#### Modelling and simulation of Cl2 plasma and mixtures : Application to material etching

Guillaume Le Dain<sup>1</sup>, Ahmed Rhallabi<sup>1</sup>, Christophe Cardinaud<sup>1</sup>, Aurélie Girard<sup>1</sup>

<sup>1</sup>Institut des MatériauxJean Rouxel (IMN), UMR CNRS 6502, 2 rue de la Houssinière, Nantes



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- Plasma etching
- Kinetic modelling
- Cl<sub>2</sub> plasma
- Cl<sub>2</sub> plasma : Addition of Ar



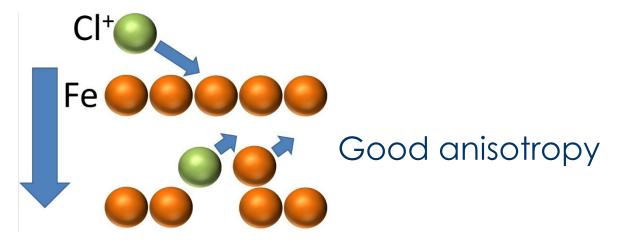


### Plasma etching

Chemical etching : Creation of volatil molecules due to adsorption of reactive neutrals onto surface atoms



Physical etching : Expulsion of surface atoms due to impact of energetic ions









# Plasma etching



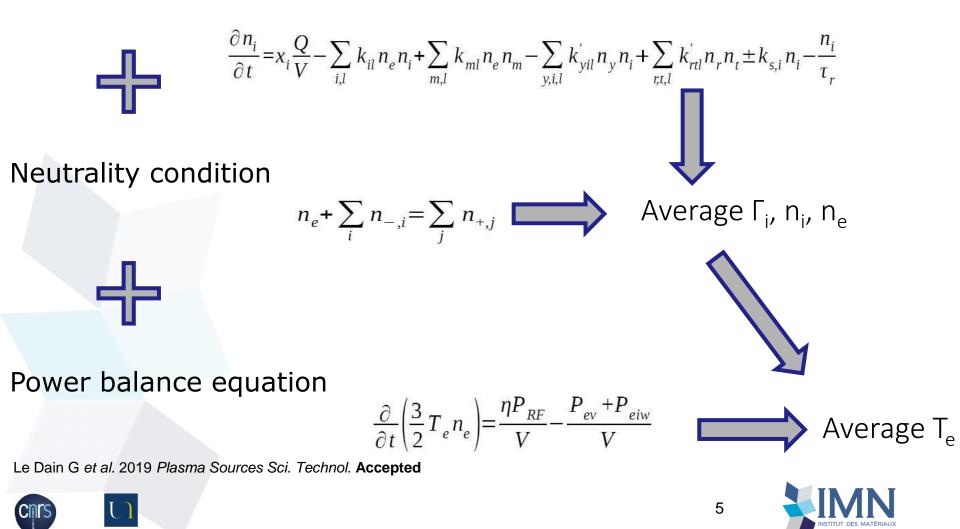
Cédric Doutriaux [2018]





### Kinetic modelling : Global 0D model

Density balance equation



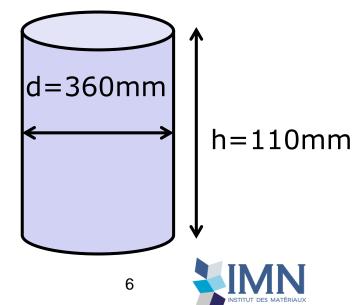


Follow densities in Cl<sub>2</sub> plasma versus power, pressure and flow rate

Aim : From operating conditions provided by experiments, study their variations to be able to provided better operating conditions

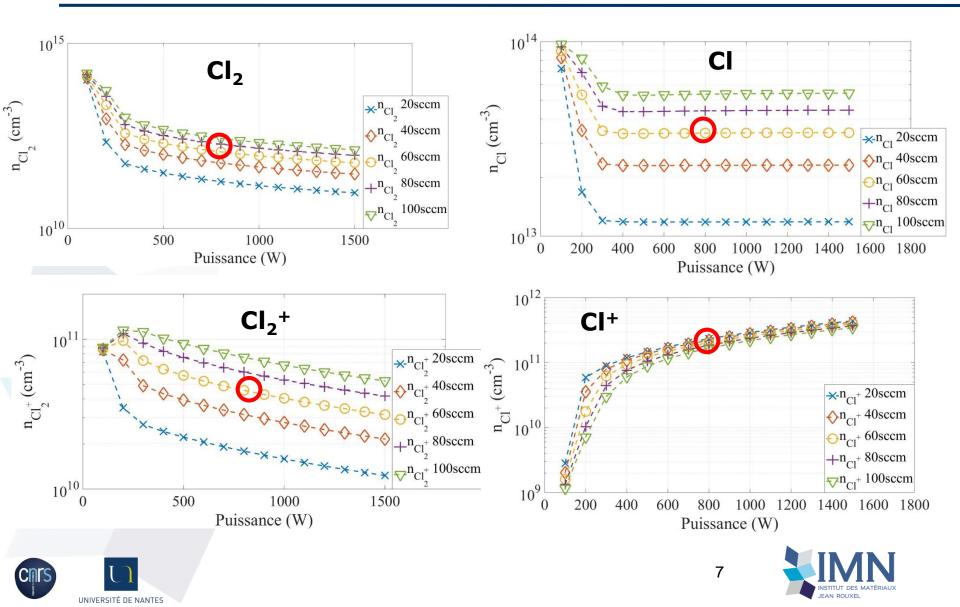
| Pression<br>(mTorr) | Puissance (W) | Débit (sccm) |
|---------------------|---------------|--------------|
| 5-20                | 100-1500      | 20-100       |
| (Pas de 20)         | (Pas de 100)  | (Pas de 20)  |

Simplified geometry of an ICP/RIE SENTECH





## Cl<sub>2</sub> Plasma : Densities evolution 10mTorr



## Cl<sub>2</sub> plasma : Conclusion

Cl<sub>2</sub> plasma has high dissociation rate in these operating conditions

Increasing flow rate (100sccm) with moderate pressure (20mTorr) and low power (400W) allows to increase reactive neutral densities

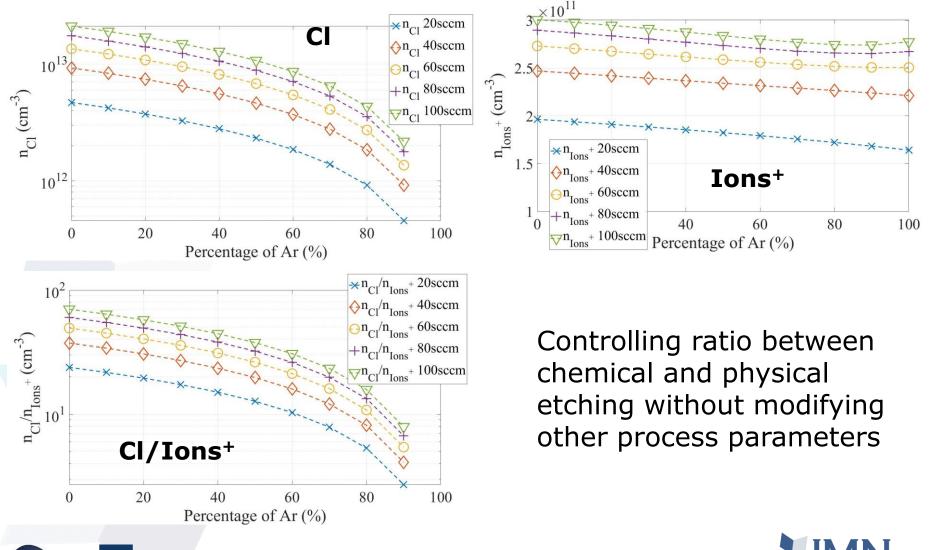
Moderate flow rate (60sccm) and moderate pressure (20mTorr) associated to a high power (1200W) allows to increase positive ions densities





## Cl<sub>2</sub> Plasma : Addition of Ar

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### Plasma modelling : Conclusion

Global kinetic modelling helps to study densities all over a large range of operating conditions

Not affected by reactor changes

Depending of the experiment to confirm the results

Simplified representation of a plasma





#### Acknowledgments

Shank you for your

attention



