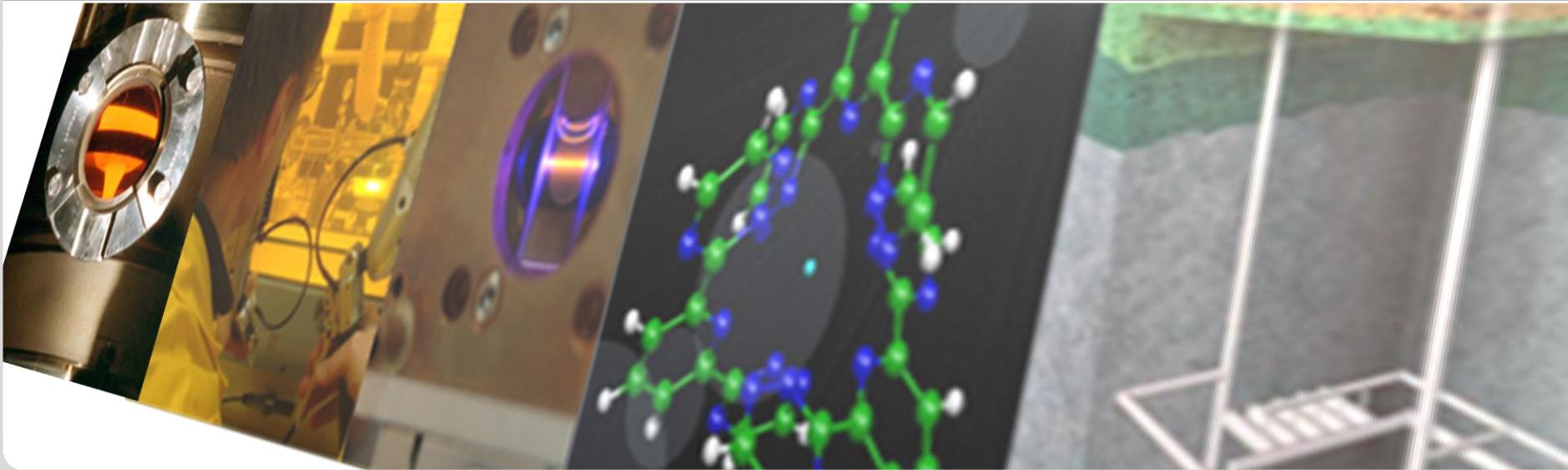


Visit to the Institute for Nuclear Waste Disposal (INE)

European Summer School Radiochemistry and Nuclear Instrumentation

Horst Geckeis (Email: horst.geckeis@kit.edu)

INSTITUT FÜR NUKLEARE ENTSORGUNG (INE)



European Summer School
Radiochemistry and nuclear instrumentation (low level radioactivity)



Visit at the Institute for Nuclear Waste Disposal (INE),
Karlsruhe Institute of Technology (KIT)

23 august 2017

Programme

09:00 h	Arrival at KIT-Campus North entrance	
09:15 h	Welcome at INE and introduction	<i>H. Geckeis</i>
09:30 h	Research towards radioactive waste disposal at KIT-INE	<i>H. Geckeis</i>
10:15 h	Coffee break	
10:30 h	Ultratrace analysis of radionuclides by AMS	<i>F. Quinto</i>
11:30 h	Radionuclide speciation by laser spectroscopy	<i>C. Garcia-Perez</i>
12:30 h	Lunch	
13:30 h	Synchrotron based X-ray spectroscopic characterization of radionuclides and Visit of INE-Beamlines at ANKA (ANKA buiding)	<i>T. Vitova</i>
15:00 h	Visit of INE laboratories <ul style="list-style-type: none">• Hot cells• Mass spectrometry• Laser spectroscopy• Geochemistry	<i>M. Herm</i> <i>F. Quinto</i> <i>C. Garcia-Perez</i> <i>C. Joseph</i>
17:00 h	End of visit	

Topics

- What is KIT?
- Nuclear waste in Germany and worldwide
- Research at the Institute for Nuclear Waste Disposal

KIT – The Research University in the Helmholtz Association

Status: January 2017

PRESIDENTIAL COMMITTEE

Scientists in teaching and research *Cultural diversity* *Research infrastructure*

DEVELOPING SCIENTIFIC CAREER PATHS

The Research University in the Helmholtz Association

TRANSPARENT SERVICES *KIT thinks and acts as ONE institution*
FOR RESEARCH; TEACHING, AND INNOVATION *Energy*

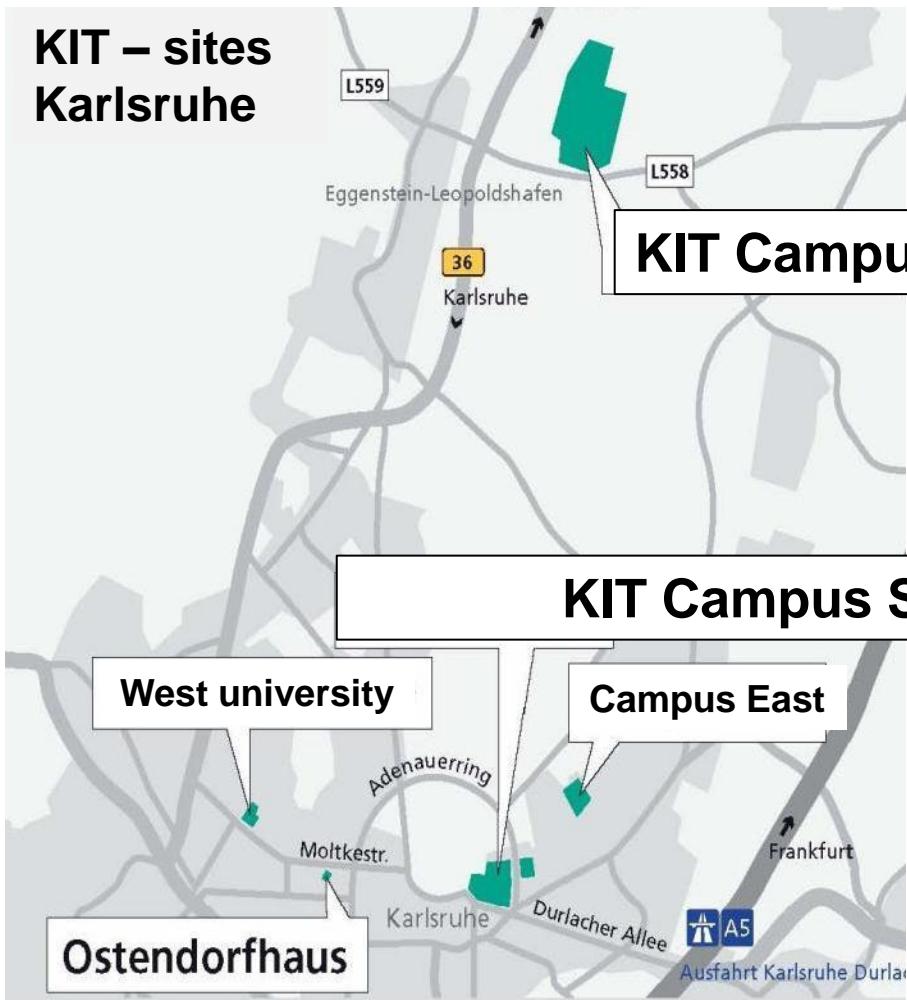
*Research-based teaching
and learning*

INNOVATION AS A STATUTORY MISSION

TOWARDS A LEADING POSITION IN EUROPE

Mobility
Information

Two KIT sites



Branch Offices



<http://www.freepik.com/free-photos-vectors/map> "Map" vector designed by Freepik

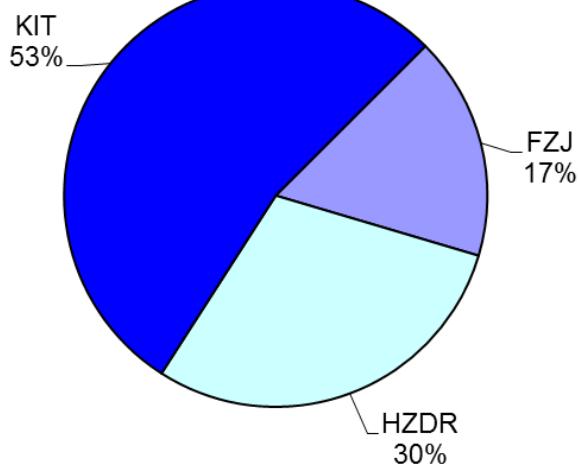
The research programme NUSAFFE

Nuclear Waste Management and Safety as well as Radiation Research

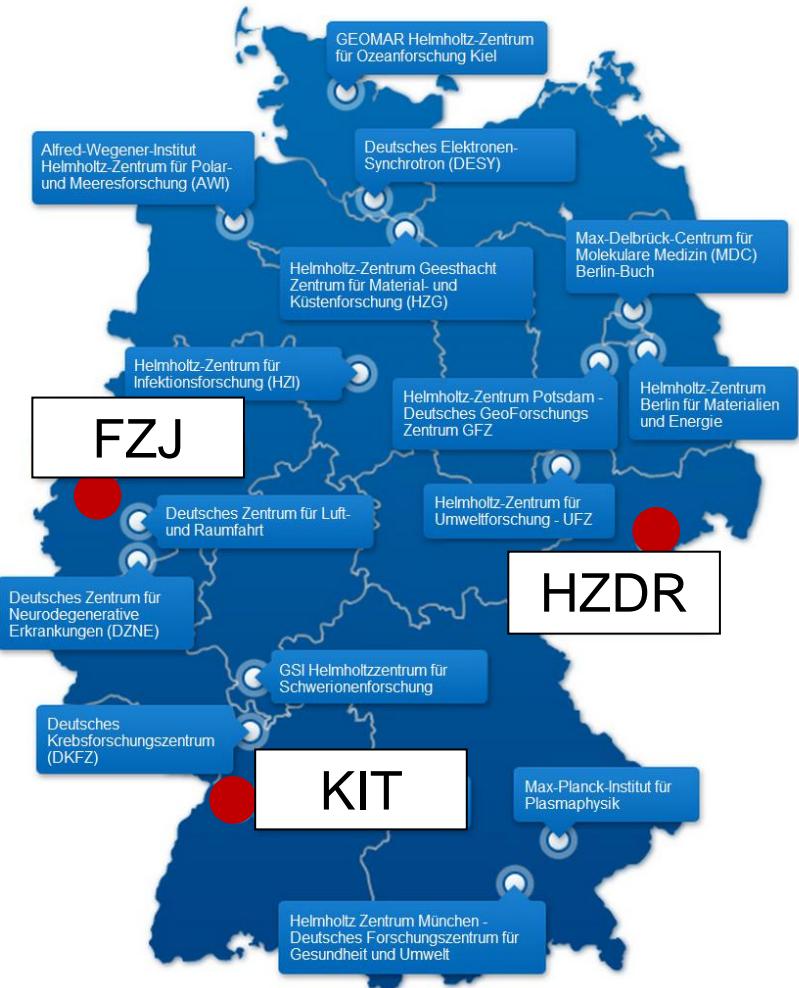


NUSAFFE resources:

234 FTE



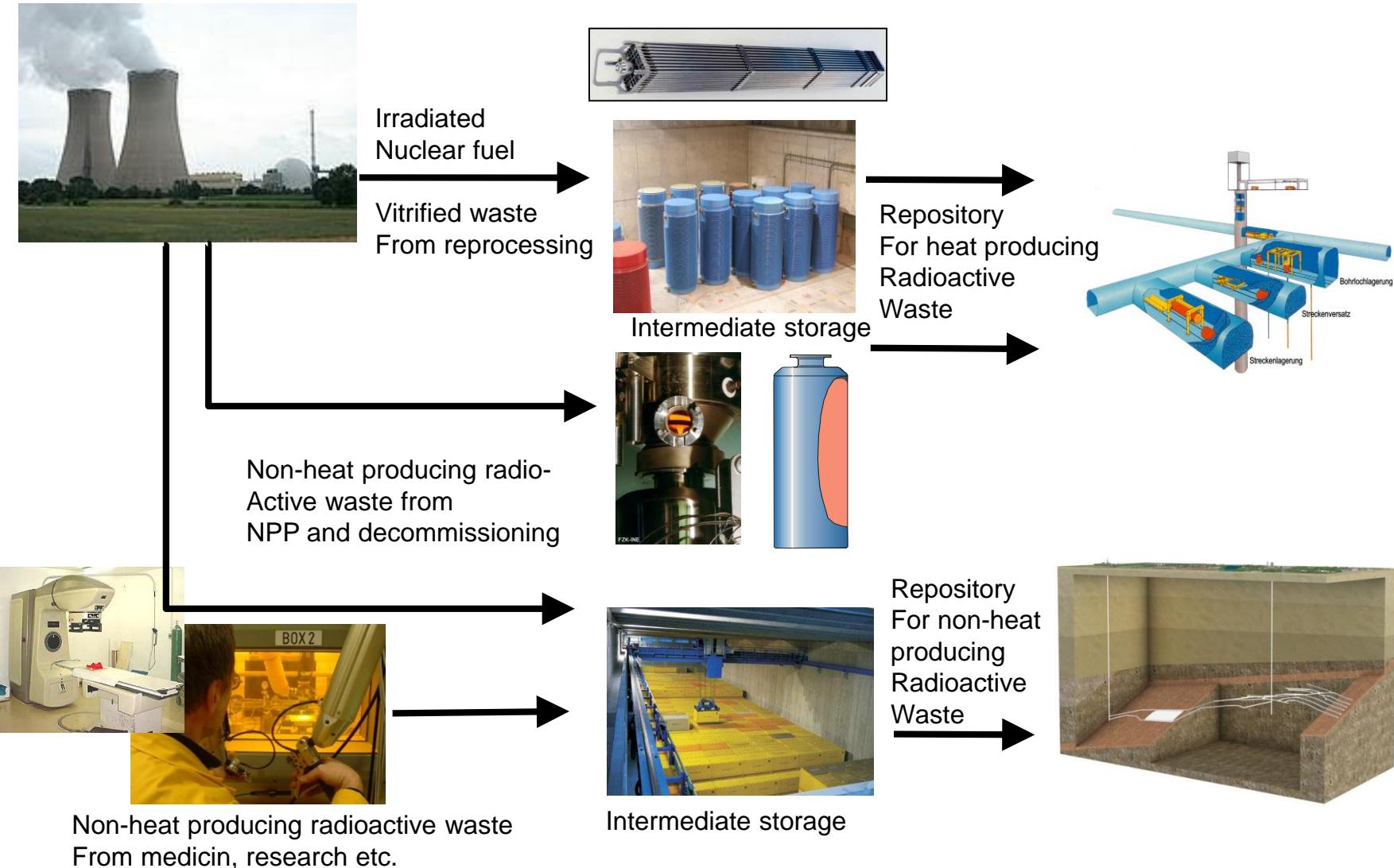
67% Nuclear Waste Management
33% Reactor Safety



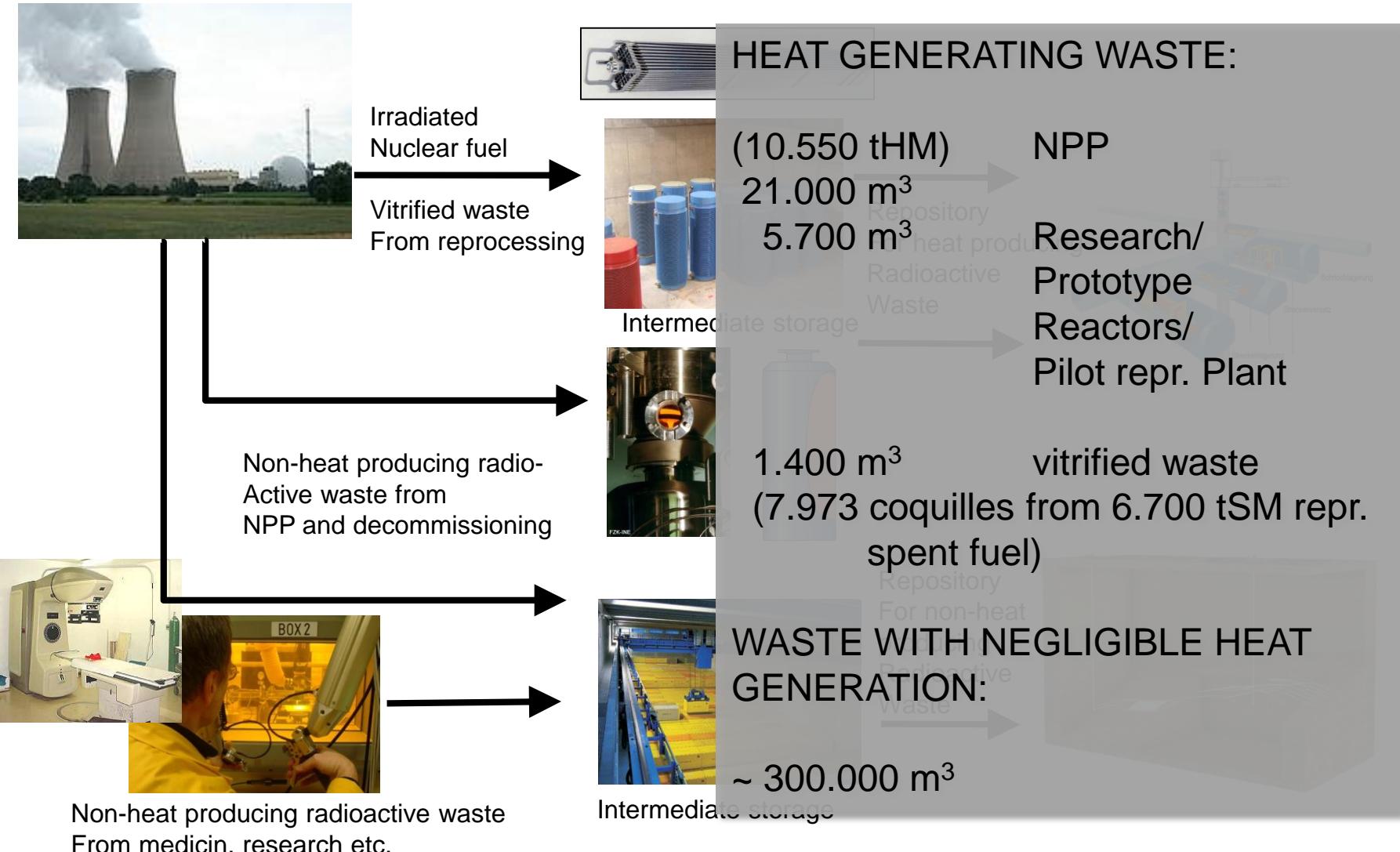
Topics

- What is KIT?
- Nuclear waste in Germany and worldwide
- Research at the Institute for Nuclear Waste Disposal

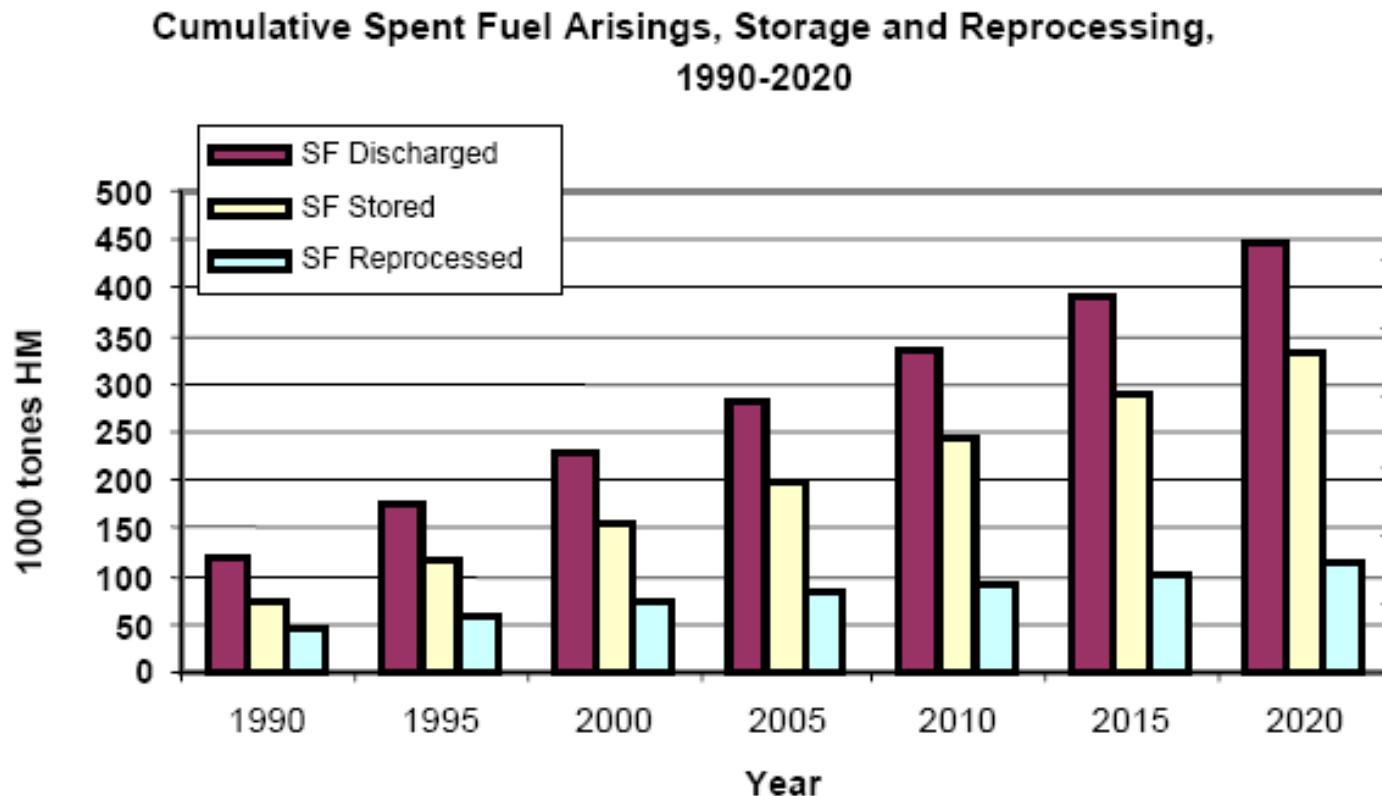
Nuclear waste in Germany



Nuclear waste in Germany



Spent fuel arising world wide



Per year:

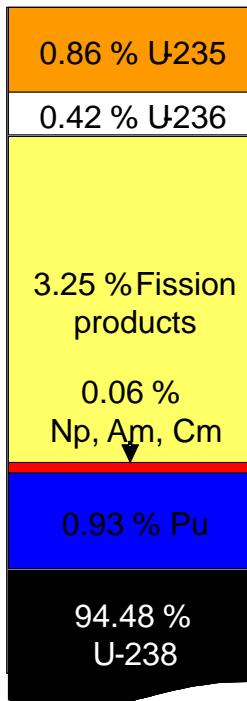
Cumulative until 2020:

10.500 t spent nuclear fuel

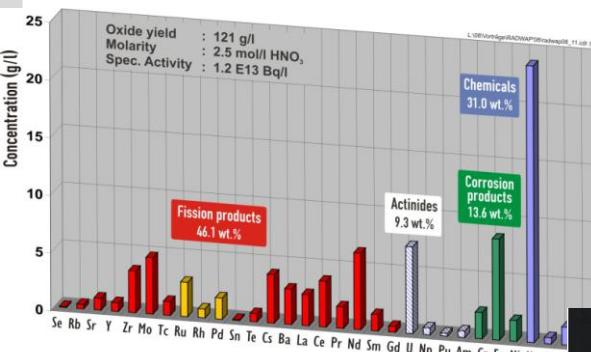
~ 445.000 t spent nuclear fuel

~ 1/3 reprocessed → Vitrification of
remaining high level liquid waste

Content of high-level nuclear waste spent fuel



PWR; Burn-up 33GWd/tU



Vitrified liquid waste

PERIODE	Hauptgruppen	
	I	II
	1 1.01	
	H	
1.	Wasserstoff -259 0.09 -253 2.1	
	3 6.94	4 9.01
2.	Li	Be
	Lithium 181 0.53 1330 1.0	Beryllium 177 1.85 2970 1.5
	11 22.99	12 24.31

Some long-lived fission and activation products

C-14	$(t_{1/2} \sim 5.700 \text{ a})$
Cl-36	$(t_{1/2} \sim 300.000 \text{ a})$
Se-79	$(t_{1/2} \sim 360.000 \text{ a})$
Tc-99	$(t_{1/2} \sim 213.000 \text{ a})$
I-129	$(t_{1/2} \sim 15,6 \text{ Ma})$

3. **Fission products:**
Cs-137, Sr-90 ($t_{1/2} \sim 30$ a)

Fission products: Cs-137, Sr-90 ($t_{1/2} \sim 30$ a)												open								
												VIII								
												Fe	Co	Ni	C					
4.	Kalium	Calcium	Scandium	Titan	Vanadium	Chrom	Mangan	Eisen	Cobalt	Nickel	Kupfer									
4.	64 760	86 1440	1.55 1.0	1539 2730	3.0 1.3	1669 3260	4.54 1.5	1900 3450	6.1 1.6	1875 2200	7.19 1.6	1245 2097	7.43 1.5	1536 3000	7.88 1.8	1495 2900	8.9 1.9	1453 2730	8.9 1.9	1083 2595
5.	Rb	Sr	Y	Zr	Nb	Mo	Tc*	Ru	Rh	Pd	A									
5.	Rubidium 39 689	Strontium 768 0.8	Yttrium 2.6 1.0	Zirconium 1509 2927	Niob 4.47 1.2	Technetium 649 1.4	Molybdenum 2469 4827	Ruthenium 8.57 1.6	Rhenium 2610 5560	Rhodium 10.2 1.8	Palladium 11.5 1.9	Si								
6.	Cs	Ba	Lanthanoide	Hf	Ta	W	Re	Os	Ir	Pt	A									
6.	Caesium 122.91 69.0	Barium 1.90 164.0		Hafnium 2222 5400	Tantal 13.31 3.5425	Wolfram 2996 1.5	Rhenium 16.6 5930	Osmium 19.3 1.7	Iridium 3410 5900	Platin 21.0 1.7	Gold 22.6 5500	G								
7.	Fr*	Ra*	Actinoide	Ku*	Ha*															
7.	Francium (27) 677	Radium 1100 0.7		Kurkatschium -	Hahnium -															
	87 (223)	88 226.03	89-103	104 (261)	105 (262)	106	107	108	109	110	111									

Hauptgruppen							
II	IV	V	VI	VII	VIII		
						2 He	4.00 Helium
						-269	0.18
						10.81	10.18
		C	N	O	F	Ne	Neon
		Kohlenstoff	Stickstoff	Sauerstoff	Fluor		
		6 Oxymethan 4839	7 -210 1.25 196 3.0	8 -219 1.43 -183 3.5	9 -220 1.7 -188 4.0		
		13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
		Aluminium	Silicium	Phosphor	Schwefel	Chlor	Argon
I	II	6691 2450 2.0	1410 2.5 2689 2.9	44 1.82 280 2.1	119 2.07 445 2.5	103 3.2 30 3.0	189 1.78 -186
63.55	30	65.38	31	69.72	32	72.59	33
Cu	Zn	Ga	Ge	As	Se	Br	Kr
Spf	Zink	Gallium	Germanium	Arsen	Selen	Brom	Krypton
89.48	470	7.13	36 5.91	93.7 5.32	Subl. 5.72	217 4.79	7.3 -157 3.7

Important Transuranium elements:

Pu-238 ($t_{1/2} \sim$ 88 a)

Pu-239 ($t_{1/2} \sim 24.110$ a)

Am-241 ($t_{1/2} \sim 432$ a)

Cm-243($t_{1/2} \sim$ 29 a)

Np-237 ($t_{1/2} \sim 2.1$ Ma)

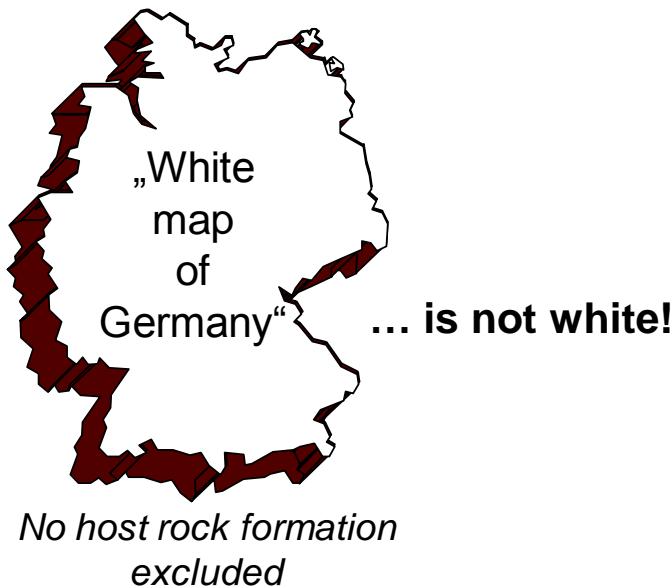
Element	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	Lawrencium
Symbol	Cf	Es	Fm	Mv	No	Lr

N	Neptunium	237	$t_{1/2} \sim$	2.1 Ma		
U	Californium	Einsteini	Fermium	Mendelevium	Nobelium	Lawrencium

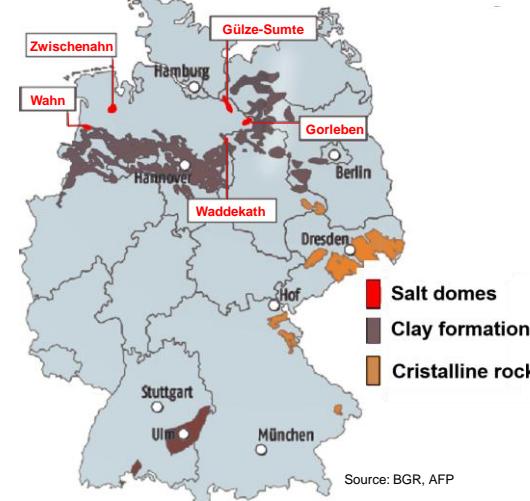
Lanthanoide	La	Ce	Pr	Nd	Pm*	Sm	Eu	Gd	Tb								
Lanthan	Cer	Praseodym	Neodym	Promethium	Samarium	Europium	Gadolinium										
920	8.17	795	6.67	935	6.77	1024	7.06	(1021)	7.22	1023	7.54	826	5.36	1312	7.98		

	Actinoide	89 Ac* Actinium 1050	90 Th* Thorium 101	91 Pa* Protactinium (1230) 11.7	92 U* Uranium (1230) 15.4	93 Np* Neptunium 1132 19	94 Pu* Plutonium 1994	95 Am* Americium 13.7	96 Cm* Curium (1240) 12.51	97 Bk* Berkelium -
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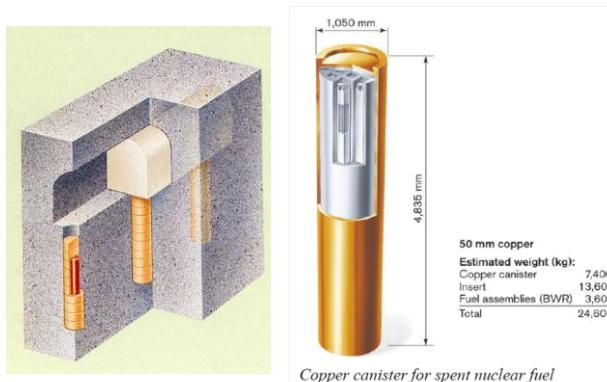




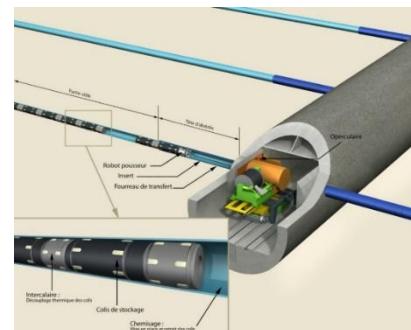
Potential host rocks for nuclear waste repositories in Germany



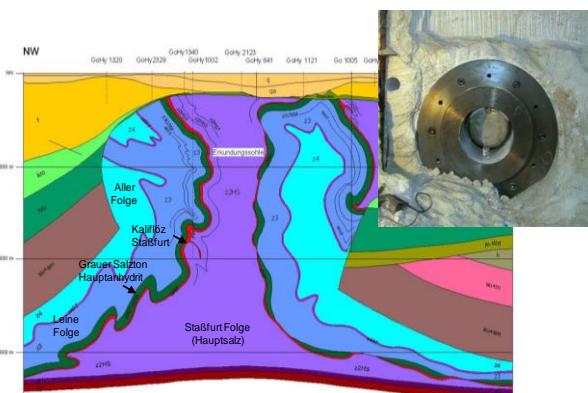
→ Site selection does not only require a science based selection of host rocks but also a selection of repository concepts!



Crystalline rock



Argillaceous rock



Rock salt

Topics

- What is KIT?
- Nuclear waste in Germany and worldwide
- Research at the Institute for Nuclear Waste Disposal



Research and development activities at KIT-INE are an integral component of national provident research and encompass the following areas:

- **Long-term safety research for nuclear waste disposal**
- **Actinide coordination chemistry**
- **Decommissioning of nuclear facilities**
- **Geothermal energy**
- **Research for radiation protection**
- **(Immobilisation of high level radioactive waste by vitrification)**

NUSAFFE

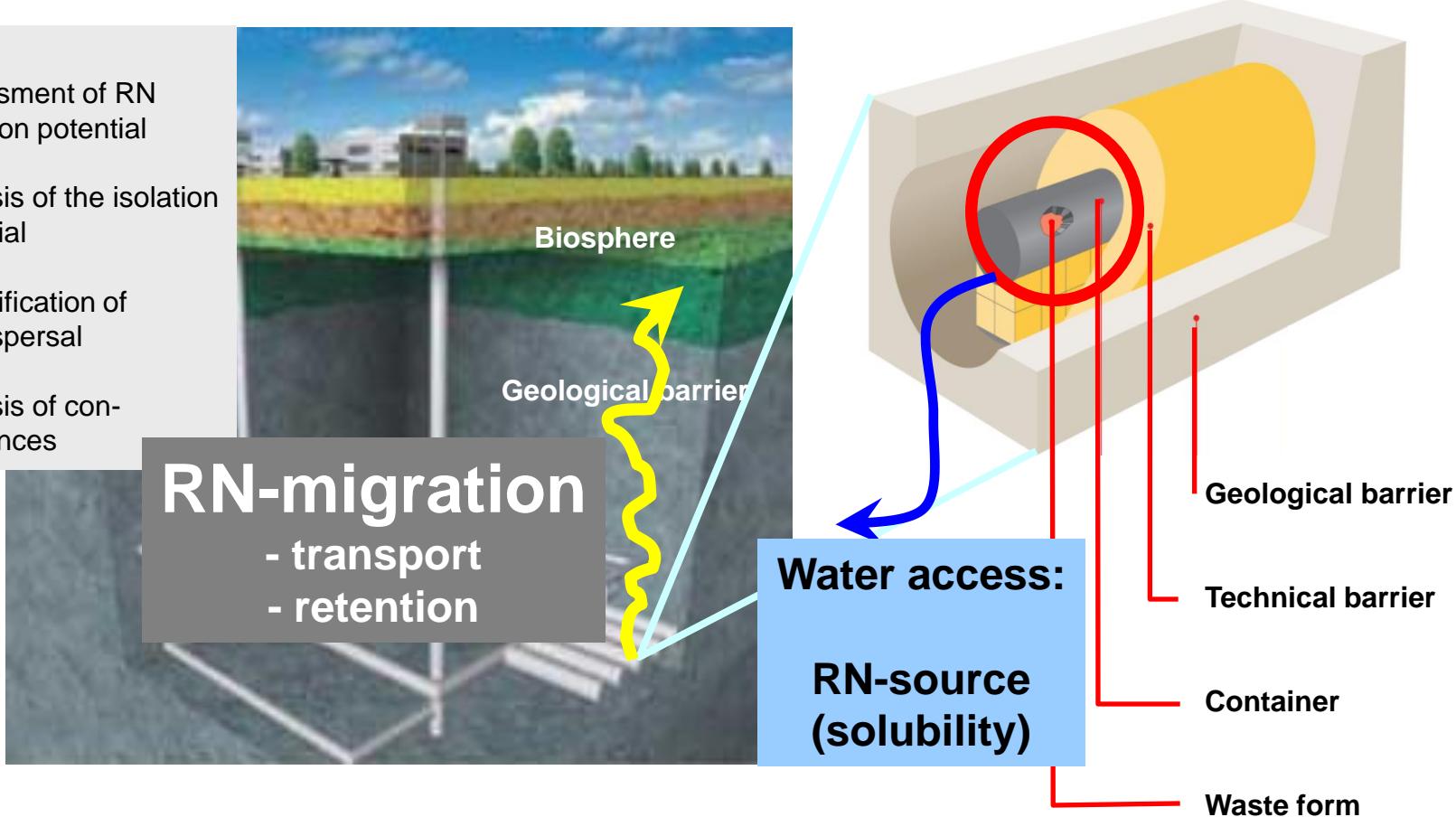


HELMHOLTZ
ASSOCIATION

Processes in the multibarrier system

- Chemical and geochemical aspects -

- Assessment of RN retention potential
- Analysis of the isolation potential
- Quantification of RN dispersal
- Analysis of consequences



Source term:

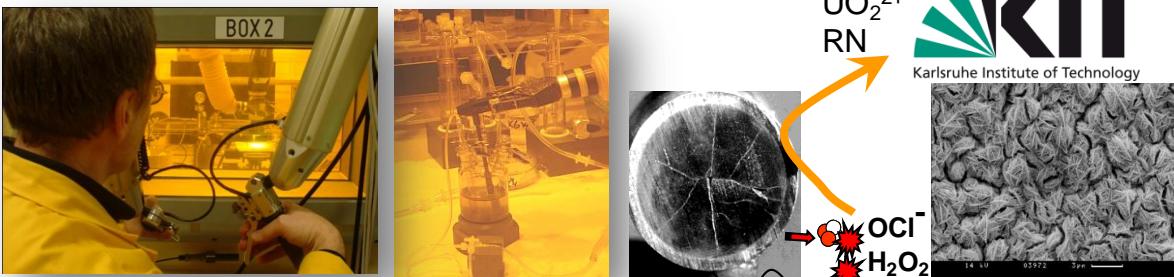
$$r = k_0 \times A_{\min} \times e^{-E_a/RT} a_{H^+}^{n_{H^+}} g(I) \prod_i a_i^{n_i} f(\Delta G_r)$$

Transport + retention:

$$-u \frac{\partial c_i}{\partial t} + D \frac{\partial^2 c_i}{\partial x^2} - \frac{\rho}{\theta} K_d \frac{\partial c_i}{\partial t} = \frac{\partial c_i}{\partial t}$$

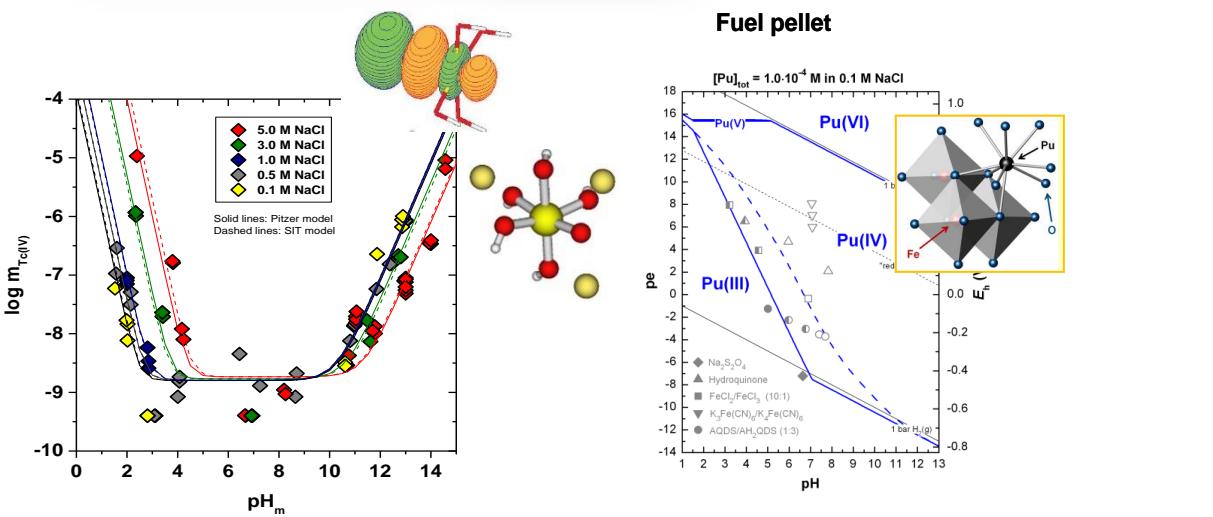
Nuclear waste form behavior

- spent nuclear fuel
- HLW glass
- Cementitious waste



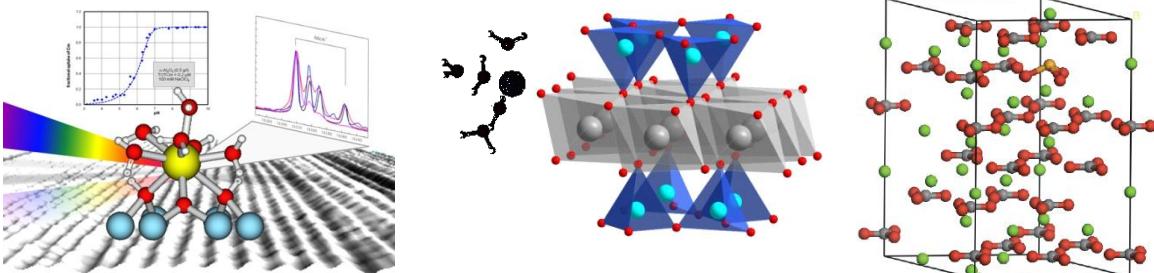
Aquatic chemistry

- Solubility
- Redox
- Complexation



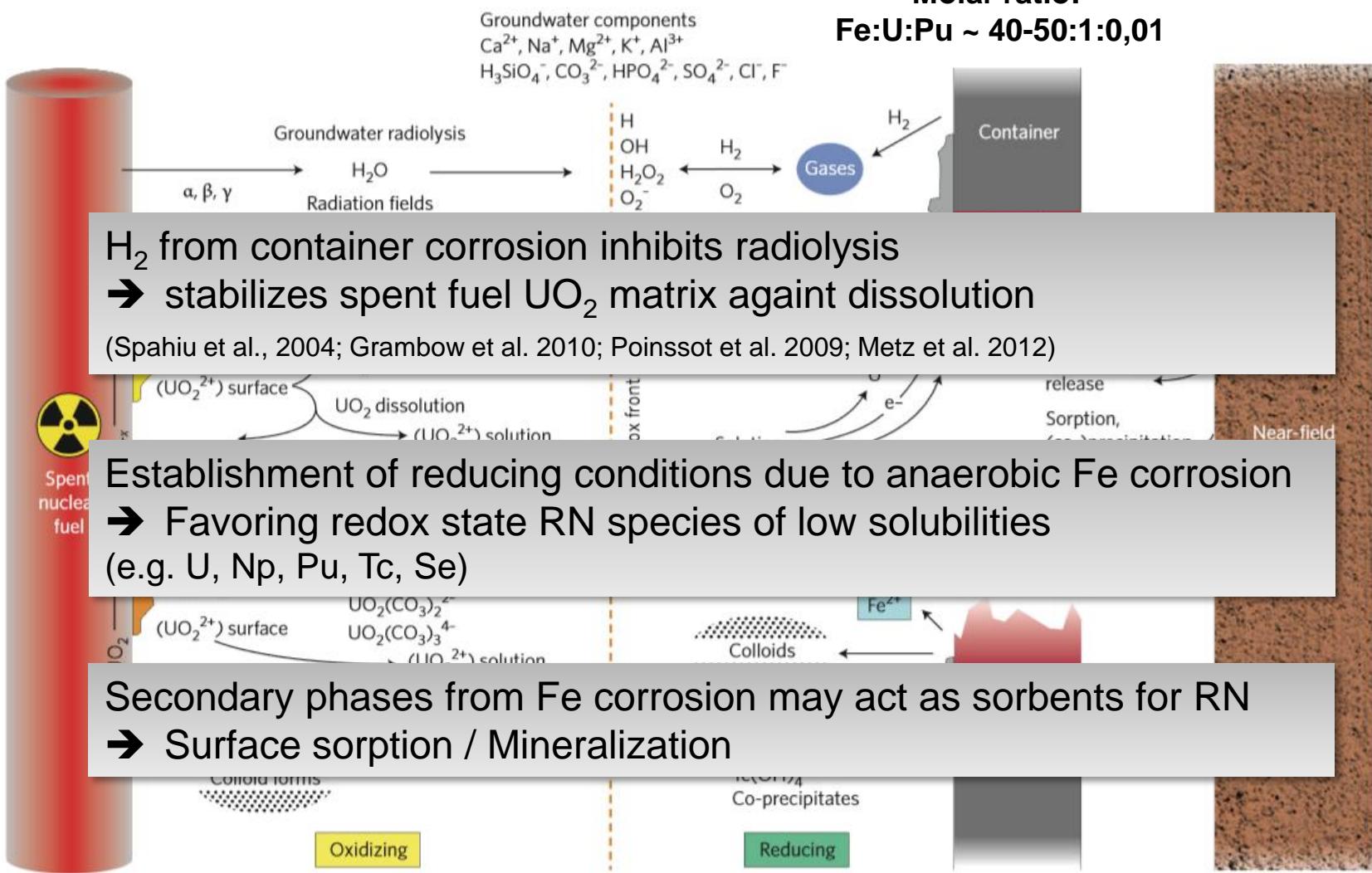
Solid/liquid interactions

- Sorption to mineral surfaces
- solid-solution formation
- Colloid formation



Processes in the near field of a repository for spent nuclear fuel

Molar ratio:
Fe:U:Pu ~ 40-50:1:0,01

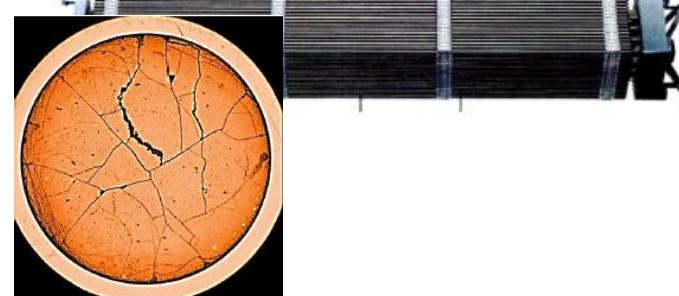
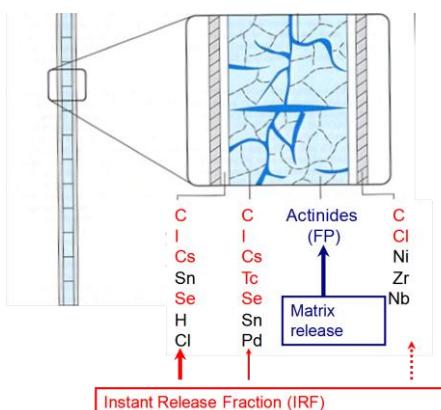
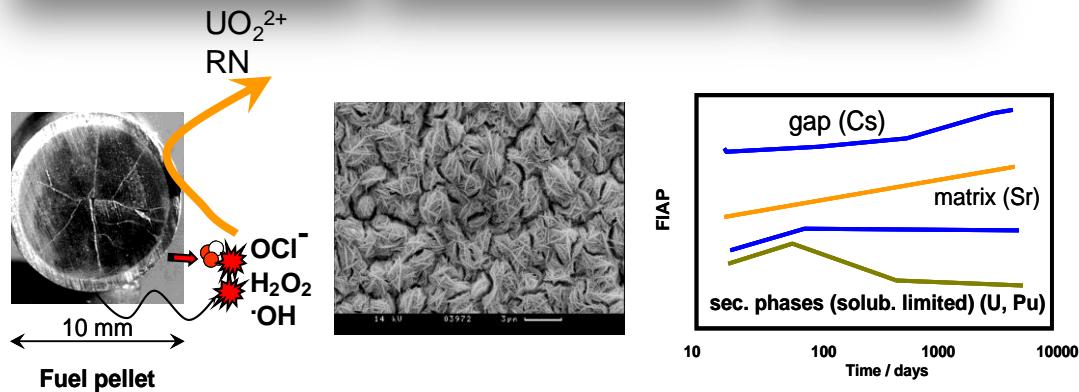


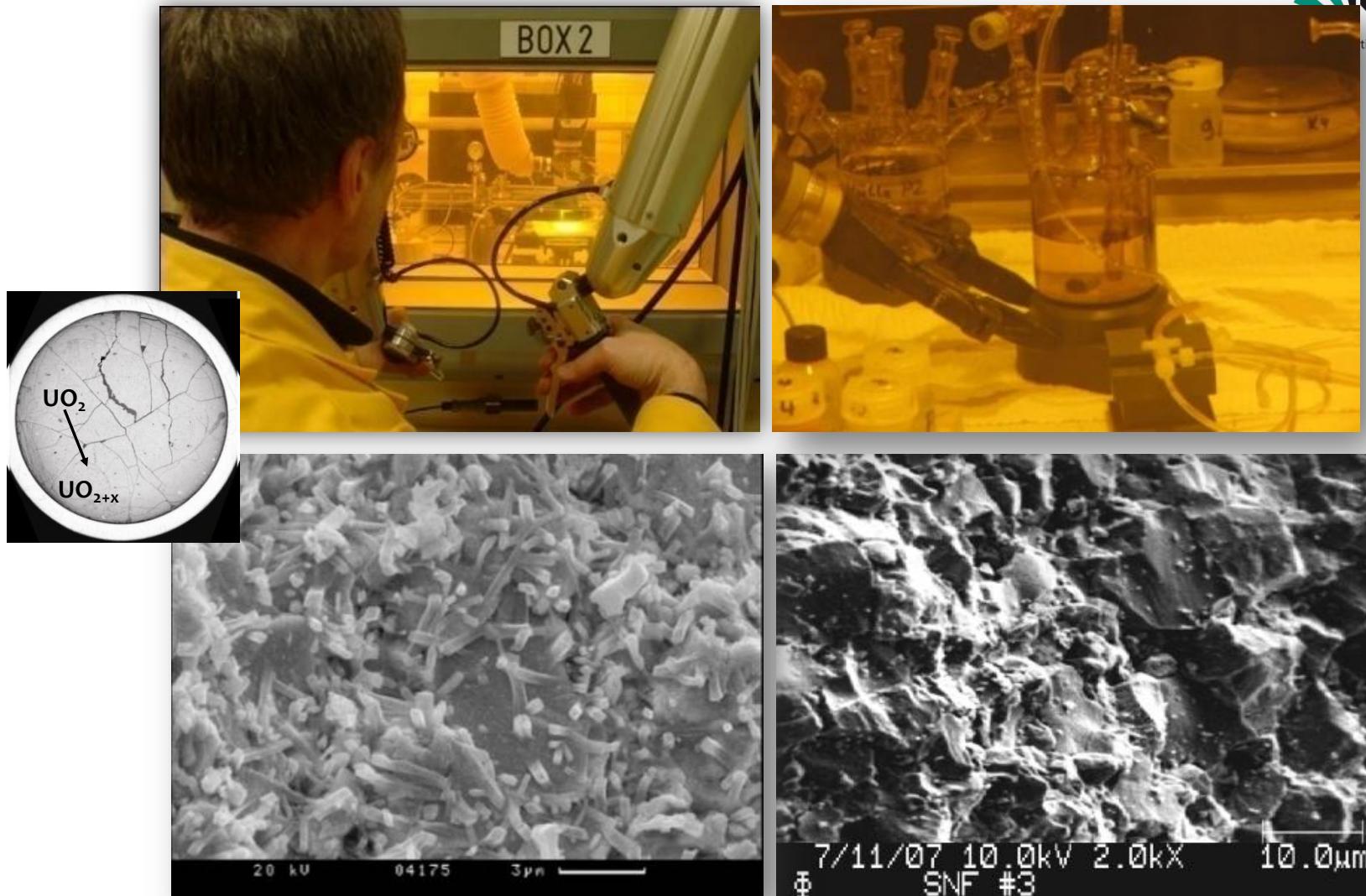
R. Ewing, Nature Mat., 2015

The role of the waste form:

- Spent nuclear fuel -

- Impact of radiolysis on fuel behavior in a repository
- Long term stability of fuel in the repository („H₂-effect“)
- „Instant release“ fraction
- New issue: degradation behavior of heat generating radioactive waste under extended interim storage conditions?





Anoxic corrosion of UOX fuel

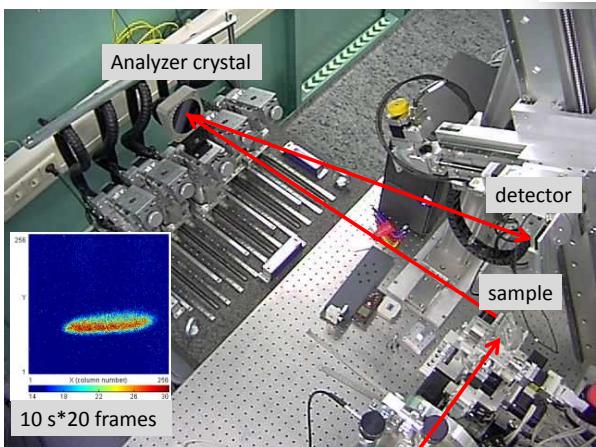
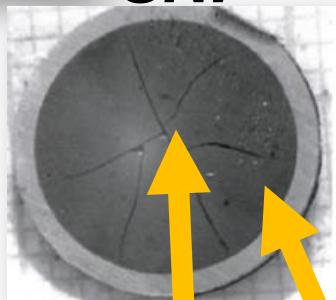
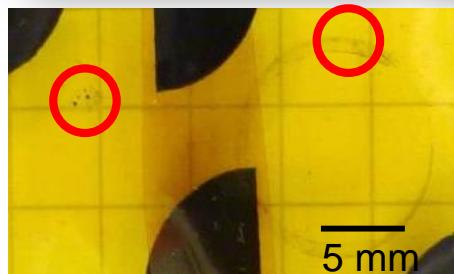
Anoxic corrosion of UOX fuel in presence of H₂

Source: Loida et al. (2007) MRS SBfNWM 30, vol. 985, 14-20; Metz et al. RCA, 2012

HR-XANES analysis of Uranium oxidation state in spent UO₂ fuel



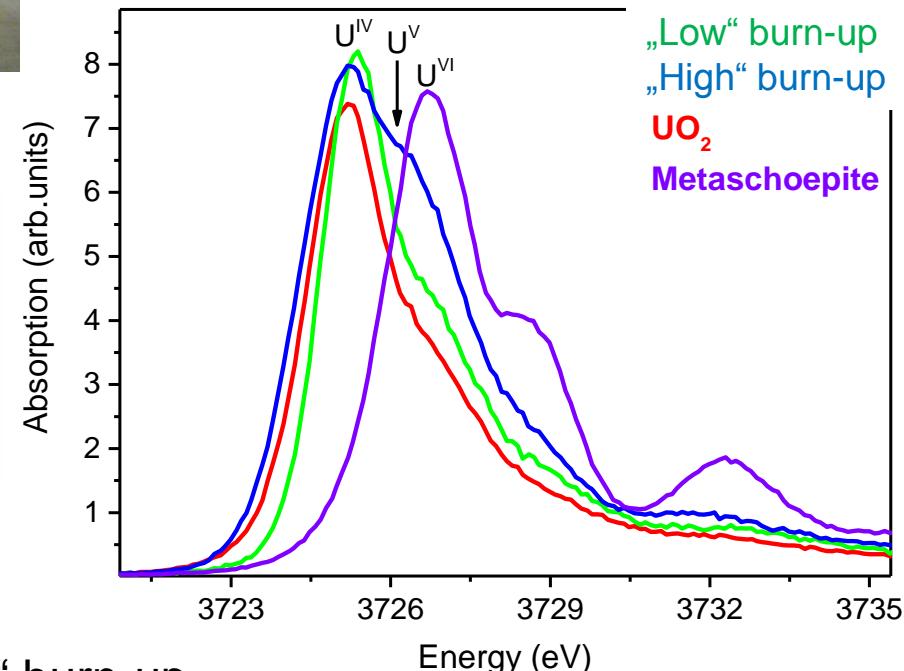
SNF



HR-XAS spectrometer at INE-Beamline

X-ray beam

S. Bahl et al., to be submitted to Environ. Sci. Technol.

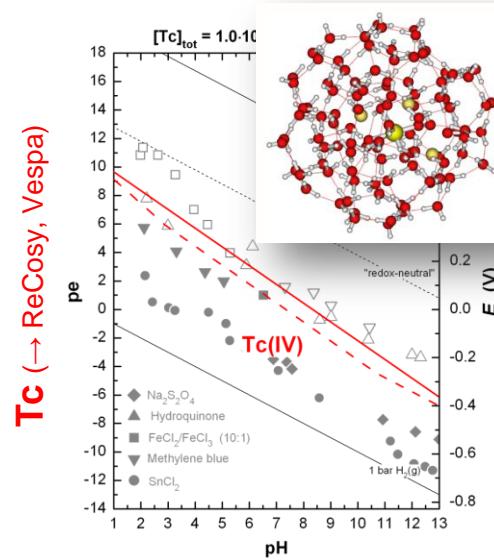
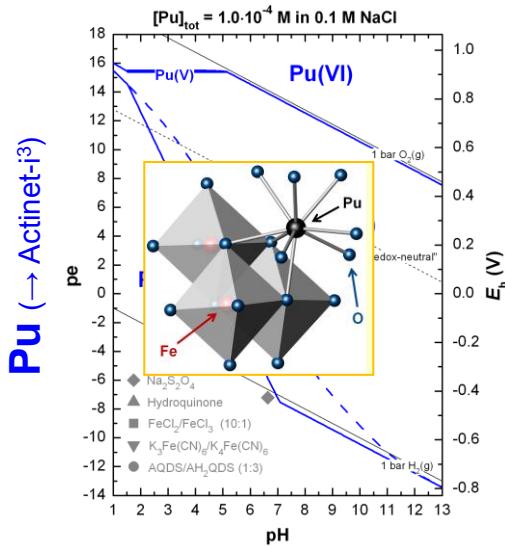


„High“ burn-up
rim
„Low“ burn-up
center

1st time:
Detection of U(V) in spent fuel

Fundamental research

Example: redox chemistry of actinides/fission products



mechanistic understanding, thermodynamics, kinetics
(coop.: Kyoto University, KAIST, LANL - Carlsbad office)

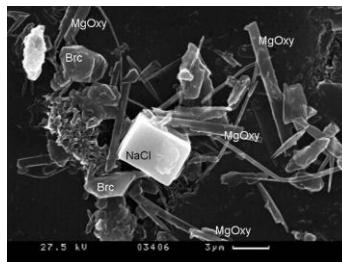


Applied research

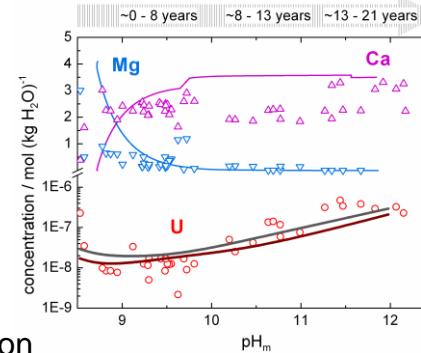
Example: radionuclide retention in cementitious phases



1:1 scale experiments from the Asse mine

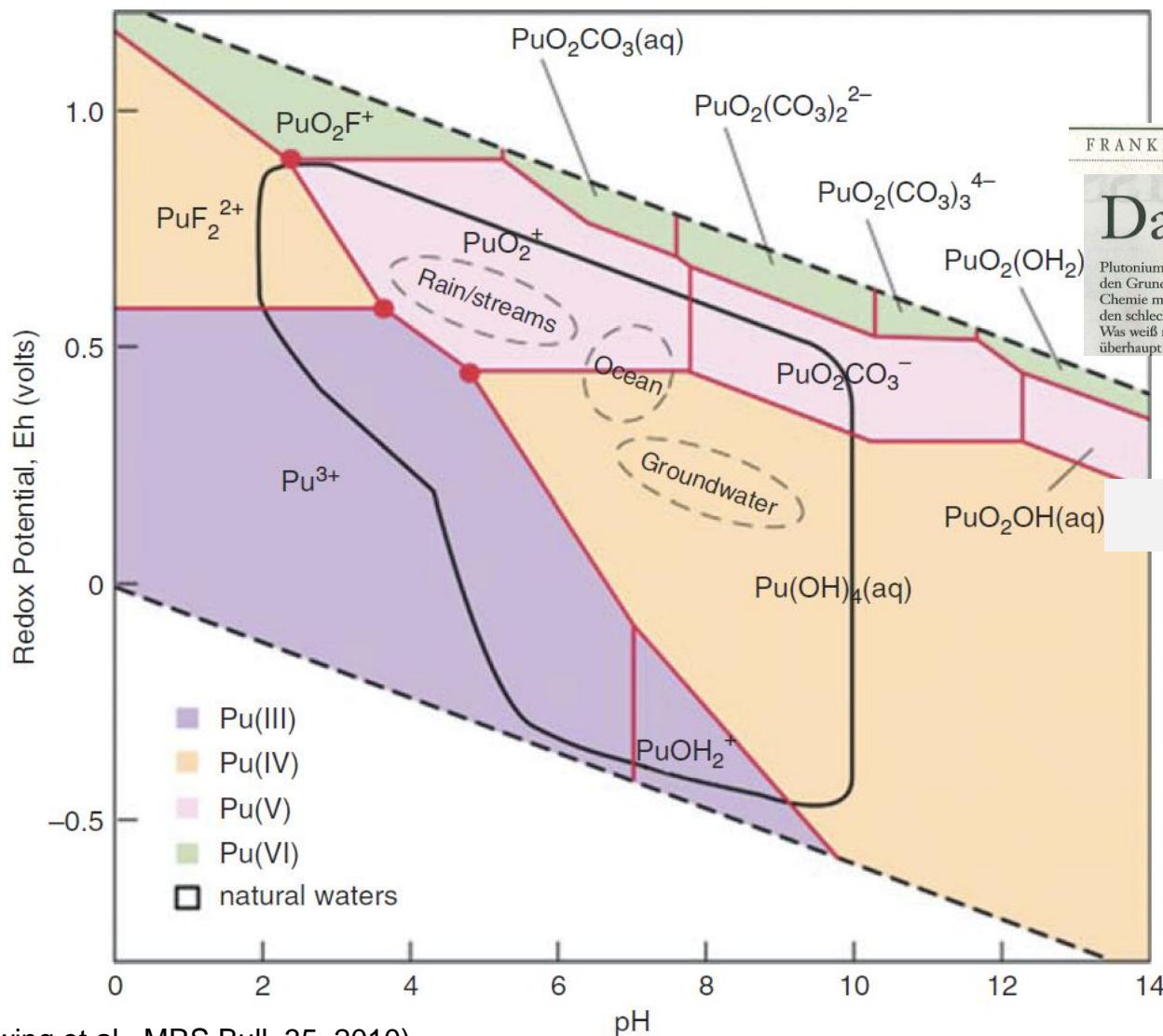


Secondary phase formation During cement corrosion



Geochemical description

Redox chemistry of Plutonium



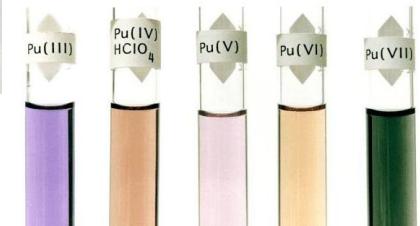
(Ewing et al., MRS Bull. 35, 2010)

Fellhauer, Altmaier, Fanghänel, 2010

FRANKFURTER ALLGEMEINE SONNTAGSZEITUNG

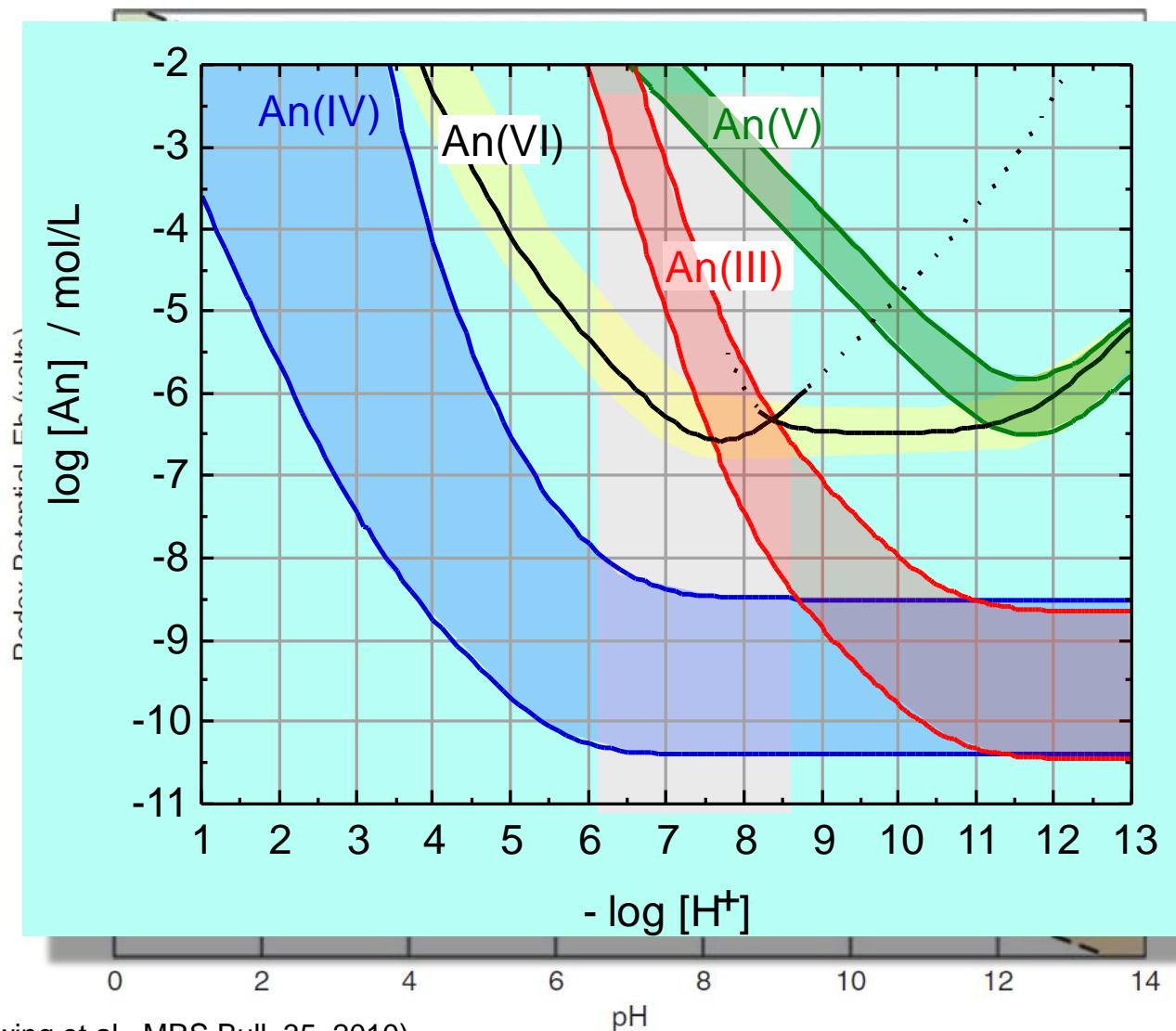
Das elementare Böse

Plutonium hat unter den Grundstoffen der Chemie mit Abstand den schlechtesten Ruf. Was weiß man überhaupt darüber?

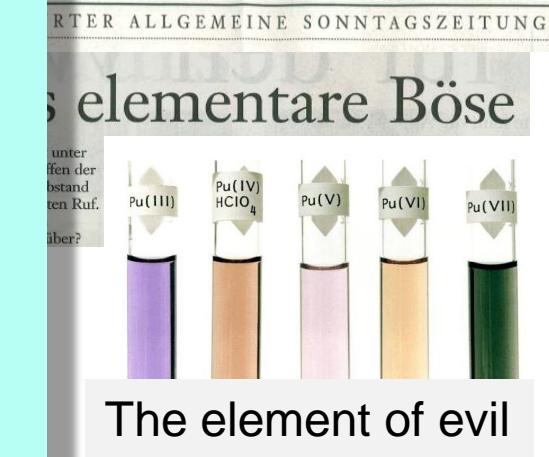


The element of evil

Redox chemistry of Plutonium



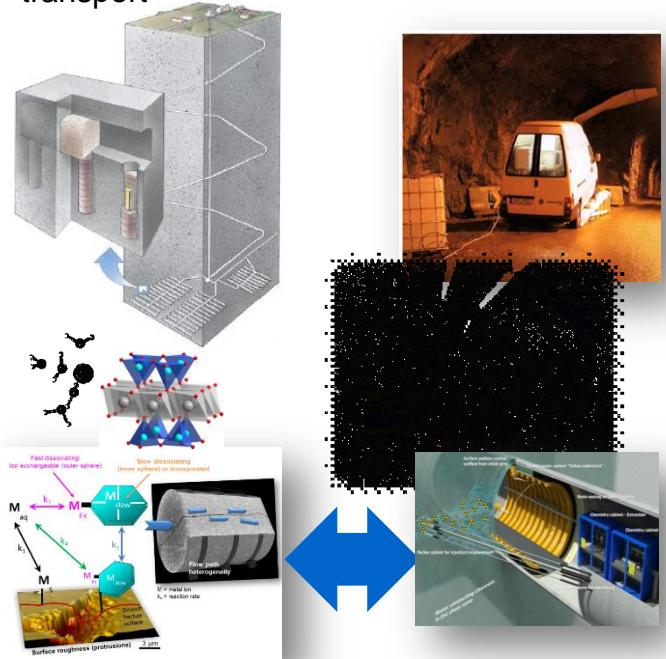
(Ewing et al., MRS Bull. 35, 2010)



Fellhauer, Altmaier, Fanghänel, 2010

Host rock: crystalline

Impact of glacial meltwater onto the erosion of bentonite barriers in a repository in granite and onto the clay nanoparticle borne radionuclide transport



Laboratory

underground rock laboratory

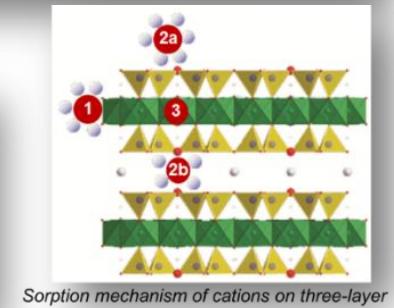
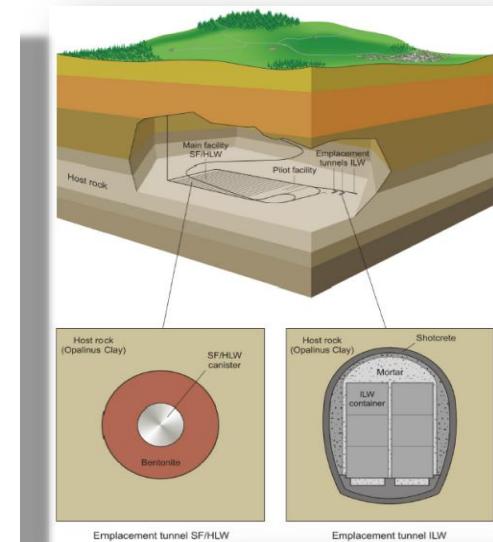
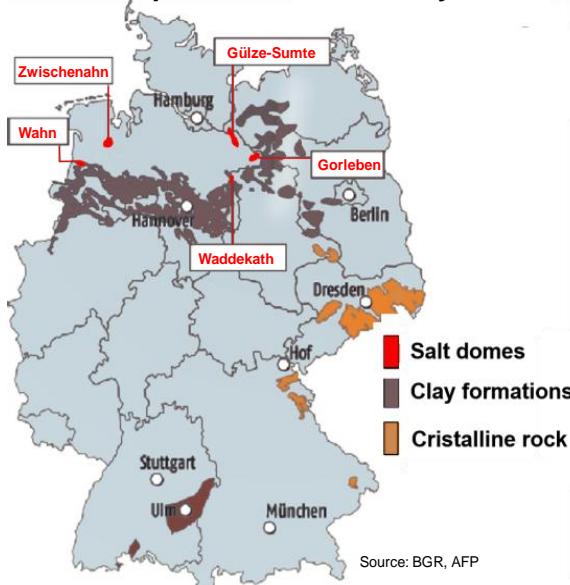


Host rock: clay rock

Diffusion and retention of radionuclides in clay rock: Impact of Pore water composition (I, pH, Eh, CO_3^{2-} , T...)



Potential host rocks for nuclear waste repositories in Germany



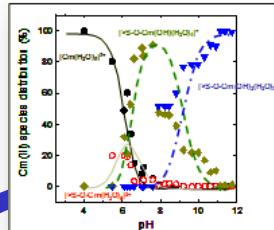
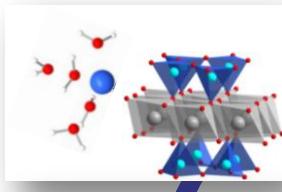
US-DOE



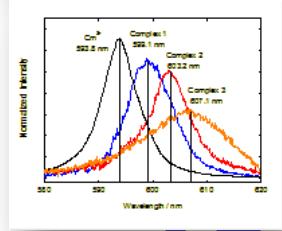
Comprehensive approach: from the nanoscale to the field

Example: Impact of glacial melt water induced erosion of bentonite barriers in a repository in crystalline rock

Geochemical
speciation,
Reaction kinetics

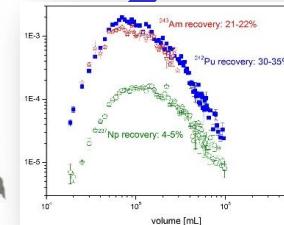
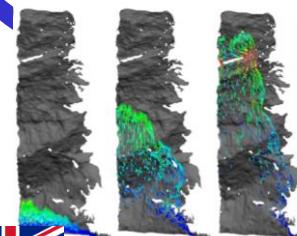


Spectroscopy,
Quantum
Chemistry



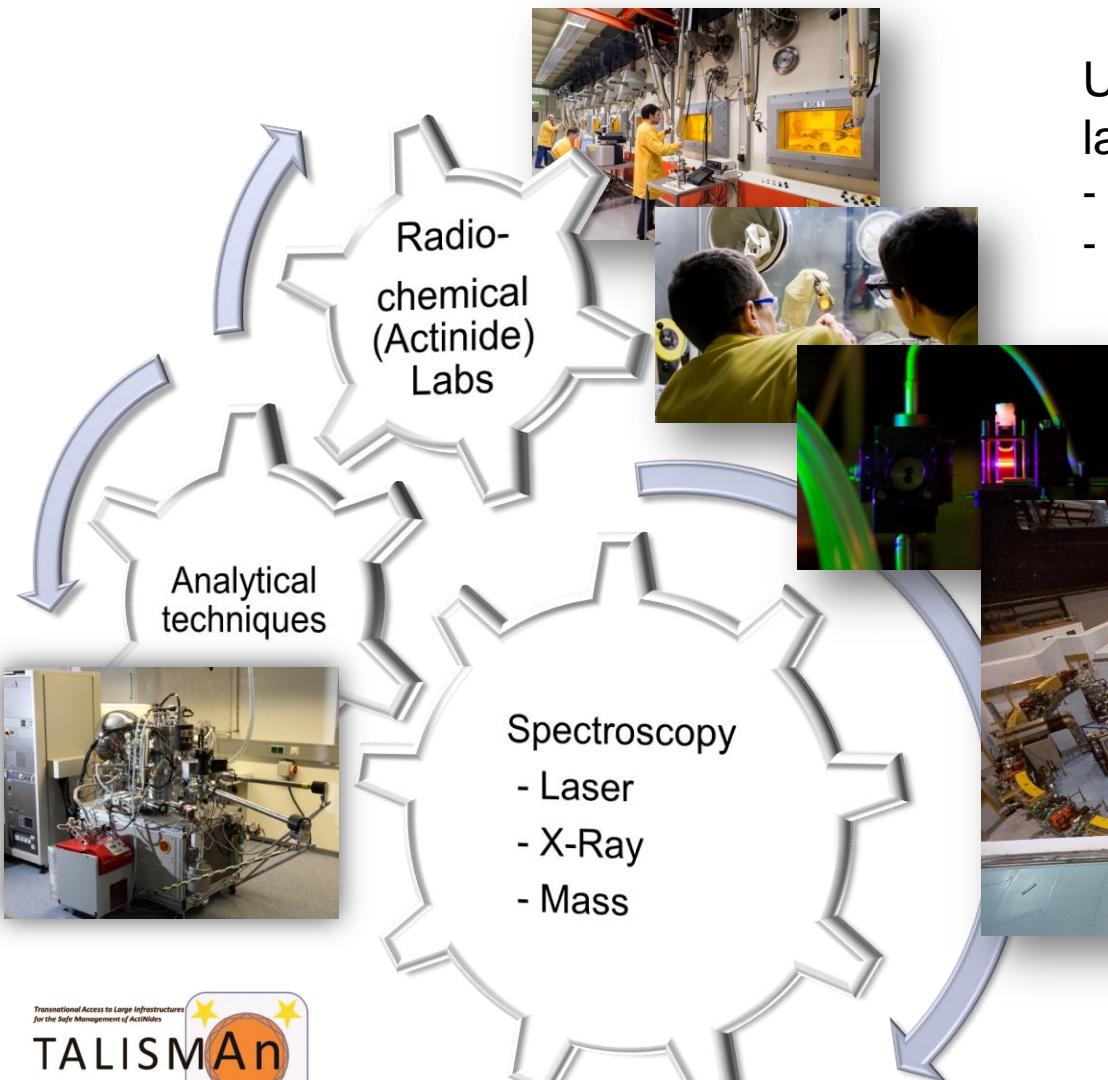
**Radionuclide
behaviour in the
geosphere**

In-situ experiments
(laboratory; underground rock
laboratories - Grimsel, Äspö)
→ upscaling



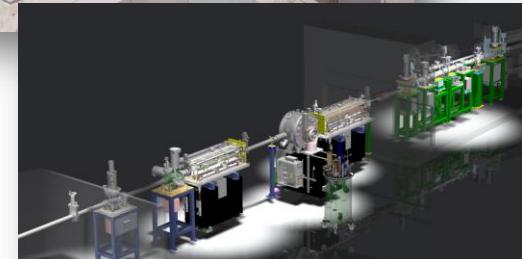
Simulation of radionuclide
migration by reactive
transport modeling

Laboratory platform

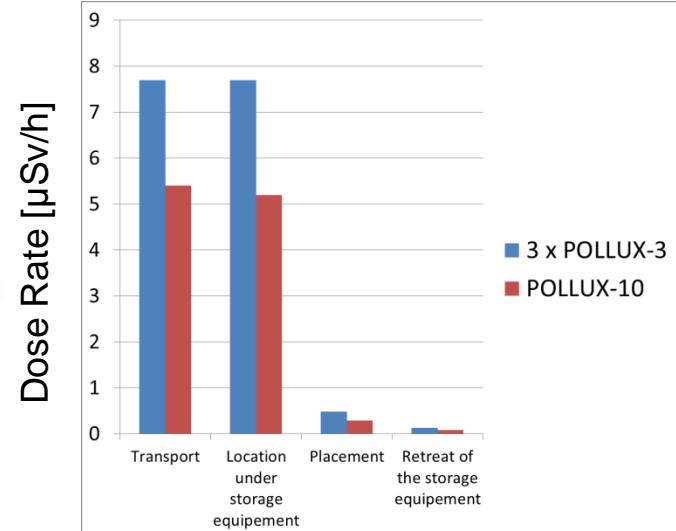
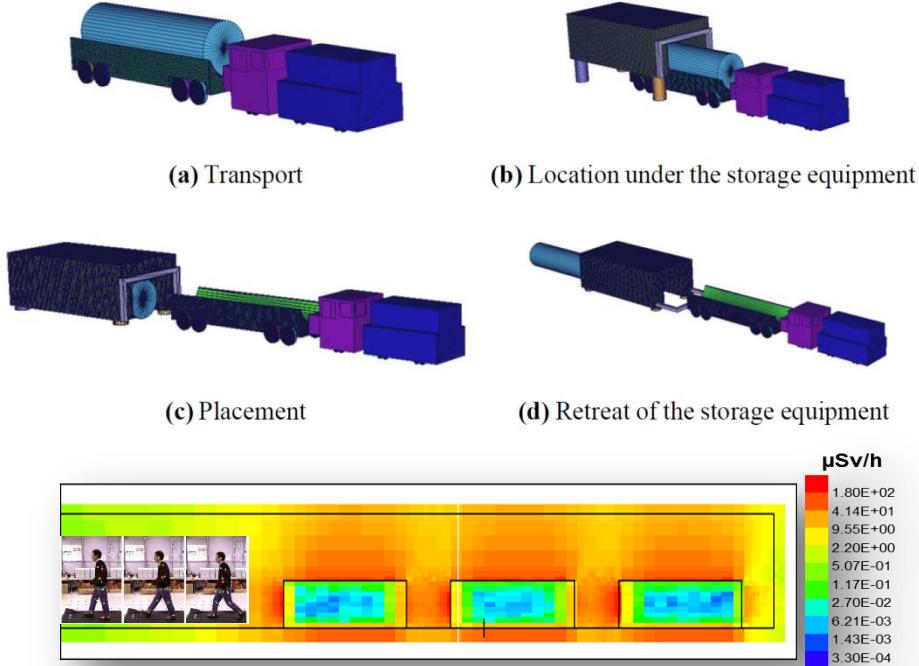


Unique combination of actinide laboratories and spectroscopies

- Fundamental science
- „European Actinide Pooled Facility“ Transnational Access Initiatives „Seed crystal“ for international cooperation (EU, USA, Asia)



Investigation of POLLUX storage/disposal scenarios (rock salt and clay stone)



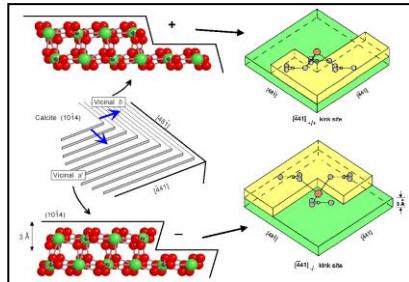
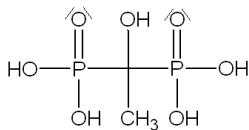
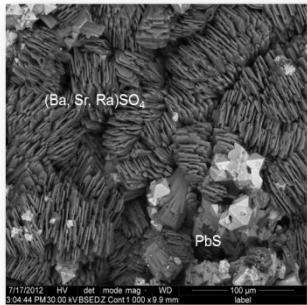
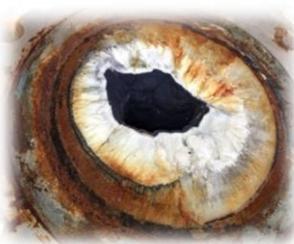
Comparison - a similar amount of SNF should be disposed:

- Germany: 90% UOX + 10% MOX
 - POLLUX-10 casks in **rock salt**
 - POLLUX-3 casks in **claystone (temperature constraint)**
- => 1 POLLUX-10 vs. 3 POLLUX-3**

NORM residues from geothermal power plants

Data: Scheiber et al. (2012)

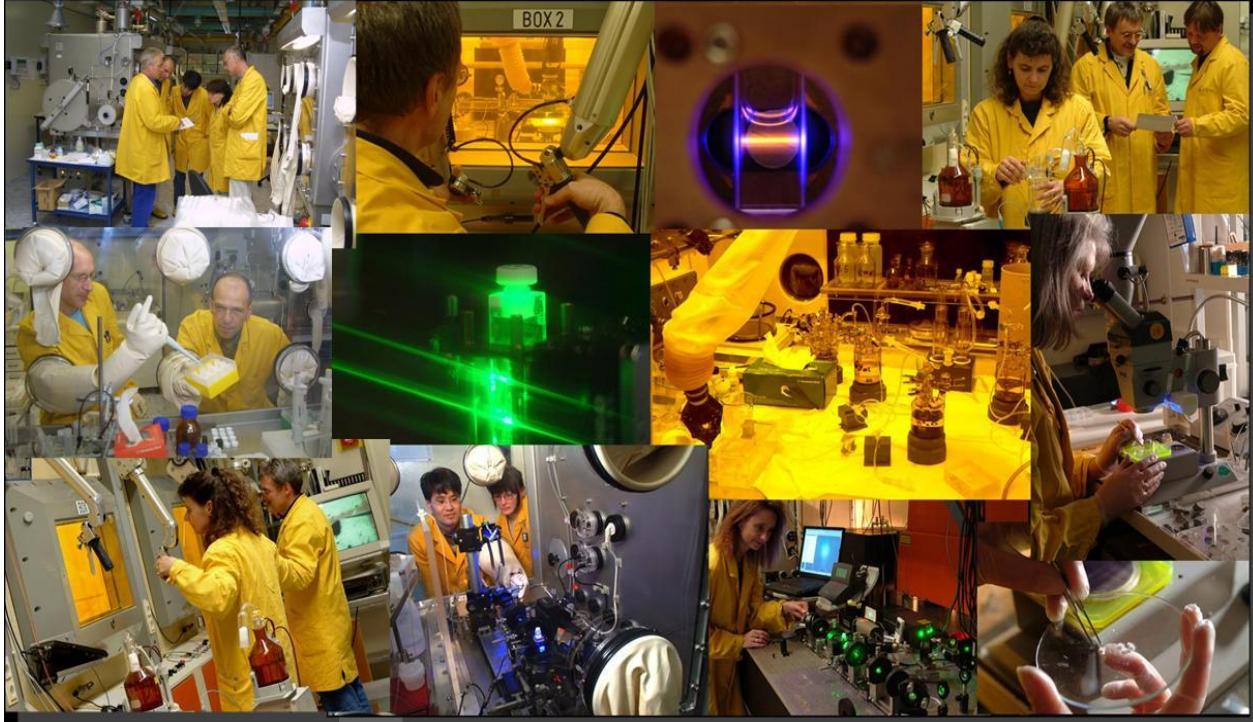
NORM residues C [Bq /g]	Dose rate [heat exchanger] (exemplarily)
up to 1,350 (Ra-226)	12 µSv/h in 1m distance
up to 1,100 (Pb-210)	> 34 µSv/h in 0.1m distance



HEDP as crystal growth inhibitor
(hydroxyethylidene-diphosphonate,
 $\text{C}_2\text{H}_8\text{O}_7\text{P}_2$):



- Formation of scalings particularly in the cold part of the facility ($\Delta T 90^\circ \text{ C}$)
- Primarily Ba/Sr-sulfate & PbS
- ^{226}Ra in Ba/Sr-Sulfate, ^{210}Pb PbS.
- **Development of Inhibitor concept to avoid scale formation**



INE:

- employees: ~ 105
- Radiochemical laboratories
- Hot cell laboratories
- Glove boxes
- „State of the Art“-instrumentation

• Research is embedded in European and International projects and cooperations

• INE is engaged in training and education of students

• Close cooperation with various universities