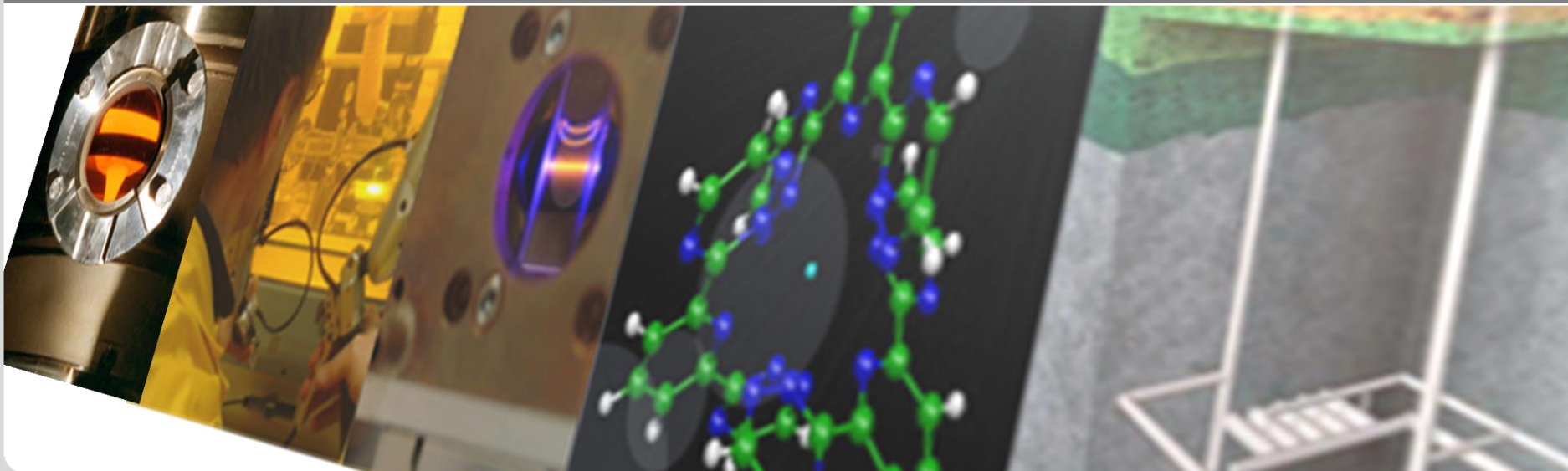


Research at the Institute for Nuclear Waste Disposal (INE)

European Summerschool:
Radiation Measurements and Radiochemistry in Environment
and Decommissioning

01.-05.07.2024

INSTITUT FÜR NUKLEARE ENTSORGUNG (INE)



European Summer School

Radiation Measurements and Radiochemistry in Environment and Decommissioning

Visit at the Institute for Nuclear Waste Disposal (INE),
Karlsruhe Institute of Technology (KIT)

03 July 2024

Programme

07:15 h	Bus to KIT	
09:00 h	Arrival at KIT-Campus North entrance	
09:30 h	Welcome at INE and Introduction	H. Geckeis
09:45 h	Research towards radioactive waste disposal at KIT-INE	H. Geckeis
10:30 h	Accelerator Mass spectrometry (AMS) for radionuclide analysis	F. Quinto
11:15 h	Decommissioning of nuclear facilities	C. Krauss
12:00 h	Transfer INE - Casino	
12:15 h	Lunch	

2

13:45 h Transfer Casino - KARA

14:00 h Visit: Synchrotron based X-ray spectroscopic characterization of radionuclides

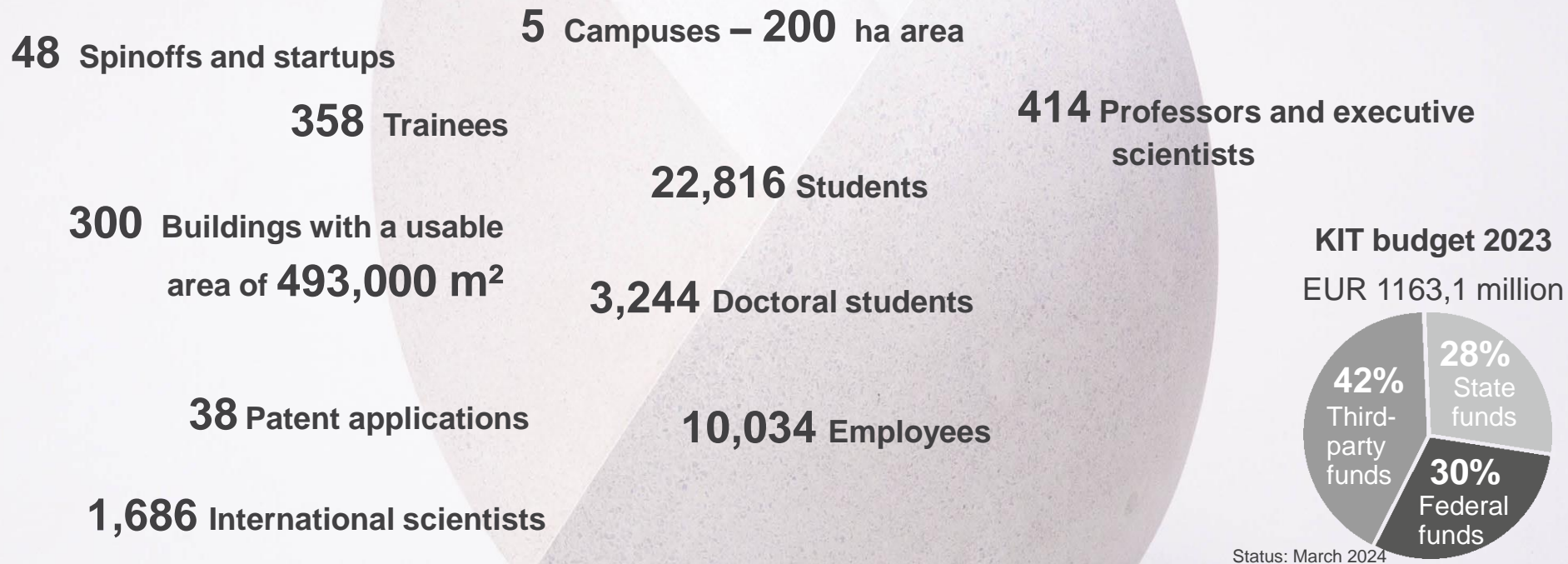
J. Rothe

17:00 h End of visit

Topics

- What is KIT?
- Nuclear waste disposal in Germany
- Research at the Institute for Nuclear Waste Disposal

Figures and Facts 2023



KIT – Research and Innovation at 6 Locations



Campus North



Campus South



Campus East



Campus West



Campus Alpine



Helmholtz Institute Ulm

Five Discipline-focused Divisions

Division I
Biology, Chemistry,
and Process
Engineering

Division II
Informatics,
Economics, and
Society

Division III
Mechanical
Engineering and
Electrical Engineering

Division IV
Natural and Built
Environment

Division V
Physics und
Mathematics

Nine Interdisciplinary KIT Centers



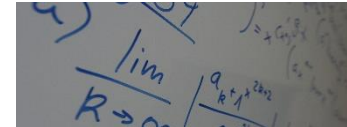
Energy



Mobility Systems



Materials



Mathematics



Humans and Technology



**Climate and
Environment**



**Elementary Particle and
Astroparticle Physics**



**Information - Systems -
Technology**

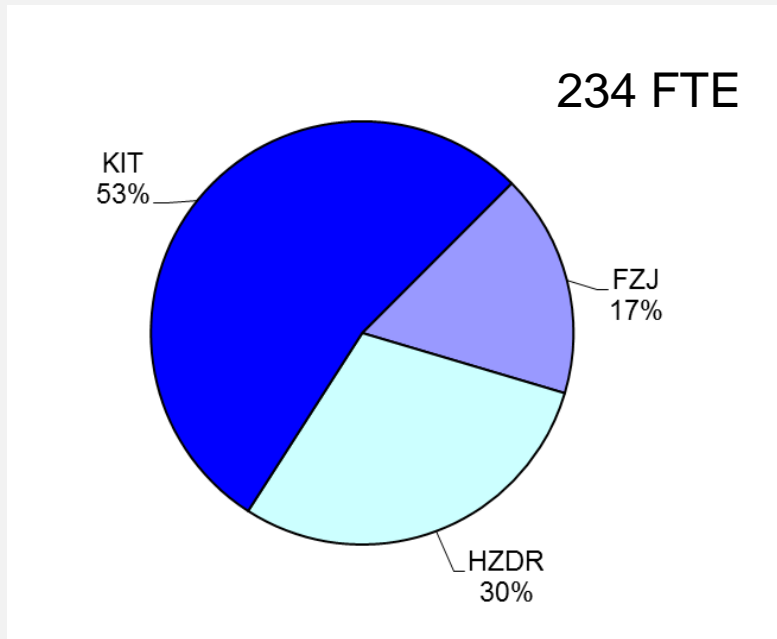


Health Technologies

The research programme NUSAFE

Nuclear Waste Management and Safety as well as Radiation Research

NUSAFE resources:



67% Nuclear Waste Management
33% Reactor Safety



Topics

- What is KIT?
- Nuclear waste disposal in Germany
- Research at the Institute for Nuclear Waste Disposal

15.04.2023

Atomausstieg Neckarwestheim 2

RHEIN-NECKAR-ZEITUNG



Letztes Atomkraftwerk im Südwesten ist vom Netz gegangen

Aus Baden-Württemberg fließt kein Atomstrom mehr ins deutsche Netz. Mit zwei anderen Meilern schreibt das Kernkraftwerk Neckarwestheim 2 Geschichte.

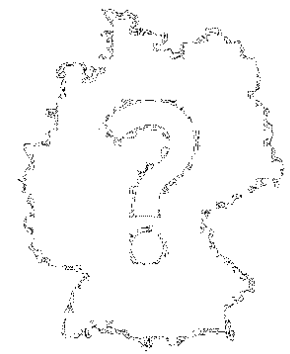
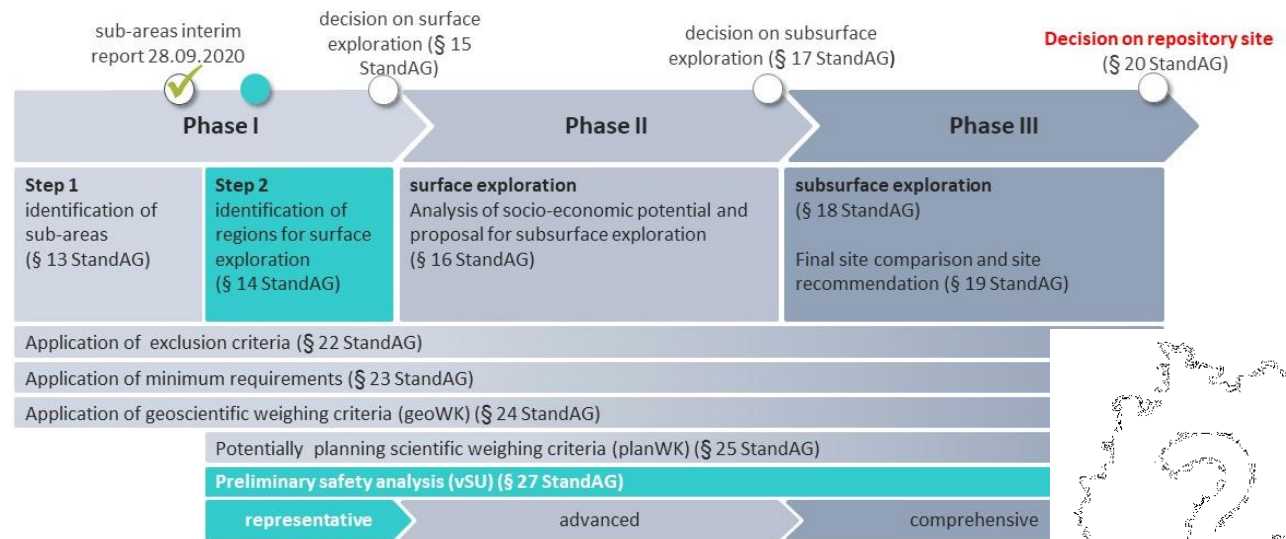
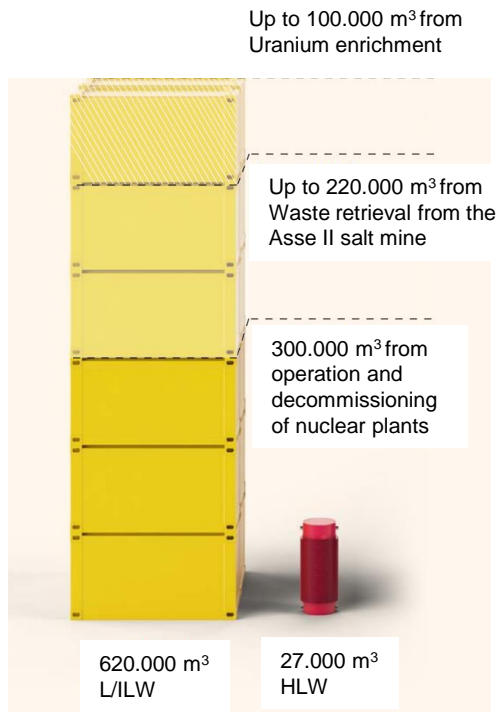


Aus dem Kühlturm am Kernkraftwerk Neckarwestheim steigt Wasserdampf. Mit der Trennung der Kernkraftwerke Isar 2, Neckarwestheim und Emsland vom Stromnetz endete am Samstag die Ära der kommerziellen Stromerzeugung mit Atomkraftwerken in Deutschland. Foto: dpa

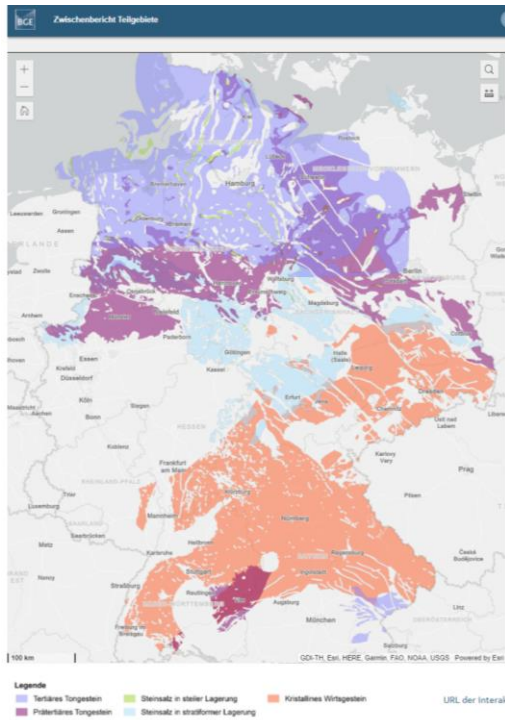
Implementation of the German Site Selection Procedure

27.000 m³ highly radioactive waste (spent nuclear fuel, vitrified waste)

Up to 620.000 m³ low and intermediate level waste (L/ILW) → licensed repository Konrad mine
+ another L/ILW repository to be decided



The site selection process (acc. to the German site selection bill, StandAG, 2017)



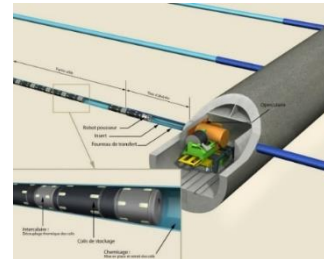
- 54% of the area in Germany is considered worth to be explored
- 90 site areas (60 salt domes, 14 stratiforme rock salt layers, 9 clay rock, 7 crystalline rock areas)



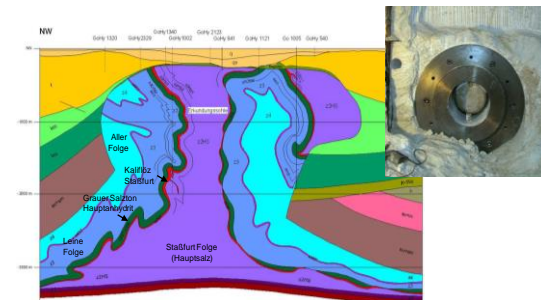
50 mm copper
 Estimated weight (kg):
 Copper canister 7.400
 Insert 13.600
 Fuel assemblies (BWR) 3.600
 Total 24.600

Copper canister for spent nuclear fuel

Crystalline rock



Argillaceous rock



Rock salt

- What is KIT?
- Nuclear waste disposal in Germany
- Research at the Institute for Nuclear Waste Disposal

Institute for Nuclear Waste Disposal (INE) @ KIT



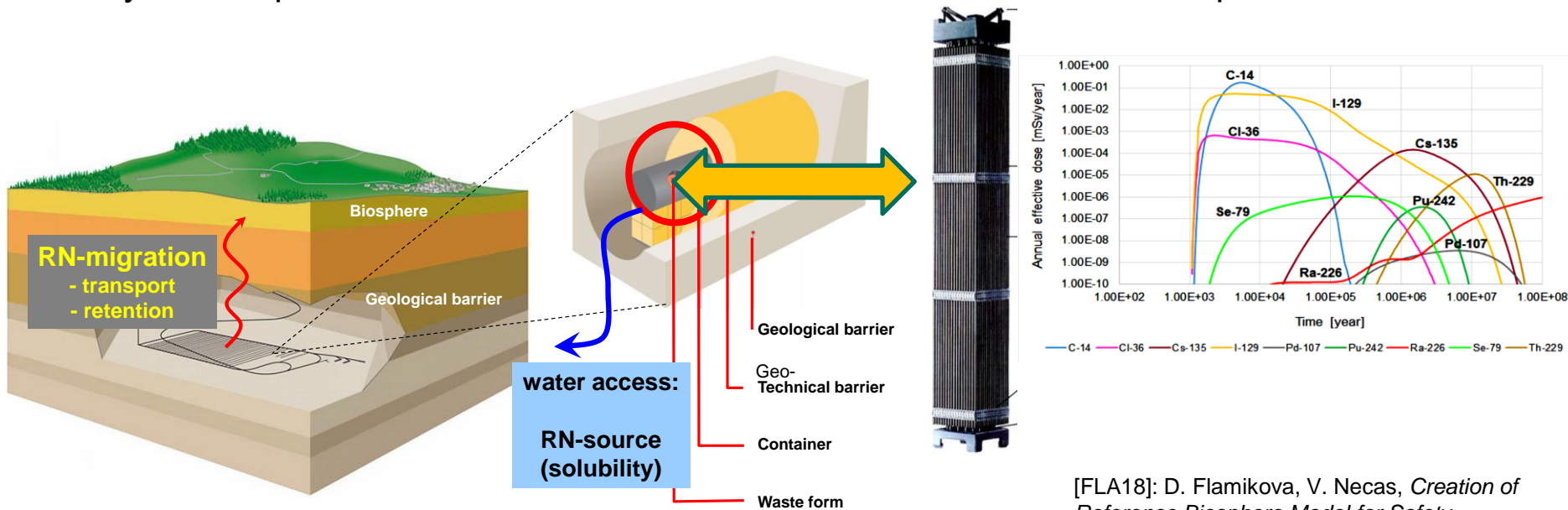
Research and development activities at KIT-INE are an integral component of national provided research and encompass the following areas:

- **Long-term safety research for nuclear waste disposal**
- **Predisposal** – *decommissioning, extended interim storage*
- **Fundamental actinide chemistry**
- **Geothermal energy**



How do radionuclides and radioactive waste behave in a deep geological repository?

- Safety concept of repositories for nuclear waste disposal relies on the multibarrier system → prevent / minimize the release of radionuclides into the biosphere.



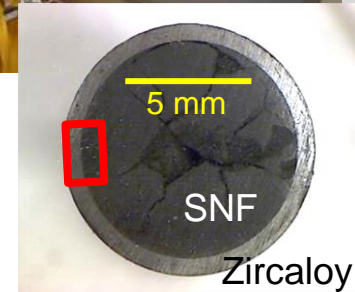
Nagra (2014)

[FLA18]: D. Flamikova, V. Necas, *Creation of Reference Biosphere Model for Safety Assessment of Deep Geological Repository*, AIP Conference Proceedings 1996, 020009 (2018).

Project examples

Research on spent nuclear fuel (SNF) assemblies

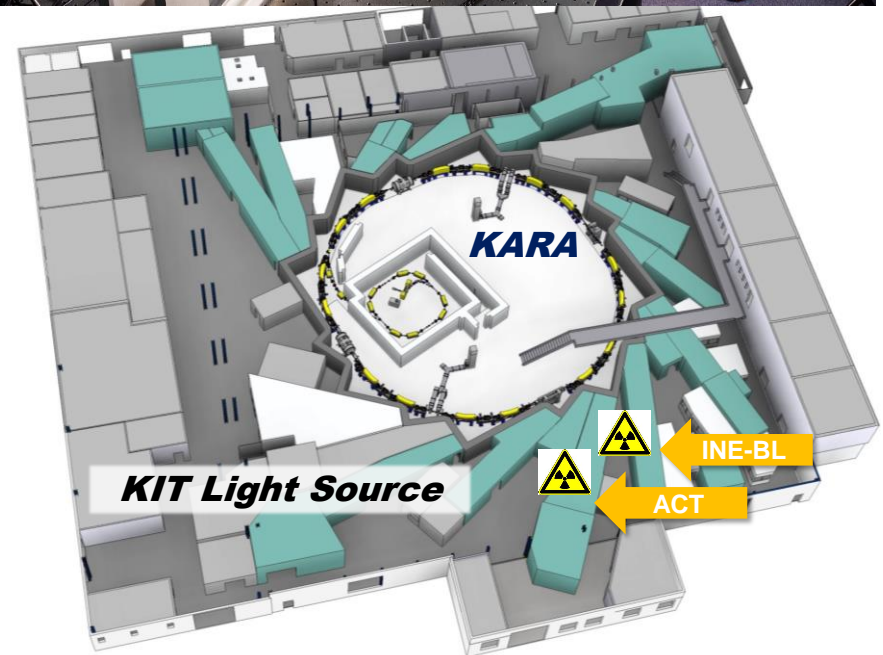
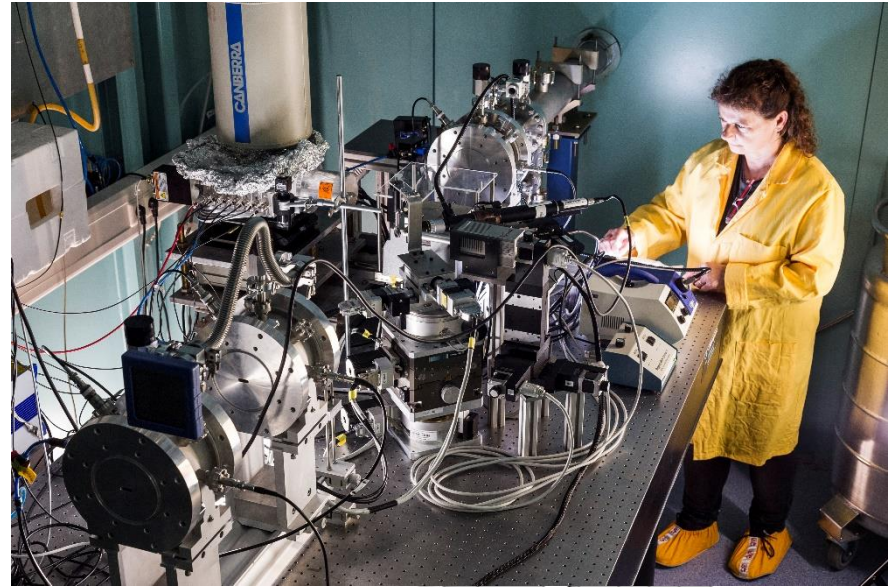
- Quantification of ^{129}I and ^{36}Cl inventories in irradiated SNF and zircaloy:
 - for the 1st time radio-chemical analysis of ^{36}Cl in irradiated Zircaloy, MOX and UO_2 fuels → good agreement between analyses and MCNP calculations for ^{36}Cl in claddings.



PhD thesis: T. König

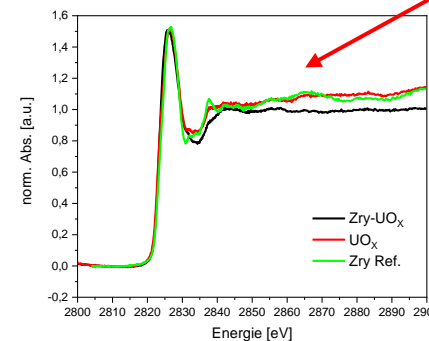
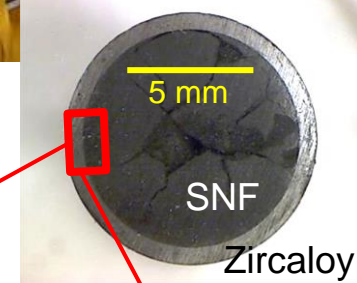
Project examples

Analysis by X-Ray Vision

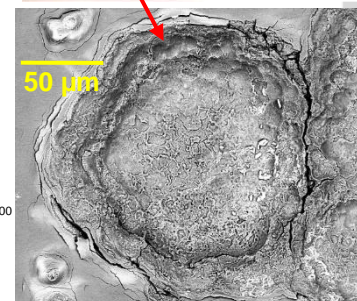


Research on spent nuclear fuel (SNF) assemblies

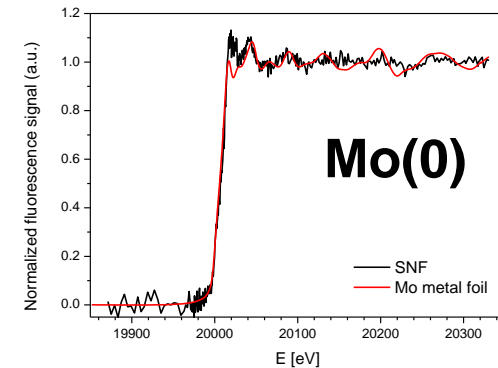
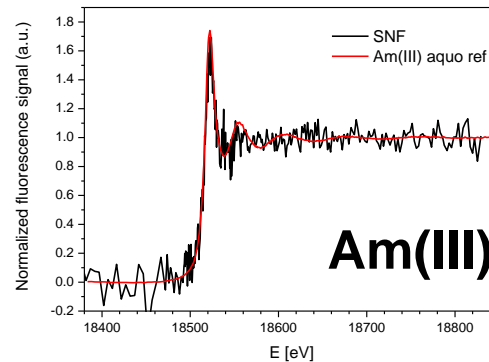
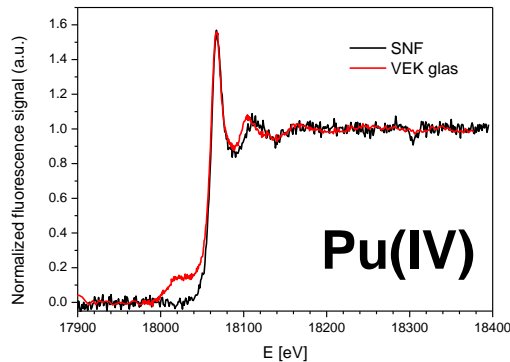
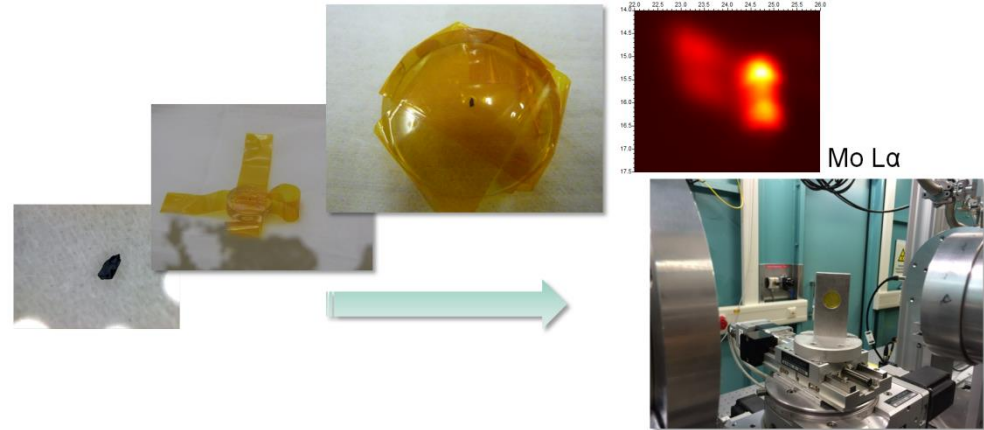
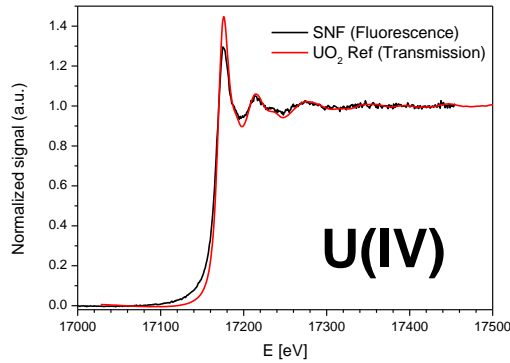
- **Quantification of ^{129}I and ^{36}Cl inventories in irradiated SNF and zircaloy:**
 - for the 1st time radio-chemical analysis of ^{36}Cl in irradiated Zircaloy, MOX and UO_2 fuels → good agreement between analyses and MCNP calculations for ^{36}Cl in claddings.
 - $^{36}\text{Cl}/^{129}\text{I}$ exists in mixed solid phases (Cs, Te, Ba, U, Pu)
 - Cl K-edge XANES characterization points to chemical environment different to CsCl
- **Pellet-cladding interactions (simulated)**
 - Indications for **formation of complex Cs-U-O-Zr-Cl-I compounds** and consecutive **pitting corrosion in cladding** observed at $<400^\circ\text{C}$ (interim dry storage conditions)



Chlorine K-XANES



XAFS/XRF investigation of a SNF particle sampled from a test fuel rod irradiated at the Gösgen (CH) PWR

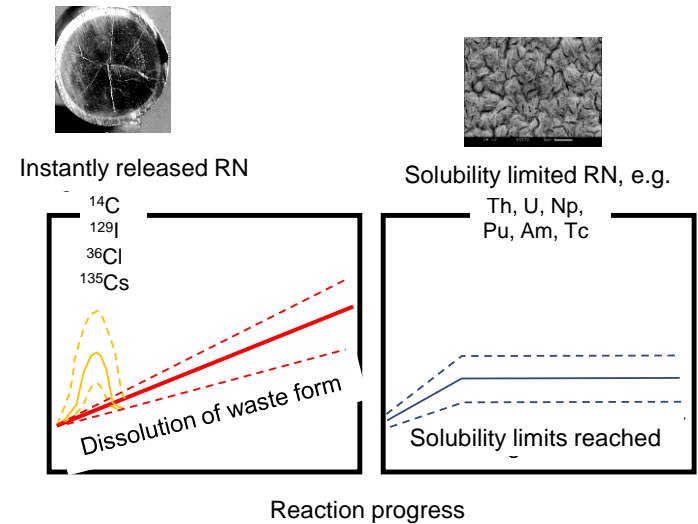
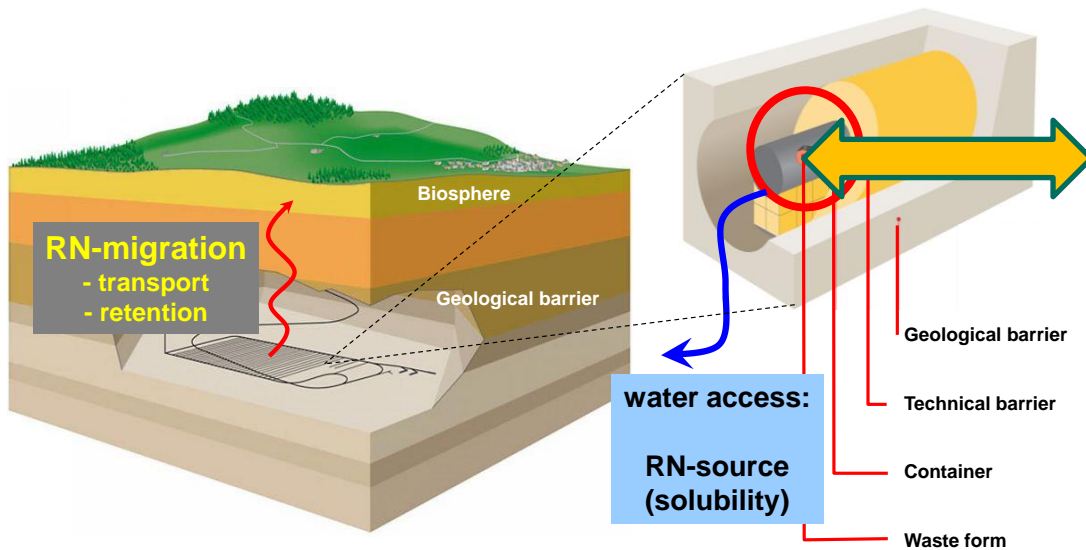


XAFS:

- Zr and Mo are found as Zr(IV) and Mo(0) (metallic state) in the SNF particle – the actinides in the oxidation states U(IV), Pu(IV), Am(III)

Nuclear waste disposal research

- Safety concept of repositories for nuclear waste disposal relies on the multibarrier system → prevent / minimize the release of radionuclides into the biosphere.



Nagra (2014)

Project examples

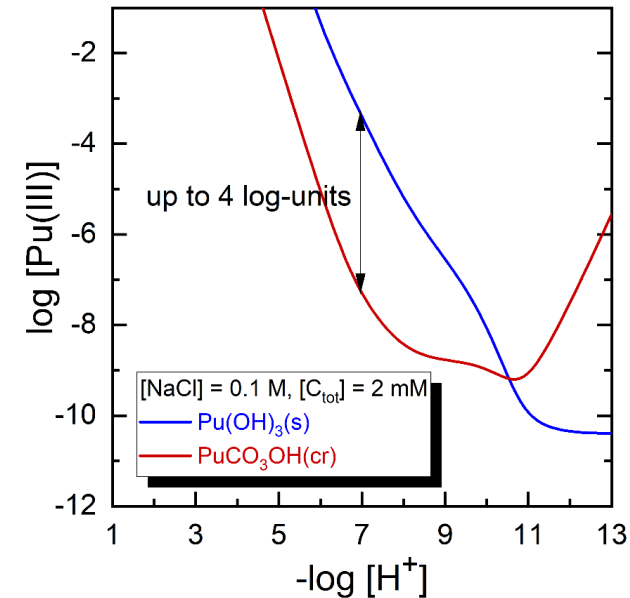
The role of $\text{PuCO}_3\text{OH}(\text{cr})$ to limit Pu-solubility

- Reduction of PuO_2^{2+} to Pu^{3+} (electrolysis), and precipitation as $\text{Pu}(\text{OH})_3(\text{s})$
- Equilibration of $\text{Pu}(\text{OH})_3(\text{s})$ in $\text{Na-HCO}_3\text{-CO}_3$ solution at $\text{pH} \approx 8$
- After 10 months: transformation of $\text{Pu}(\text{OH})_3(\text{s})$ into a **blue fine crystalline solid**

- * “sink” for **Pu(III)** at near-neutral pH
- * forms **spontaneous** from $\text{Pu}(\text{OH})_3(\text{s})$

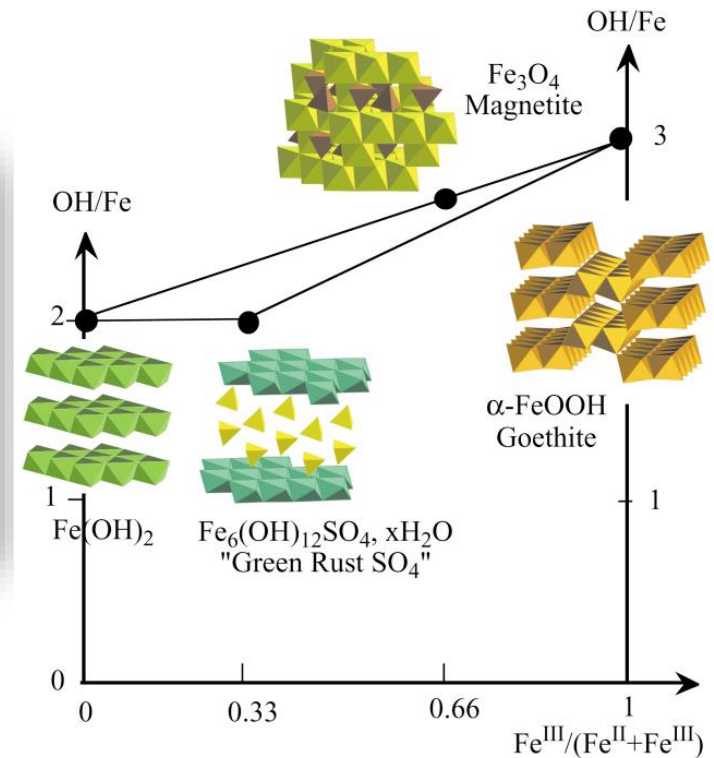
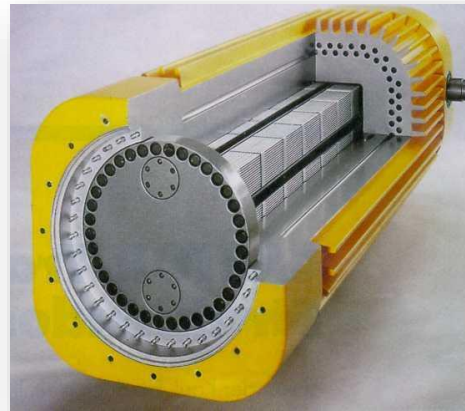
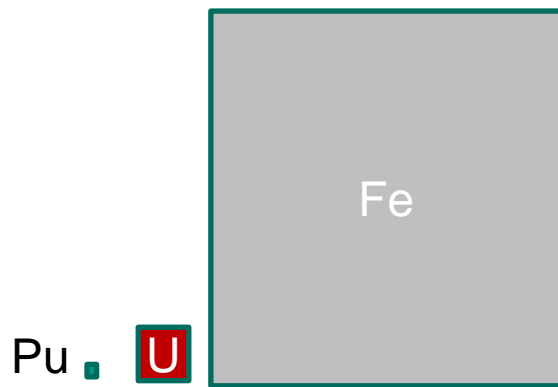


PhD thesis: P. Müller



Barrier function of container corrosion products

- Repository for high-level nuclear waste inventory:
molar ratio Fe:U:Pu ~ 40-50:1:0,01

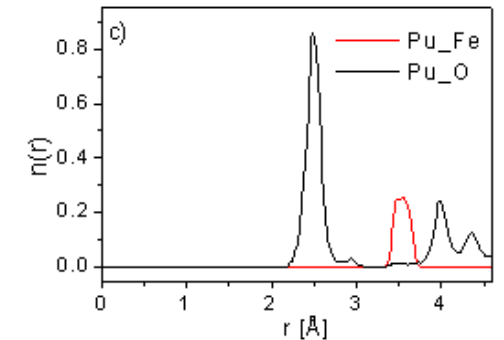
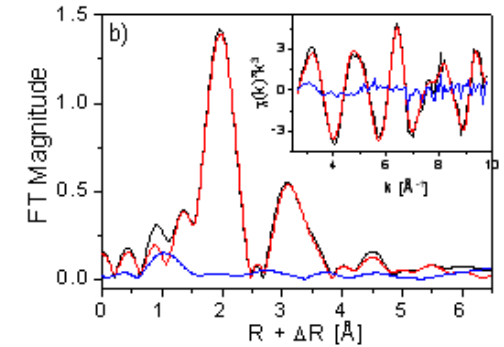
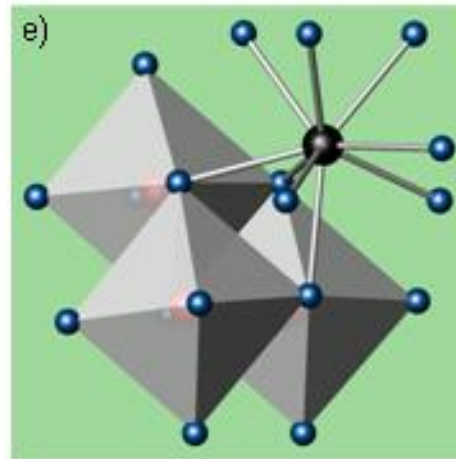
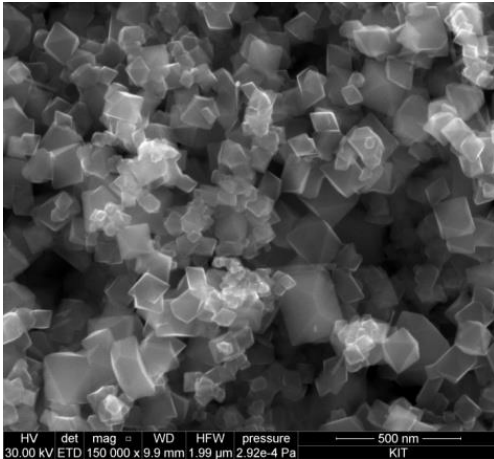


Jolivet *et al.*, Chem. Commun. **2004**, 481-487

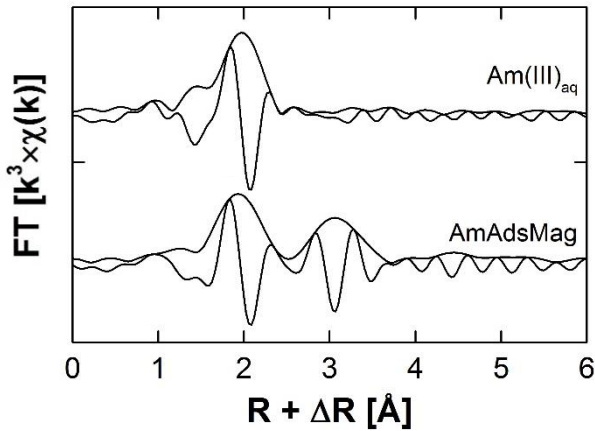
Barrier function of container corrosion products

- Interaction of trivalent actinide ions with magnetite -

Pu interaction with magnetite (Kirsch, Scheinost et al. ES&T, 2011)



Am interaction with magnetite (Finck et al., 2015)



Am-O: 2.47 Å

Am-Fe: 3.50 Å

Finck, Nedel, Dideriksen, Schlegel, 2015

Pu(III)-O: 2.48 Å

Pu(III)-Fe: 3.54 Å

Kirsch, Scheinost et al. ES&T, 2011

Pu-LIII EXAFS investigation of the Pu+Magnetite system

Laser technology for analysis ?



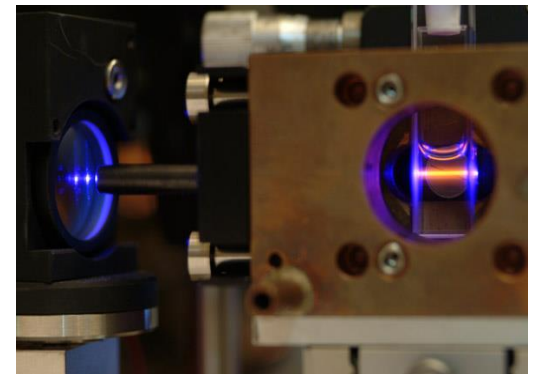
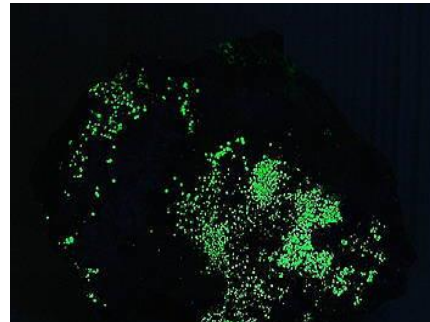
Times are changing



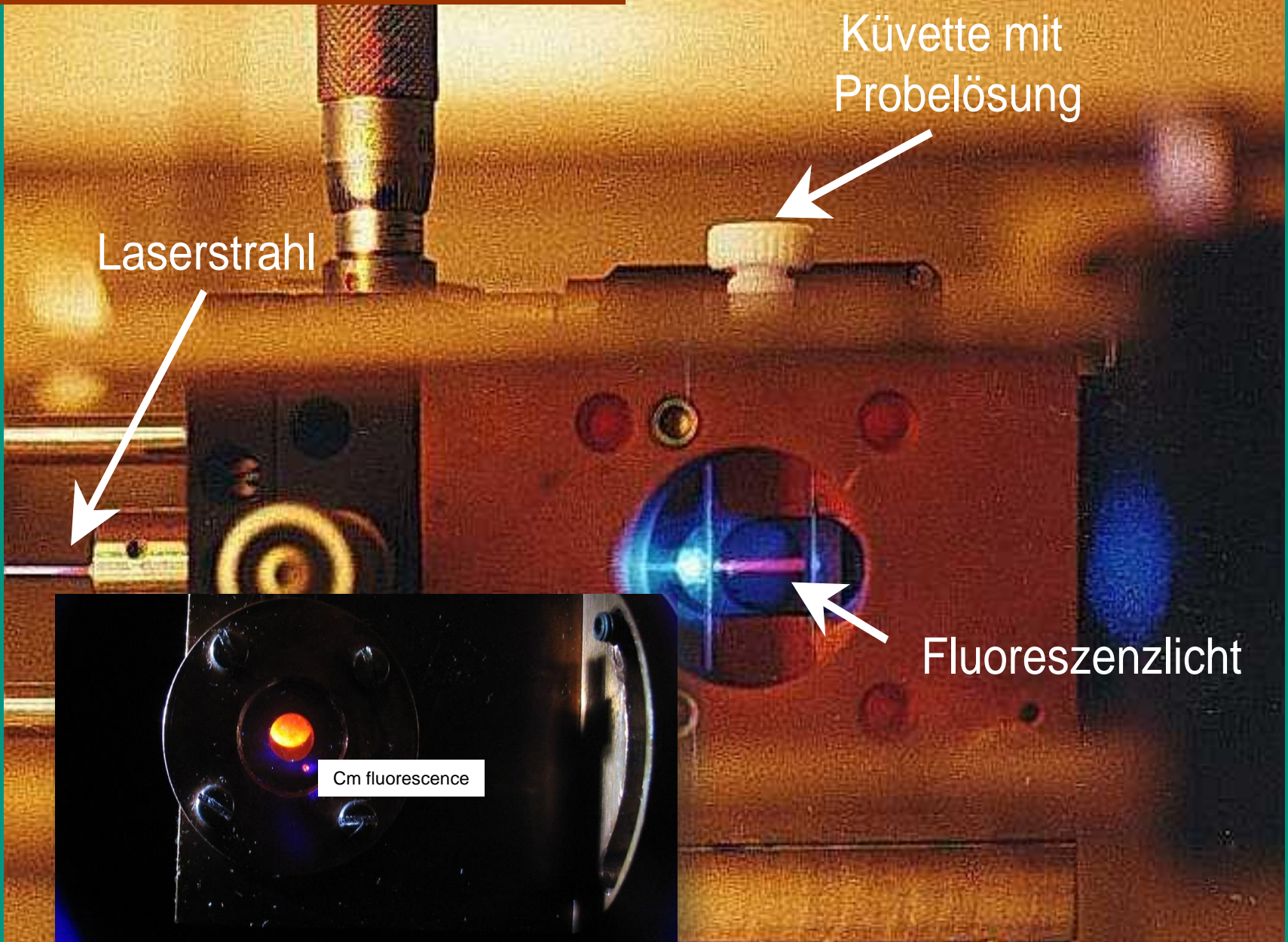
Antoine Henri Becquerel (1852-1908)

Becquerel studied 1896 the luminescence of $K_2UO_2(SO_4)_2 \cdot 2H_2O$ and discovered radioactivity

Today, we study chemistry of radioactive elements by observing their luminescence

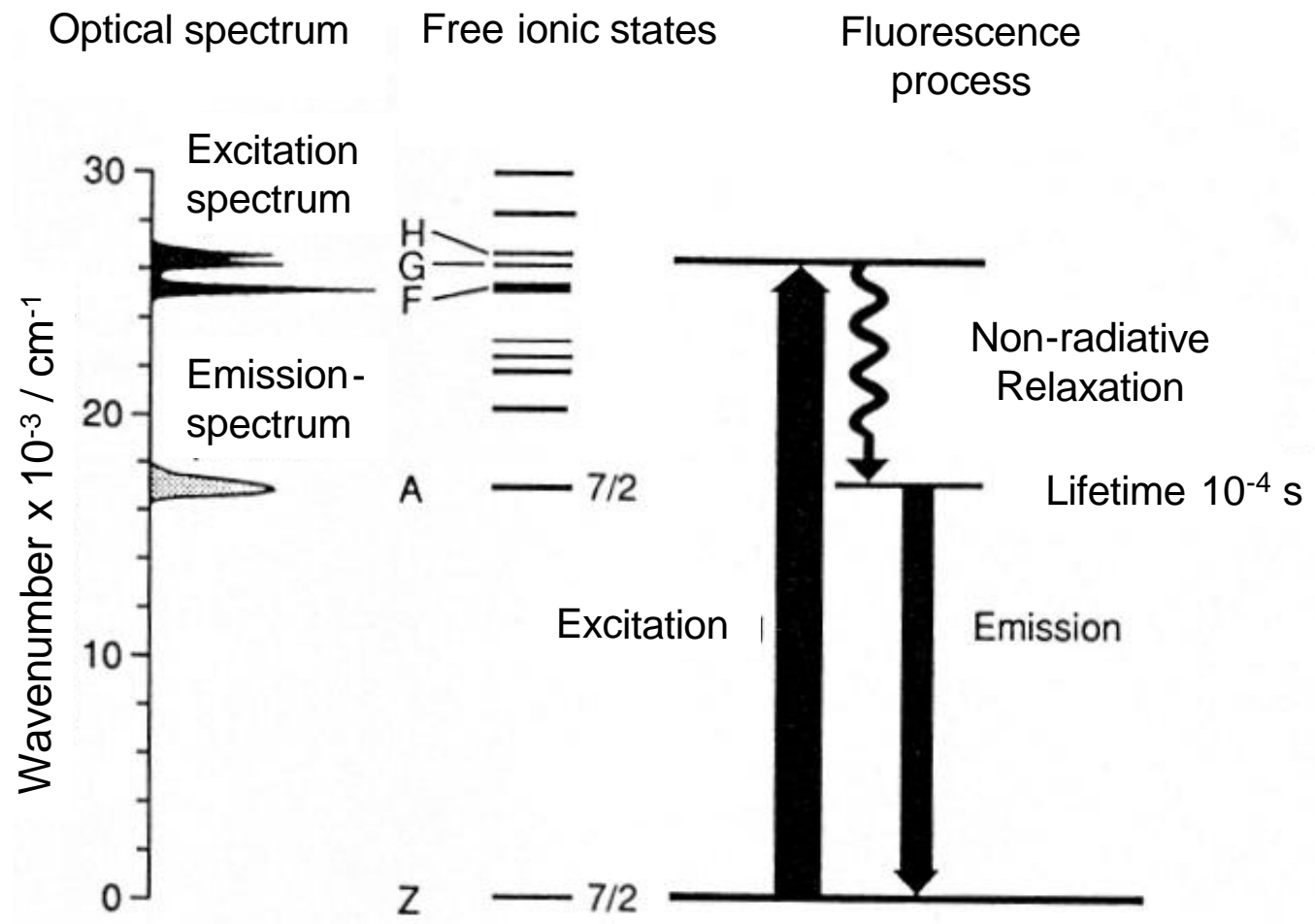


Laser luminescence spectroscopy

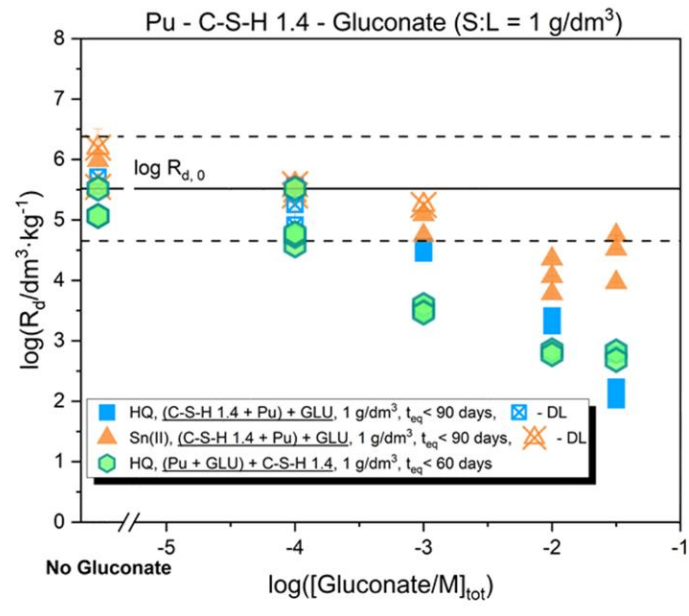
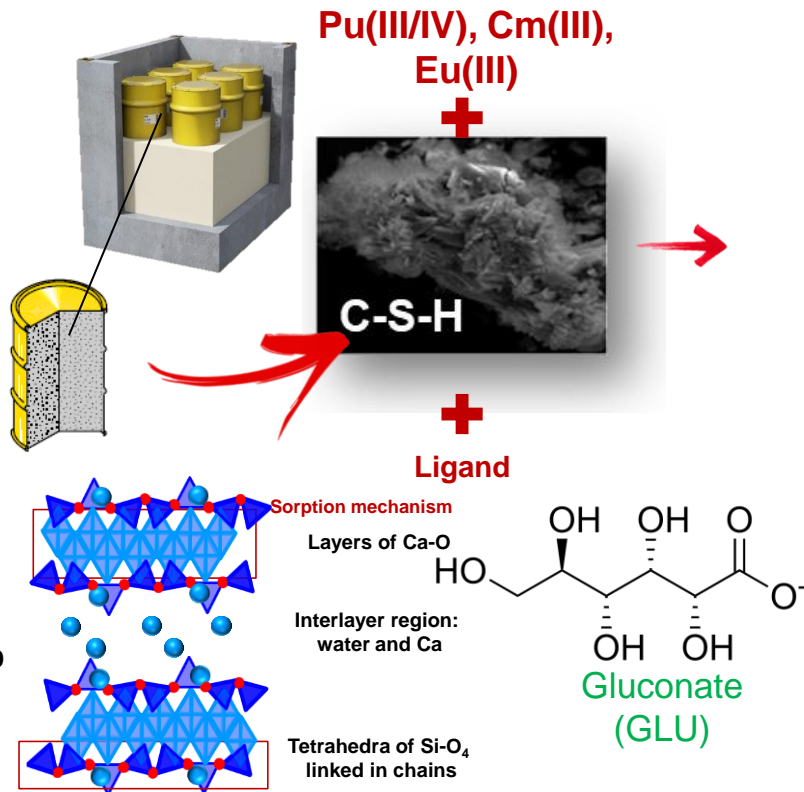


The Time Resolved Laser-Fluorescence Spectroscopy

- The basic principle -



Project example: Impact of organic cement plasticisers on actinide solubility

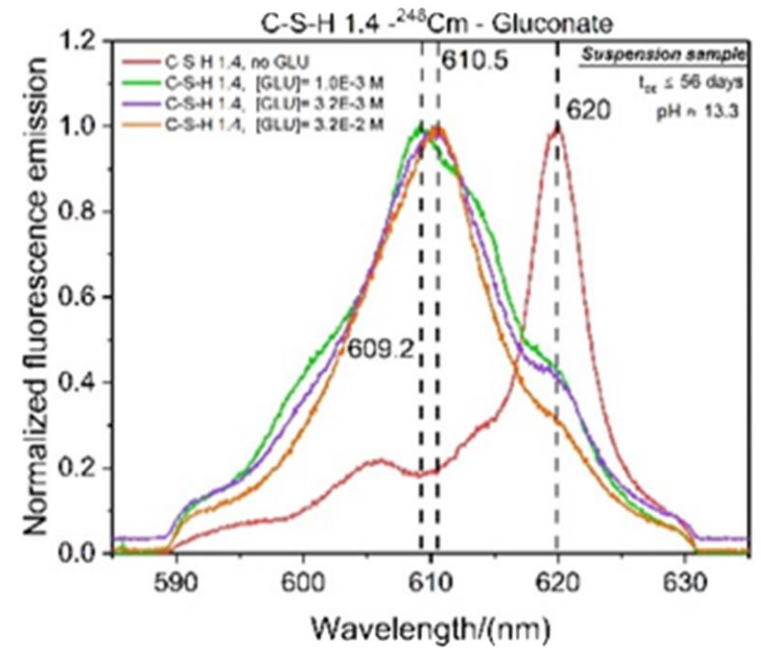
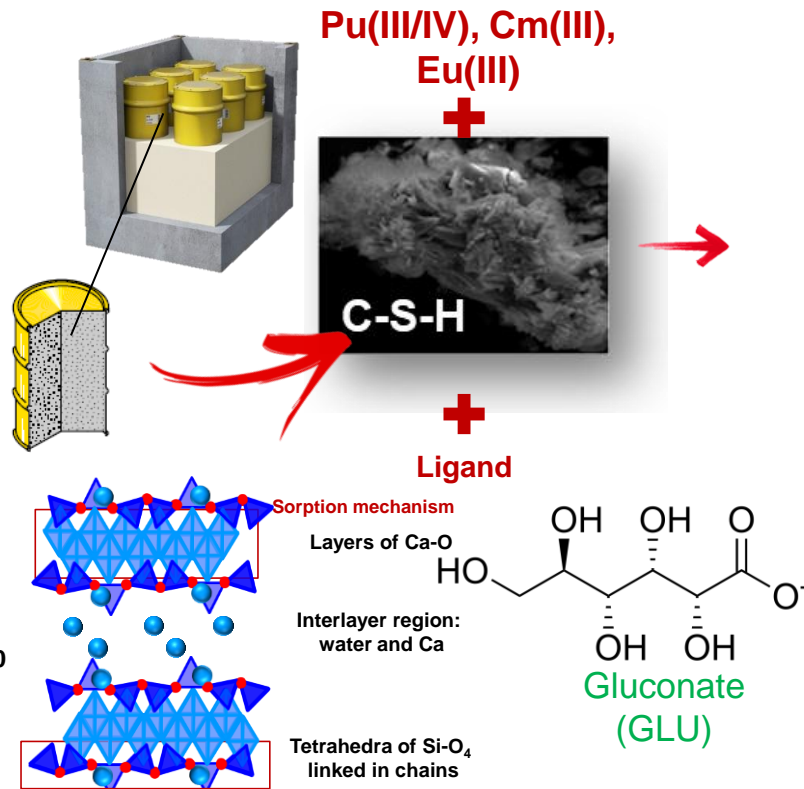


Retention behavior

PhD thesis: R. E. Guidone

Project example:

Impact of organic cement plasticisers on actinide solubility

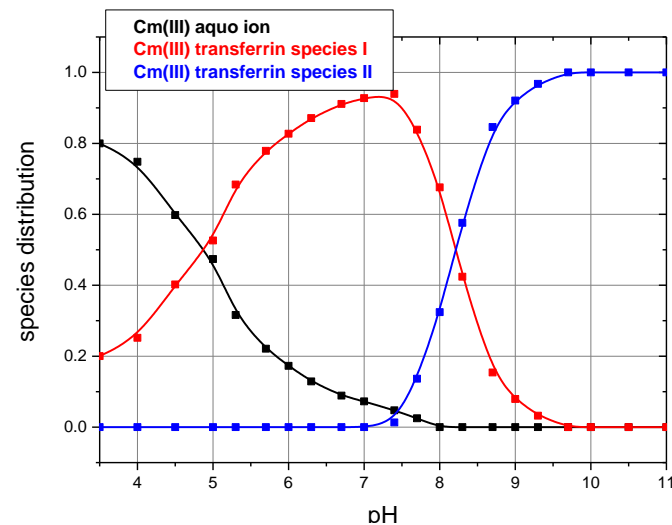
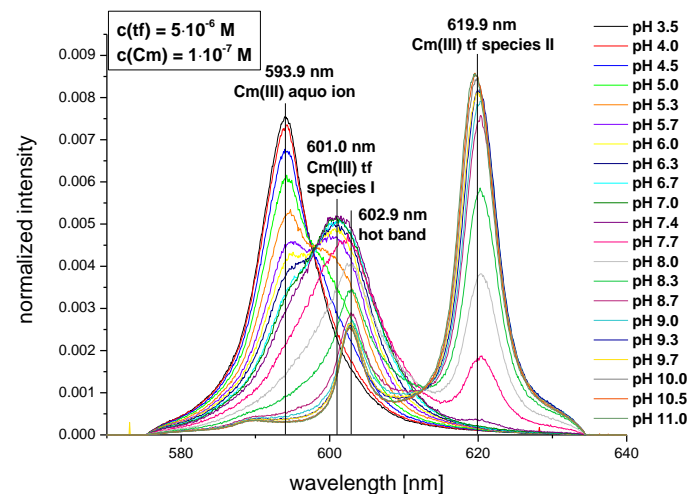
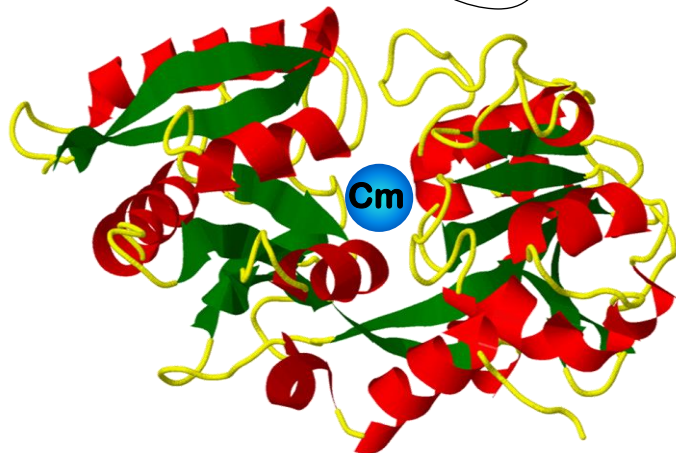
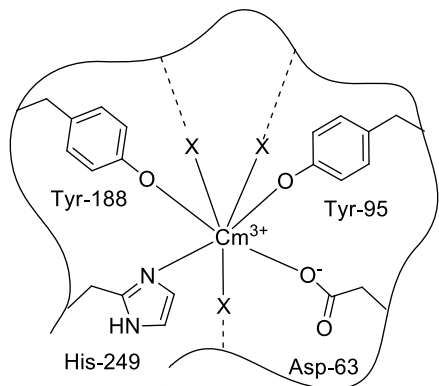


Cm-TRLFS spectra C-S-H suspension

PhD thesis: R. E. Guidone

Complexation of transferrin with Cm(III)

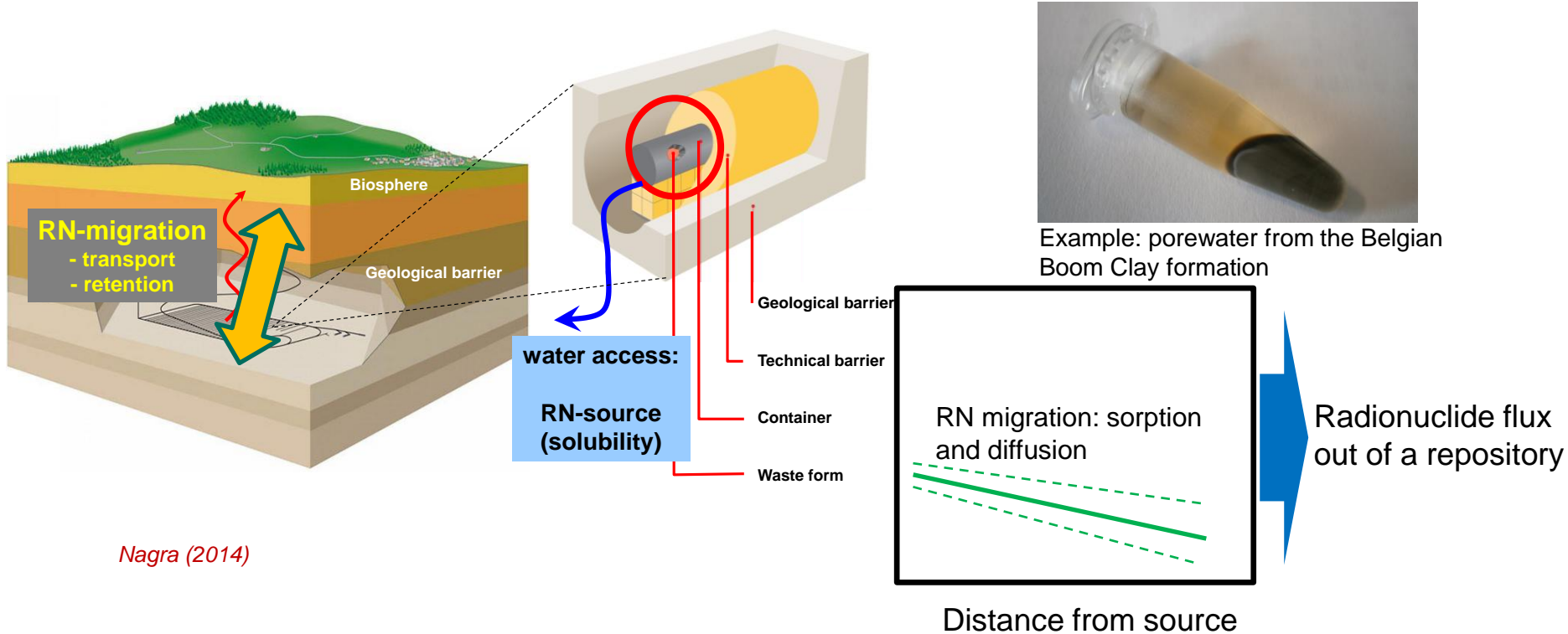
A complex with Cm(III) bound at the C-terminal binding site and another metal ion coordinated at the N-Lobe can potentially bind to the receptor and be brought into cells via endocytosis



N. Bauer, thesis Uni Heidelberg, 2015

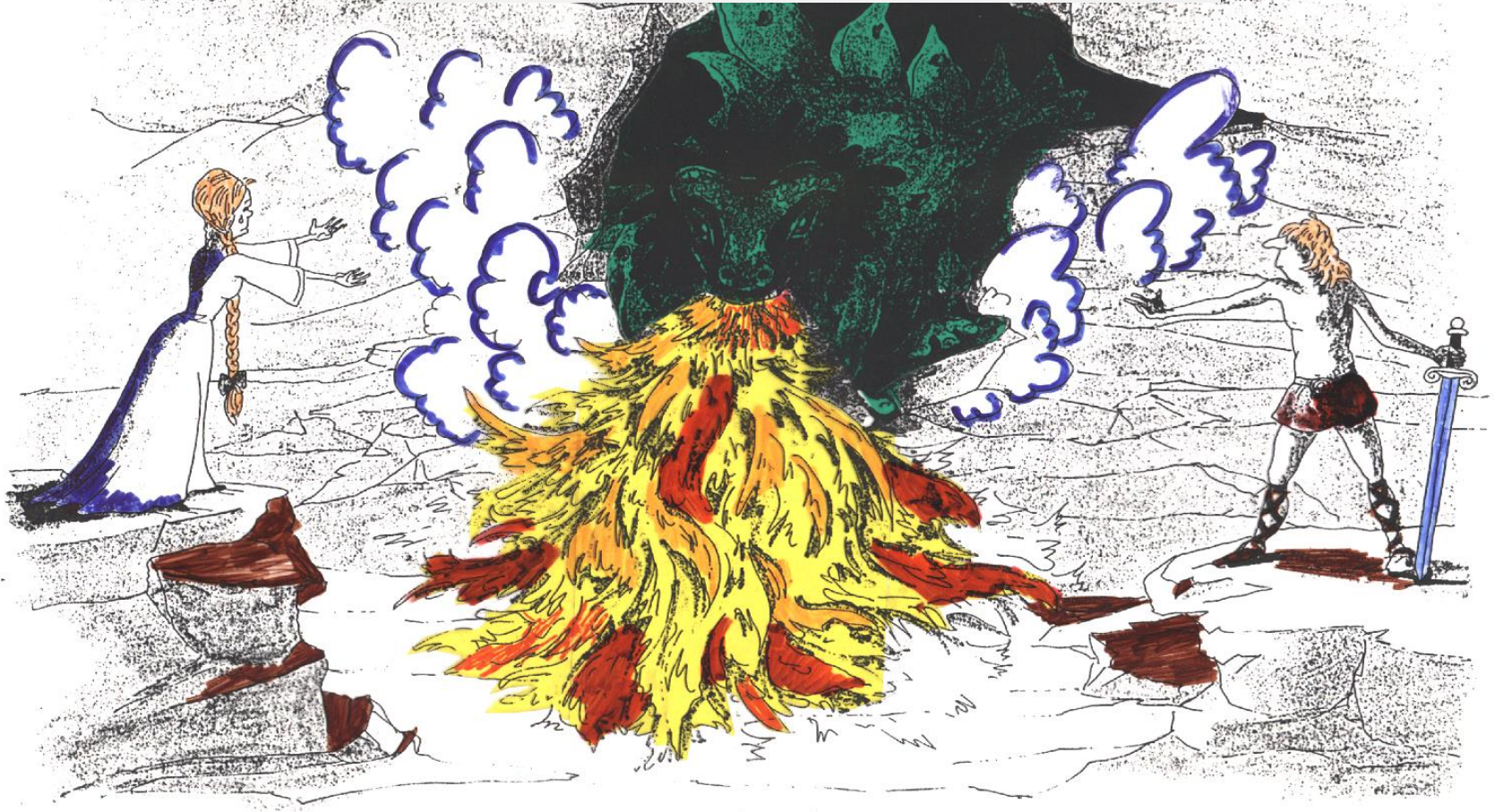
Nuclear waste disposal research

- Safety concept of repositories for nuclear waste disposal relies on the multibarrier system → prevent / minimize the release of radionuclides into the biosphere.

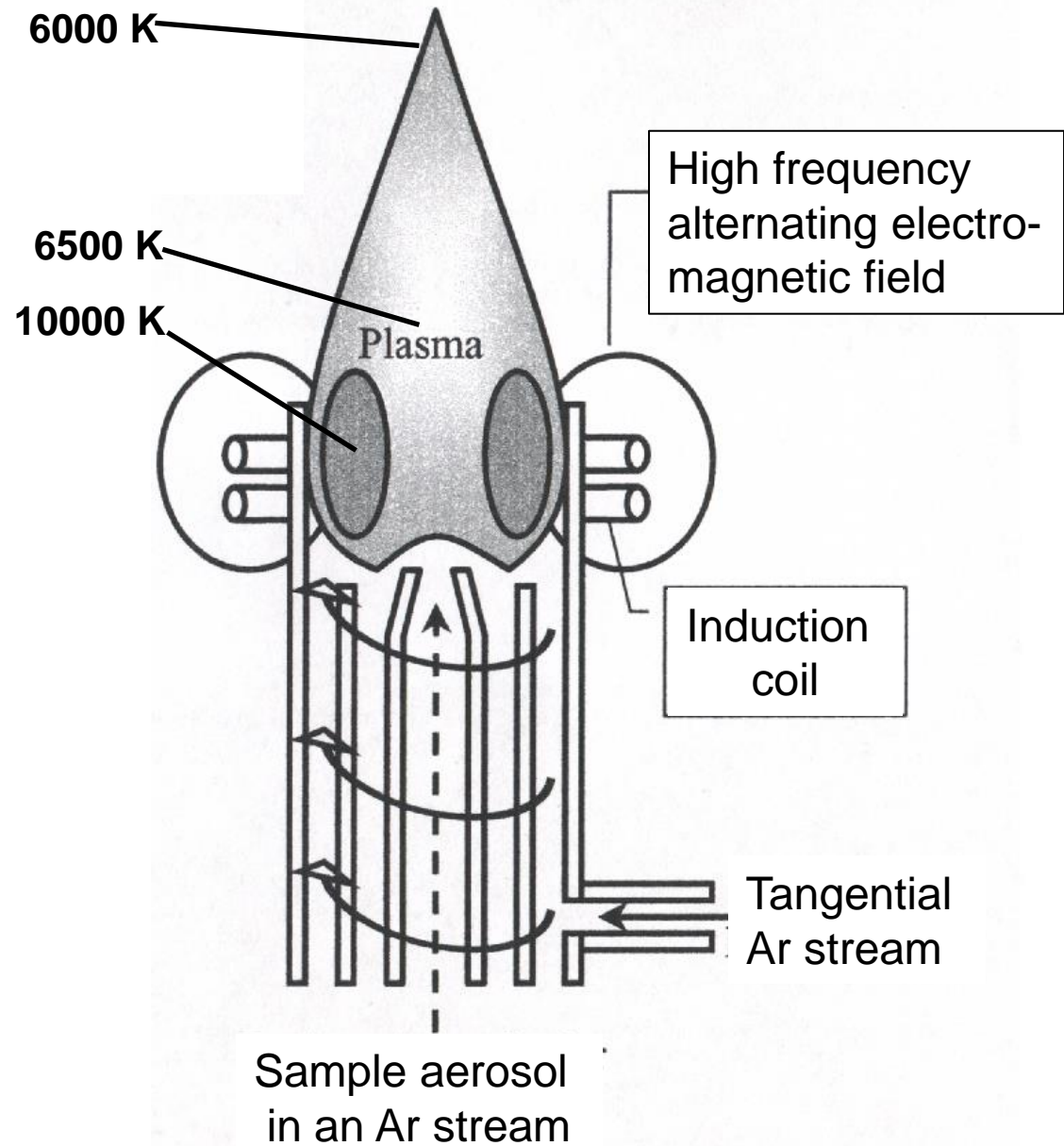


Nagra (2014)

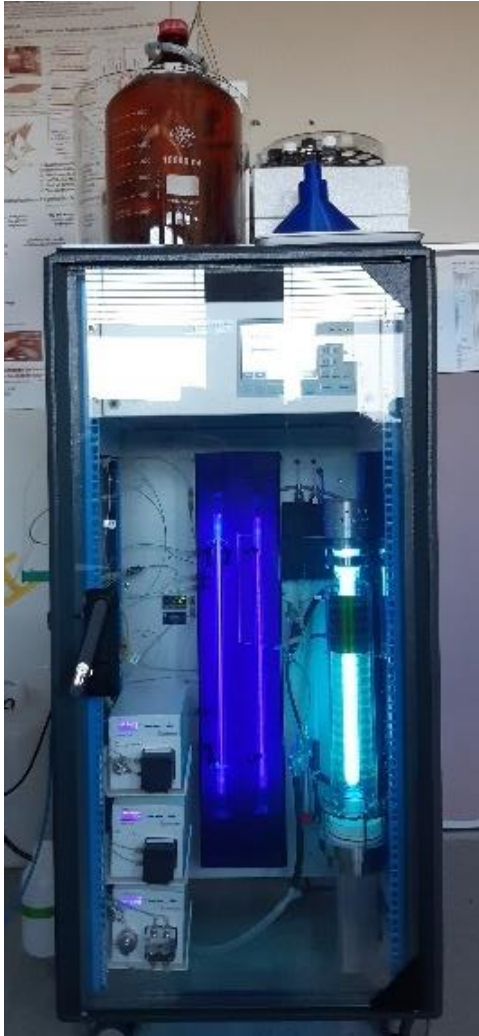
Atomic spectrometry



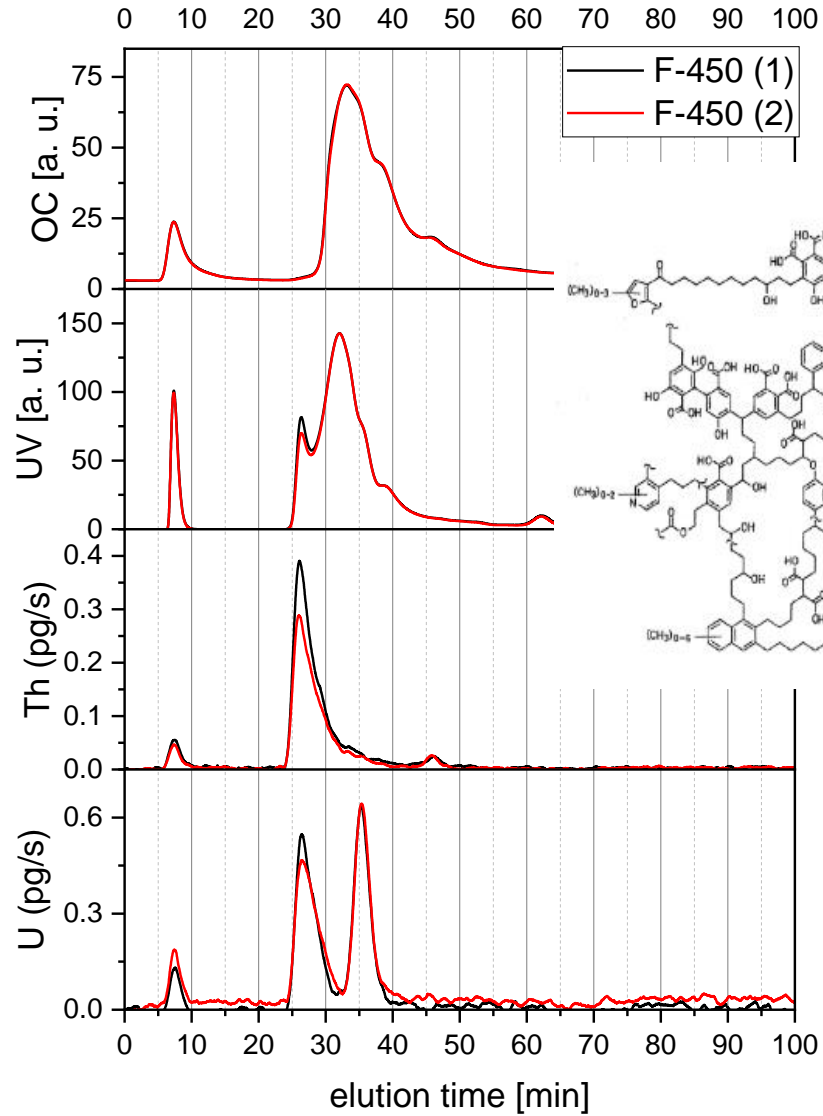
Thermische Dissoziation
Thermal dissociation



Colloid-borne naturally abundant U, Th in Boom-Clay porewater

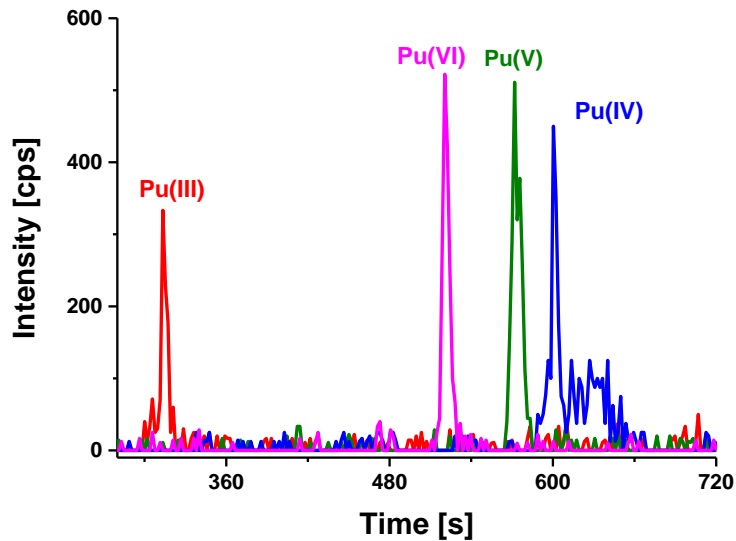


Coupling: LC-OCD-OND-UVD – ICP-MS

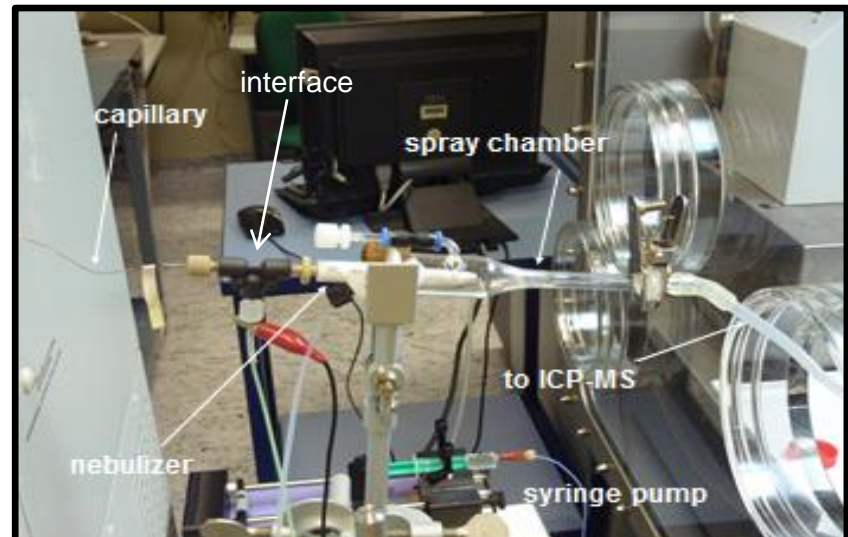
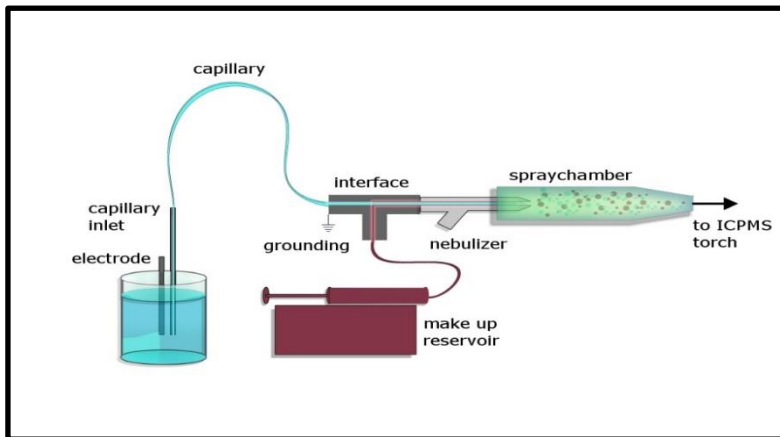


A. Lunz, Masterthesis, 2022

Hyphenation of CE and SF-ICP-MS



Species	Limit of detection [mol·L ⁻¹] (3 σ)
²⁴² Pu	2 · 10 ⁻¹²
²³⁸ Pu	5 · 10 ⁻¹¹
²³⁷ Np	5 · 10 ⁻¹²
⁵⁶ Fe	5 · 10 ⁻⁹



(Graser, Lagos, Marquardt et al., Anal.Chem., 2015)



Thank you for your attention