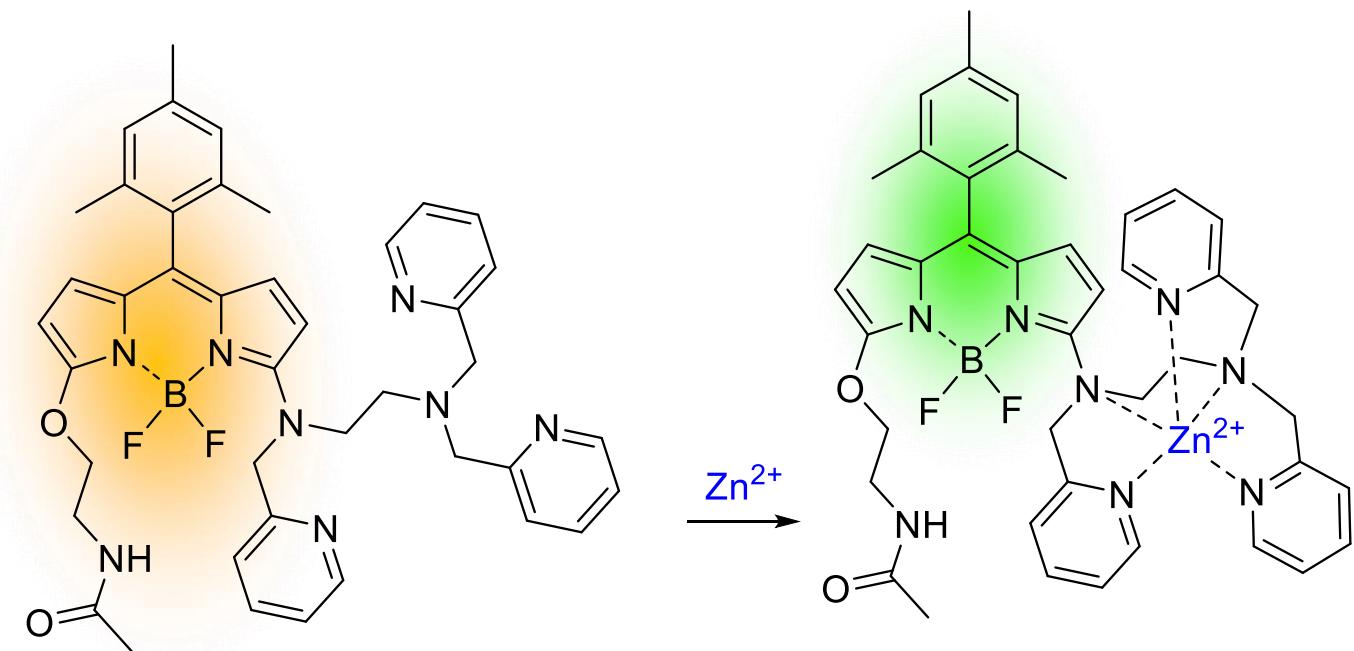


FMIs (Fluorescent Monocolor Indicators) and **F**BIIs (Fluorescent Bicolor Indicators)

- Dry phase
- Selectivity (Ba^{2+})
- Background free
- 10^{27-28} years

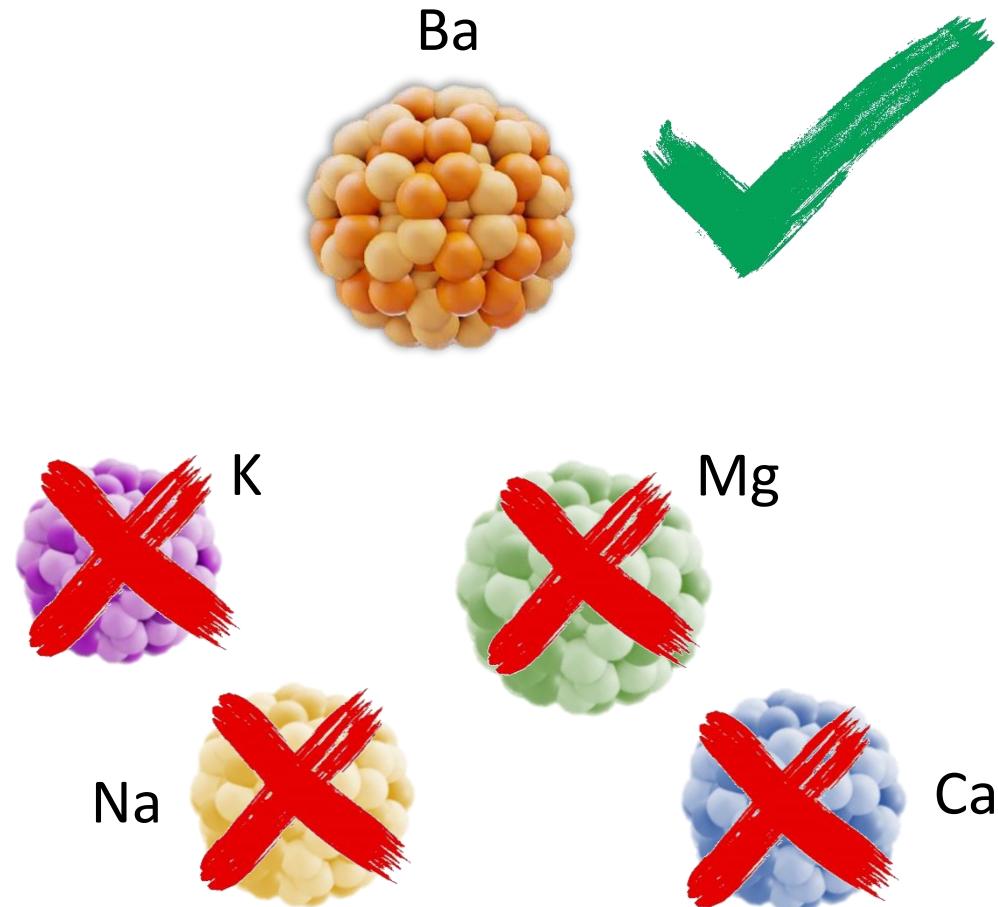
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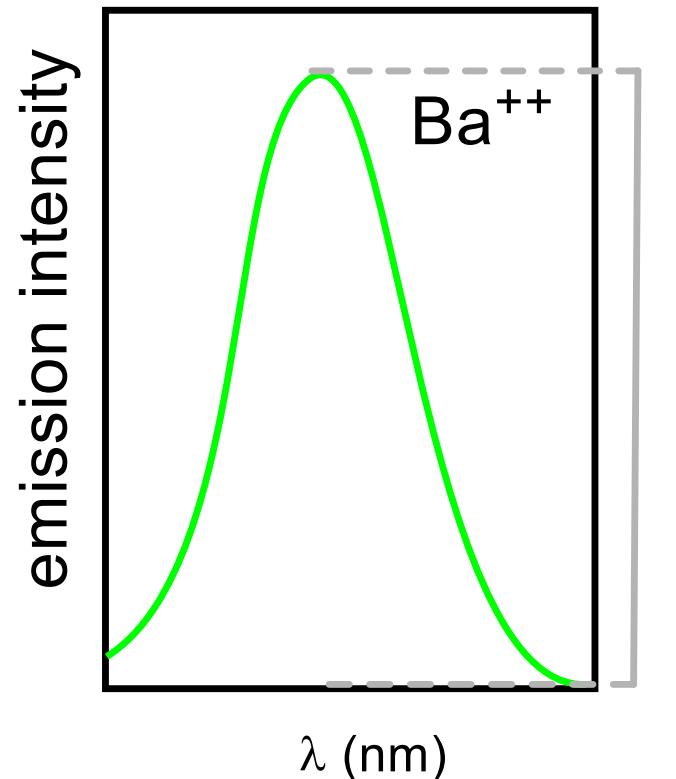
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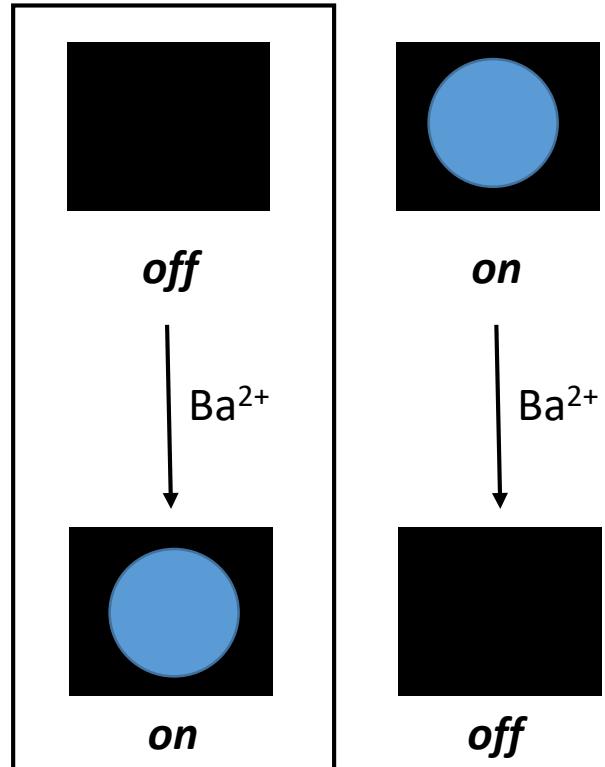


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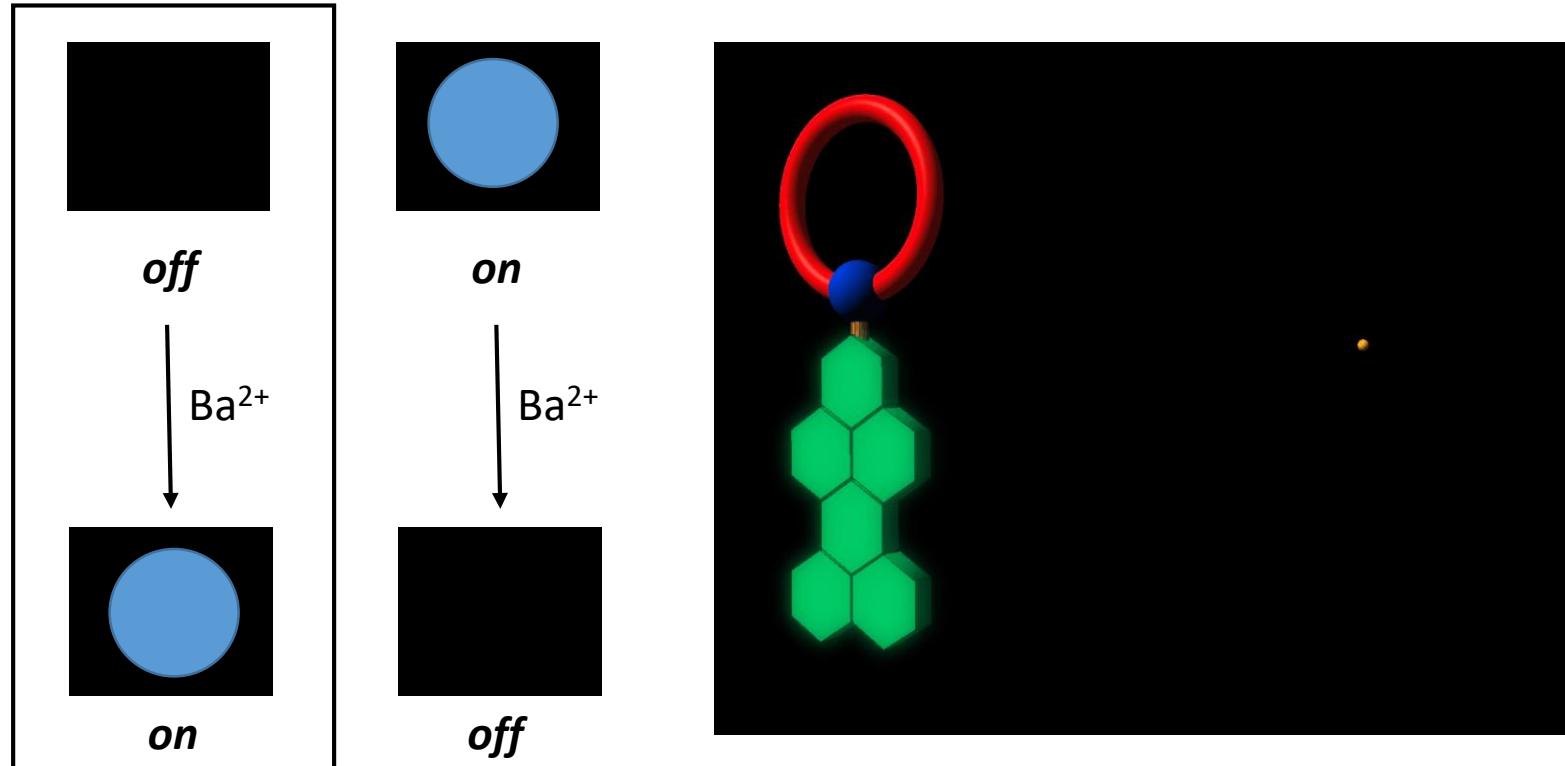


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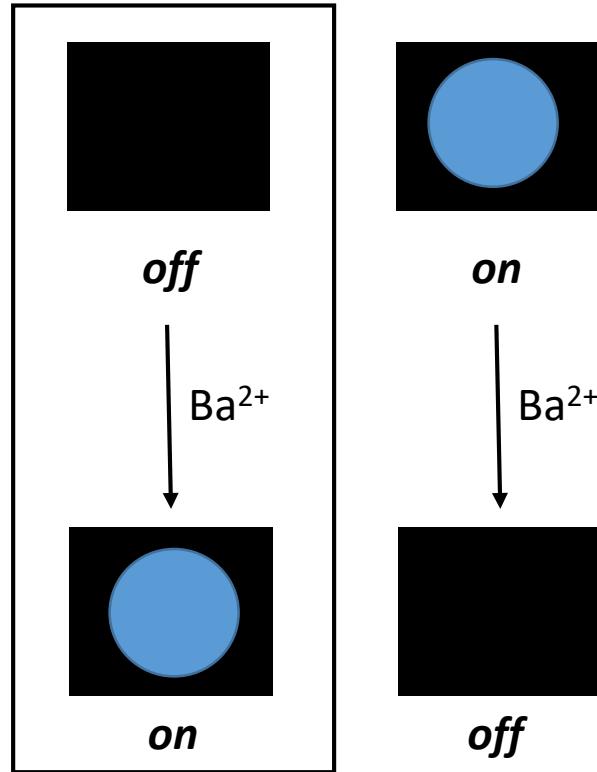
FMIs

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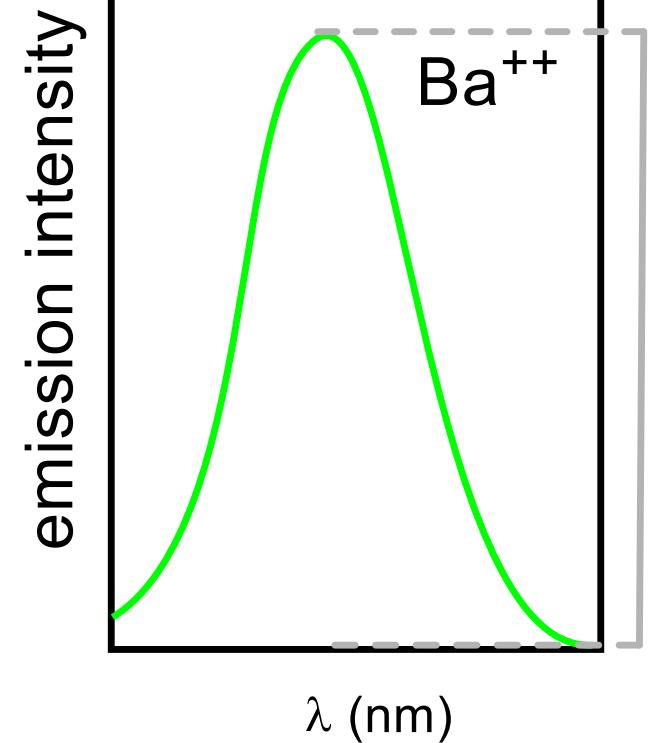
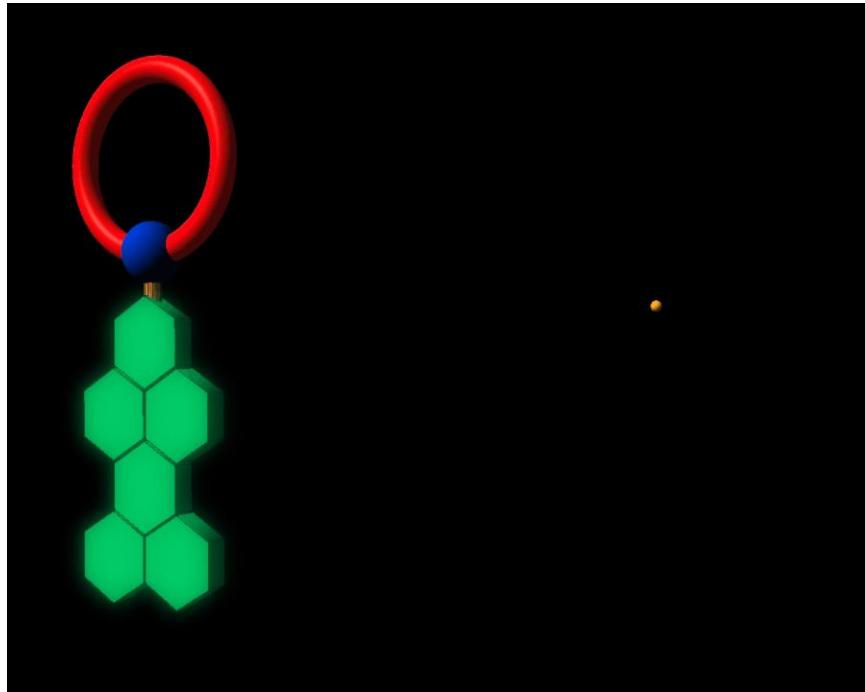


FMIs

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FMIs



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DOI: 10.1021/acssensors.1c00001

Demonstration of Selective Single-Barium Ion Detection with Dry Diazacrown Ether Naphthalimide Turn-on Chemosensors

Pawan Thapa,* Nicholas K. Byrnes,* Alena A. Denisenko, James X. Mao, Austin D. McDonald, Charleston A. Newhouse, Thanh T. Vuong, Katherine Woodruff, Kwangho Nam, David R. Nygren, Benjamin J. P. Jones,* and Frank W. Foss, Jr.*

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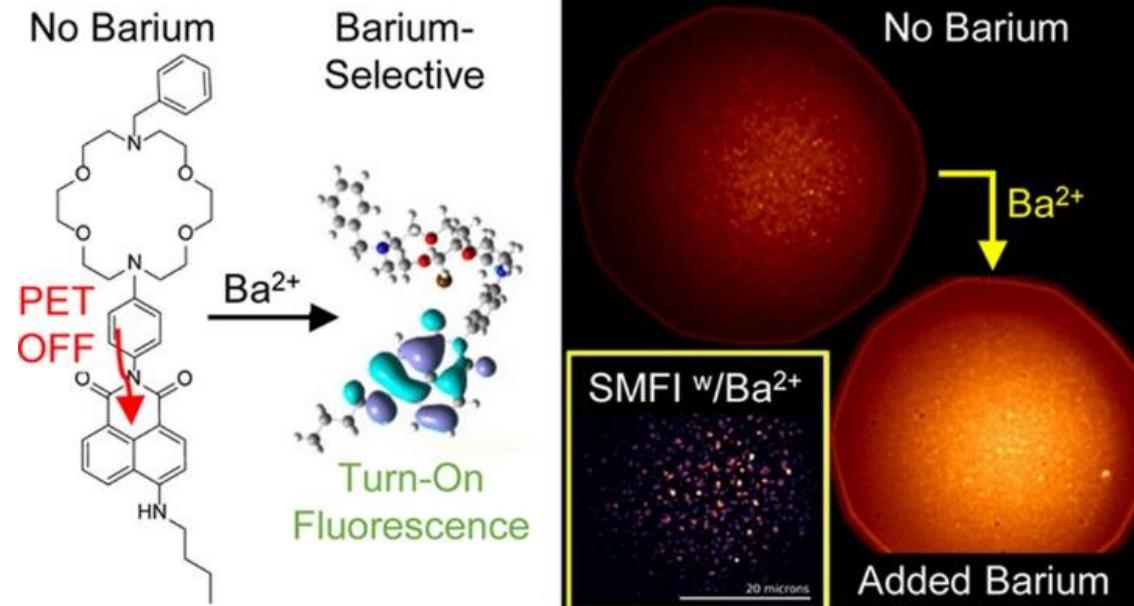
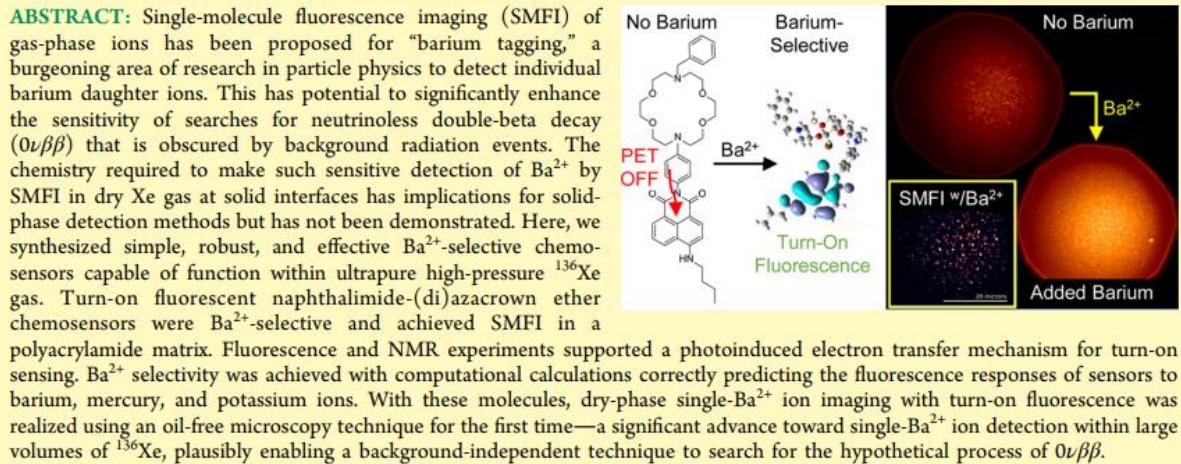
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FMIs (Fluorescent Monocolor Indicators) and FBIs (Fluorescent Bicolor Indicators)

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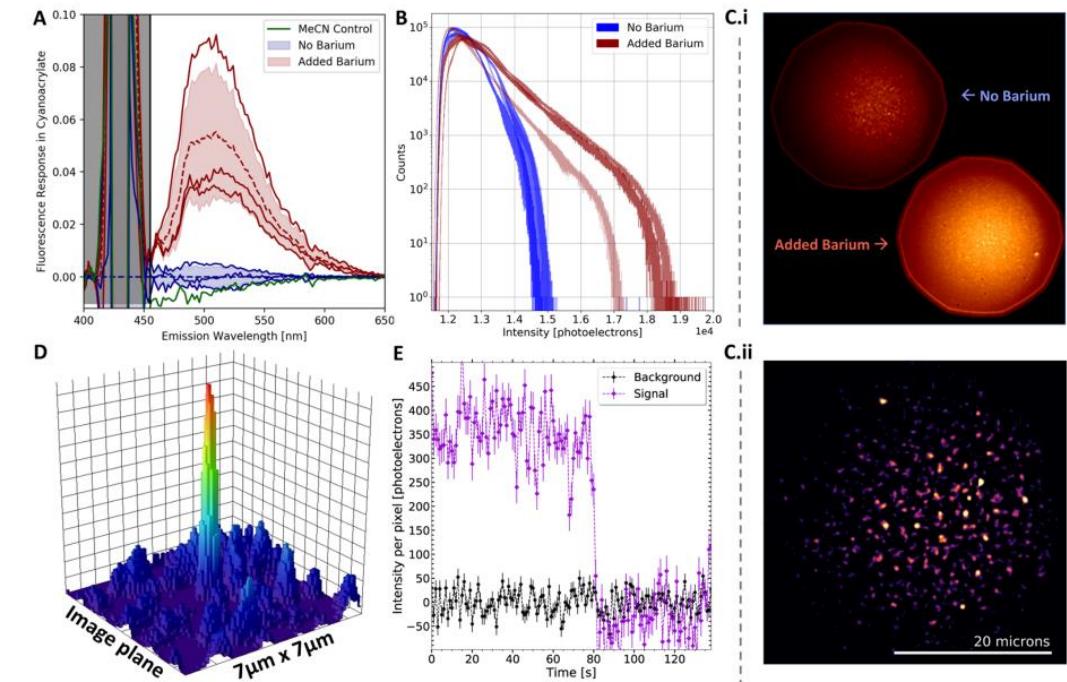
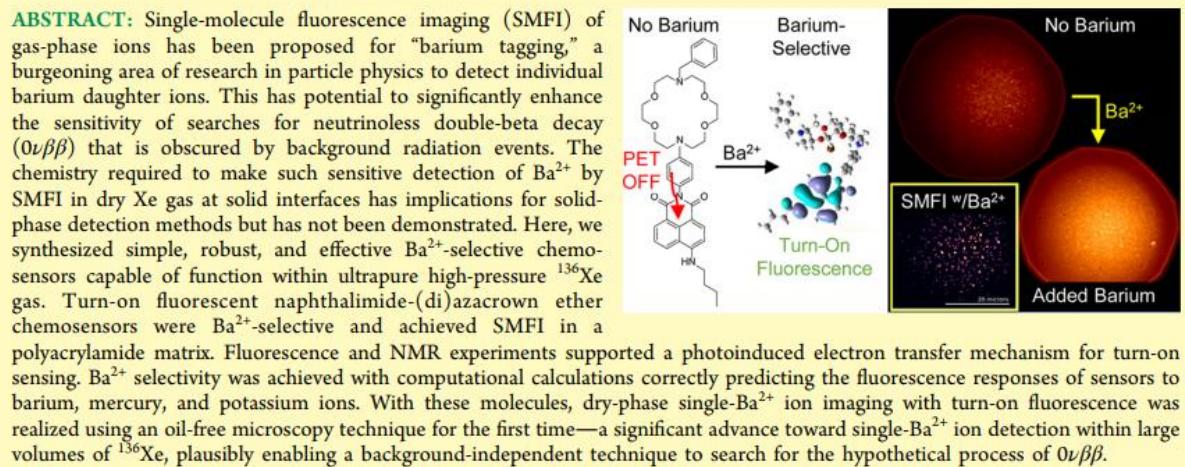
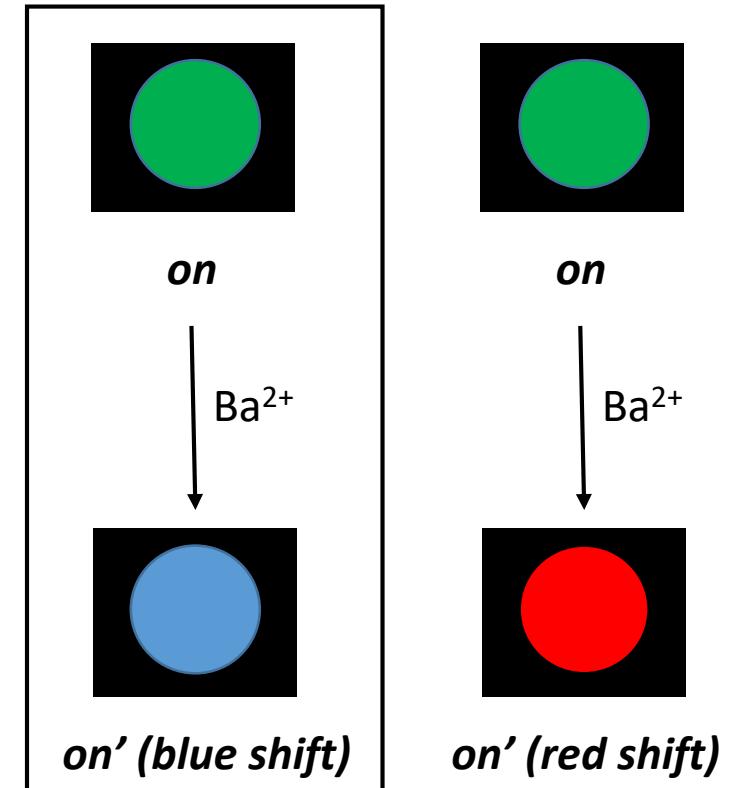


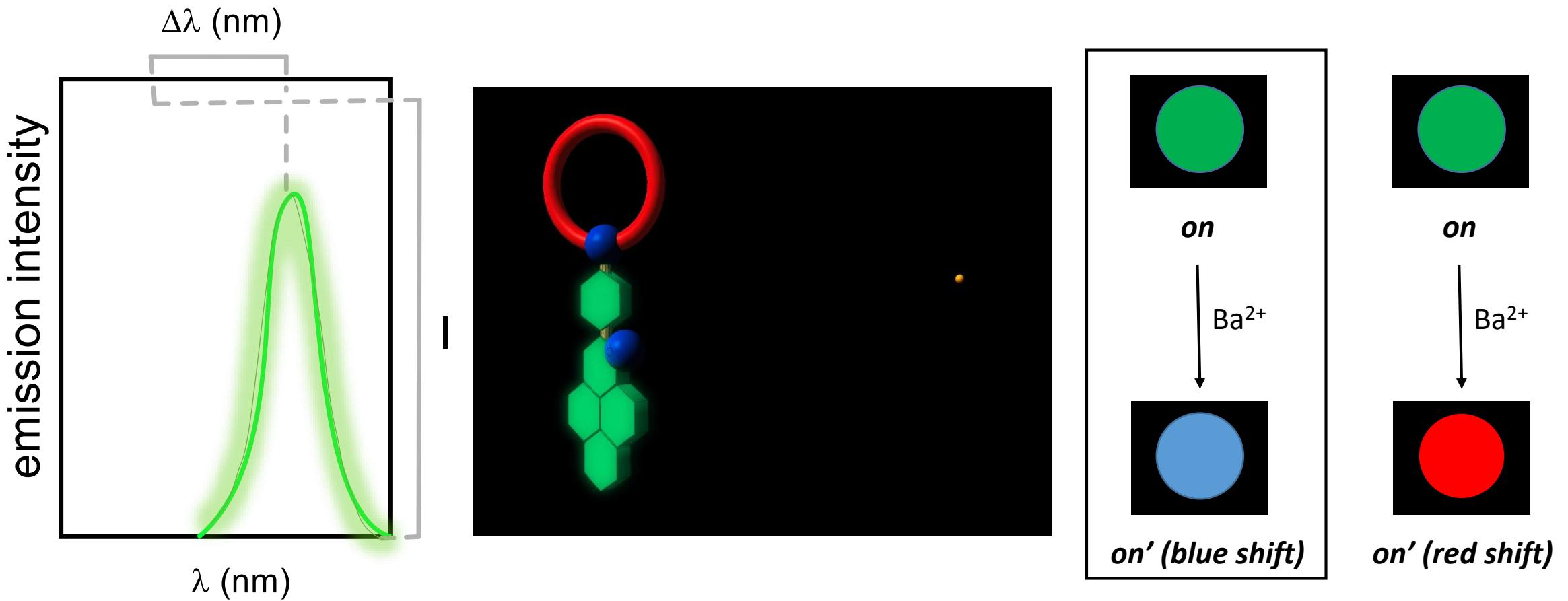
Figure 7. Solid-matrix measurements of barium-sensing fluorophore 1D. (A) Bulk spectrophotometry in a dried cyanoacrylate matrix; (B) raw pixel histogram of three 1D-coated slides imaged at the single-molecule level; (C) single-molecule level microscopy images of the 1D layer with and without Ba^{2+} (i) and with Ba^{2+} after background removal (ii); (D) specific single reconstructed barium-chelated 1D molecule; and (E) barium-chelated 1D molecule fluorescence time trajectory showing a single-step photobleaching characteristic of SMFI detection.

FMIs (Fluorescent Monocolor Indicators) and FBIs (Fluorescent Bicolor Indicators)



FBIs

FMIs (Fluorescent Monocolor Indicators) and FBIs (Fluorescent Bicolor Indicators)



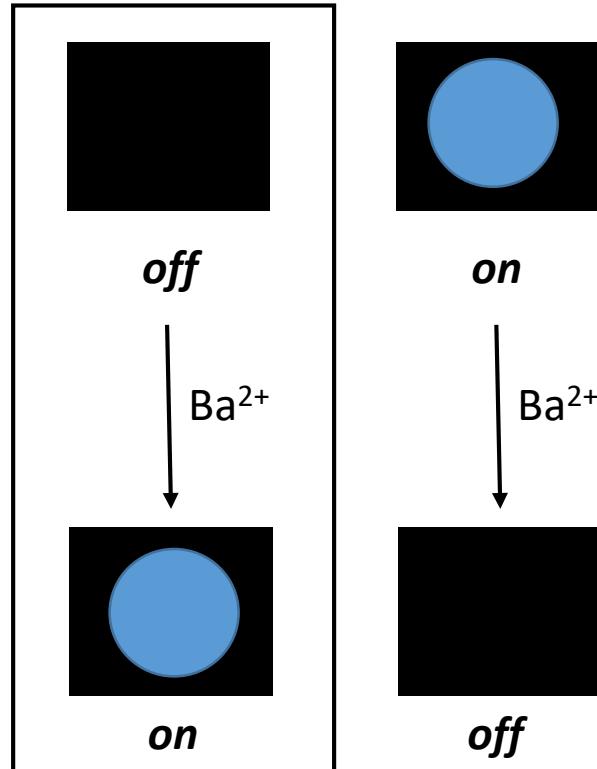
FBIs

FMIs (Fluorescent Monocolor Indicators) and FBIs (Fluorescent Bicolor Indicators)

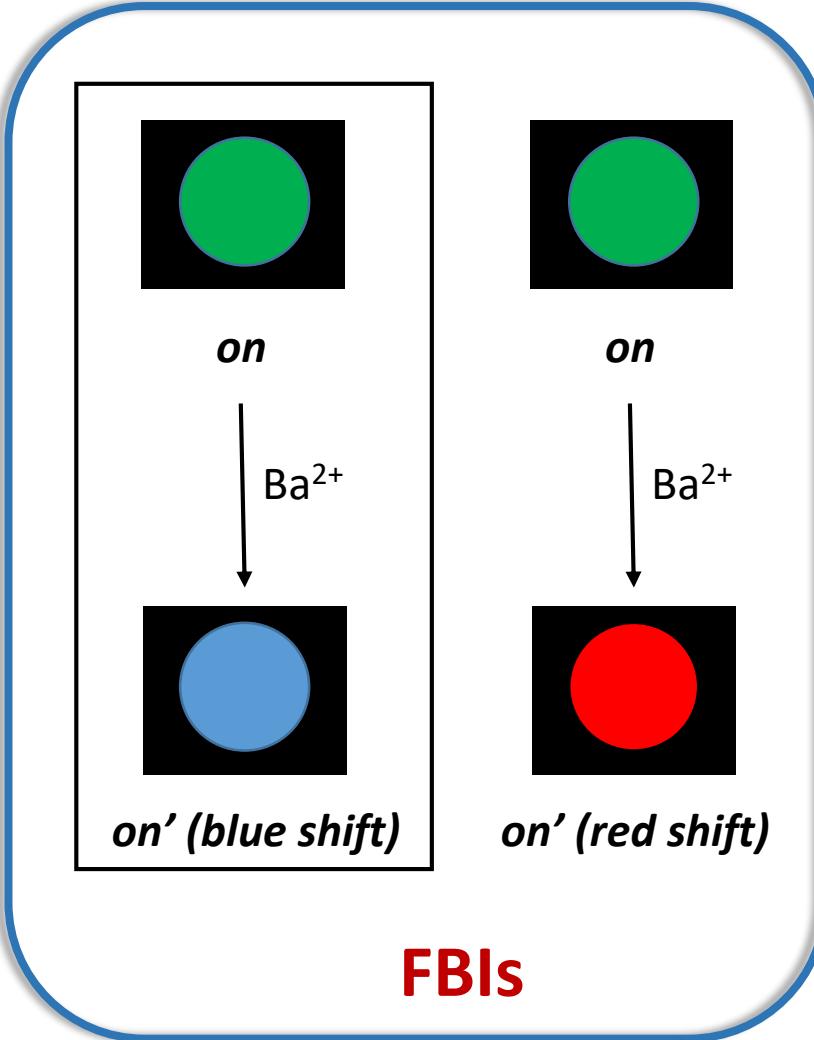


FBIs

FМИs (Fluorescent Monocolor Indicators) and FBIs (Fluorescent Bicolor Indicators)

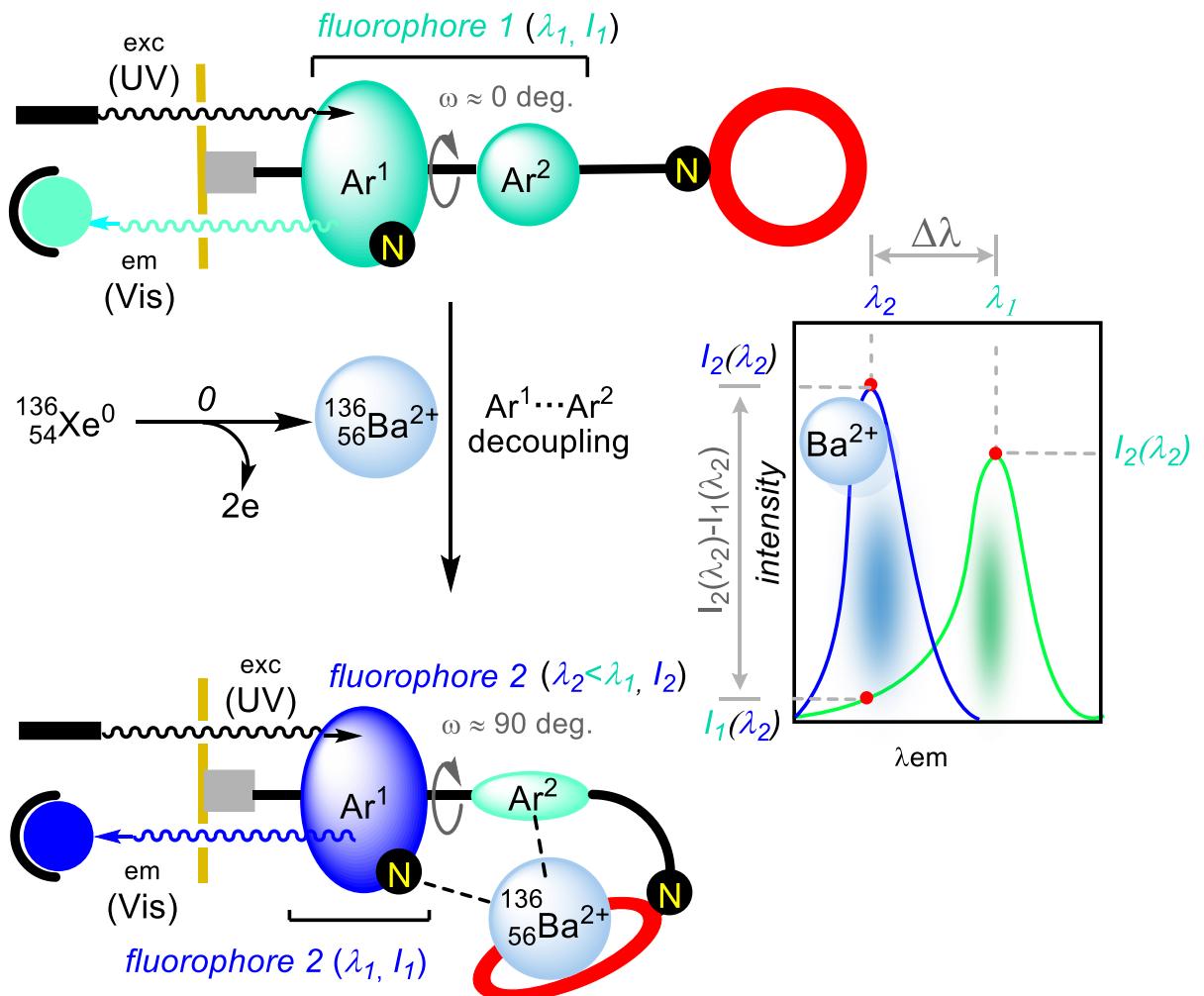


FМИs



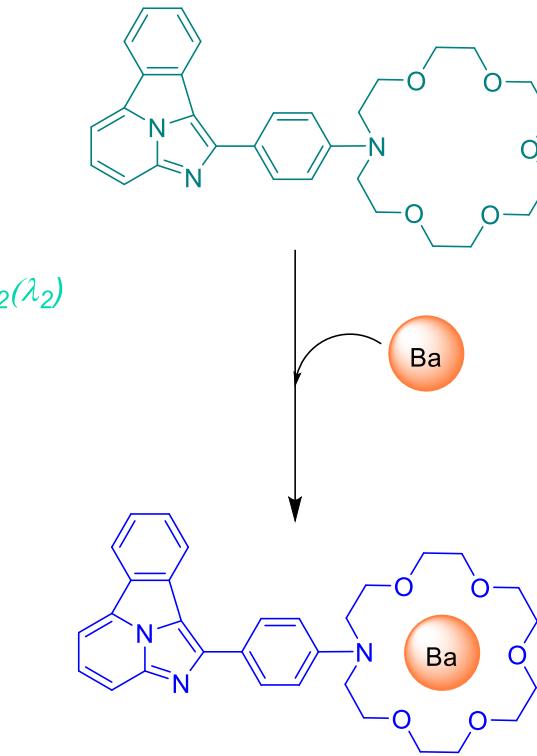
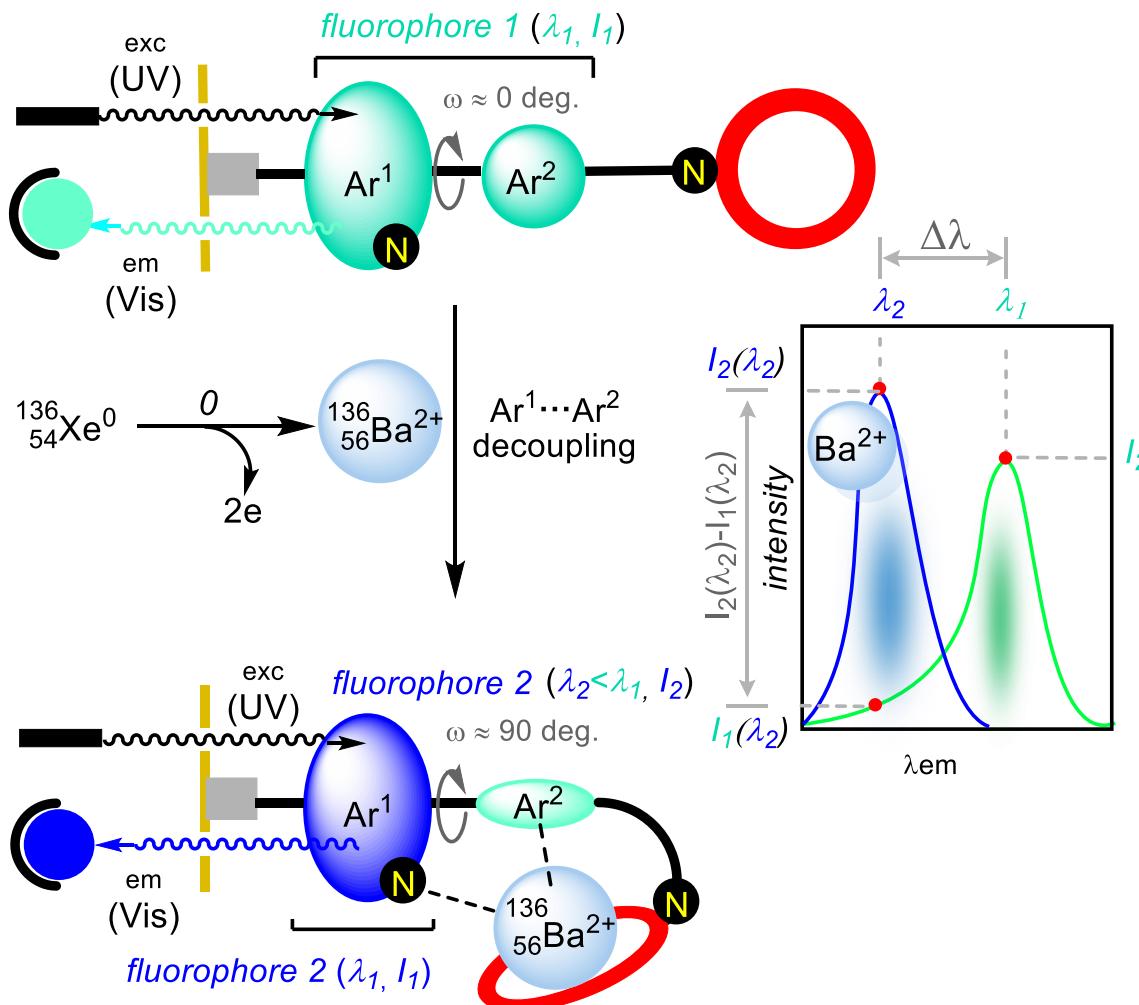
FBIs

1st generation FBIs: Two Fluorophores in One Molecule

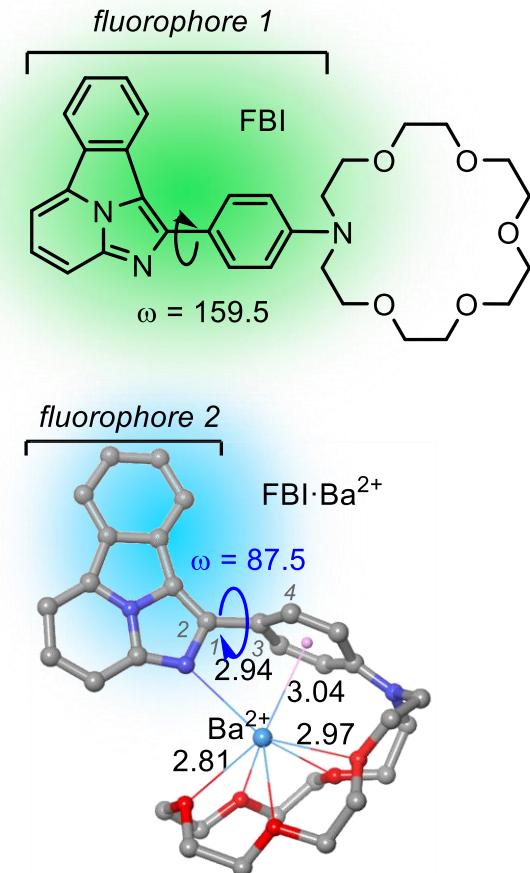


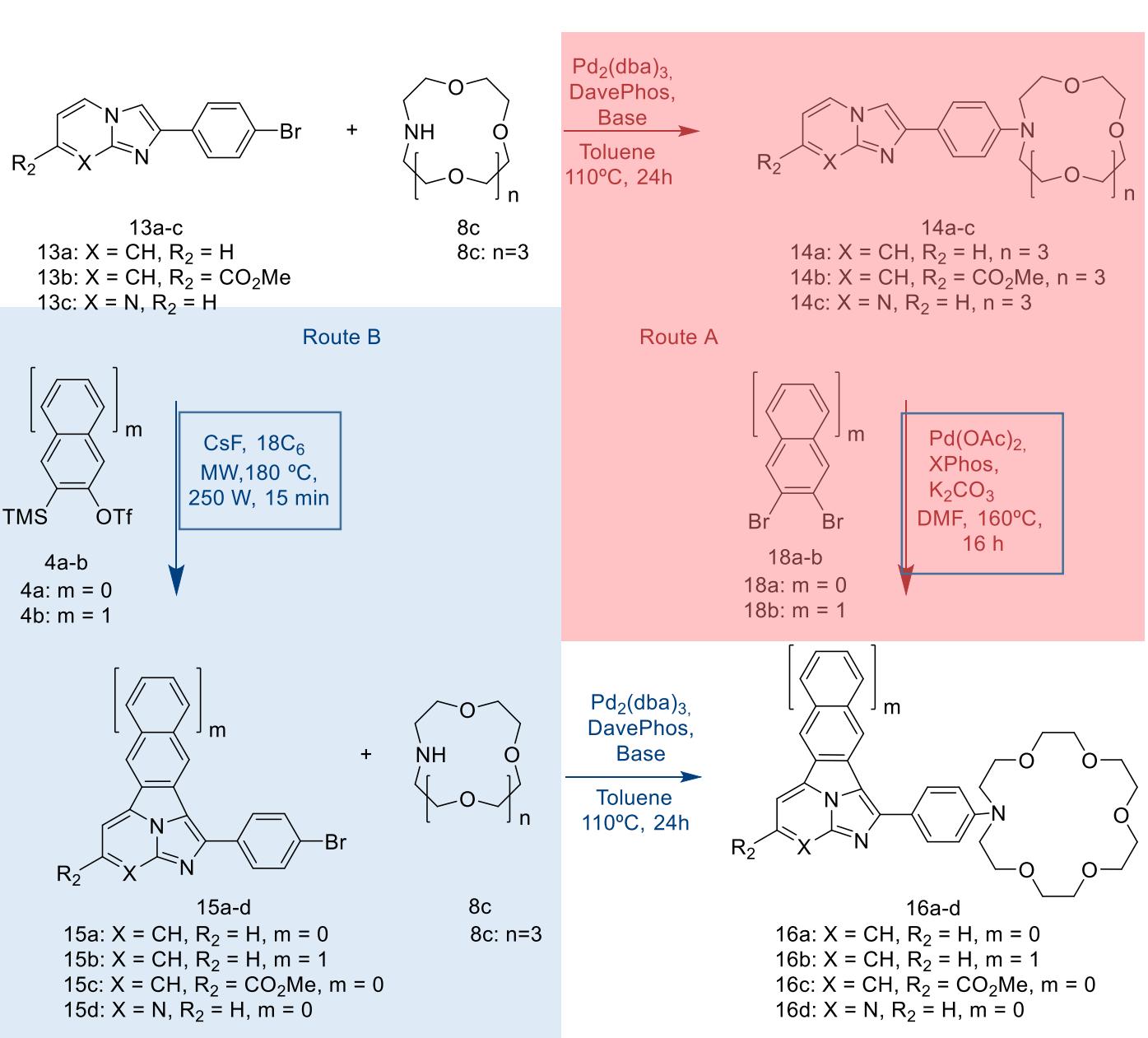
- Two fluorophores (Ph or Hetero-Aromatic-Cyclic system) in one molecule
- Crown Ether (Chelating agent)
- Two responses depending on the coordinator state of the sensor

1st generation FBIs: Two Fluorophores in One Molecule

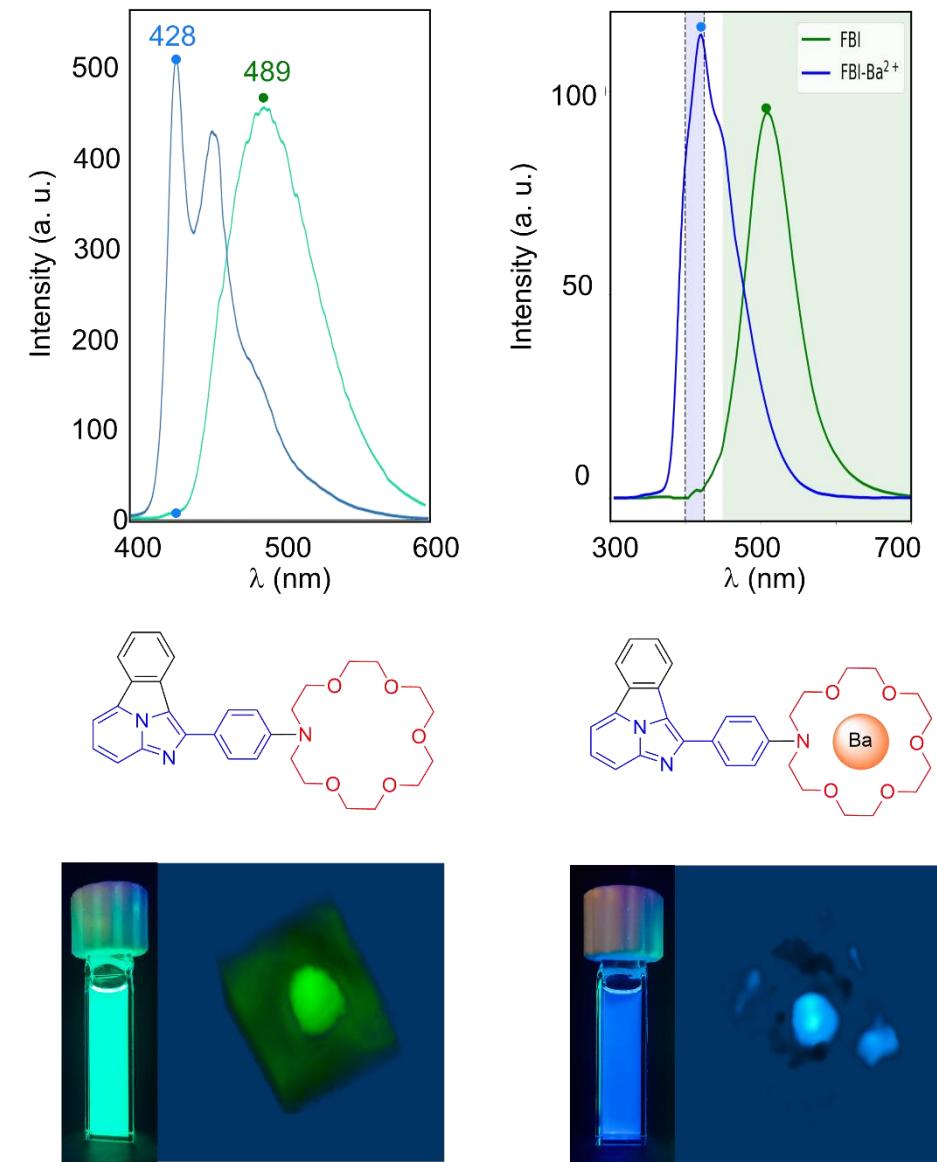
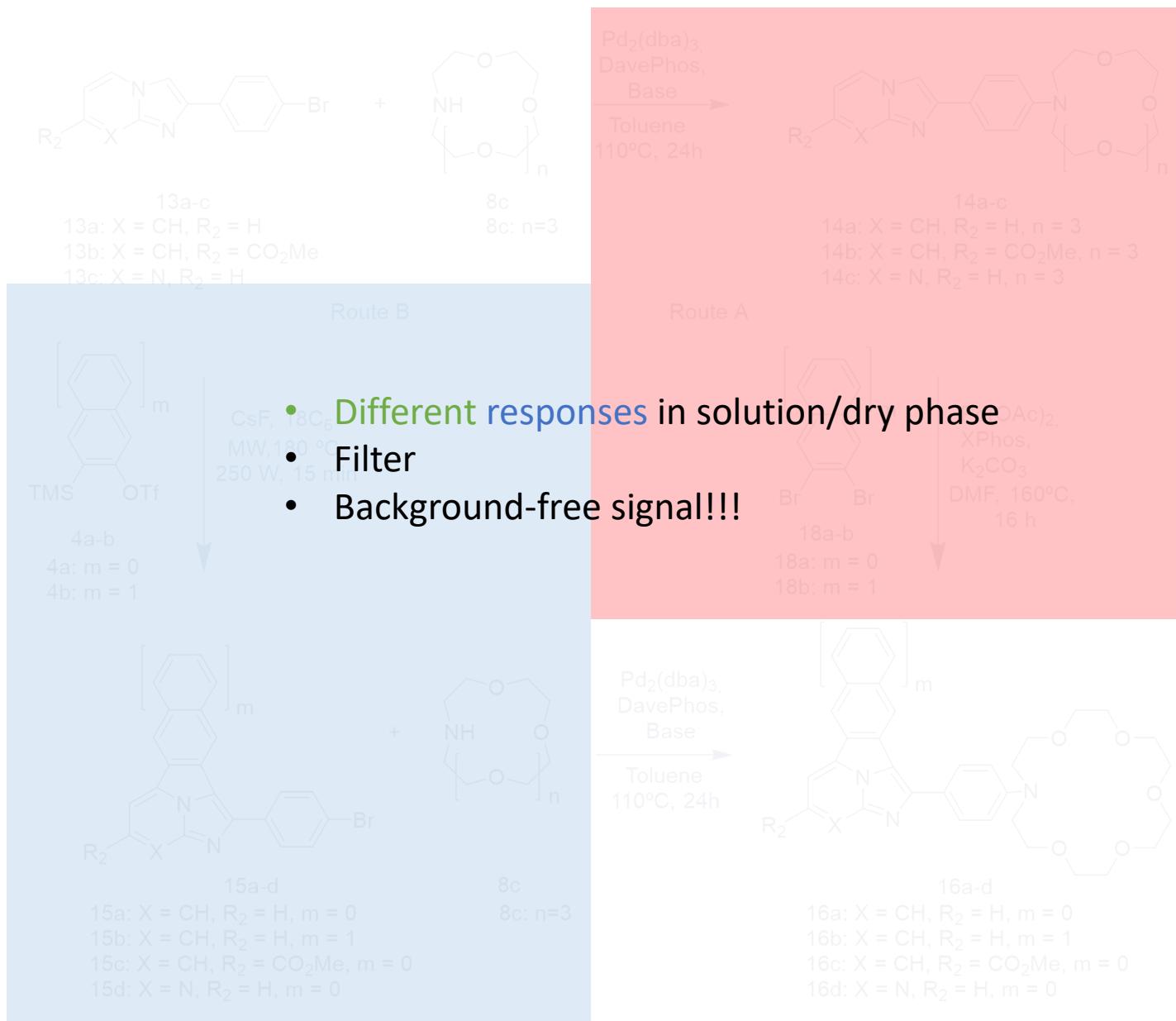


- Green (free)---Blue (Ba^{++})





- C-C or C-N coupling reactions
- High order 8+2 cycloadditions reactions
- Library of sensors or candidates



Article

Fluorescent bicolour sensor for low-background neutrinoless double β decay experiments

<https://doi.org/10.1038/s41586-020-2431-5>

Received: 15 September 2019

Accepted: 3 April 2020

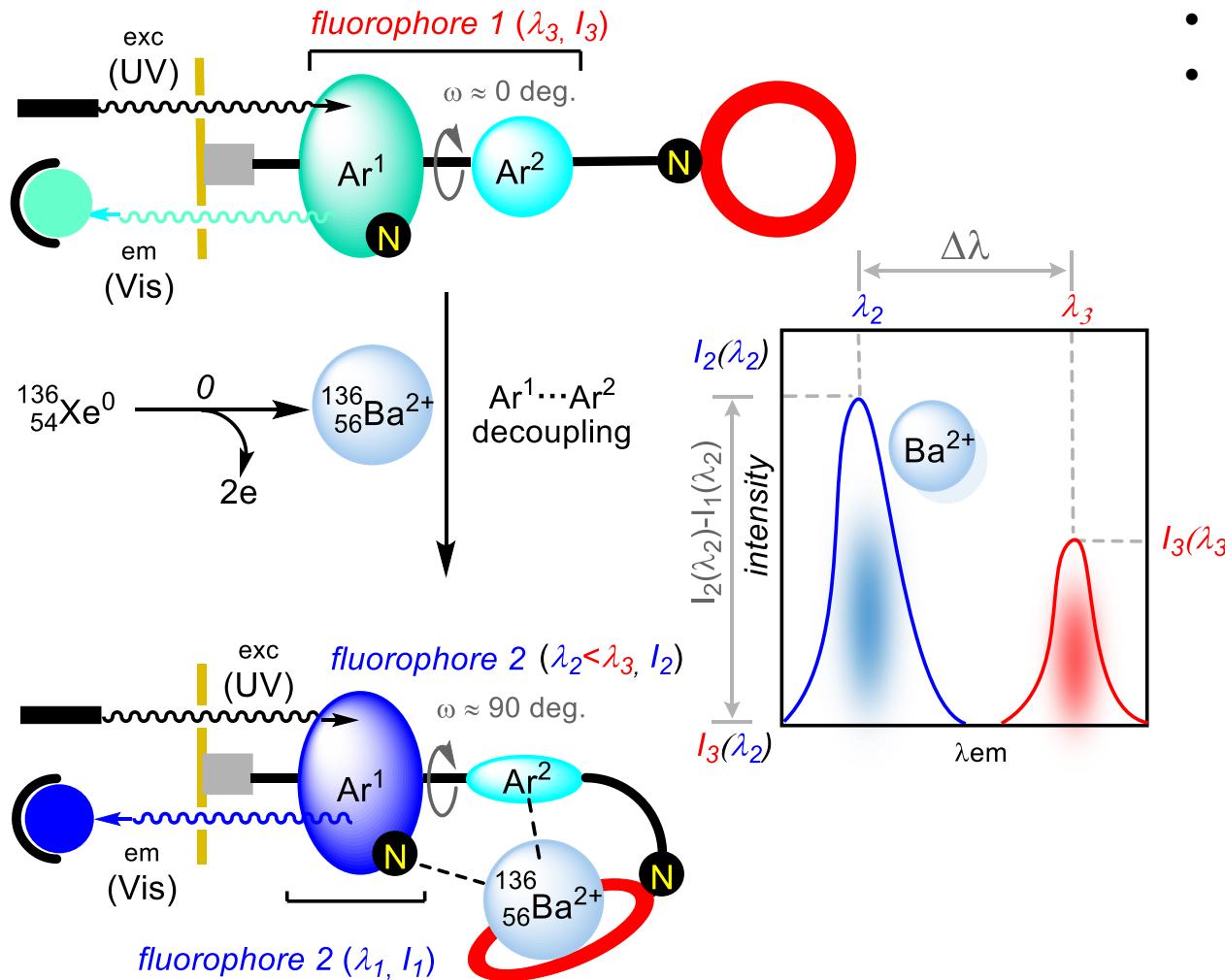
Published online: 22 June 2020

Iván Rivilla¹, Borja Aparicio², Juan M. Bueno³, David Casanova^{1,4}, Claire Tonnellé¹, Zoraida Freixa^{4,5}, Pablo Herrero¹, Celia Rogero^{1,6}, José I. Miranda⁷, Rosa M. Martínez-Ojeda³, Francesc Monrabal^{1,4}, Beñat Olave⁸, Thomas Schäfer^{4,8}, Pablo Artal³, David Nygren⁹, Fernando P. Cossío^{1,2}✉ & Juan J. Gómez-Cadenas^{1,4}✉

https://www.youtube.com/watch?v=W4_Qjd3h2PU

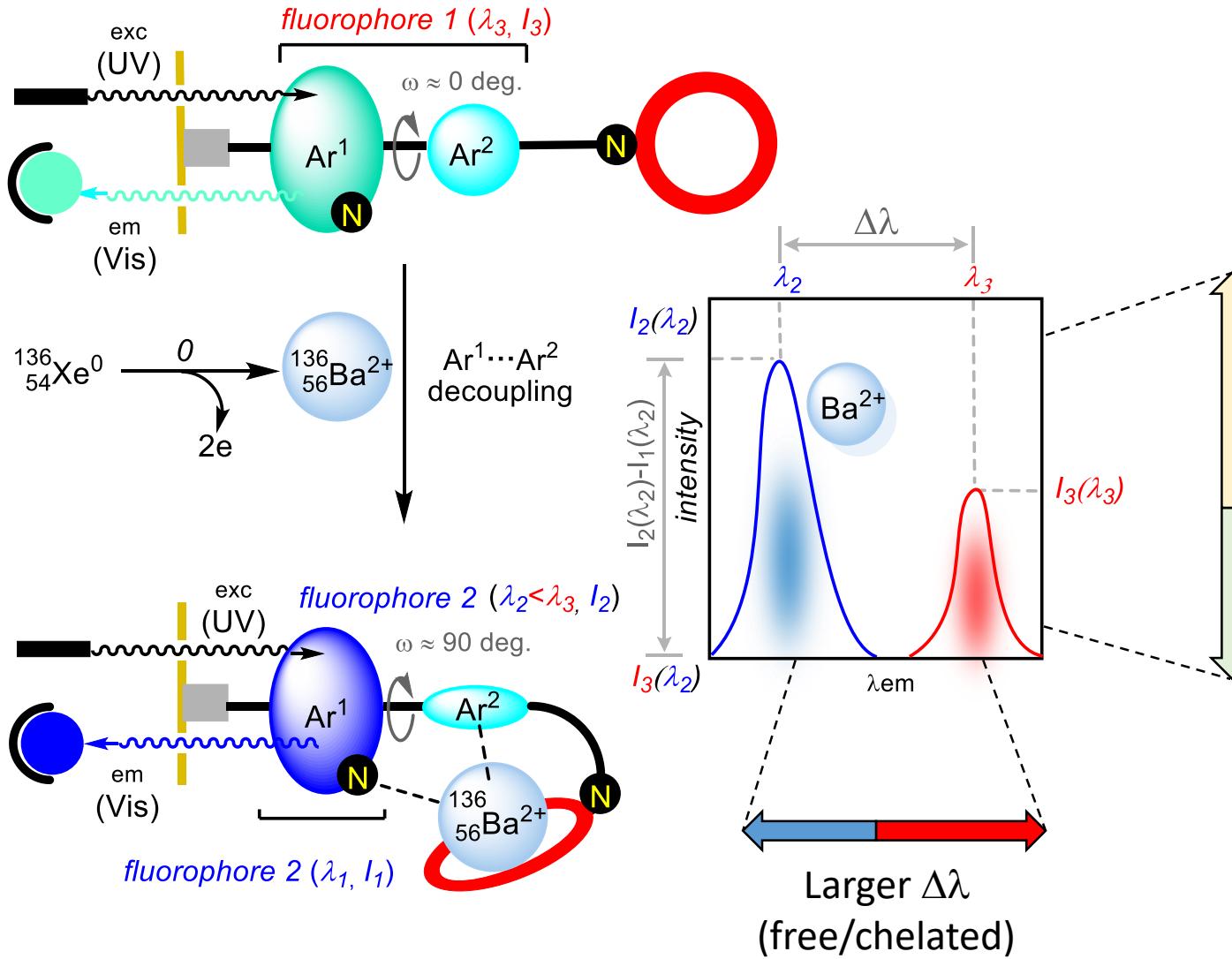


Optimizing the photophysics of FBIs: Design Criteria for Next Generations



- ...a lot of room for further improvement!!!
- Strict rules...NEXT/BOLD experiments
- The pipeline pharma strategy

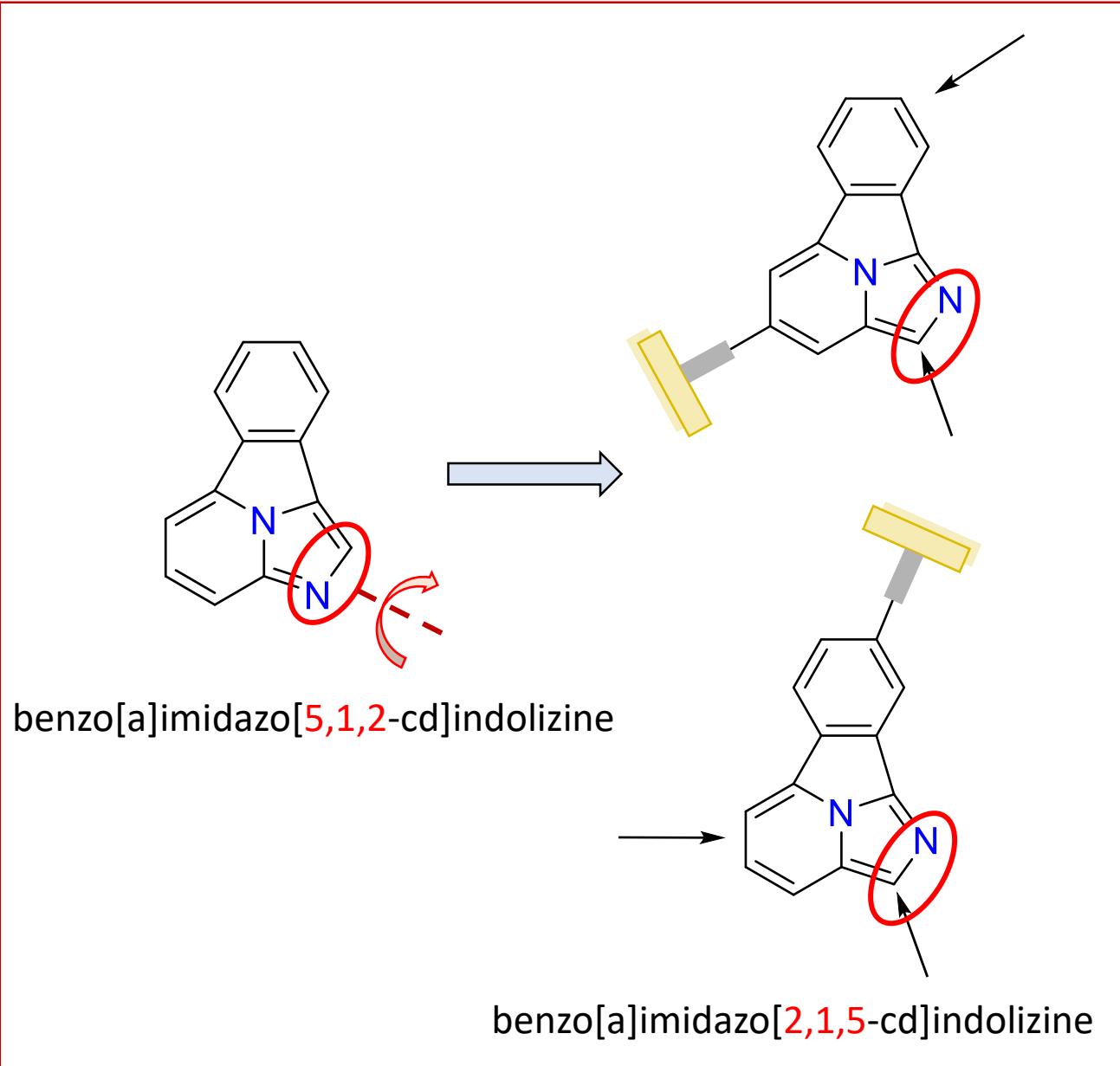
Optimizing the photophysics of FBIs: Design Criteria for Next Generations



To increase (Neutrinoless $\beta\beta 0\nu$ extremely rare event)

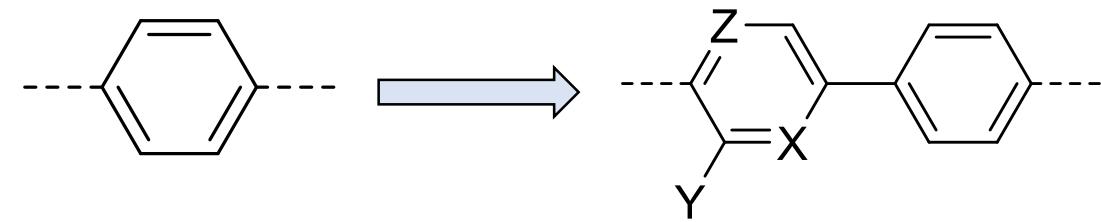
- Higher brightness
- Larger discrimination factor
- Bigger binding constant

Next generation FBIs: Novel Ar¹ and Ar² Units



Different candidates (Design)

- Rotation ... changing the nitrogen by carbon,
- 5,1,2 to 2,1,5
- Configurational space
- Decorate options
- Anchor to surfaces

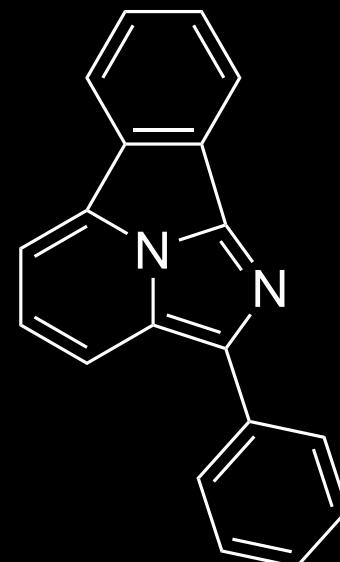
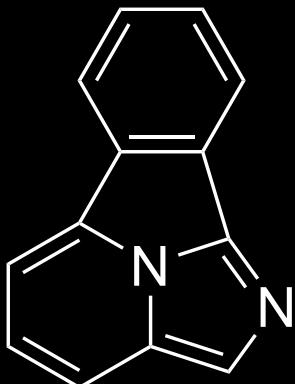
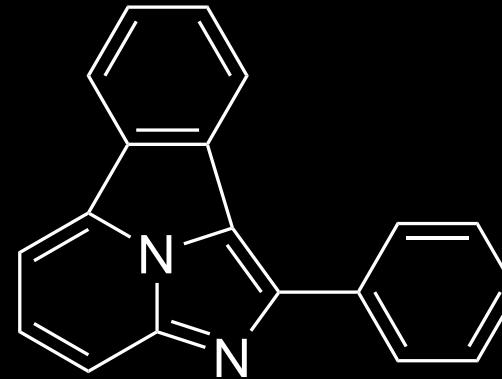
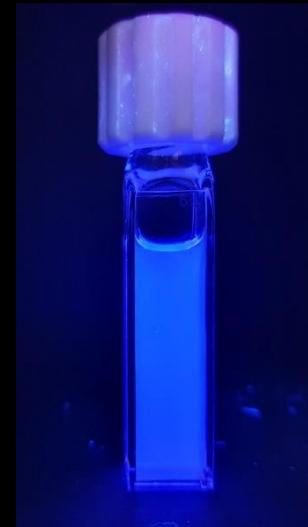
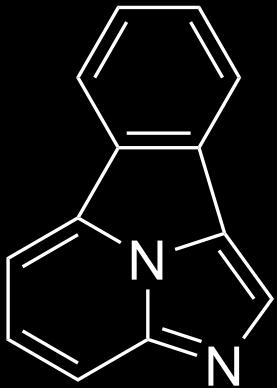


- a: X=Z=CH; Y=H
b: X=N; Y=CH₃; Z=CH
c: X=Z=N; Y=H

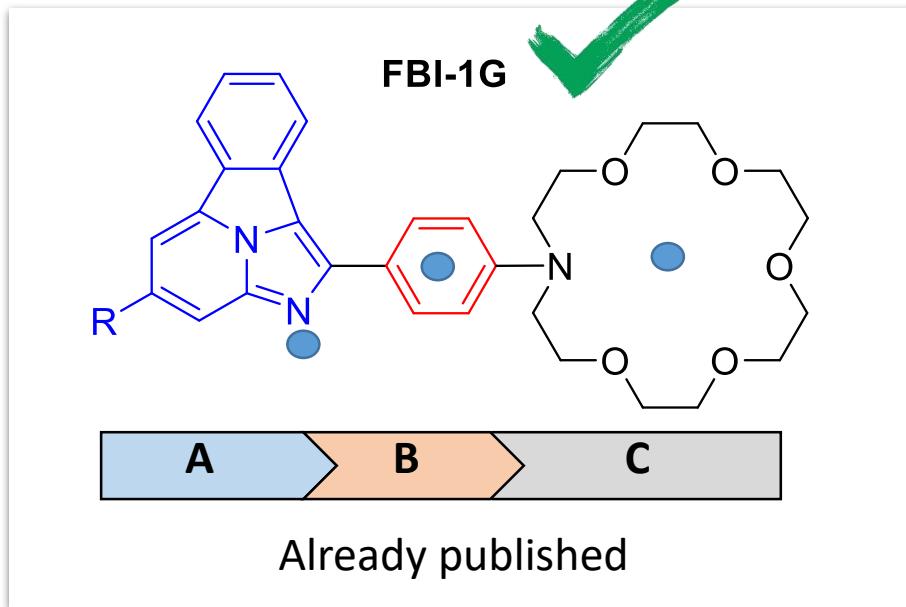
$\lambda_{\text{exc}} = 365 \text{ nm}$

[] = $5 \cdot 10^{-5} \text{ M}$

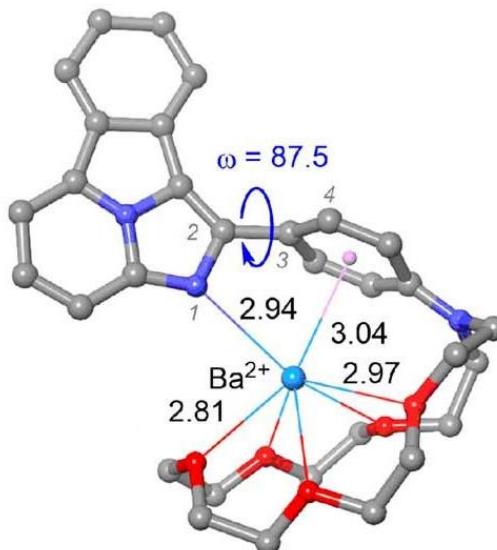
ACN



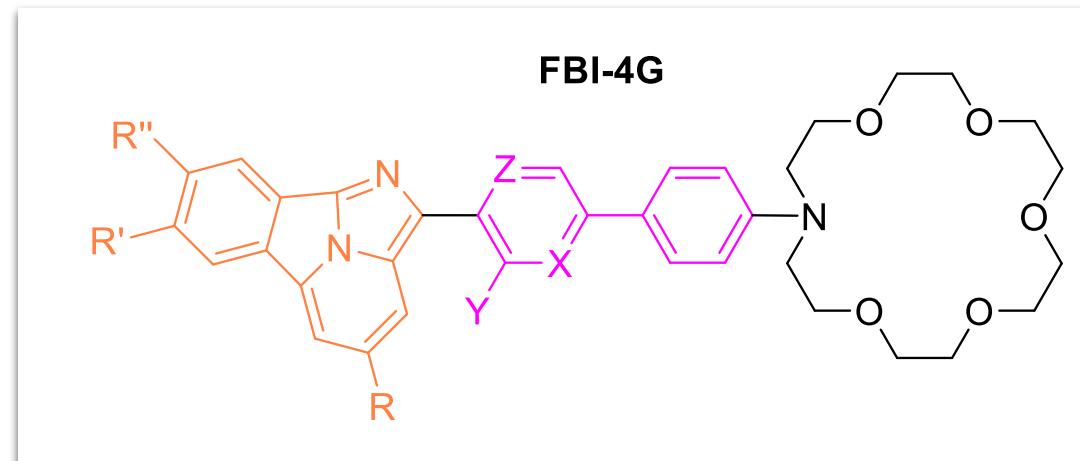
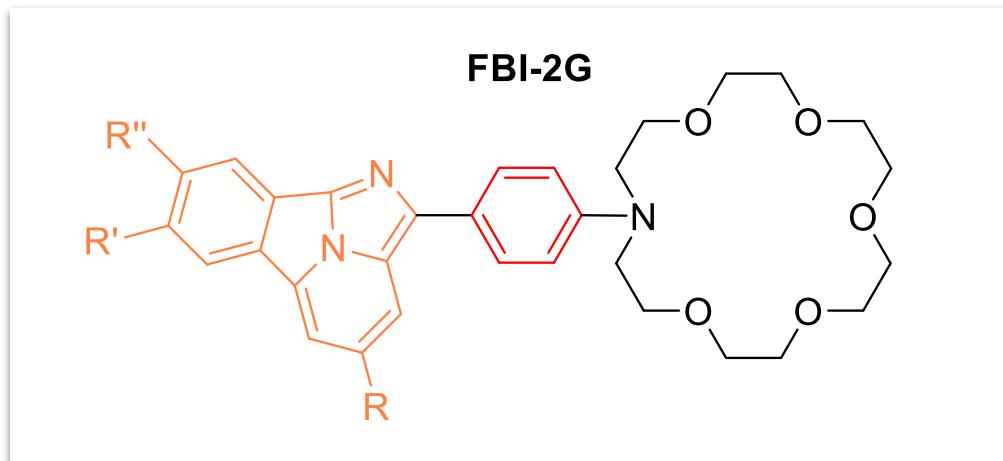
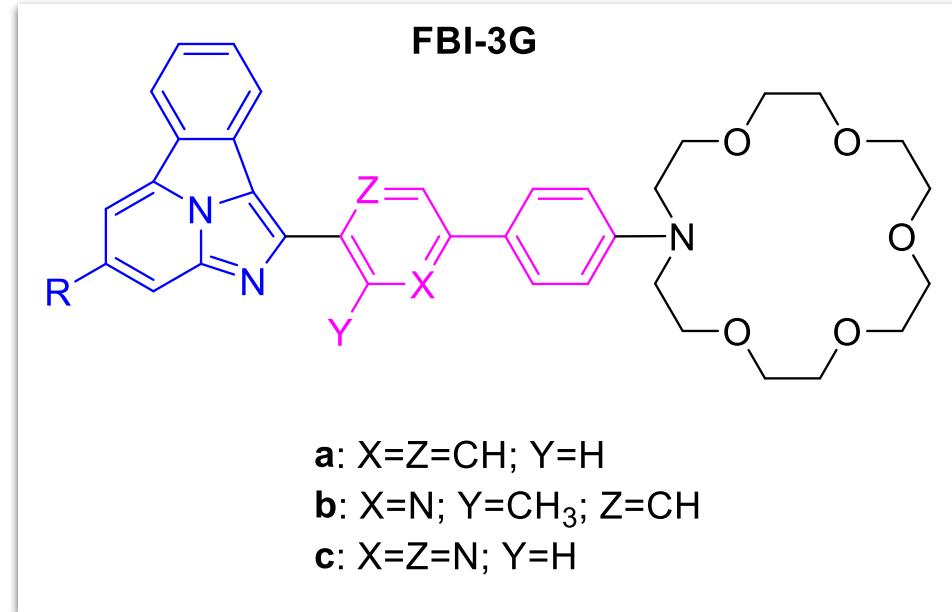
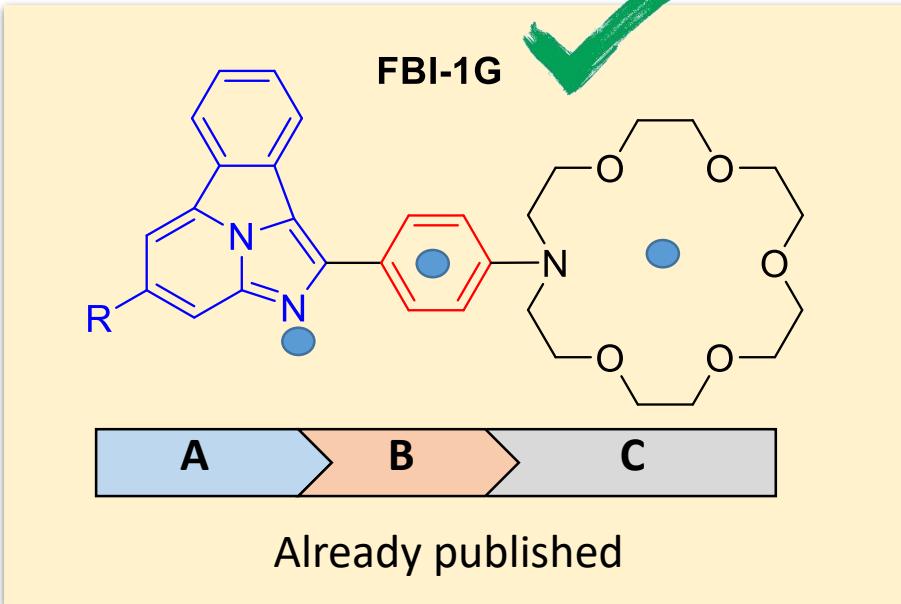
FBI Series: Four Generations

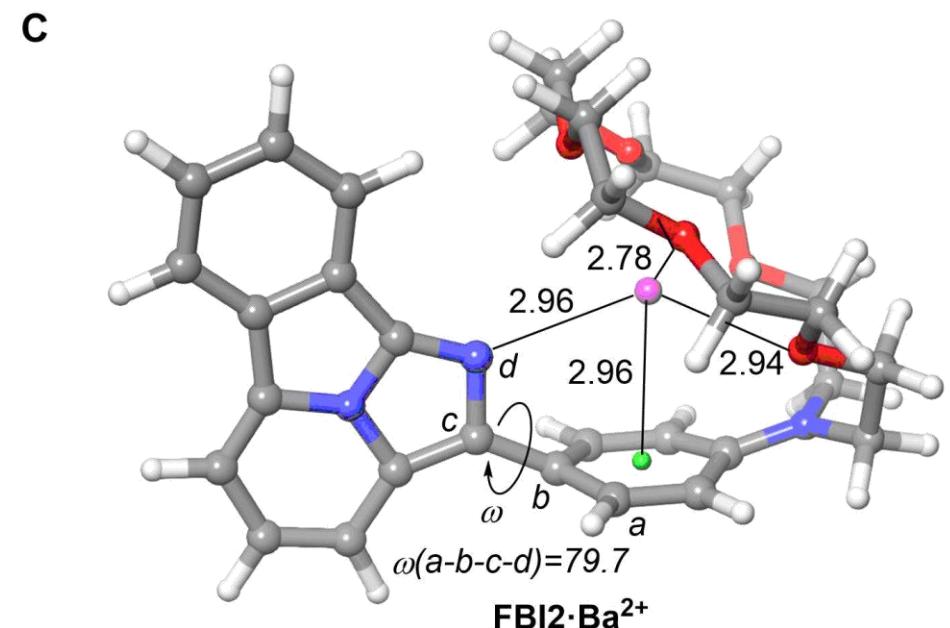
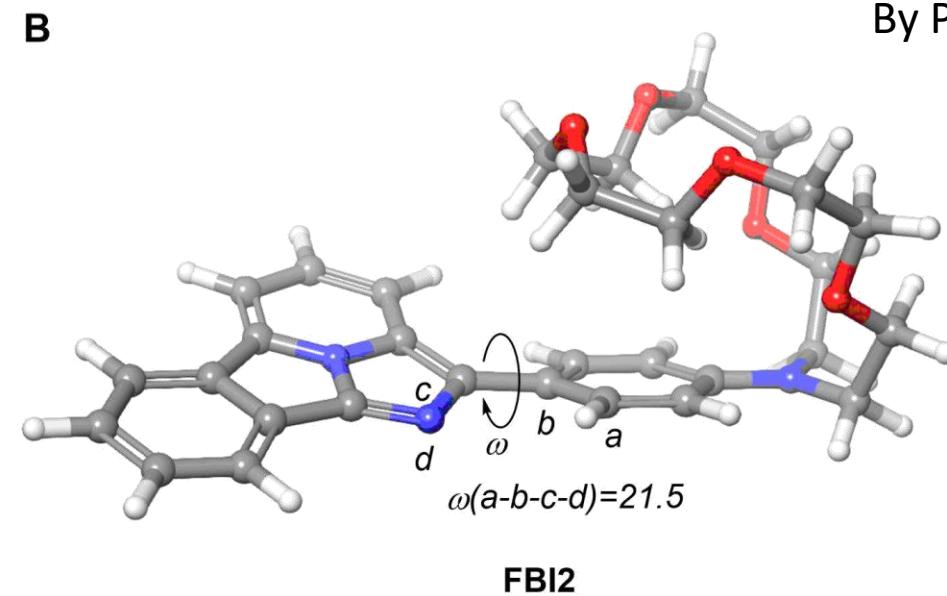
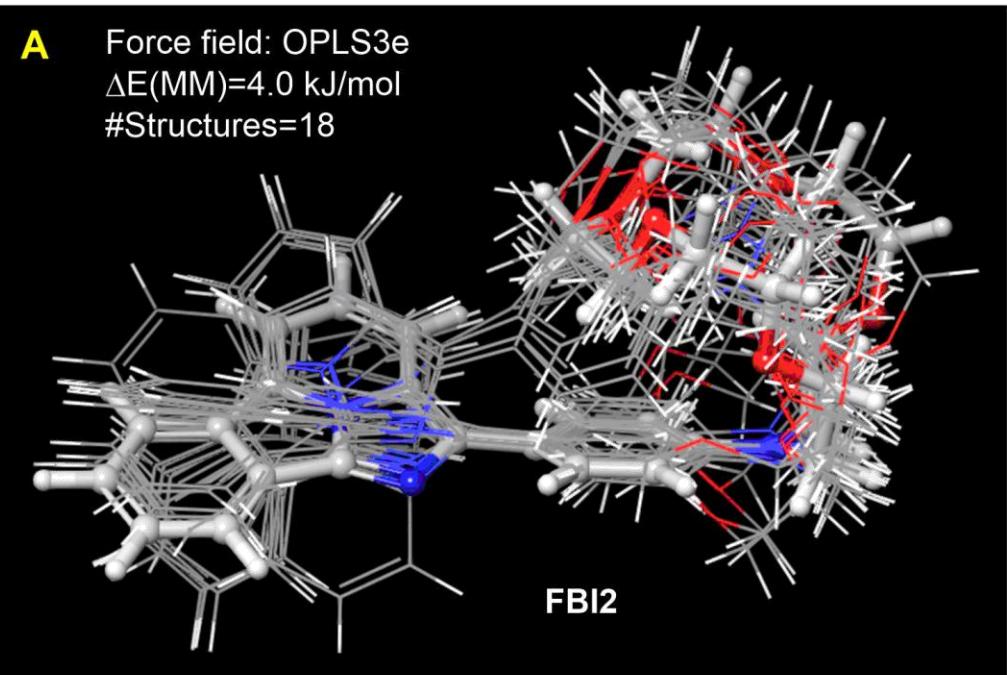


- 3 components (**A**, **B** and **C**)
- **A** Heterocycle; **B** Phenyl group and **C** Crown Ether
- Interaction with Ba⁺⁺
- Center of Crown Ether
- Pi-cation-interaction-aromatic ring
- N-from heterocycle



FBI Series: Four Generations





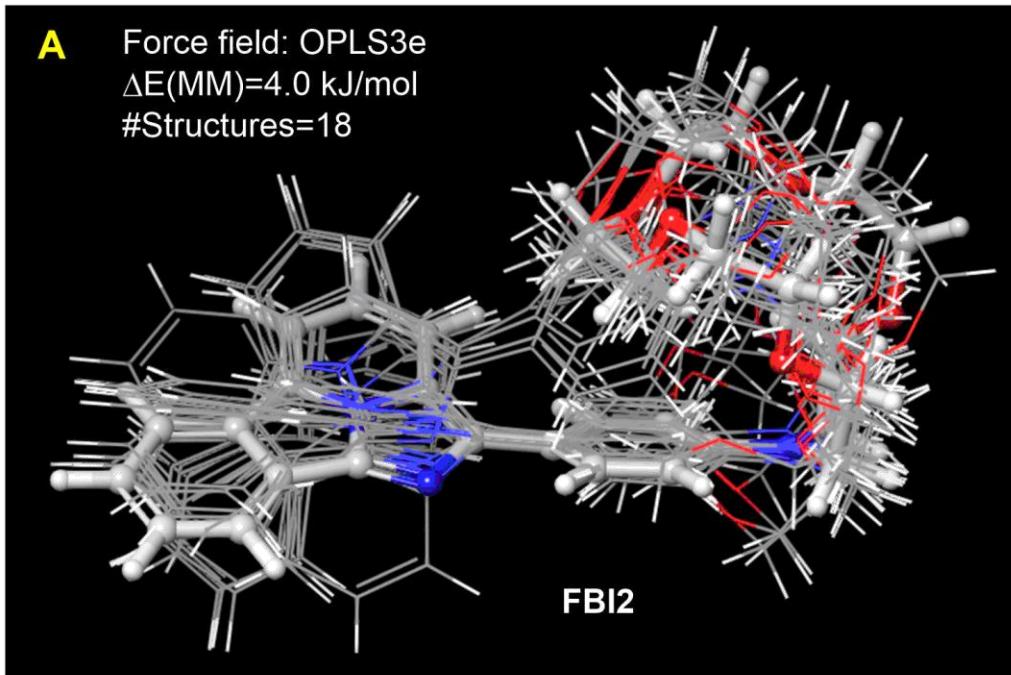
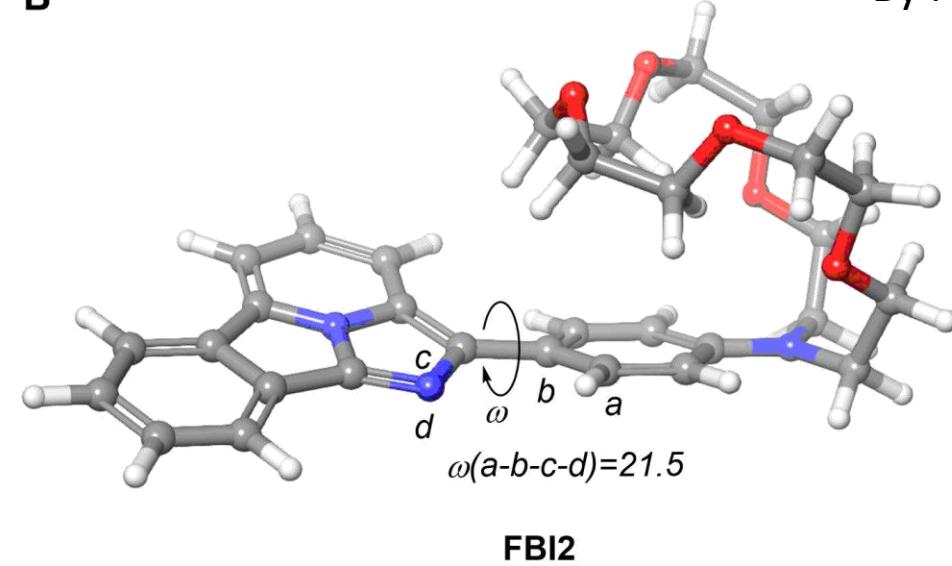
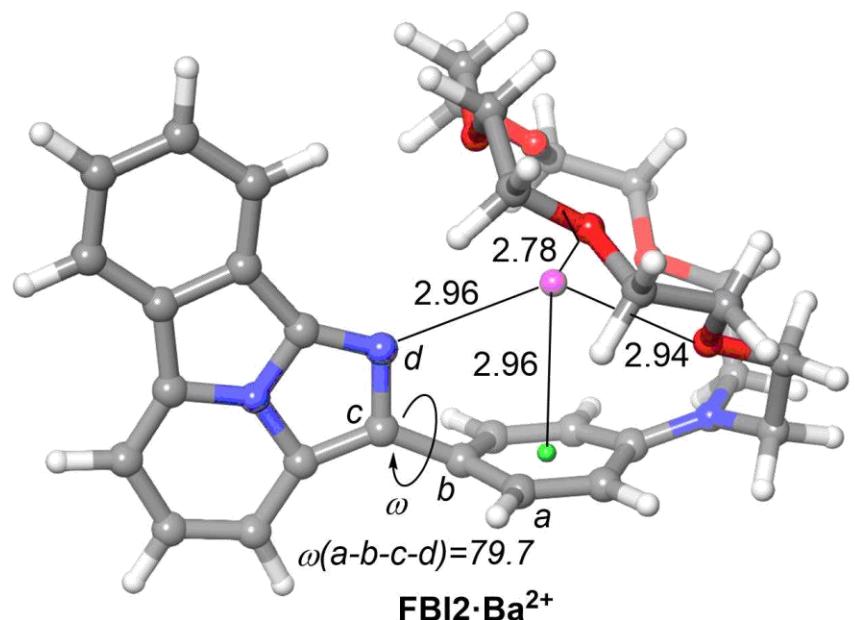
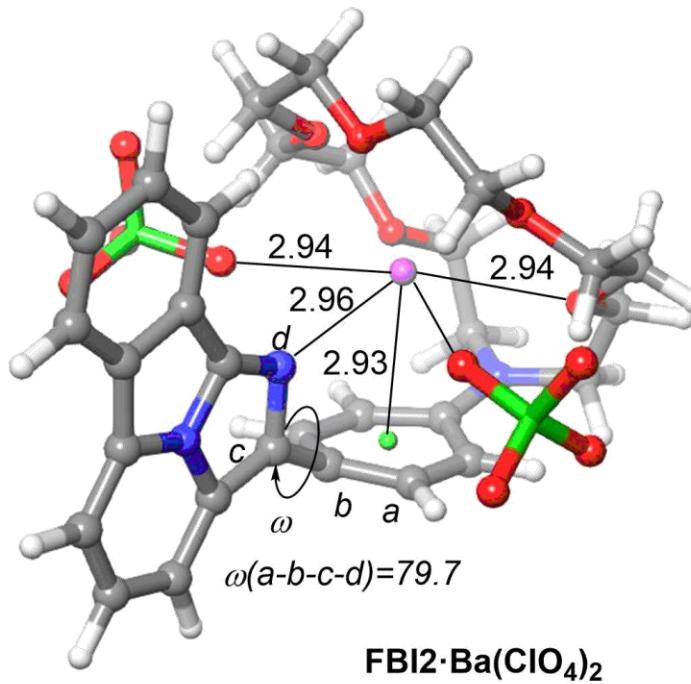
- Strong conformational flexibility
- Phenyl group coplanar with heterocycle (G1)
- Rigid conformation than G1 (Ba⁺⁺)
- The pi-cation interaction is preserved
- Disconnection----color shift

$$\Delta E_b(FBI - G2) = -172.5 \text{ kcal / mol}$$

$$\sum \Delta E_b(\text{optimal}) = -301.2 \text{ kcal / mol}$$

$$\Delta E(\text{deformation}) = +128.71 \text{ kcal / mol}$$

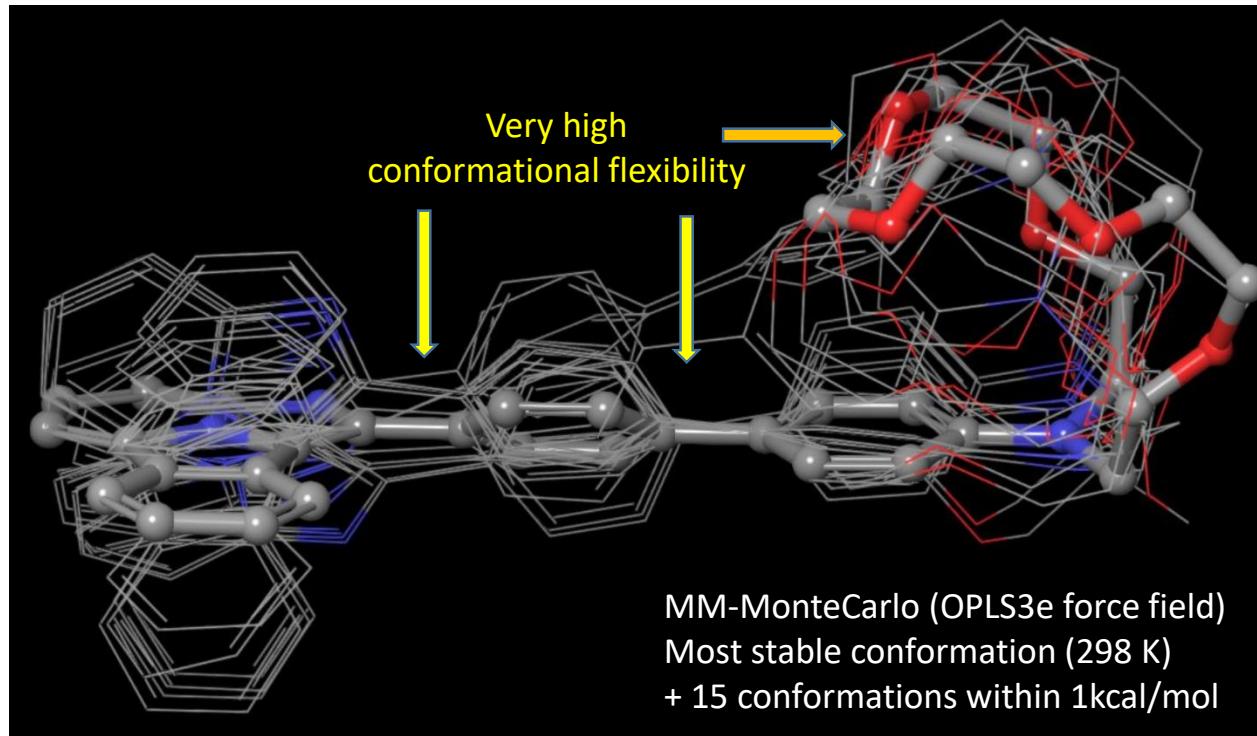
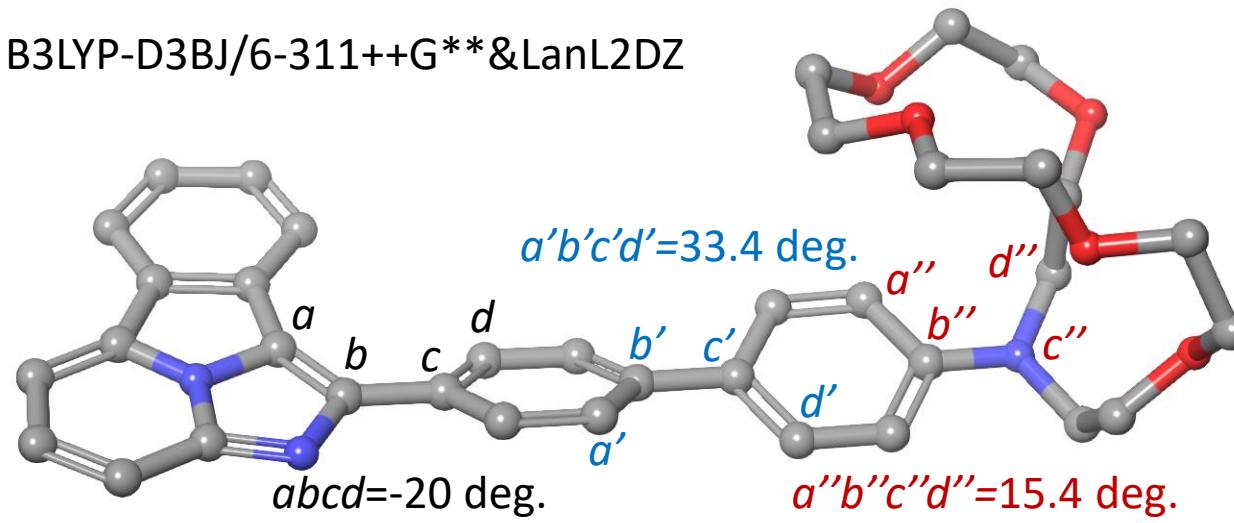
A Force field: OPLS3e
 $\Delta E(MM)=4.0 \text{ kJ/mol}$
#Structures=18

**B****C****D**

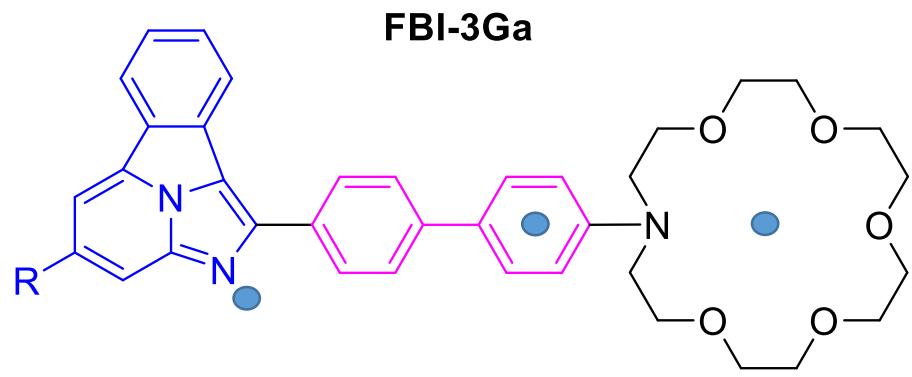
FBI-G3: The parent molecule FBI-G3a

By Prof. F Cossío

B3LYP-D3BJ/6-311++G**&LanL2DZ



- Extended system
- High flexibility respect to G1
- Broad emission spectra

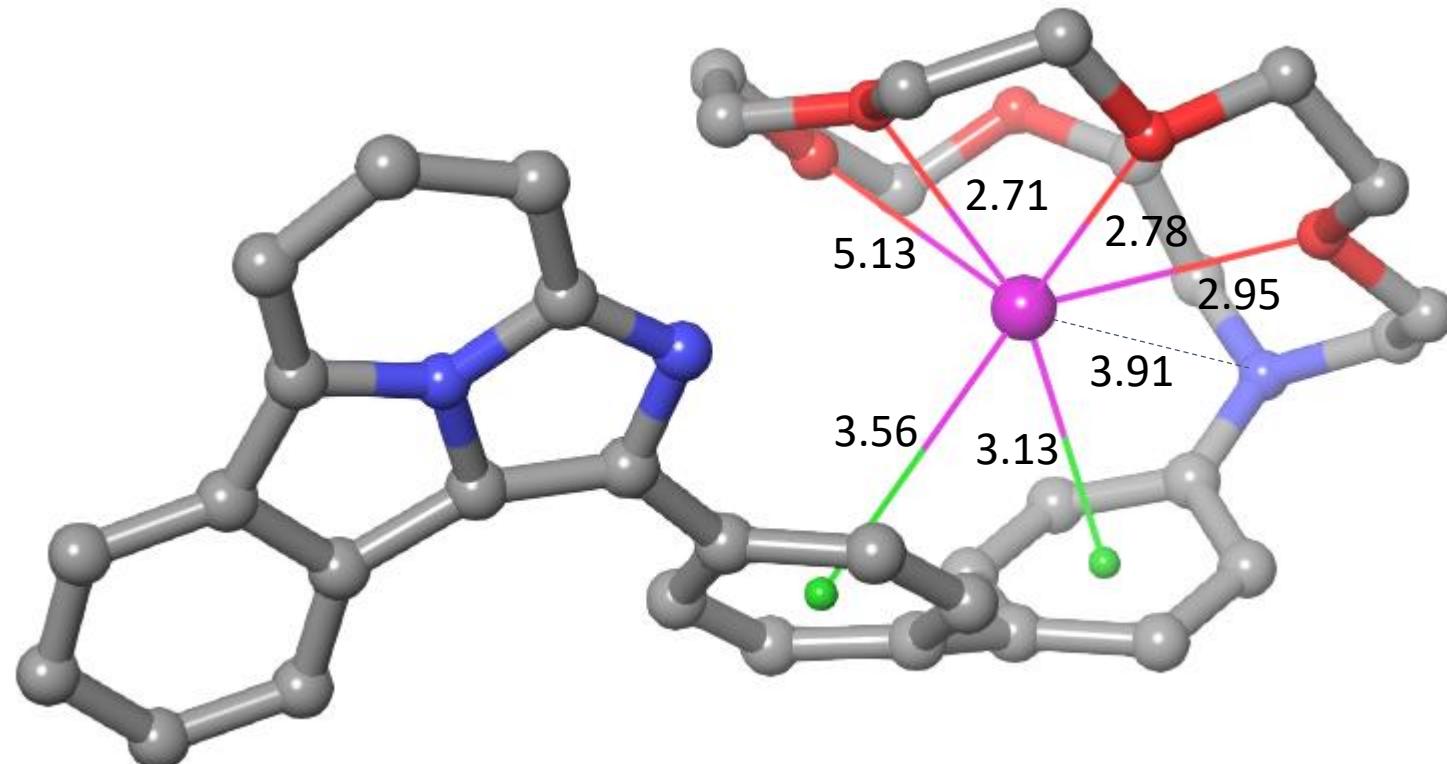


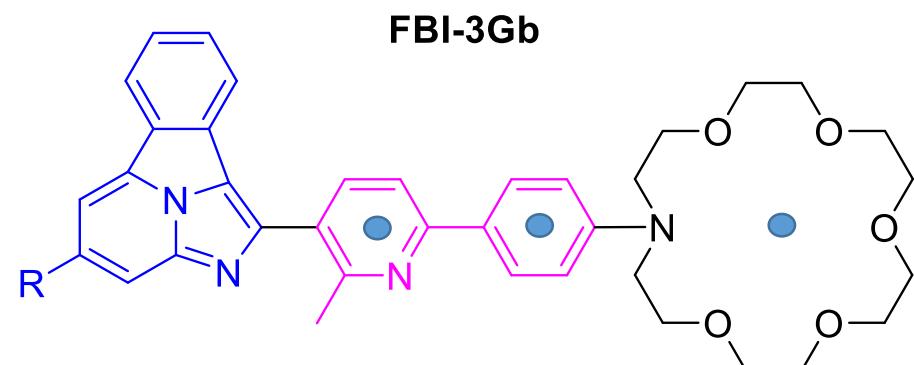
B3LYP-D3BJ/6-311++G**&LanL2DZ

$$\Delta E_b(FBI - G3a) = -157.4 \text{ kcal/mol}$$

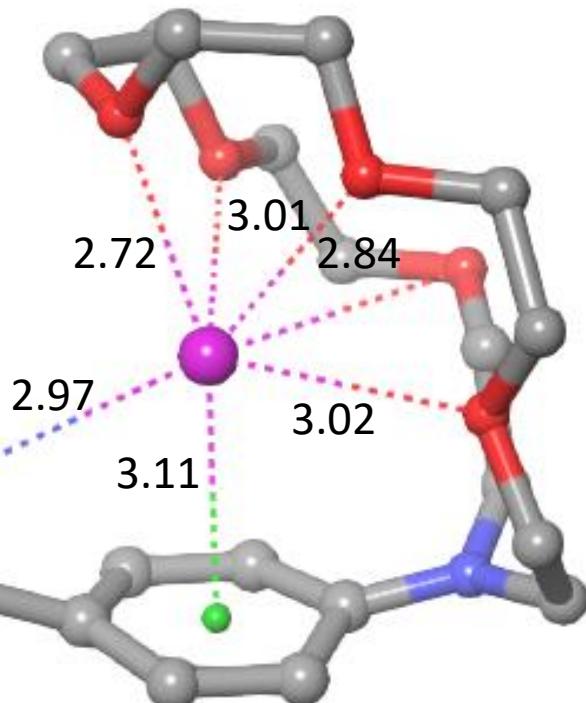
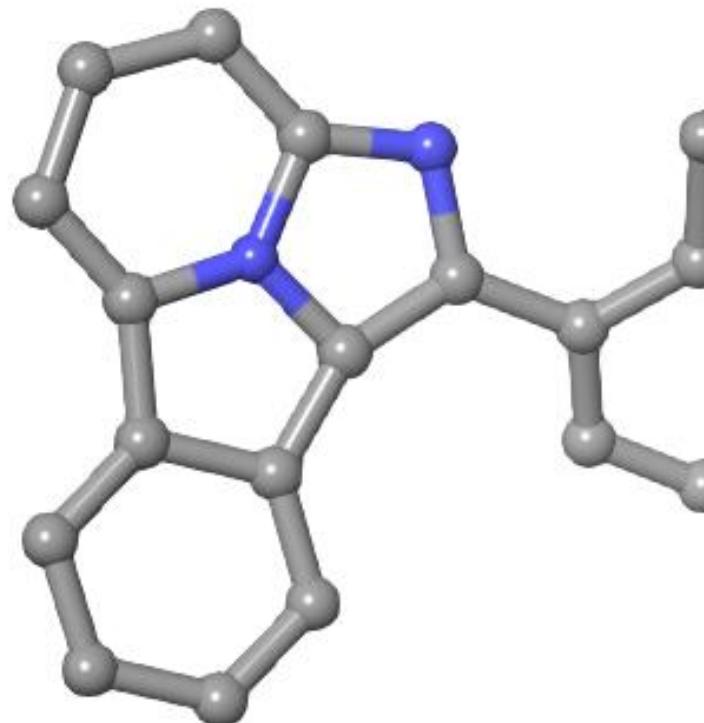
$$\sum \Delta E_b(\text{optimal}) = -315.7 \text{ kcal/mol}$$

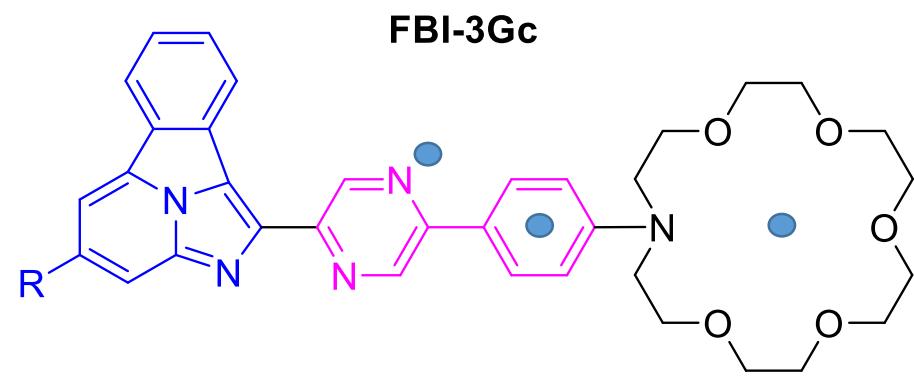
$$\Delta E(\text{deformation}) = +158.3 \text{ kcal/mol}$$



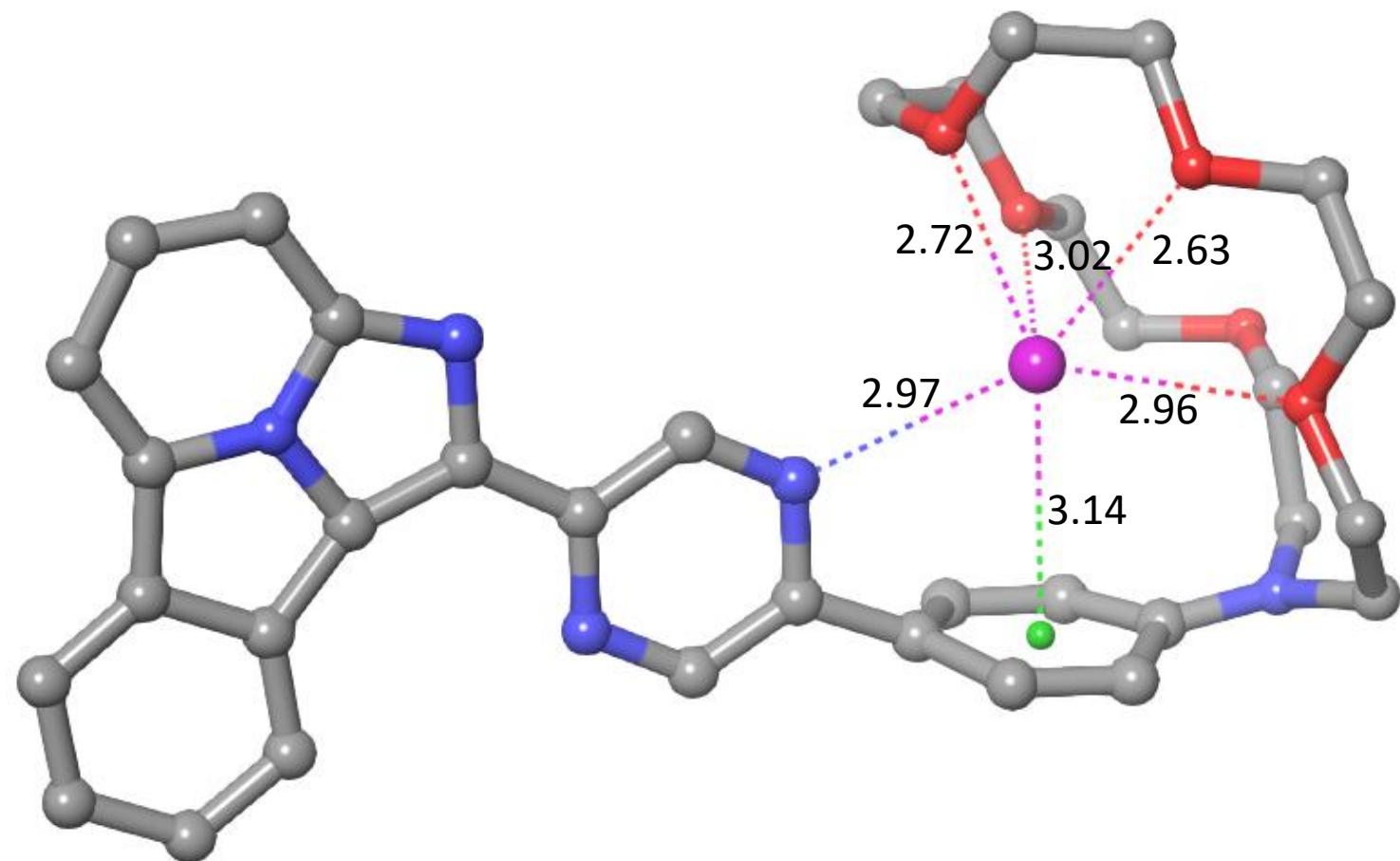


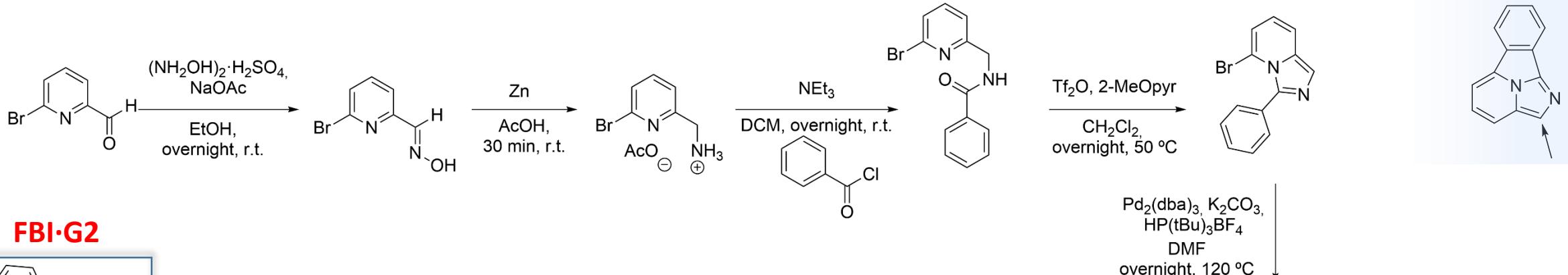
B3LYP-D3BJ/6-311++G**&LanL2DZ



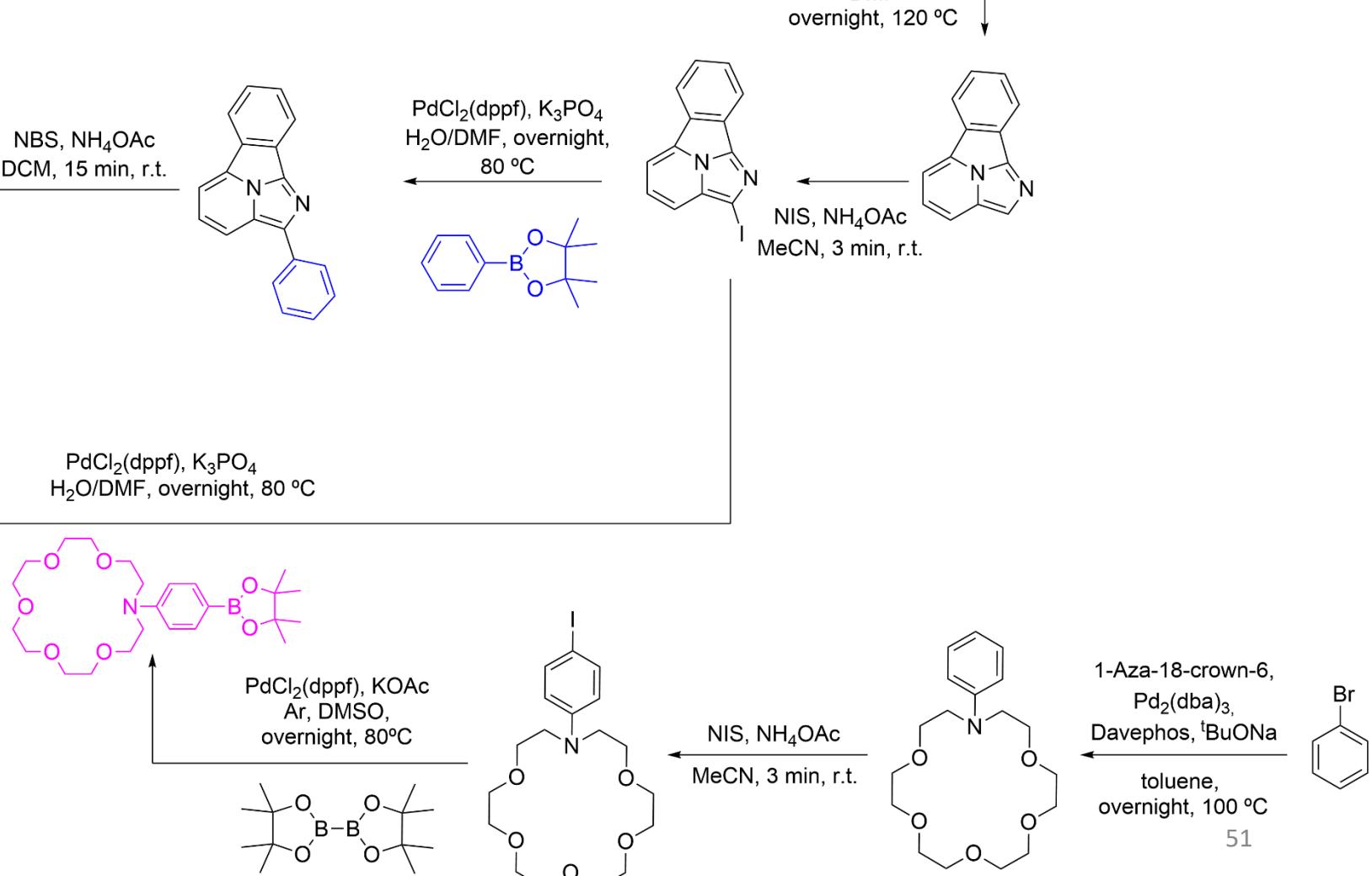
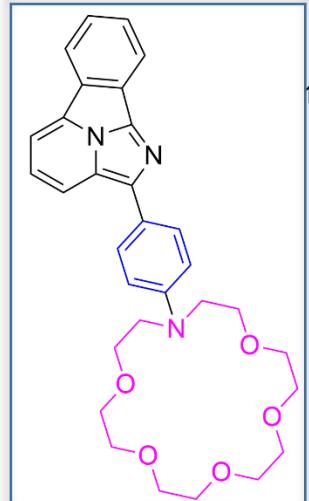


B3LYP-D3BJ/6-311++G**&LanL2DZ



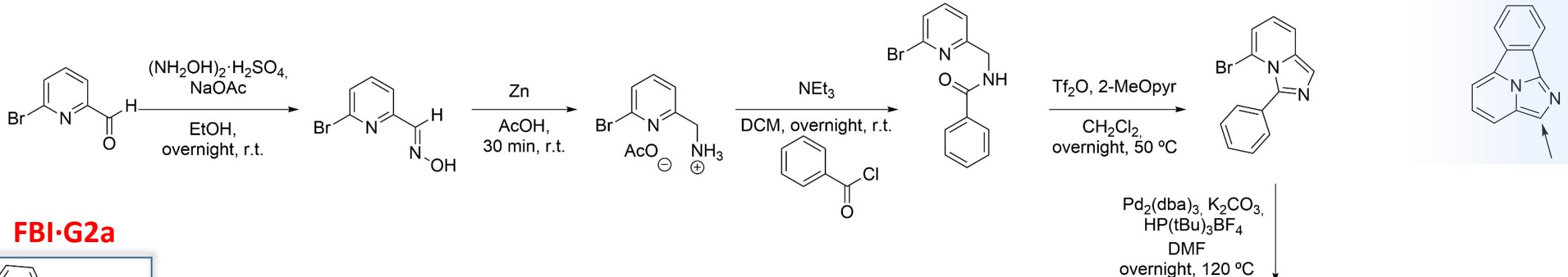


FBI·G2

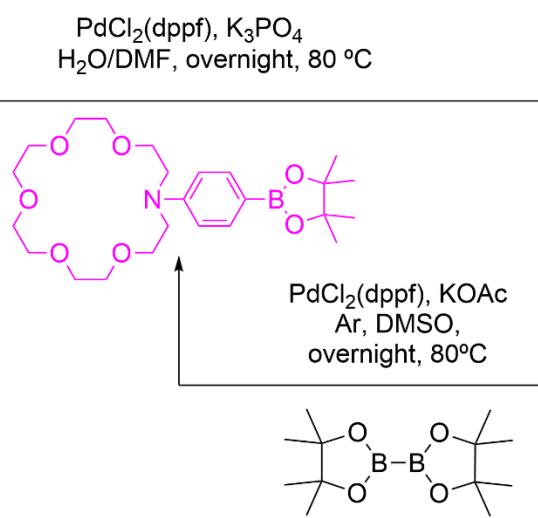
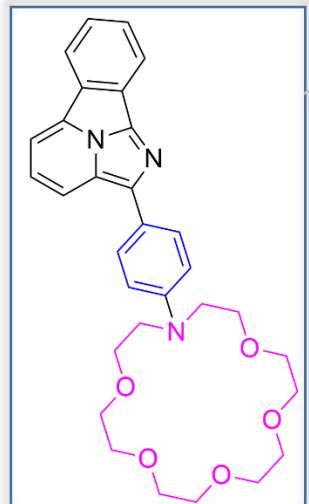


1-Aza-18-crown-6,
 $\text{Pd}_2(\text{dba})_3$,
Davephos, $t\text{BuONa}$
toluene,
overnight, 100 °C

51



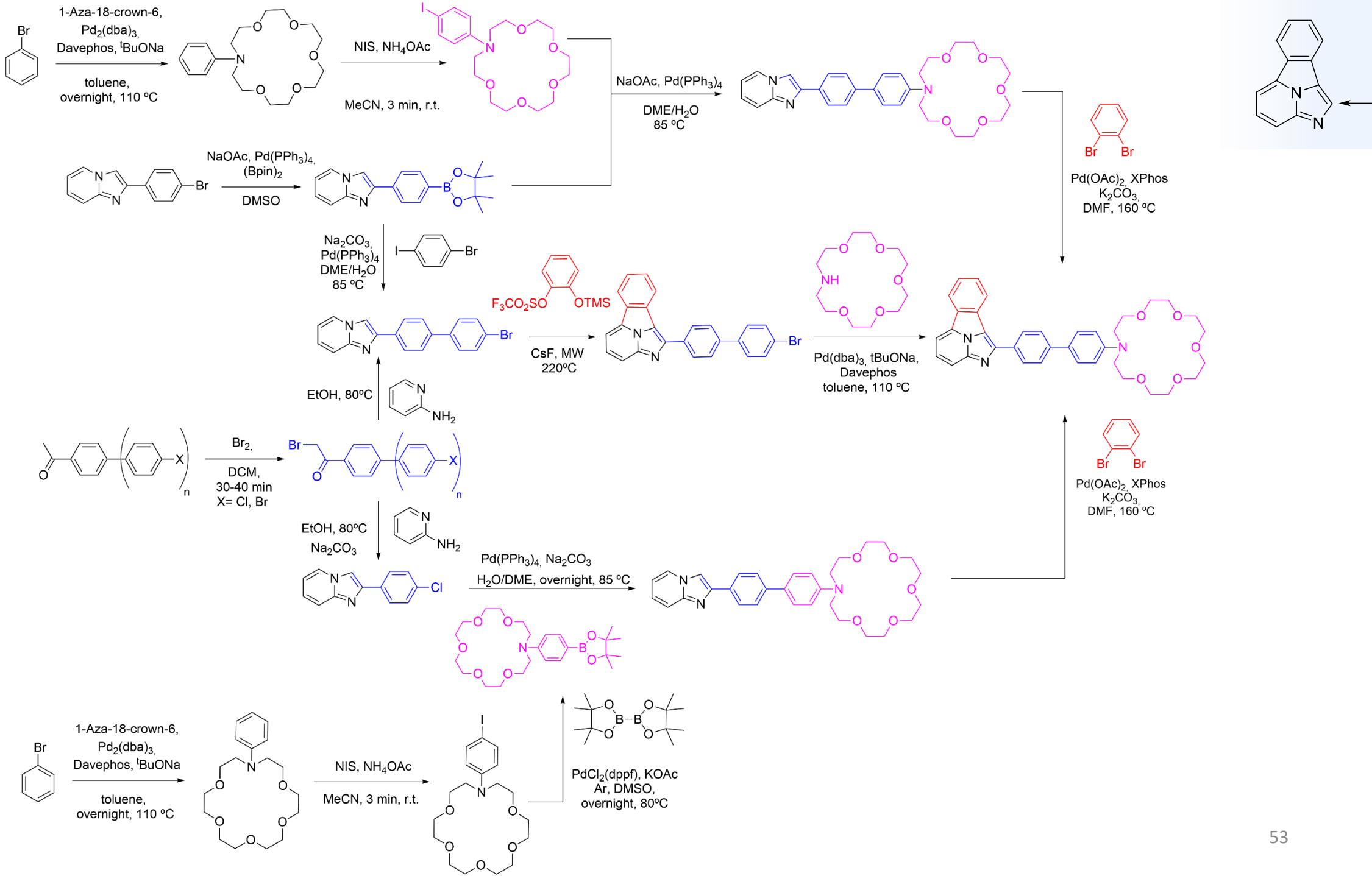
FBI·G2a



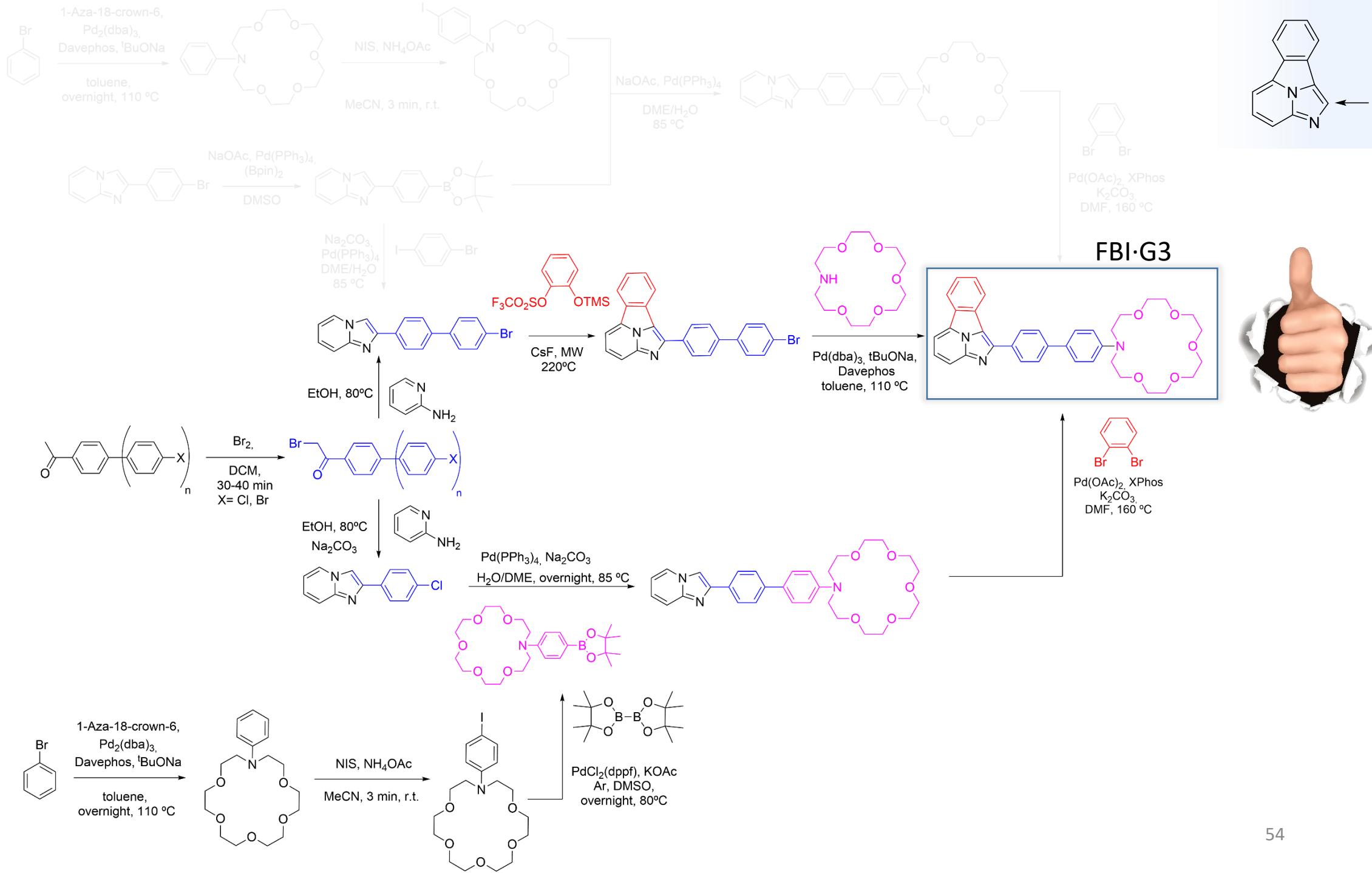
1-Aza-18-crown-6,
 $\text{Pd}_2(\text{dba})_3$,
Davephos, $t\text{BuONa}$
toluene,
overnight, 100 °C

52

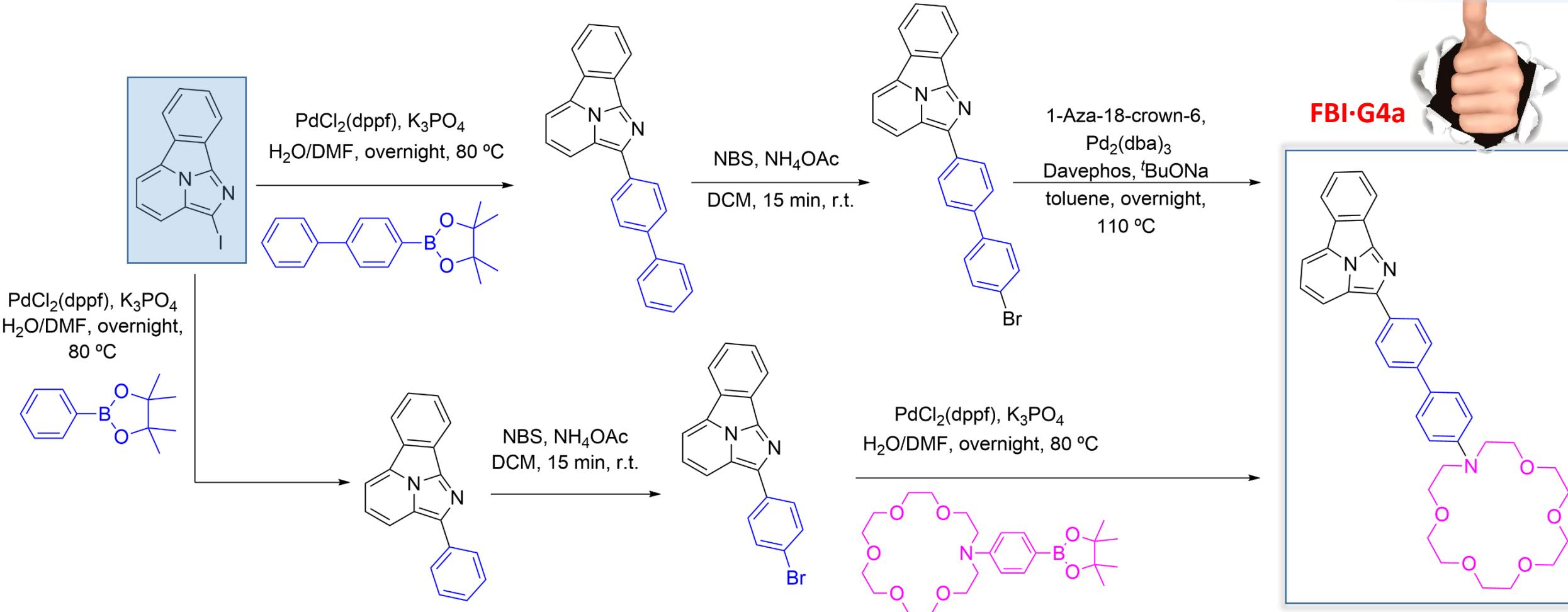
Chemical Synthesis of FBI-G3 Candidates



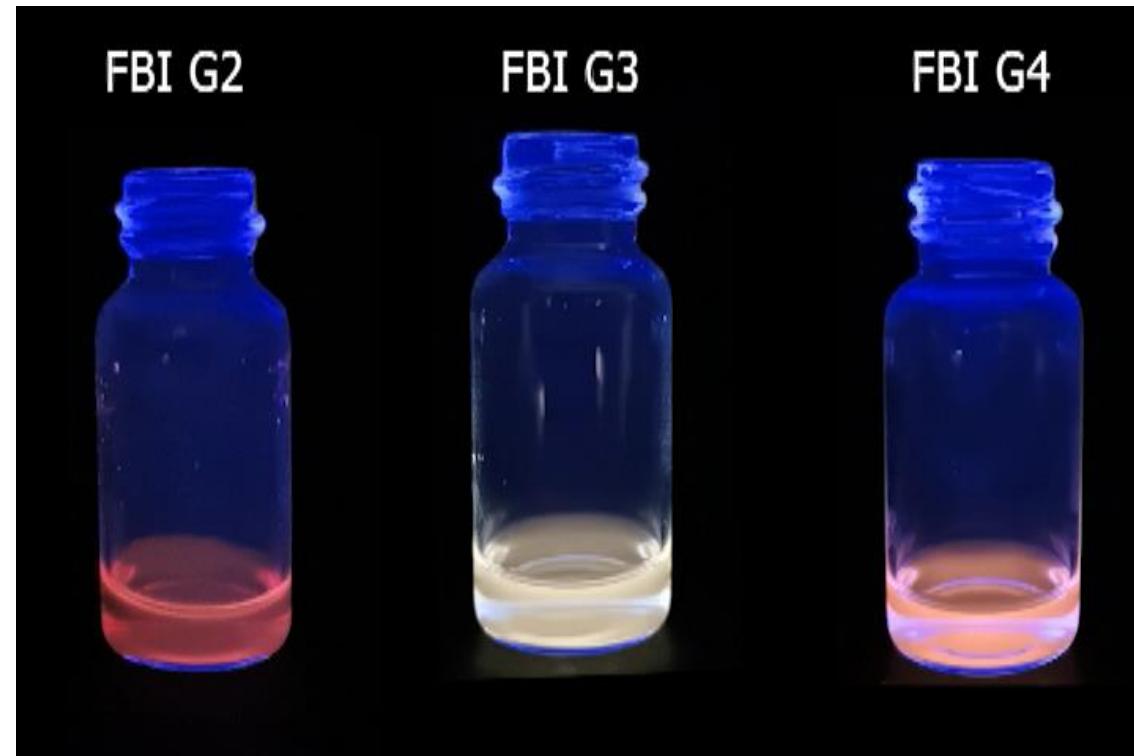
Chemical Synthesis of FBI-G3 Candidates



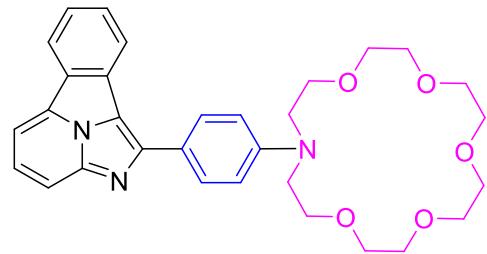
Chemical Synthesis of FBI-G4 Candidates



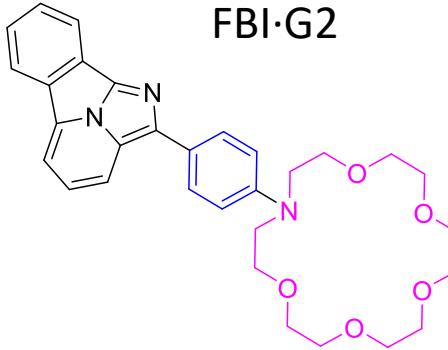
Fluorescent Bicolor Indicators G2, G3 and G4



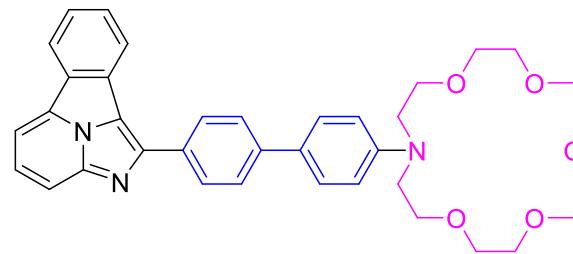
FBI·G1



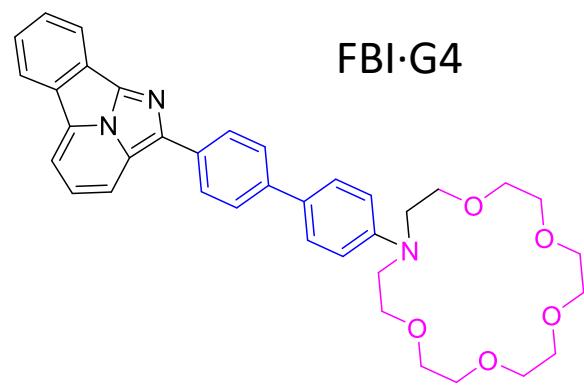
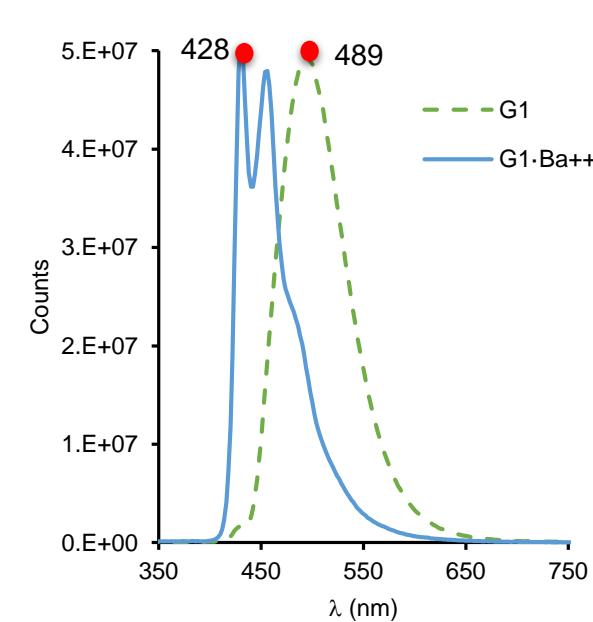
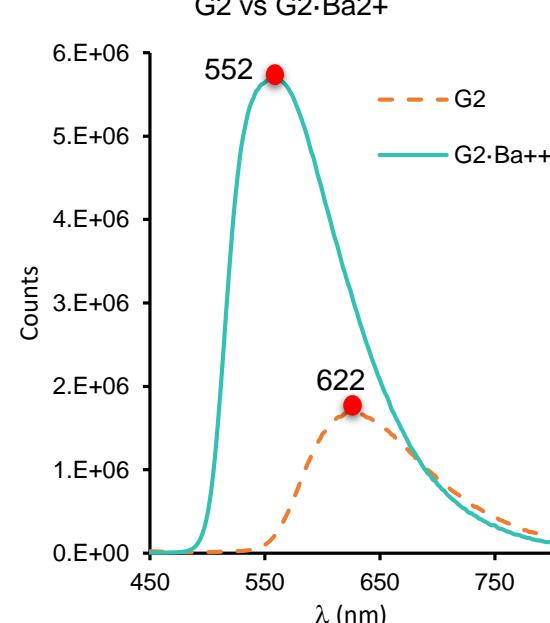
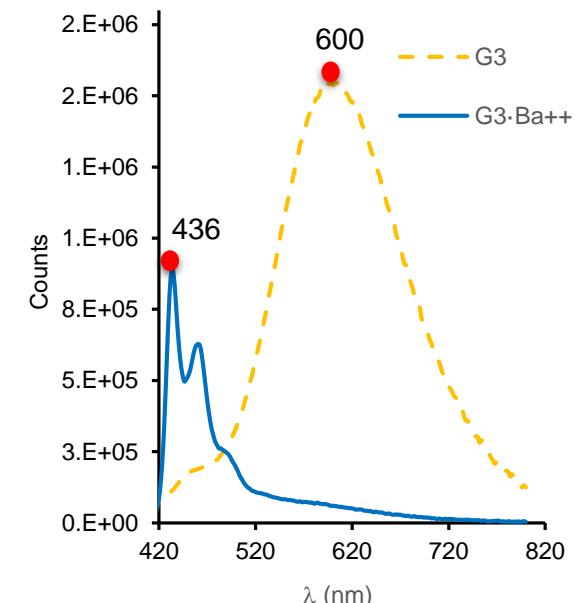
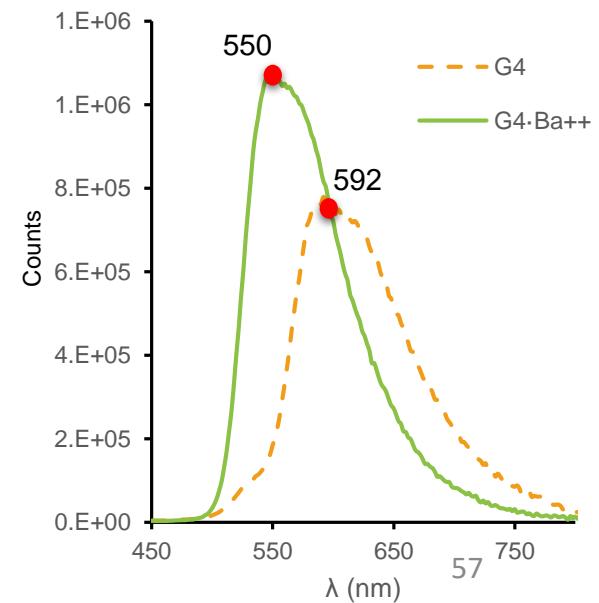
FBI·G2



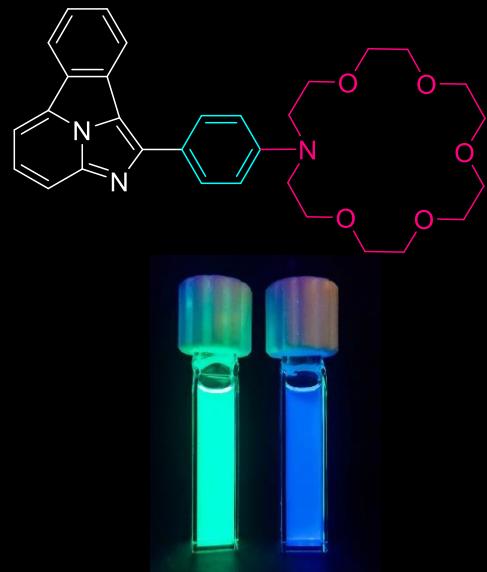
FBI·G3



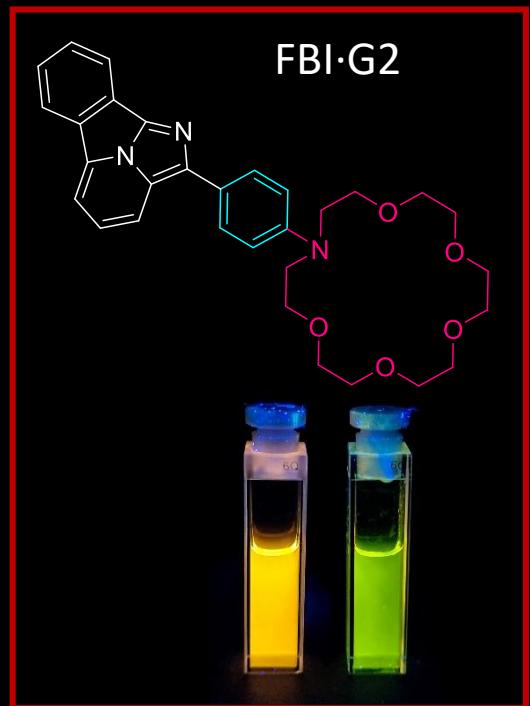
FBI·G4

G1 vs G1·Ba²⁺G2 vs G2·Ba²⁺G3 vs G3·Ba²⁺G4 vs G4 Ba²⁺

FBI·G1



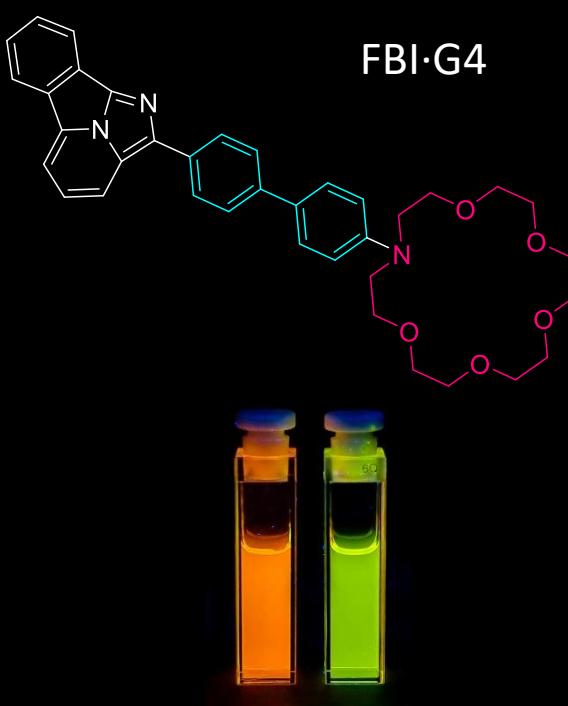
FBI·G2



FBI·G3



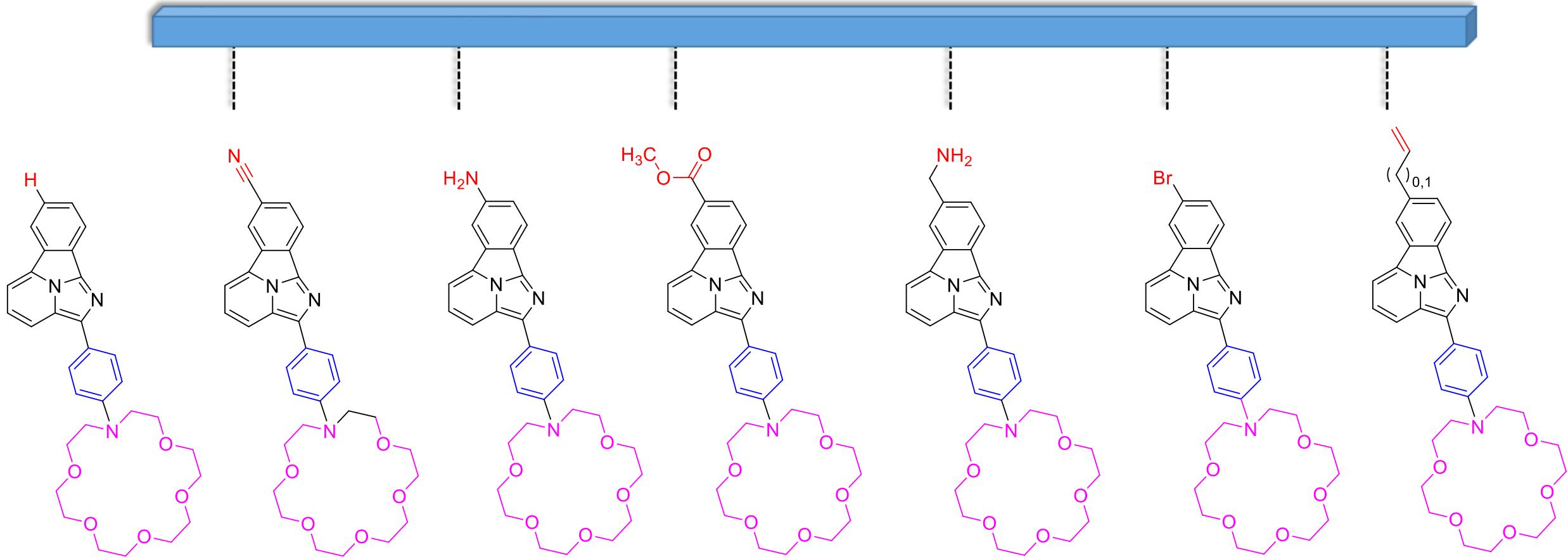
FBI·G4



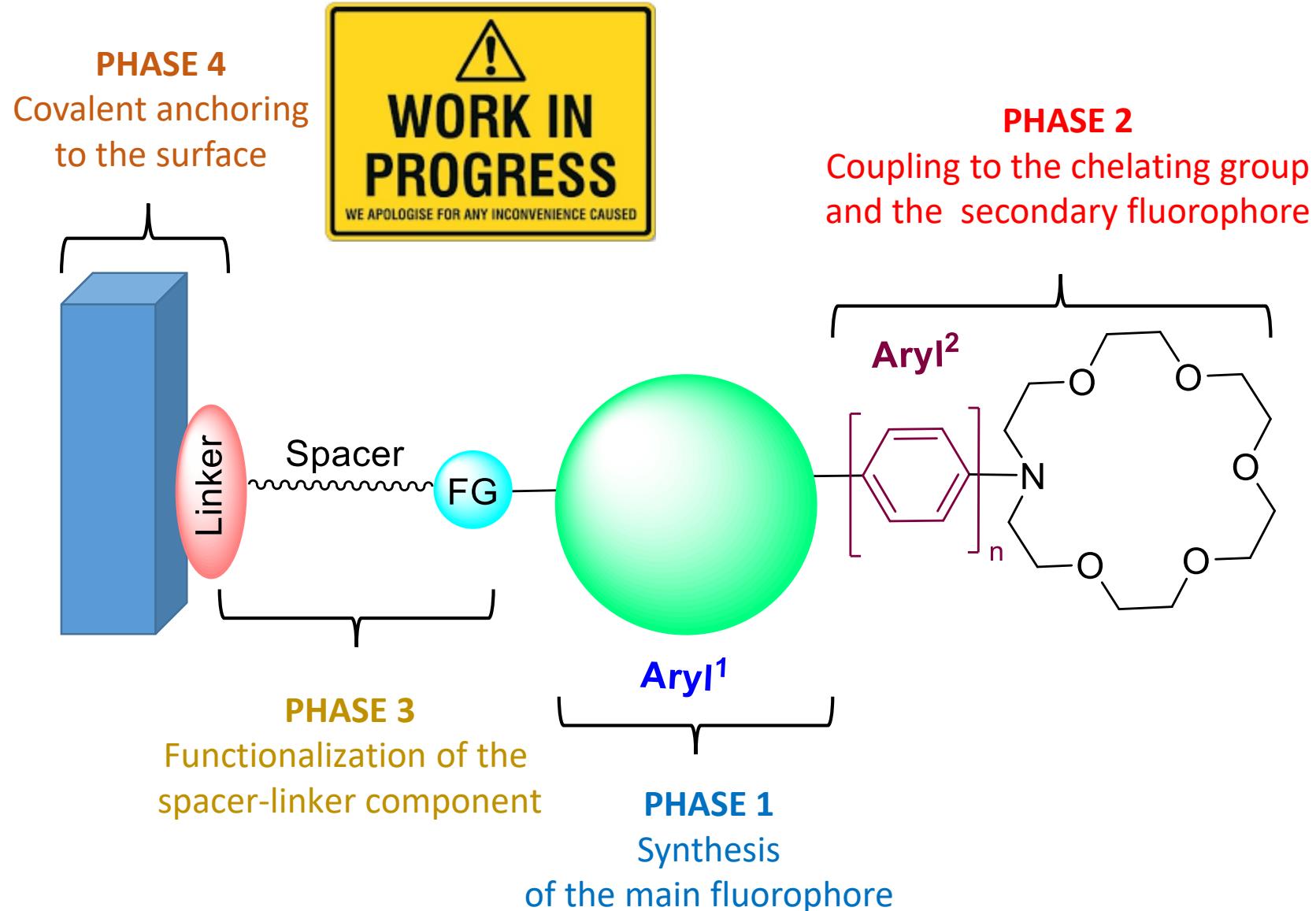
Compound	$\lambda_{\text{em}}^{\text{a}}$ (nm)		$\Delta\lambda$ (nm)	$f_{\lambda_{\text{em}}}^{\text{b}}$	$\phi_{\lambda}^{\text{c}}$		B_{λ}^{c} ($M^{-1} \text{ cm}^{-1}$)		$\varepsilon_{\lambda}^{\text{c}}$ ($M^{-1} \cdot \text{cm}^{-1}$)		$\sigma_{\lambda}^{\text{c}}$ (\AA^2)		K_a^{d} (M^{-1})
	Gn	Gn·Ba ²⁺			Gn	Gn·Ba ²⁺	Gn	Gn·Ba ²⁺	Gn	Gn·Ba ²⁺	Gn	Gn·Ba ²⁺	
G1	489	428	61	179.7	0.67	0.45	16,388	9,937	24,460	22,084	0.94	0.84	$5.3 \cdot 10^4$
G2	622	552	70	210.6	0.23	0.55	5,990	16,564	26,044	30,117	1.00	1.15	$1.3 \cdot 10^6$
G3	600	436	164	5.5	0.74	0.61	13,380	3,068	18,081	5,031	0.69	0.19	$5.0 \cdot 10^5$
G4	592	550	42	8.2	0.41	0.58	5,193	5,794	12,667	9,991	0.48	0.38	$1.0 \cdot 10^5$

^a Emission wavelengths λ_{em} at a $\lambda_{\text{exc_max}}$. ^b Peak discrimination factors (f_{λ}) with respect to unbound fluorophores Gn at λ_{em} . ^c All values for Quantum yields ϕ_{λ} , molecular brightness of the fluorescent emissions (B_{λ}), molar extinction coefficient and cross section were recorded for $\lambda_{\text{(G1)}} = 250 \text{ nm}$; $\lambda_{\text{(G2)}} = 325 \text{ nm}$; $\lambda_{\text{(G3)}} = 368 \text{ nm}$; $\lambda_{\text{(G4)}} = 354 \text{ nm}$. ^d K_a with barium perchlorate in acetonitrile at 298 K using the Benesi–Hildebrand method.

Differently Functionalized FBI-G2 compounds

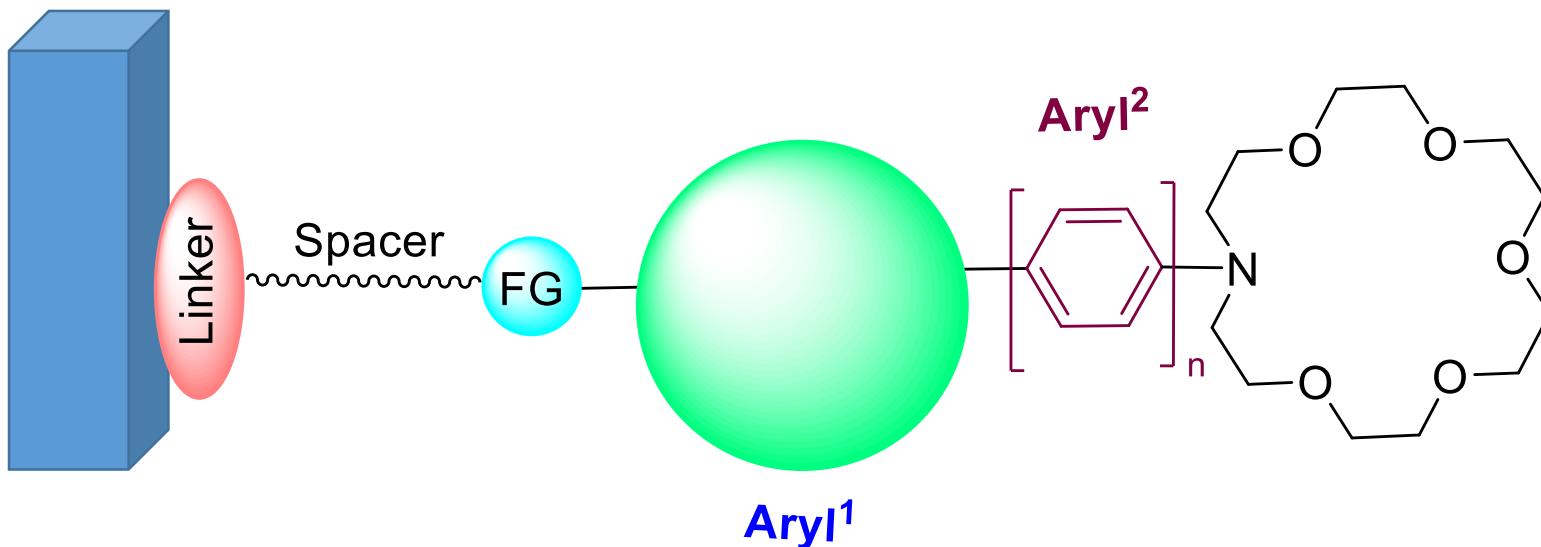


A unified synthetic scheme towards functional surfaces



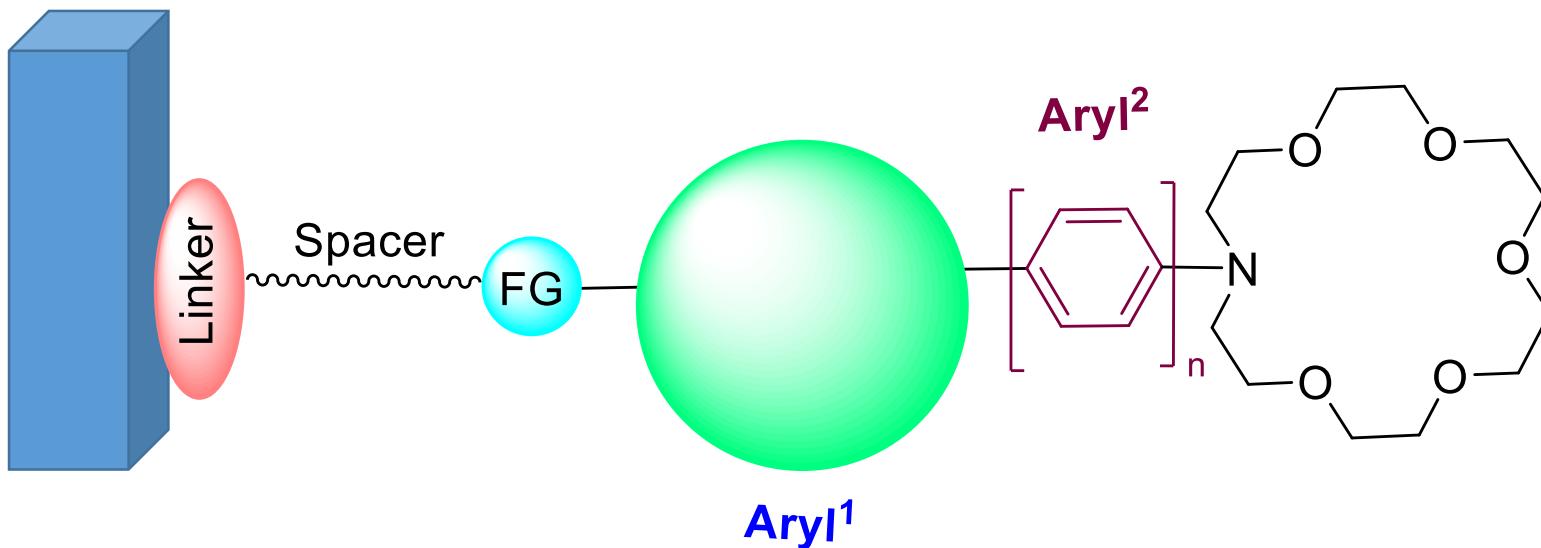
Conclusions and Outlook

- It is possible to expand and optimize the configurational space of fluorescent bicolor indicators (FBIs) by modifying the structure of the hereroaromatic polycyclic (*Aryl*¹) and the *para*-phenylene/biphenylene (*Aryl*²) components.



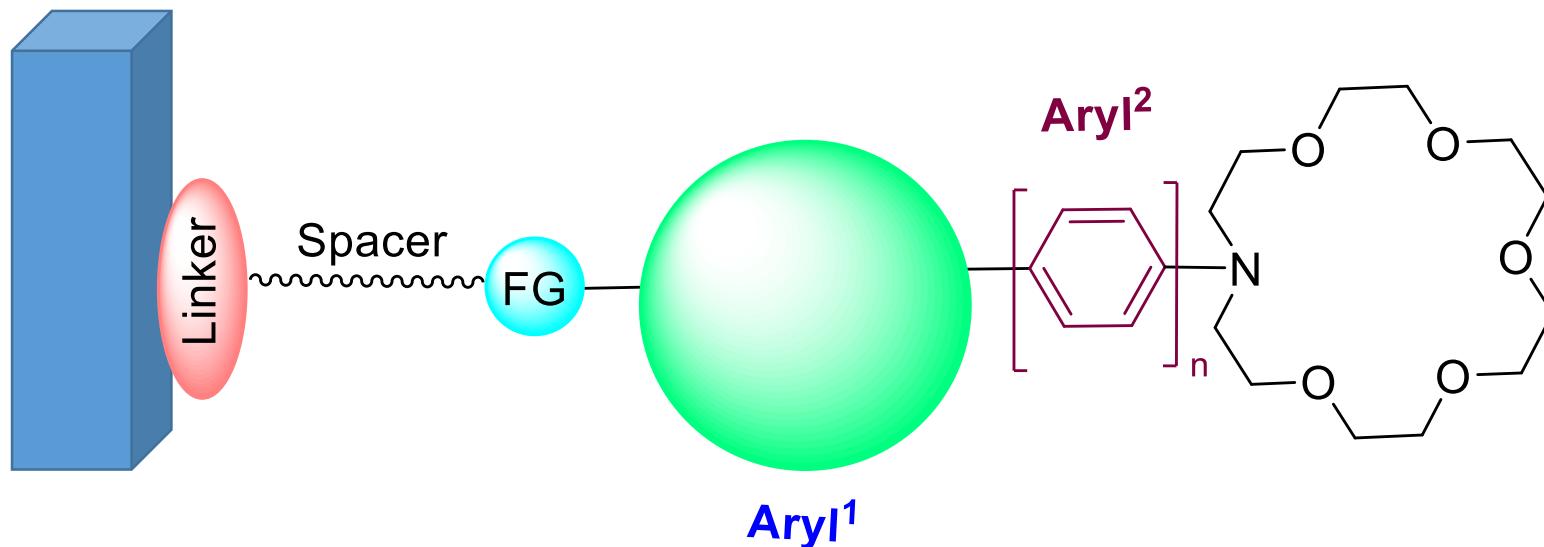
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- Work is in progress to modify this novel family of sensors in order to generate suitable functionalized surfaces.





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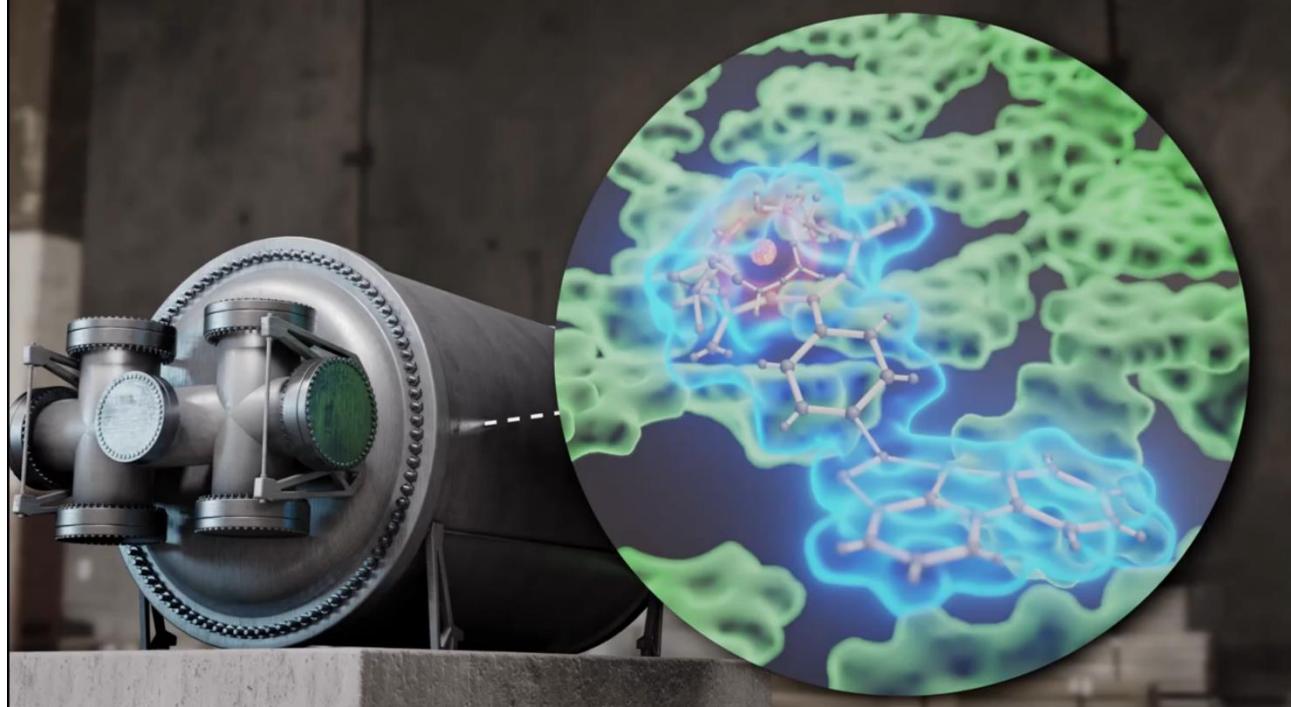
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Searching the Grail: A background free $\beta\beta^0\nu$ experiment using Ba²⁺ tagging in a High-Pressure Xenon Chamber

Iván Rivilla and on behalf of



Zoraida Freixa, Fernando Auria-Luna, Juan Molina-Canteras, Mikel Odriozola-Gimeno, Borja Aparicio, Ane I. Aranburu, Amaia Larumbe, Virginia San Nacianceno, Nerea Alberro, Pablo García, Francesc Monrabal, Pablo Herrero, Celia Rogero, Juan J. Gómez-Cadenas and Fernando P. Cossío



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