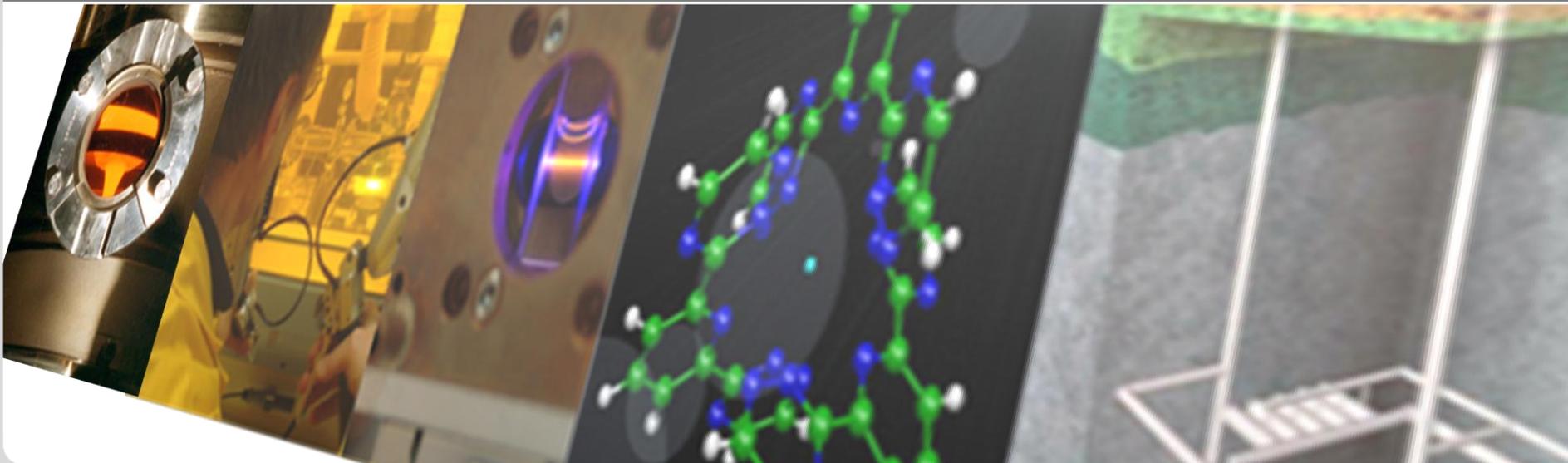


Visit to the Institute for Nuclear Waste Disposal (INE)

European Summer School Radiochemistry and Nuclear Instrumentation

Horst Geckeis (Email: horst.geckeis@kit.edu)

INSTITUT FÜR NUKLEARE ENTSORGUNG (INE)



European Summer School
Radiochemistry and nuclear instrumentation (low level radioactivity)

**Visit at the Institute for Nuclear Waste Disposal (INE),
Karlsruhe Institute of Technology (KIT)**

23 august 2017

Programme

| | | |
|---------|---------------------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------|
| 09:00 h | Arrival at KIT-Campus North entrance | |
| 09:15 h | Welcome at INE and introduction | <i>H. Geckeis</i> |
| 09:30 h | Research towards radioactive waste disposal at KIT-INE | <i>H. Geckeis</i> |
| 10:15 h | Coffee break | |
| 10:30 h | Ultratrace analysis of radionuclides by AMS | <i>F. Quinto</i> |
| 11:30 h | Radionuclide speciation by laser spectroscopy | <i>C. Garcia-Perez</i> |
| 12:30 h | Lunch | |
| 13:30 h | Synchrotron based X-ray spectroscopic characterization of radionuclides and Visit of INE-Beamlines at ANKA (ANKA building) | <i>T. Vitova</i> |
| 15:00 h | Visit of INE laboratories | |
| | <ul style="list-style-type: none">• Hot cells• Mass spectrometry• Laser spectroscopy• Geochemistry | <i>M. Herm</i> <i>F. Quinto</i> <i>C. Garcia-Perez</i> <i>C. Joseph</i> |
| 17:00 h | End of visit | |

Topics

- What is KIT?
- Nuclear waste in Germany and worldwide
- Research at the Institute for Nuclear Waste Disposal

KIT – The Research University in the Helmholtz Association

Status: January 2017

PRESIDENTIAL COMMITTEE

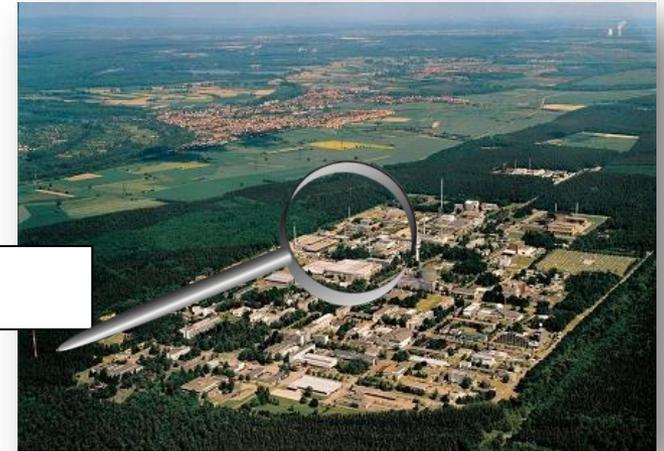
scientists in teaching and research
Cultural diversity
**DEVELOPING SCIENTIFIC
CAREER PATHS**
Research infrastructure
**The Research University
in the Helmholtz Association**
TRANSPARENT SERVICES *KIT thinks and acts as ONE institution*
FOR RESEARCH; TEACHING, AND INNOVATION
Research-based teaching and learning **INNOVATION AS A** *Energy*
STATUTORY MISSION *Mobility*
AIMING AT A LEADING POSITION IN EUROPE *Information*

Two KIT sites

KIT – sites Karlsruhe



KIT Campus North

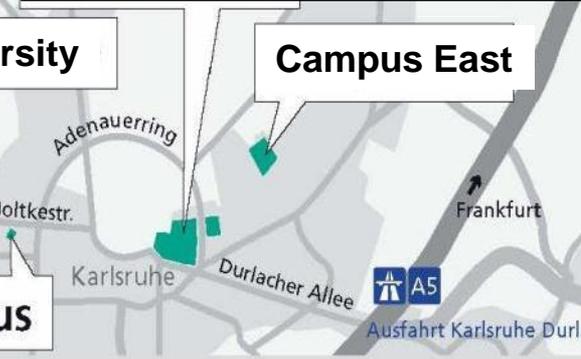


KIT Campus South

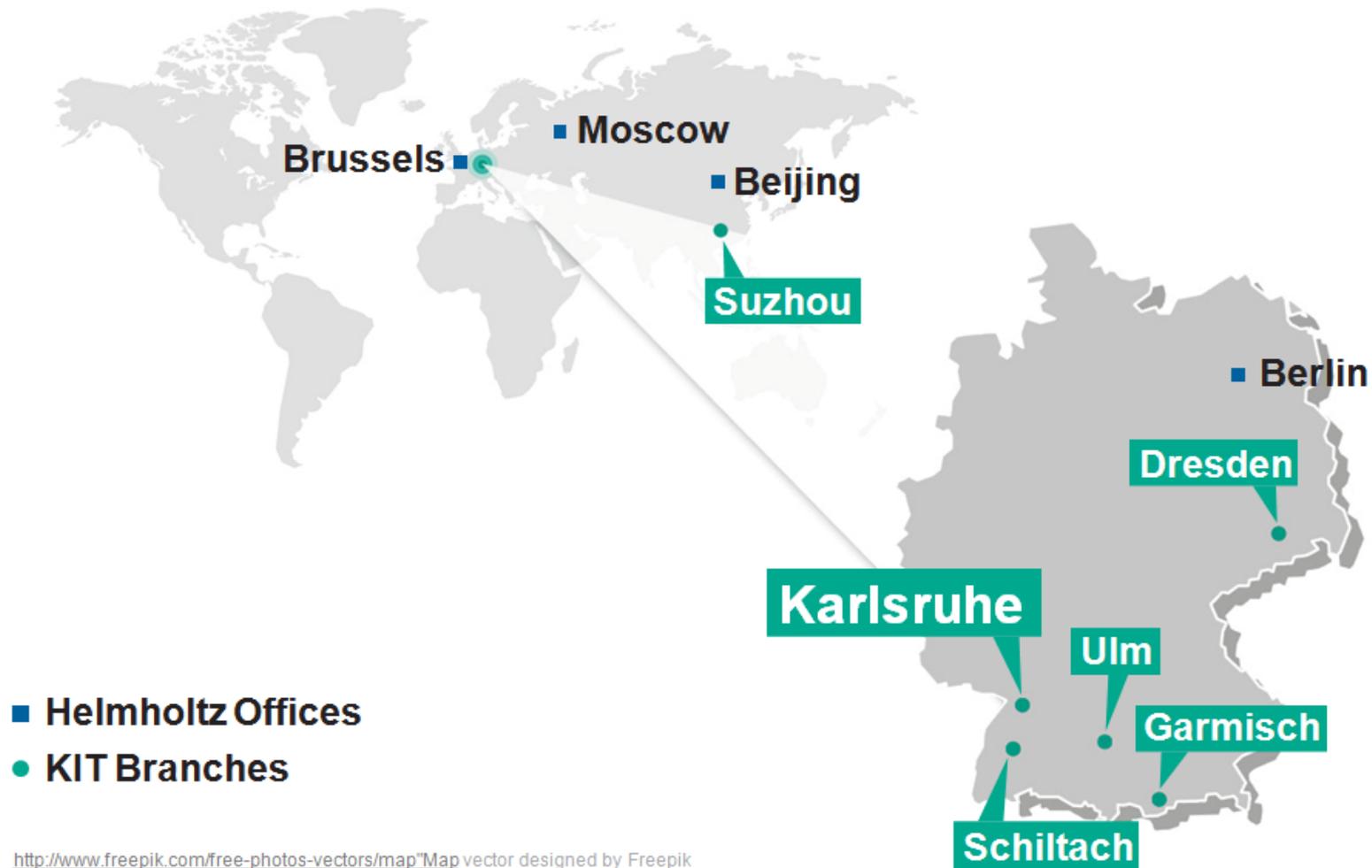
West university

Campus East

Ostendorfhaus



Branch Offices

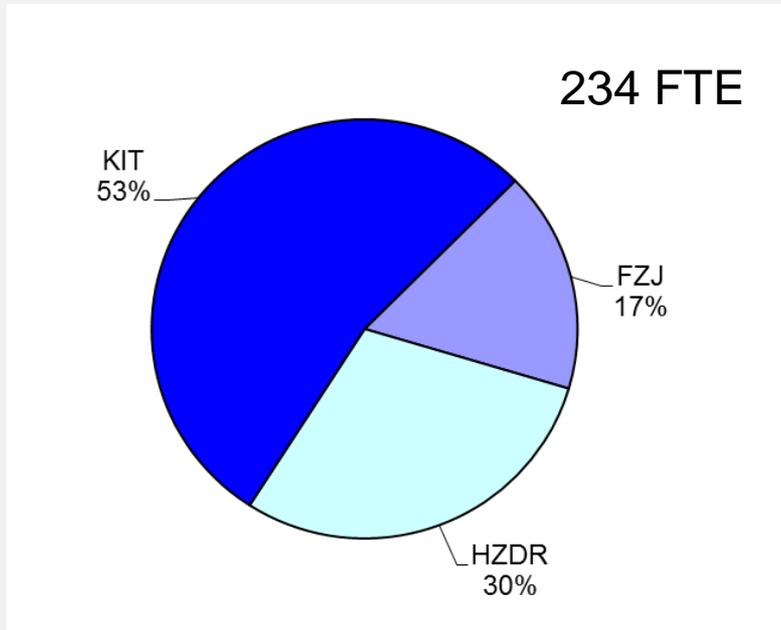


<http://www.freepik.com/free-photos-vectors/map> Map vector designed by Freepik

The research programme NUSAFE

Nuclear Waste Management and Safety as well as Radiation Research

NUSAFE resources:



67% Nuclear Waste Management
33% Reactor Safety



Topics

- What is KIT?
- Nuclear waste in Germany and worldwide
- Research at the Institute for Nuclear Waste Disposal

Nuclear waste in Germany

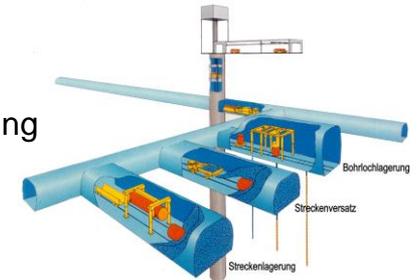


Irradiated Nuclear fuel
Vitrified waste From reprocessing

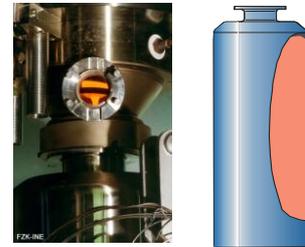


Intermediate storage

Repository For heat producing Radioactive Waste



Non-heat producing radio-Active waste from NPP and decommissioning



Repository For non-heat producing Radioactive Waste



Non-heat producing radioactive waste From medicin, research etc.



Intermediate storage



Irradiated Nuclear fuel
 Vitrified waste From reprocessing



HEAT GENERATING WASTE:

(10.550 tHM)

NPP

21.000 m³

5.700 m³

Research/
 Prototype
 Reactors/
 Pilot repr. Plant



Intermediate storage



1.400 m³

vitrified waste

(7.973 coquilles from 6.700 tSM repr. spent fuel)

Non-heat producing radio-Active waste from NPP and decommissioning



Non-heat producing radioactive waste From medicin, research etc.



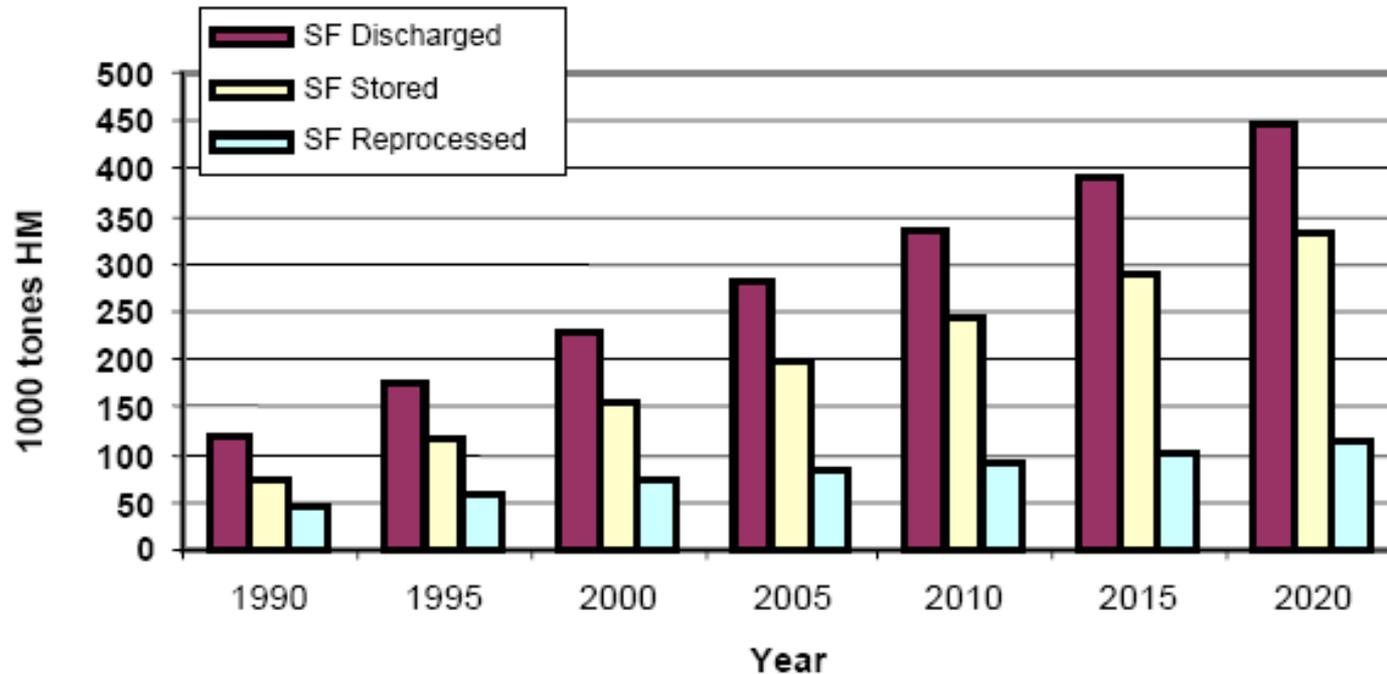
Intermediate storage

WASTE WITH NEGLIGIBLE HEAT GENERATION:

~ 300.000 m³

Spent fuel arising world wide

Cumulative Spent Fuel Arisings, Storage and Reprocessing,
1990-2020



Per year:

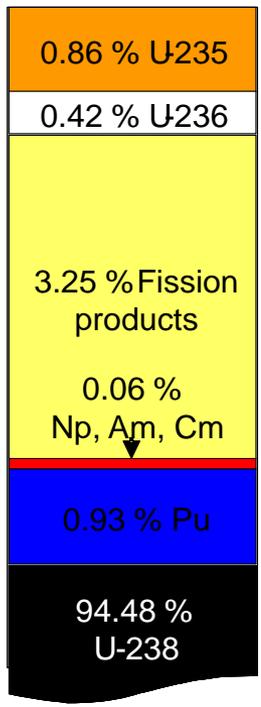
10.500 t spent nuclear fuel

Cumulative until 2020:

~ 445.000 t spent nuclear fuel

~ 1/3 reprocessed → Vitrification of remaining high level liquid waste

Content of high-level nuclear waste spent fuel



PWR; Burn-up 33GWd/tU

Some long-lived fission and activation products

- C-14 ($t_{1/2} \sim 5.700 \text{ a}$)
- Cl-36 ($t_{1/2} \sim 300.000 \text{ a}$)
- Se-79 ($t_{1/2} \sim 360.000 \text{ a}$)
- Tc-99 ($t_{1/2} \sim 213.000 \text{ a}$)
- I-129 ($t_{1/2} \sim 15,6 \text{ Ma}$)

Period

| PERIODE | Hauptgruppen | |
|---------|---------------------------------------------------------------|------------------------------------------------------|
| | I | II |
| 1. | 1 H Wasserstoff 1,01 -259 0,09 -253 2,1 | |
| 2. | 3 Li Lithium 181 0,53 1330 1,0 | 4 Be Beryllium 1277 1,65 2970 1,5 |

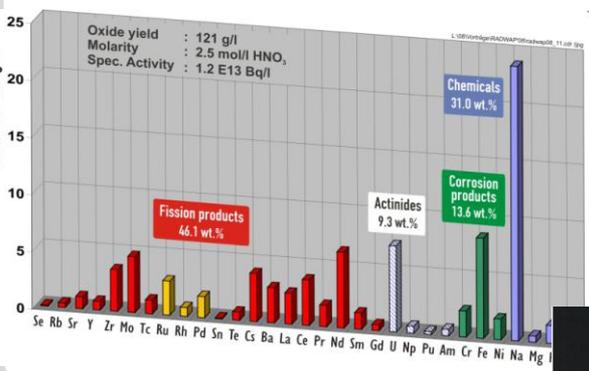
Protonenzahl (Ordnungszahl)
Elementsymbol
Schmelztemperatur (gerundet) in °C
Siedetemperatur (gerundet) in °C
Elektronenanzahl nach L. Pauling
Alle anderen Werte nach Römpp-Chemielexikon, Stuttgart, 8. Auflage

Fission products:
Cs-137, Sr-90 ($t_{1/2} \sim 30 \text{ a}$)

| PERIODE | Hauptgruppen | | | | | | | | | | | | | | | | |
|---------|------------------|------------------|-----------------------------|-------------------|-------------------|-----------------|------------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|
| | I | II | III | IV | V | VI | VII | VIII | IX | X | XI | XII | I | II | III | IV | |
| 1. | 1 H | | | | | | | | | | | | | | | | 2 He |
| 2. | 3 Li | 4 Be | | | | | | | | | | | | | | | 10 Ne |
| 3. | 11 Na | 12 Mg | | | | | | | | | | | | | | | 18 Ar |
| 4. | 19 K | 20 Ca | | | | | | | | | | | | | | | 36 Kr |
| 5. | 37 Rb | 38 Sr | 39 Y | 40 Zr | 41 Nb | 42 Mo | 43 Tc* | 44 Ru | 45 Rh | 46 Pd | 47 Ag | 48 Cd | 49 In | 50 Sn | 51 Sb | 52 Te | 54 Xe |
| 6. | 55 Cs | 56 Ba | 57-71 Lanthanoide | 72 Hf | 73 Ta | 74 W | 75 Re | 76 Os | 77 Ir | 78 Pt | 79 Au | 80 Hg | 81 Tl | 82 Pb | 83 Bi | 84 Po | 86 Rn |
| 7. | 87 Fr* | 88 Ra* | 89-103 Actinoide | 104 Ku* | 105 Ha* | | | | | | | | | | | | 118 Og |

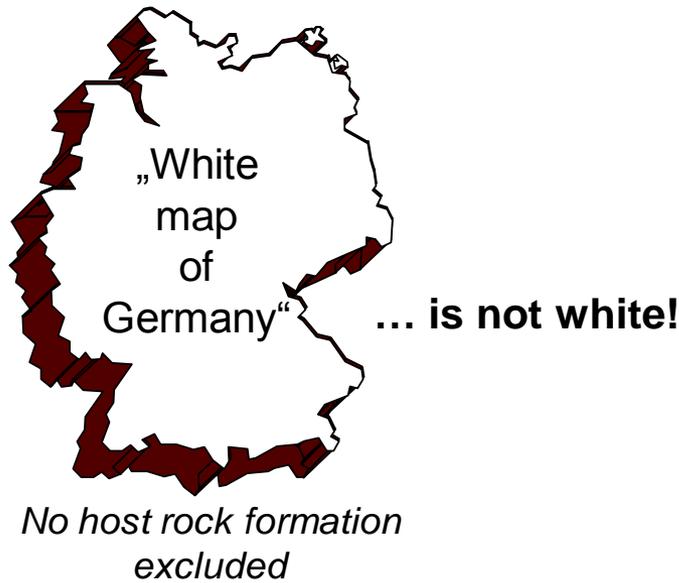
Important Transuranium elements:

- Pu-238 ($t_{1/2} \sim 88 \text{ a}$)
- Pu-239 ($t_{1/2} \sim 24.110 \text{ a}$)
- Am-241 ($t_{1/2} \sim 432 \text{ a}$)
- Cm-243 ($t_{1/2} \sim 29 \text{ a}$)
- Np-237 ($t_{1/2} \sim 2.1 \text{ Ma}$)

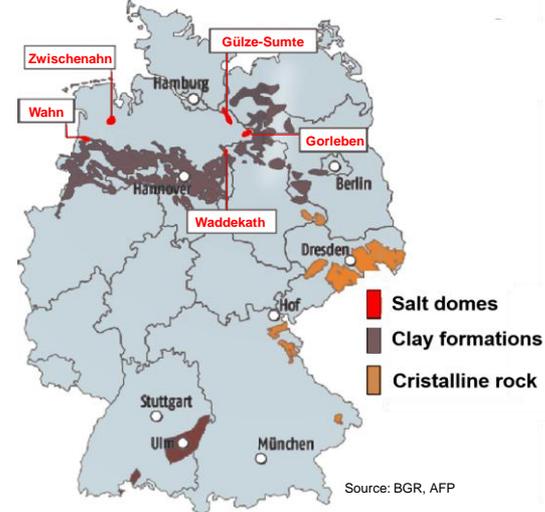


Vitrified liquid waste

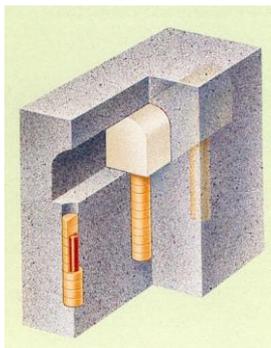




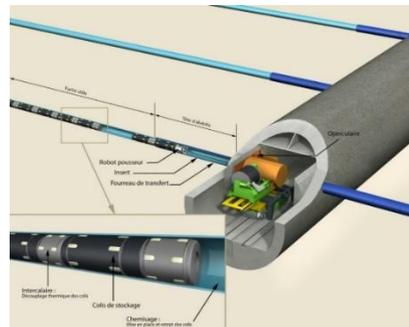
Potential host rocks for nuclear waste repositories in Germany



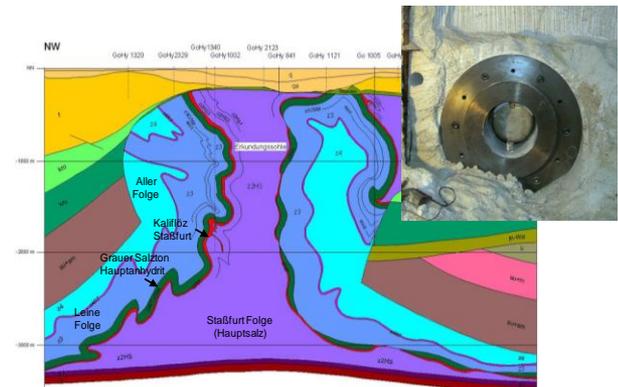
➔ Site selection does not only require a science based selection of host rocks but also a selection of repository concepts!



Crystalline rock



Argillaceous rock



Rock salt

Topics

- What is KIT?
- Nuclear waste in Germany and worldwide
- Research at the Institute for Nuclear Waste Disposal

KIT – Institute for Nuclear Waste Disposal (INE)



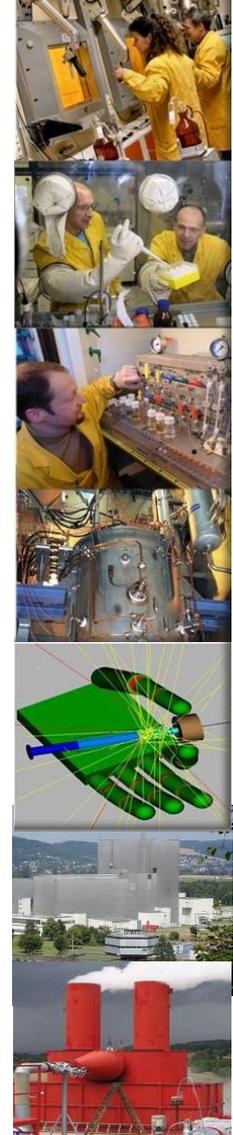
Research and development activities at KIT-INE are an integral component of national provided research and encompass the following areas:

- **Long-term safety research for nuclear waste disposal**
- **Actinide coordination chemistry**
- **Decommissioning of nuclear facilities**
- **Geothermal energy**
- **Research for radiation protection**
- **(Immobilisation of high level radioactive waste by vitrification)**

NUSAFE



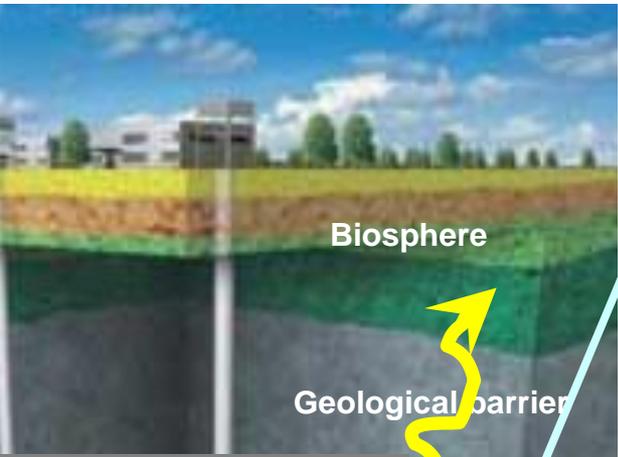
HELMHOLTZ
ASSOCIATION



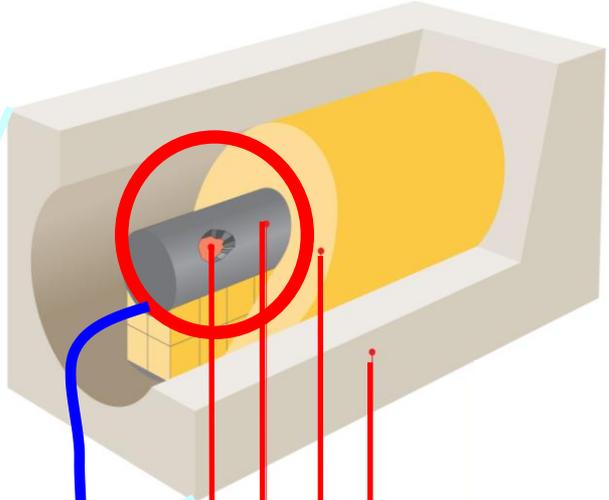
Processes in the multibarrier system

- Chemical and geochemical aspects -

- Assessment of RN retention potential
- Analysis of the isolation potential
- Quantification of RN dispersal
- Analysis of consequences



RN-migration
- transport
- retention



Water access:
RN-source (solubility)

- Geological barrier
- Technical barrier
- Container
- Waste form

| | |
|--------------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| <p>Source term:</p> $r = k_0 \times A_{\min} \times e^{-E_a/RT} a_{H^+}^{n_{H^+}} g(I) \prod_i a_i^{n_i} f(\Delta G_r)$ | <p>Transport + retention:</p> $-u \frac{\partial c_i}{\partial t} + D \frac{\partial^2 c_i}{\partial x^2} - \frac{\rho}{\theta} K_d \frac{\partial c_i}{\partial t} = \frac{\partial c_i}{\partial t}$ |
|--------------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|

Nuclear waste form behavior

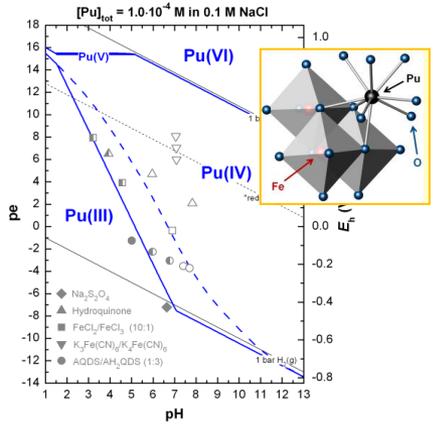
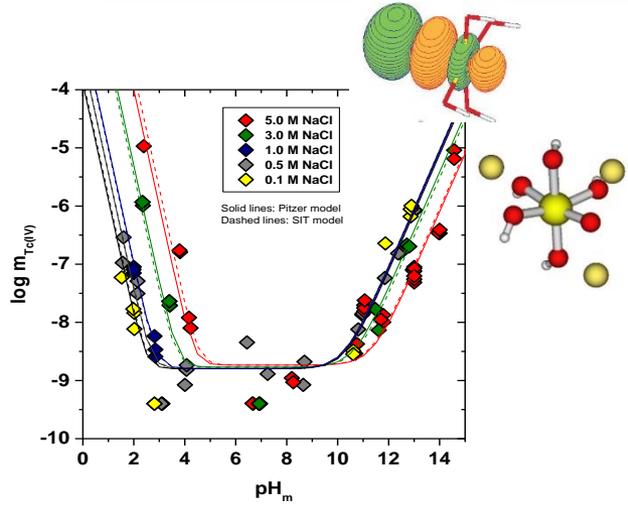
- spent nuclear fuel
- HLW glass
- Cementitious waste



Fuel pellet

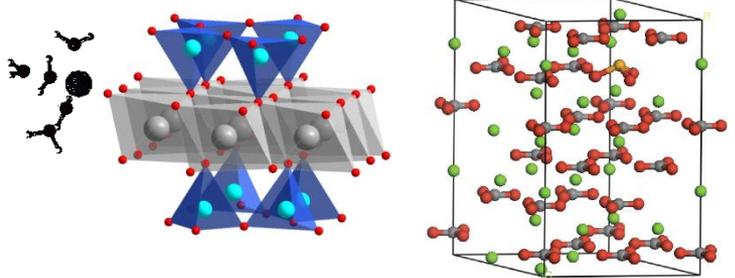
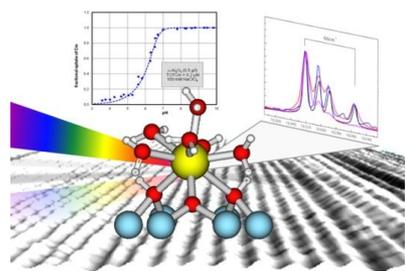
Aquatic chemistry

- Solubility
- Redox
- Complexation



Solid/liquid interactions

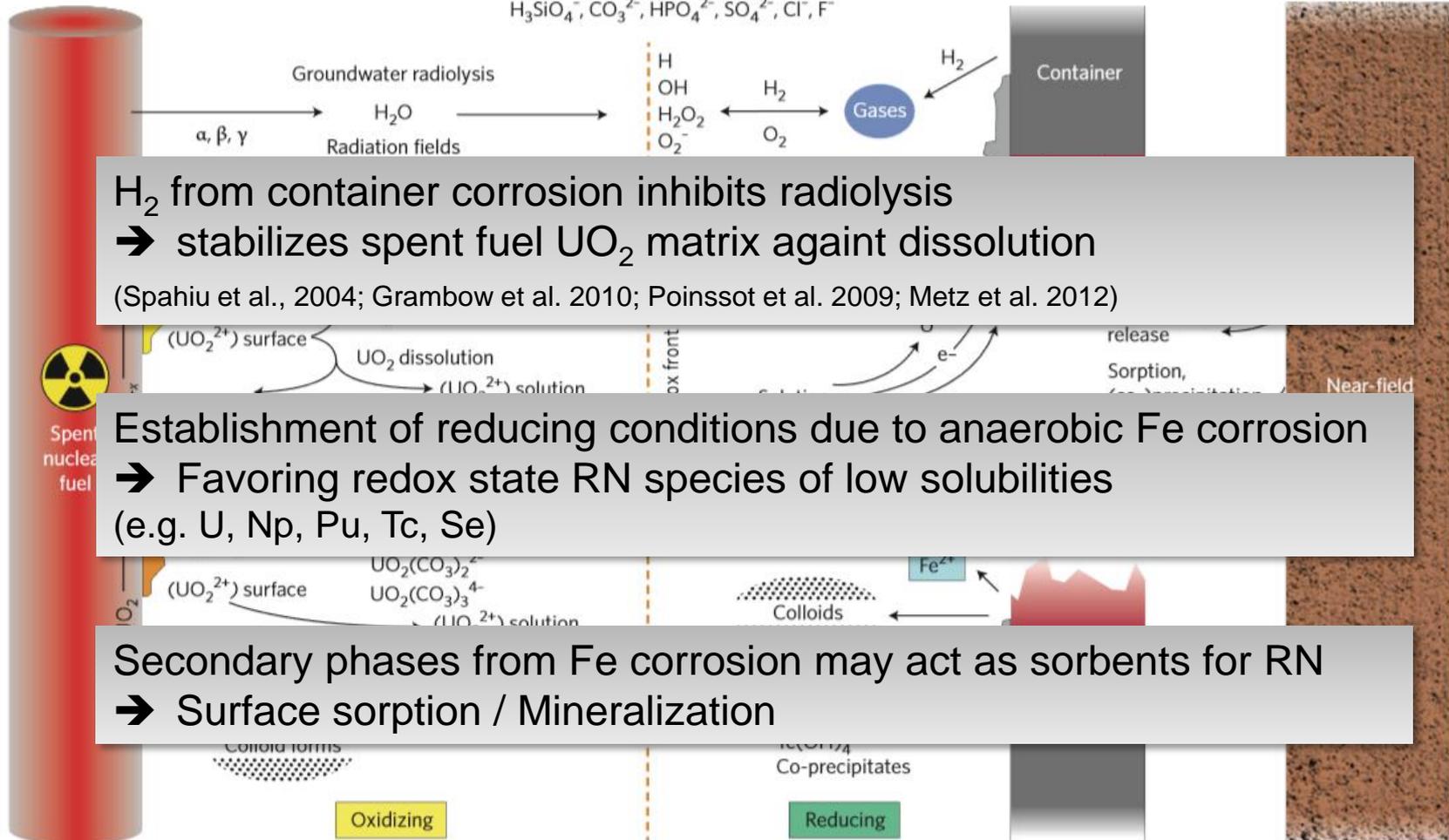
- Sorption to mineral surfaces
- solid-solution formation
- Colloid formation



Processes in the near field of a repository for spent nuclear fuel

Molar ratio:
Fe:U:Pu ~ 40-50:1:0,01

Groundwater components
Ca²⁺, Na⁺, Mg²⁺, K⁺, Al³⁺
H₃SiO₄⁻, CO₃²⁻, HPO₄²⁻, SO₄²⁻, Cl⁻, F⁻



H₂ from container corrosion inhibits radiolysis
 → stabilizes spent fuel UO₂ matrix against dissolution
 (Spahiu et al., 2004; Grambow et al. 2010; Poinssot et al. 2009; Metz et al. 2012)

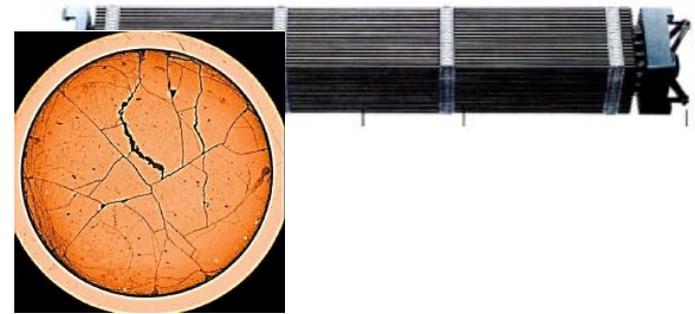
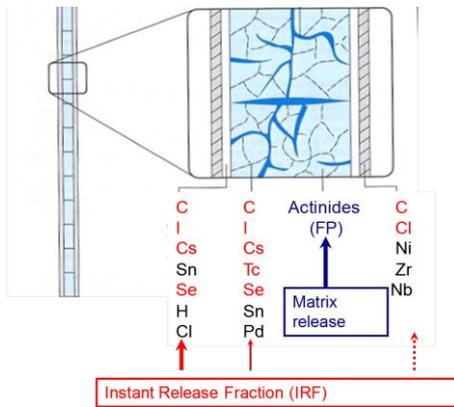
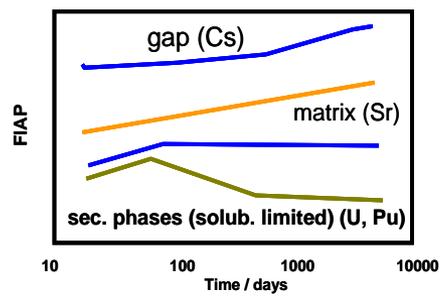
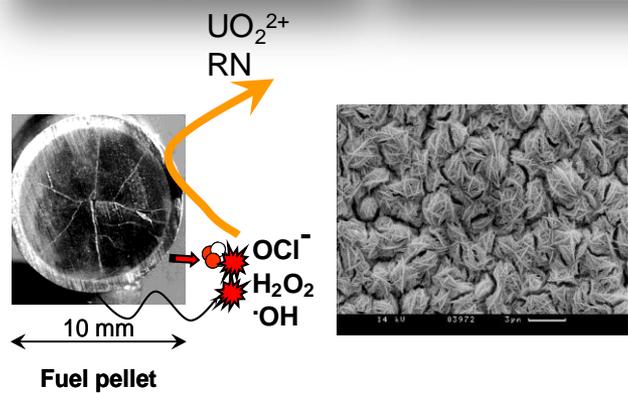
Establishment of reducing conditions due to anaerobic Fe corrosion
 → Favoring redox state RN species of low solubilities
 (e.g. U, Np, Pu, Tc, Se)

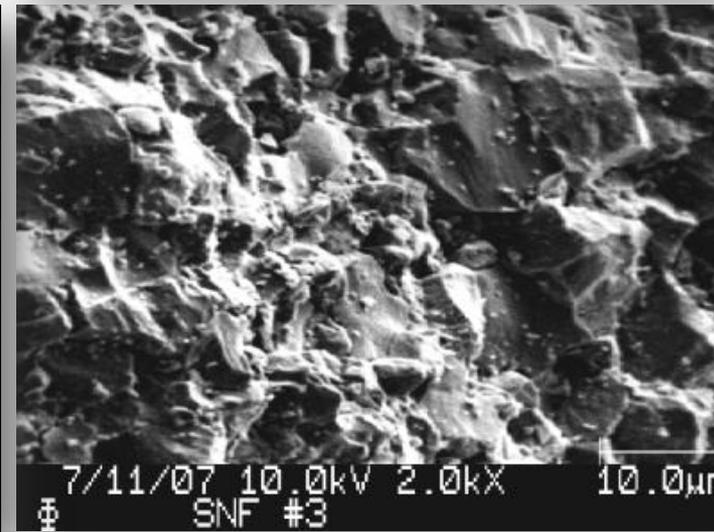
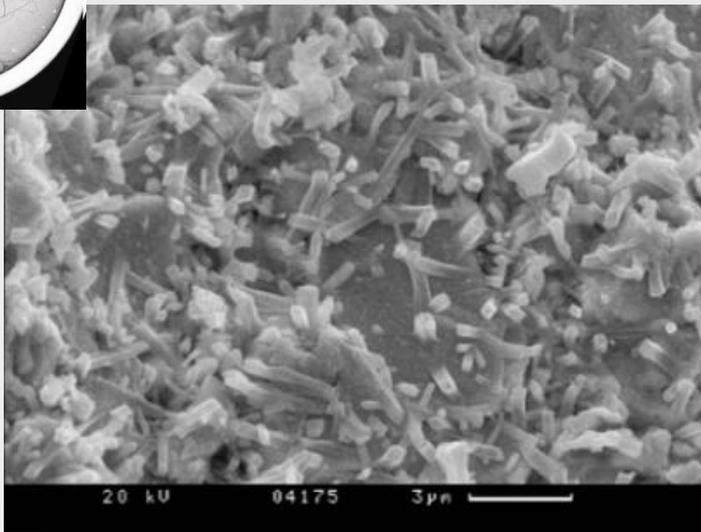
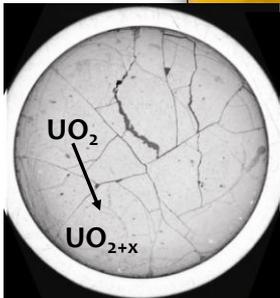
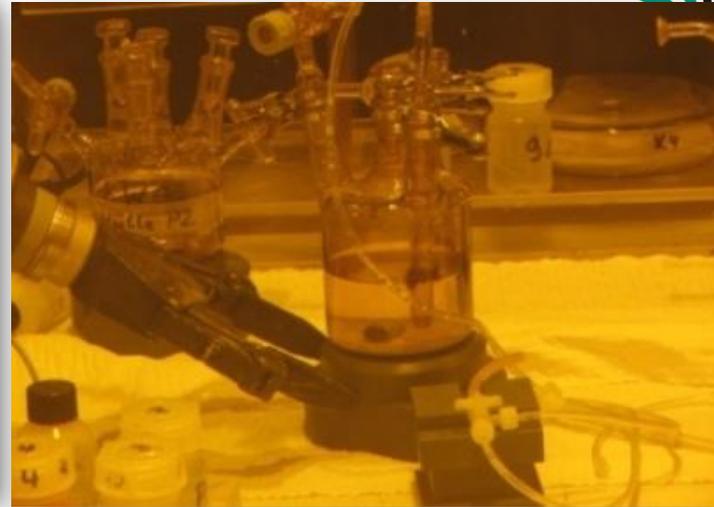
Secondary phases from Fe corrosion may act as sorbents for RN
 → Surface sorption / Mineralization

R. Ewing, Nature Mat., 2015

The role of the waste form:

- Spent nuclear fuel -
- Impact of radiolysis on fuel behavior in a repository
- Long term stability of fuel in the repository („H₂-effect“)
- „Instant release“ fraction
- New issue: degradation behavior of heat generating radioactive waste under extended interim storage conditions?





Anoxic corrosion of UOX fuel

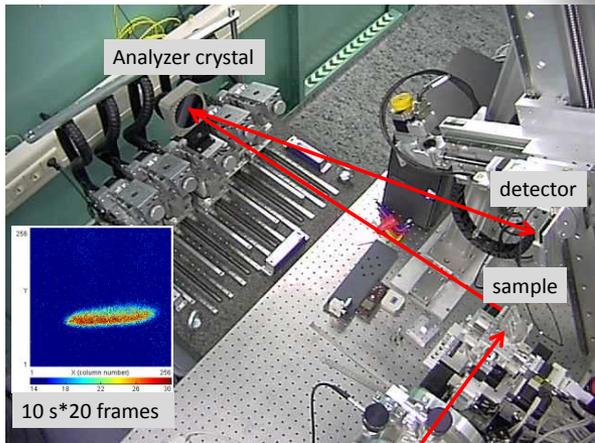
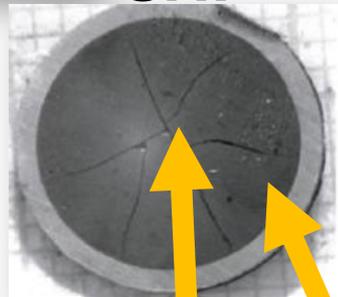
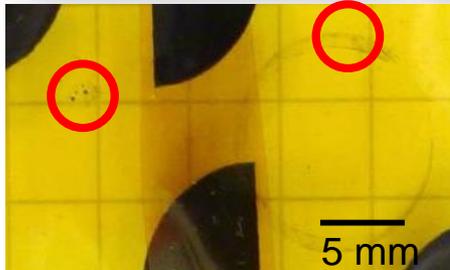
Anoxic corrosion of UOX fuel in presence of H₂

Source: Loida et al. (2007) MRS SBfNWM 30, vol. 985, 14-20; Metz et al. RCA, 2012

HR-XANES analysis of Uranium oxidation state in spent UO_2 fuel



SNF

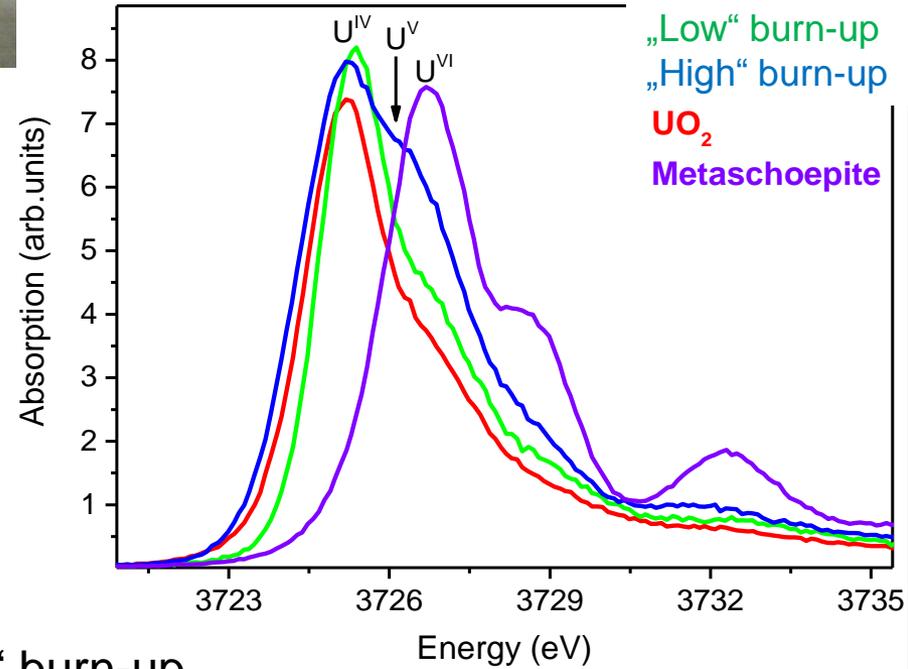


HR-XAS spectrometer at INE-Beamline

X-ray beam

„High“ burn-up rim
 „Low“ burn-up center

U M_β



„Low“ burn-up
 „High“ burn-up
 UO_2
 Metaschoepite

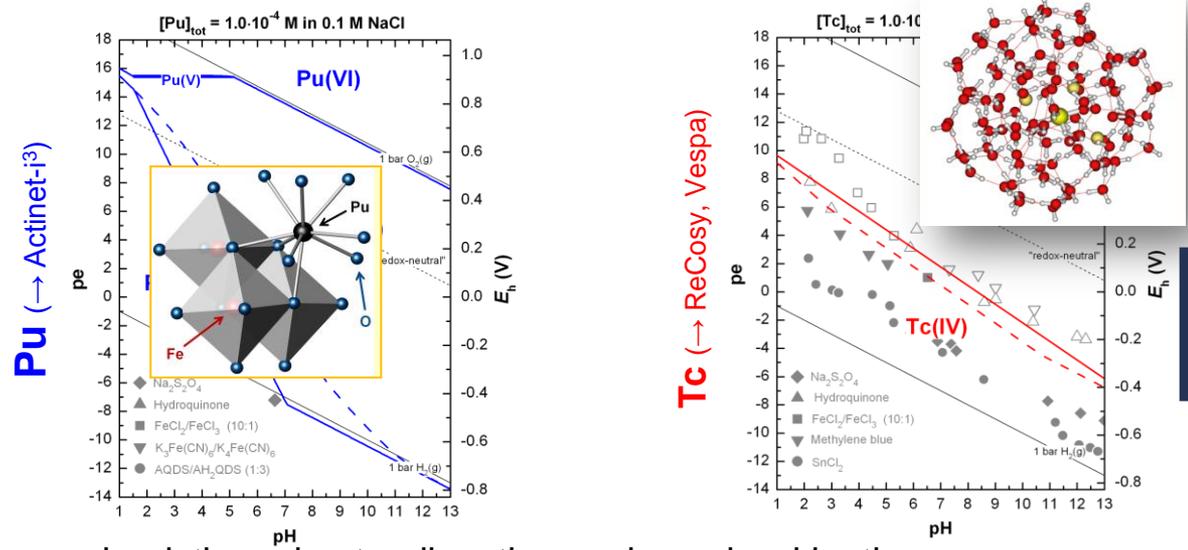
**1st time:
 Detection of U(V) in spent fuel**

S. Bahl et al., to be submitted to Environ. Sci. Technol.

Example: redox chemistry of actinides/fission products



Fundamental research



Reference data base projects

mechanistic understanding, thermodynamics, kinetics
(coop.: Kyoto University, KAIST, LANL - Carlsbad office)

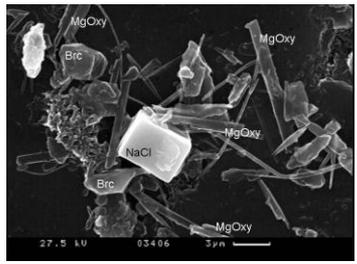


Example: radionuclide retention in cementitious phases

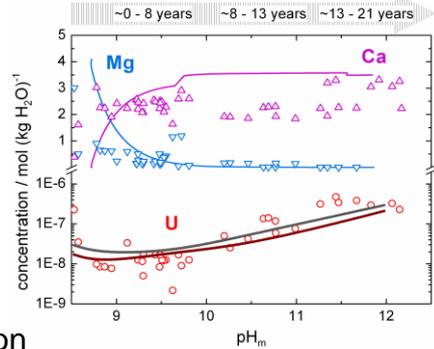
Applied research



1:1 scale experiments from the Asse mine



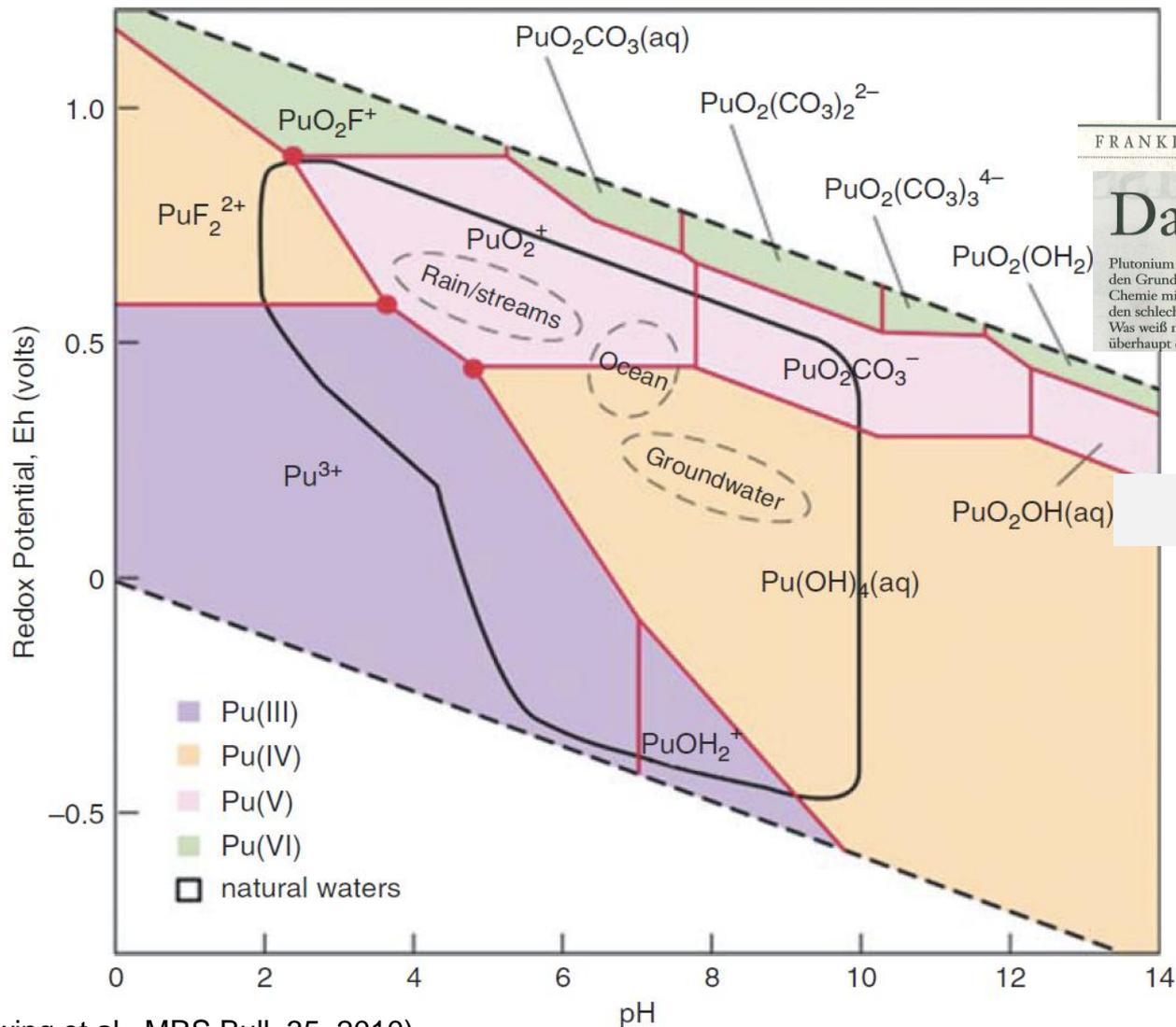
Secondary phase formation During cement corrosion



Geochemical description



Redox chemistry of Plutonium

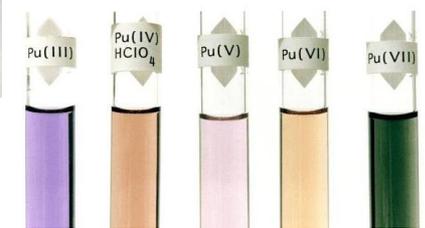


(Ewing et al., MRS Bull. 35, 2010)

FRANKFURTER ALLGEMEINE SONNTAGSZEITUNG

Das elementare Böse

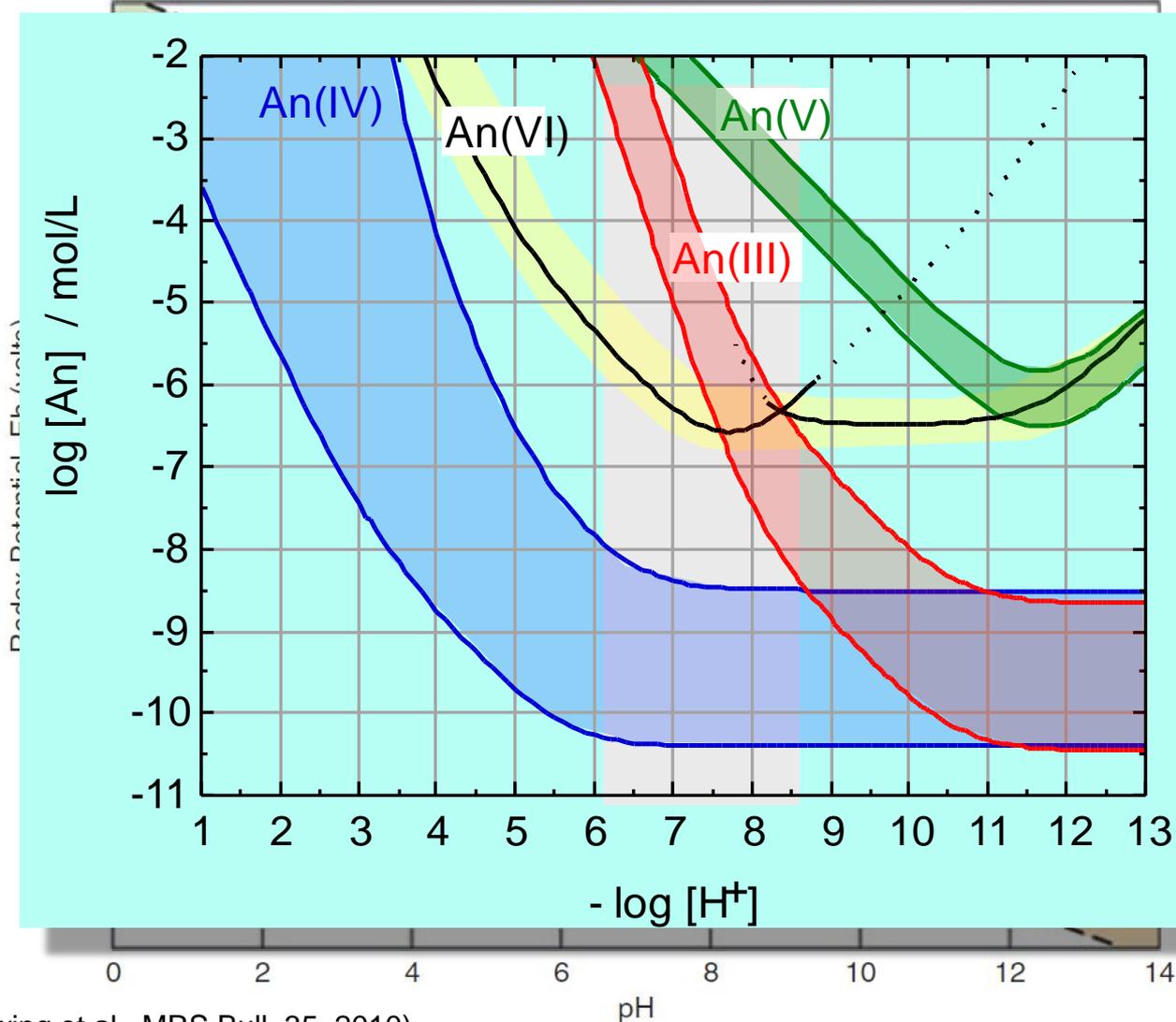
Plutonium hat unter den Grundstoffen der Chemie mit Abstand den schlechtesten Ruf. Was weiß man überhaupt darüber?



The element of evil

Fellhauer, Altmaier, Fanghänel, 2010

Redox chemistry of Plutonium

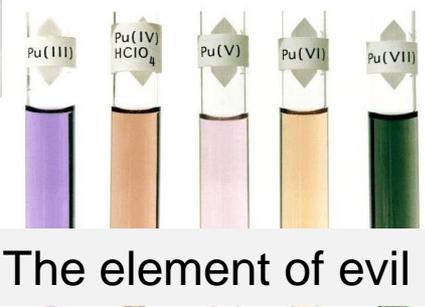


(Ewing et al., MRS Bull. 35, 2010)

ARTER ALLGEMEINE SONNTAGSZEITUNG

elementare Böse

unter
fen der
bestand
ten Ruf.
über?

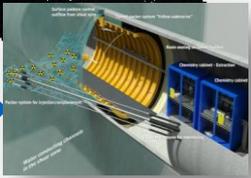
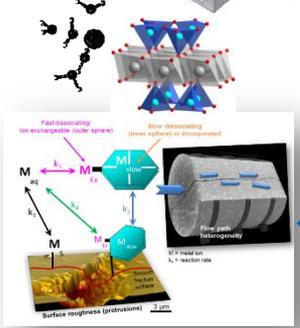
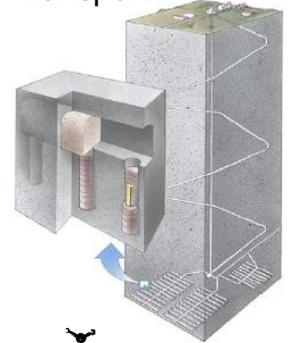


The element of evil

Fellhauer, Altmaier, Fanghänel, 2010

Host rock: crystalline

Impact of glacial meltwater onto the erosion of bentonite barriers in a repository in granite and onto the clay nanoparticle borne radionuclide transport



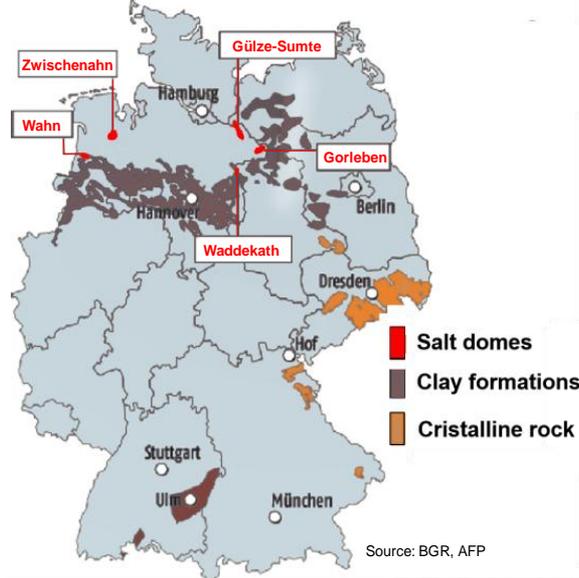
Laboratory

underground rock laboratory

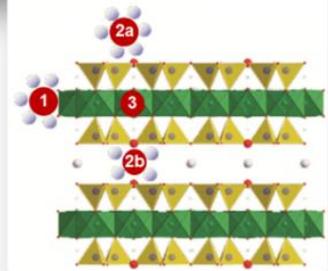
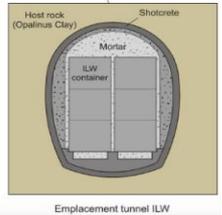
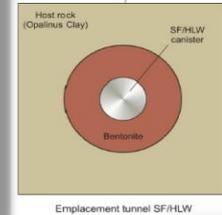
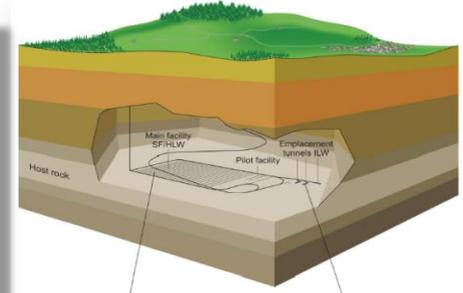
Host rock: clay rock

Diffusion and retention of radionuclides in clay rock: Impact of Pore water composition (I, pH, Eh, CO₃²⁻, T...)

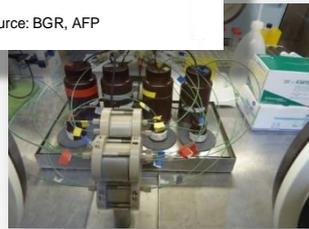
Potential host rocks for nuclear waste repositories in Germany



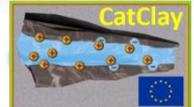
Source: BGR, AFP



Sorption mechanism of cations on three-layer sheet silicates (i.e. montmorillonite).



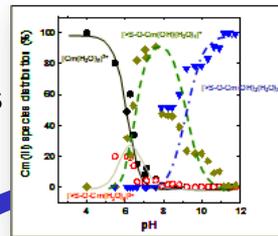
US-DOE



Comprehensive approach: from the nanoscale to the field

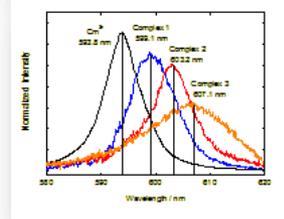
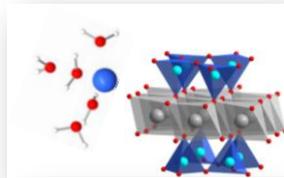
Example: Impact of glacial melt water induced erosion of bentonite barriers in a repository in crystalline rock

Geochemical speciation,
Reaction kinetics

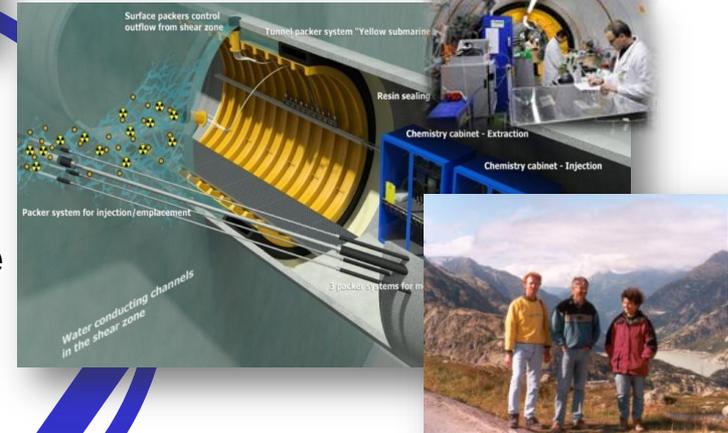


In-situ experiments
(laboratory; underground rock laboratories - Grimsel, Äspö)
→ **upscaling**

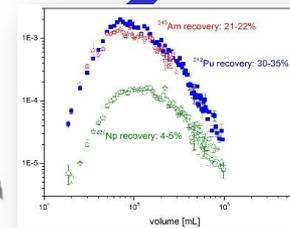
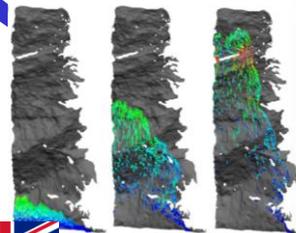
Spectroscopy,
Quantum
Chemistry



**Radionuclide
behaviour in the
geosphere**

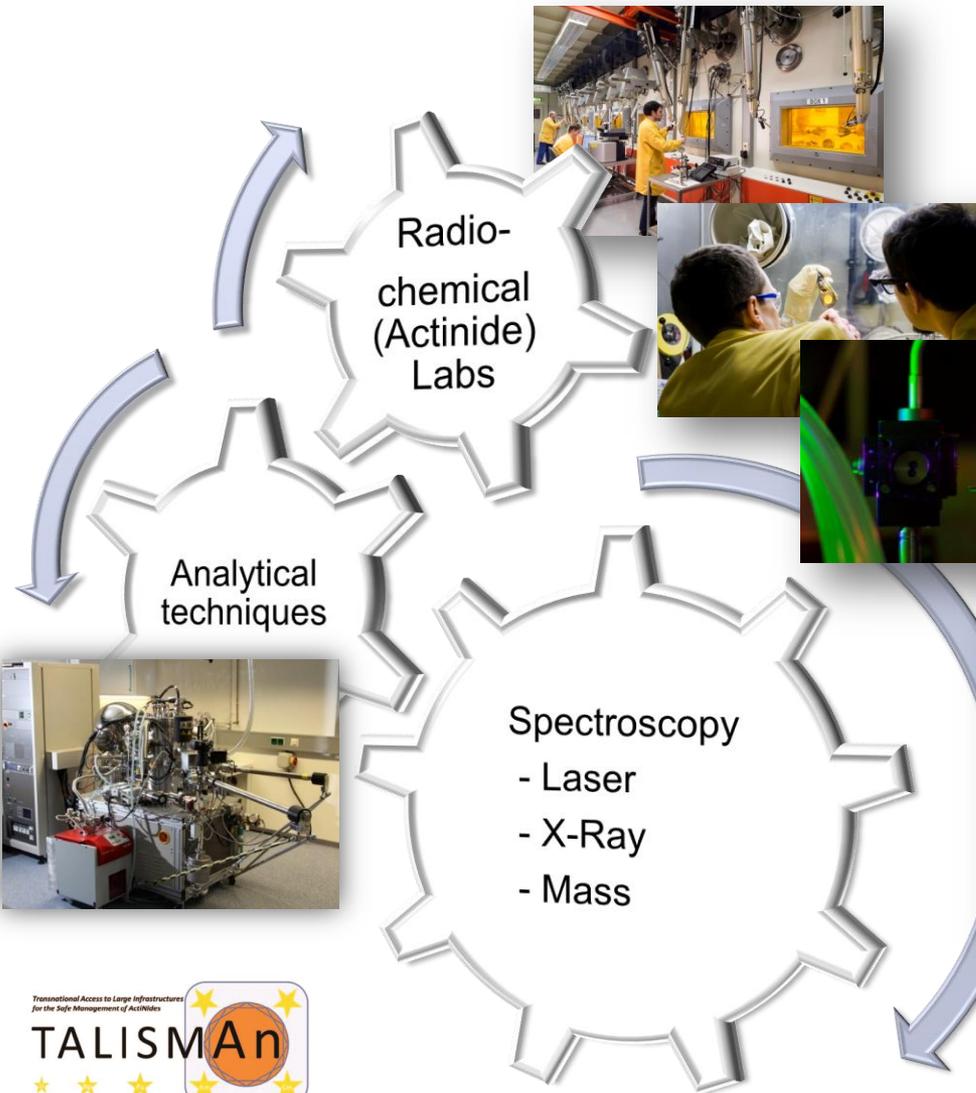


Bundesministerium für Wirtschaft und Technologie **KOLLORADOe**



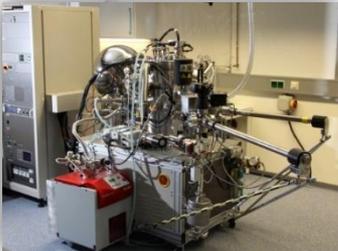
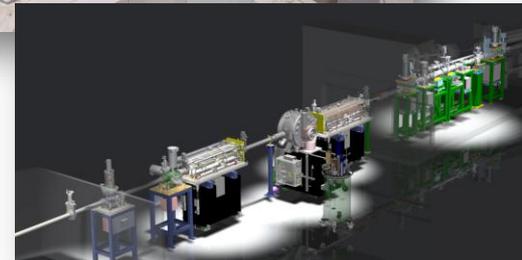
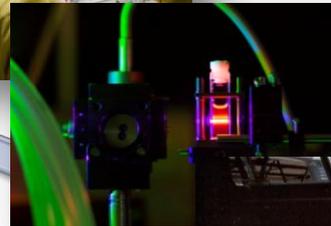
Simulation of radionuclide
migration by reactive
transport modeling



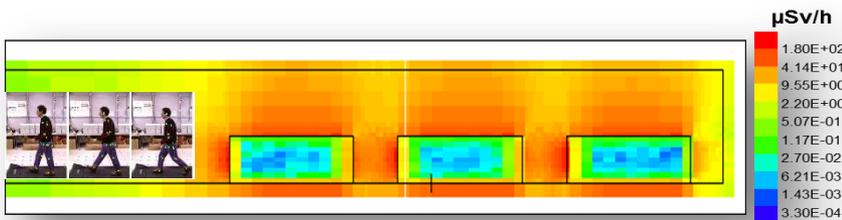
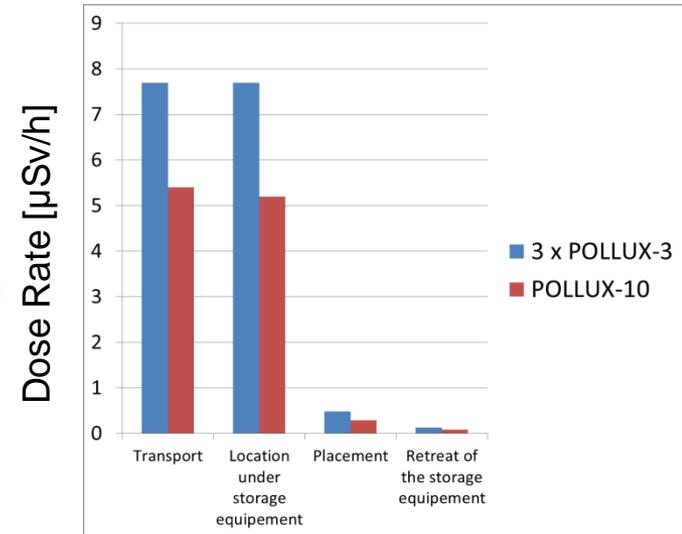
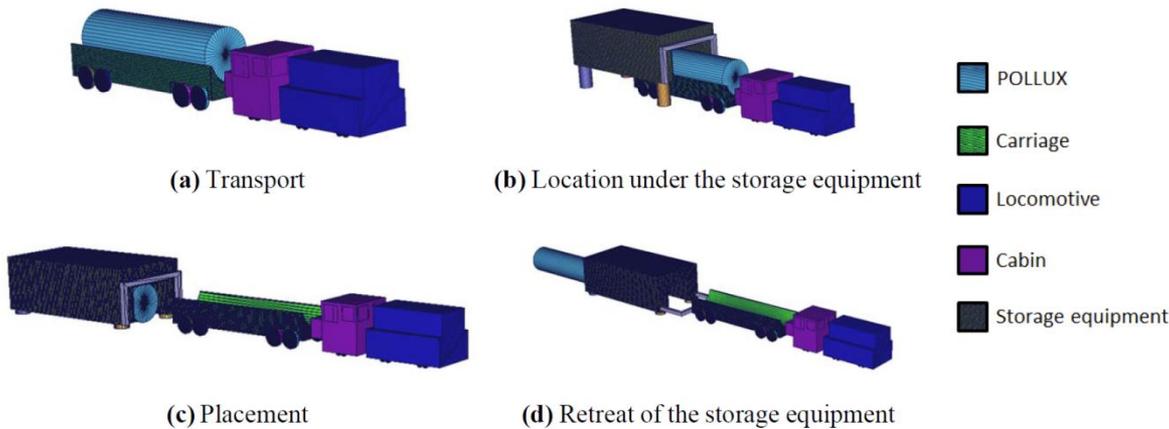


Unique combination of actinide laboratories and spectroscopies

- Fundamental science
- „European Actinide Pooled Facility“
Transnational Access Initiatives
„Seed crystal“ for international cooperation (EU, USA, Asia)



Investigation of POLLUX storage/disposal scenarios (rock salt and clay stone)



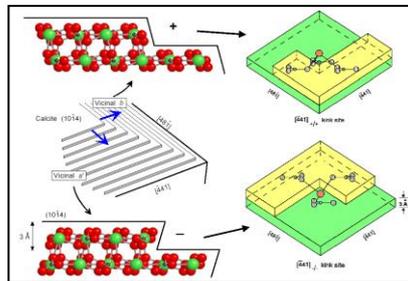
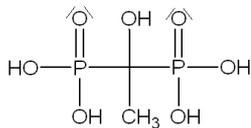
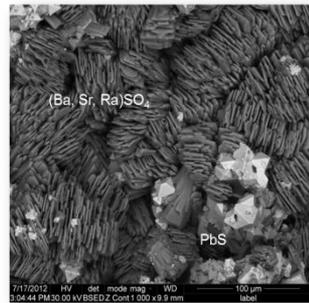
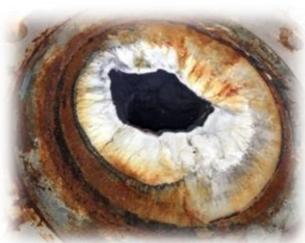
Comparison - a similar amount of SNF should be disposed:

- Germany: 90% UOX + 10% MOX
 - POLLUX-10 casks in **rock salt**
 - POLLUX-3 casks in **claystone (temperature constraint)**
- => 1 POLLUX-10 vs. 3 POLLUX-3**

NORM residues from geothermal power plants

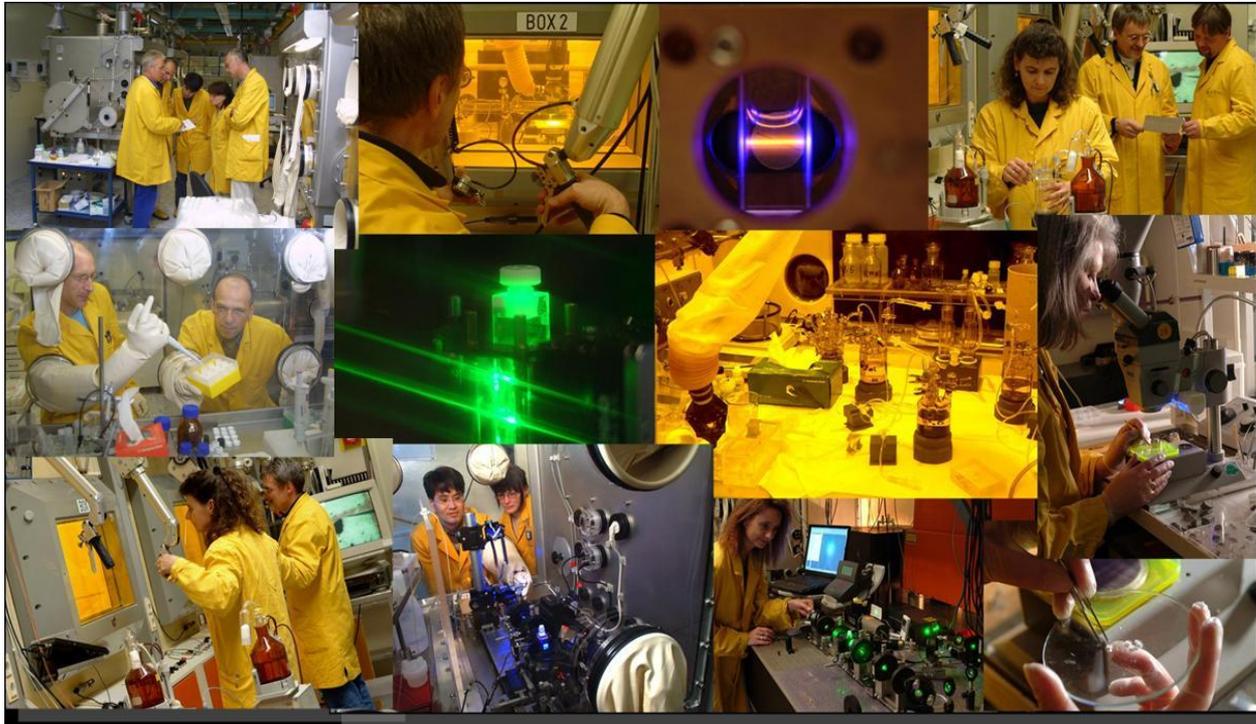
Data: Scheiber et al. (2012)

| NORM residues C [Bq /g] | Dose rate [heat exchanger] (exemplarily) |
|-------------------------|------------------------------------------|
| up to 1,350 (Ra-226) | 12 μ Sv/h in 1m distance |
| up to 1,100 (Pb-210) | > 34 μ Sv/h in 0.1m distance |



HEDP as crystal growth inhibitor
(hydroxyethylidene-diphosphonate,
 $C_2H_8O_7P_2$):

- Formation of scalings particularly in the cold part of the facility (ΔT 90° C)
- Primarily Ba/Sr-sulfate & PbS
- ^{226}Ra in Ba/Sr-Sulfate, ^{210}Pb PbS.
- **Development of Inhibitor concept to avoid scale formation**



INE:

- employees: ~ 105
- Radiochemical laboratories
- Hot cell laboratories
- Glove boxes
- „State of the Art“-instrumentation

● Research is embedded in European and International projects and cooperations

● INE is engaged in training and education of students

● Close cooperation with various universities