

Nuclear Isomers in and around various Magic Nuclei A. K. Jain

In collaboration with

Bhoomika and Swati

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James Thomason Building

Outline

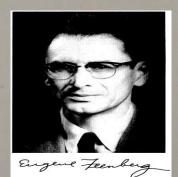


- Nuclear Isomers Feenberg's contribution
- Chart of Isomers & Global systematics related to pairing
- Seniority Isomers
- Key Features of Seniority isomers
- BE2 systematics of Z=82 isotopes and N=126 Isotones
- Extension to multi-j scheme from pure-j scheme
- Selection rules in Multi-j scheme and the new type of seniority isomers which decay by odd-tensor transitions
- BE2 puzzle in the 2+ states of Sn isotopes
- g-factors in the 2+ and 10+ states of Sn isotopes
- BE3 trends of the 3- states in Sn isotopes

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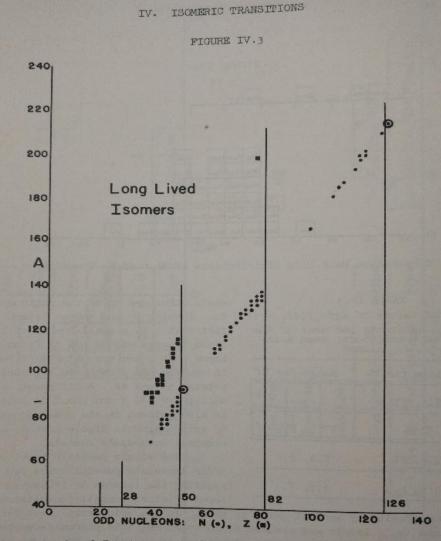
SHELL THEORY OF THE NUCLEUS

By EUGENE FEENBERG



PRINCETON PRINCETON UNIVERSITY PRESS

1955

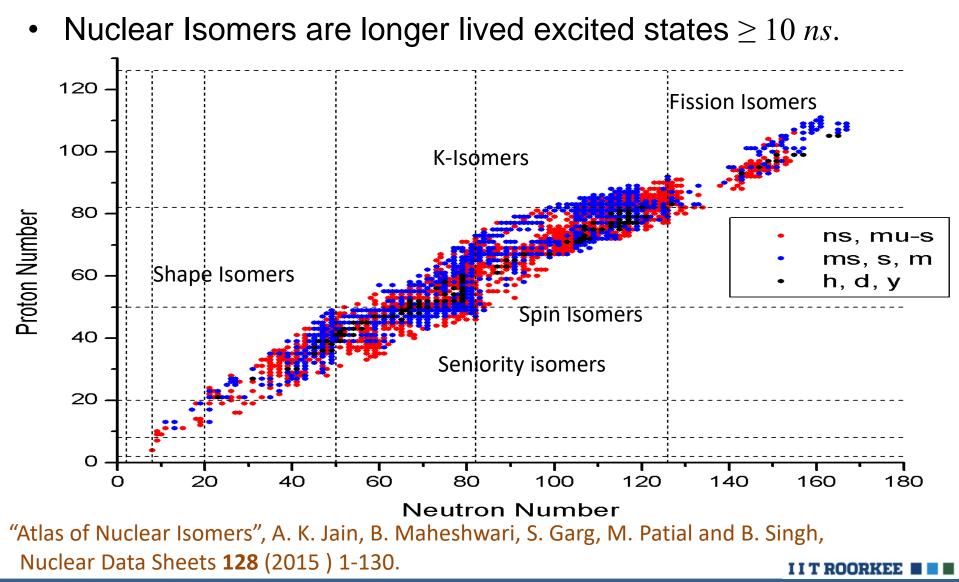


Islands of Isomerism - Distribution of Long-lived Nuclear Isomers of Odd Mass Number (Adapted from Goldhaber and Hill [52])

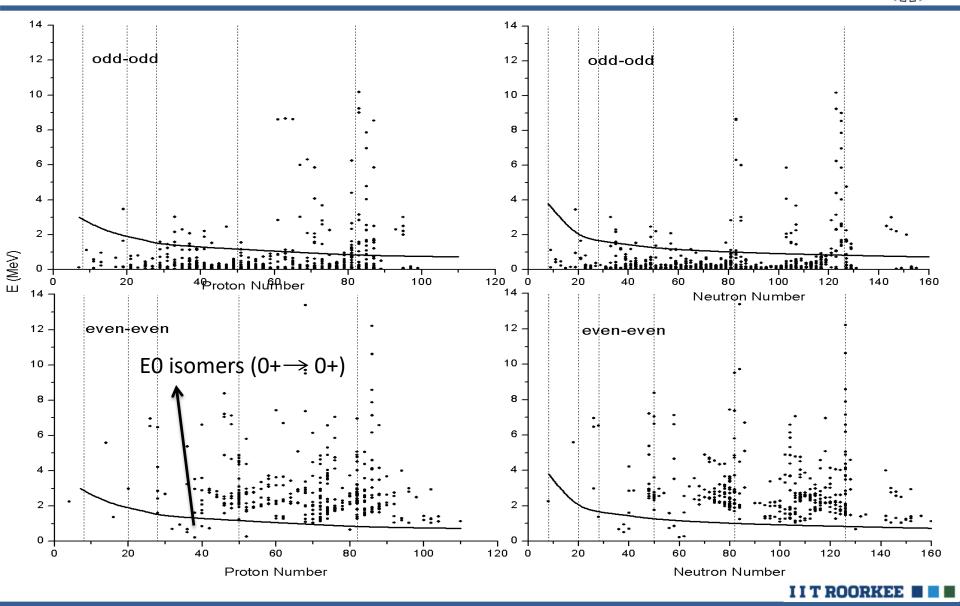
e circles at N = 51 $(_{42}Mo_{51})$ and N = 127 $(_{84}Po_{127})$ fall outside of the ecretical ranges for islands of isomerism. Goldhaber (53) interprets Mo_{51} isomerism in terms of transitions within the even-even core, the cleon taking no part in the internal rearrangements produced by the t ons.

Chart of Nuclear Isomers





Even-even & odd-odd isomers (Even-A)



Seniority and Seniority Isomers



- Seniority was introduced by Racah (1943) in atomic context
- Adopted for nuclei in a similar fashion.
- Seniority quantum number (v) may be defined as the number of unpaired nucleons.
- A set of states diagonalizable by a short range pairing interaction give rise to good or, nearly good seniority states
- Specific selection rules emerge, which may lead to isomeric states in single closed shell nuclei

Ref.: Books of Talmi, Lawson, Griener and Maruhn, and Casten carry an excellent technical/general description of the seniority quantum number.

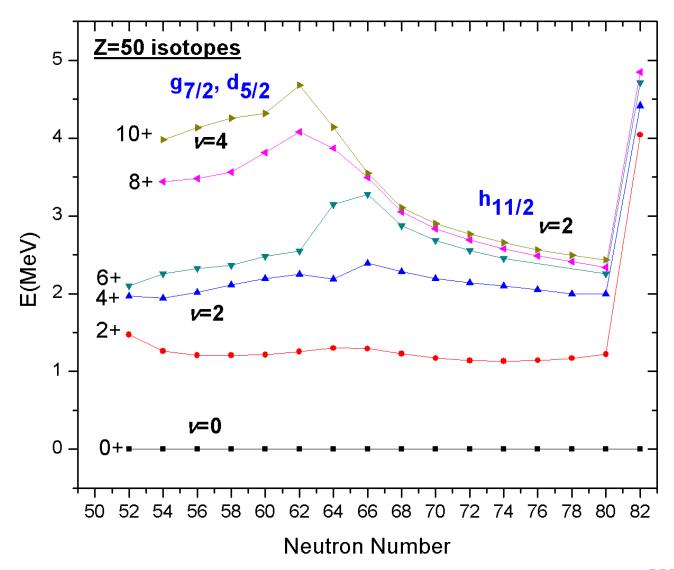
Key features



- Particle number independent energy
- Constant pairing gap
- Specific selection rules and parabolic behavior of transition probabilities
- Any interaction between identical fermions in single-*j* shell exactly conserves seniority if *j*≤7/2.
- The seniority is nearly conserved up to j=11/2 in Sn-isomers after the mid-shell, where small seniority mixing may take place, but still the features of seniority persist.

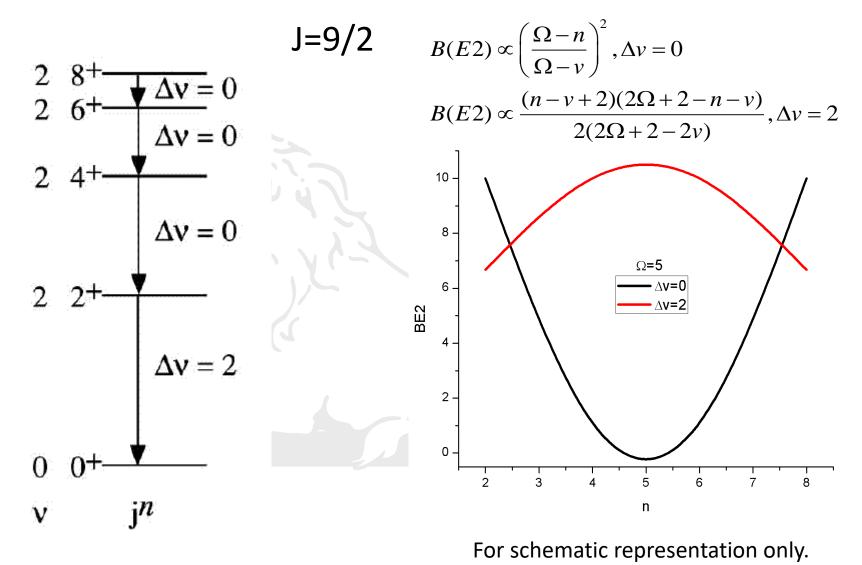
Particle number independence of energy





BE2 variation in single-j: Schematic only

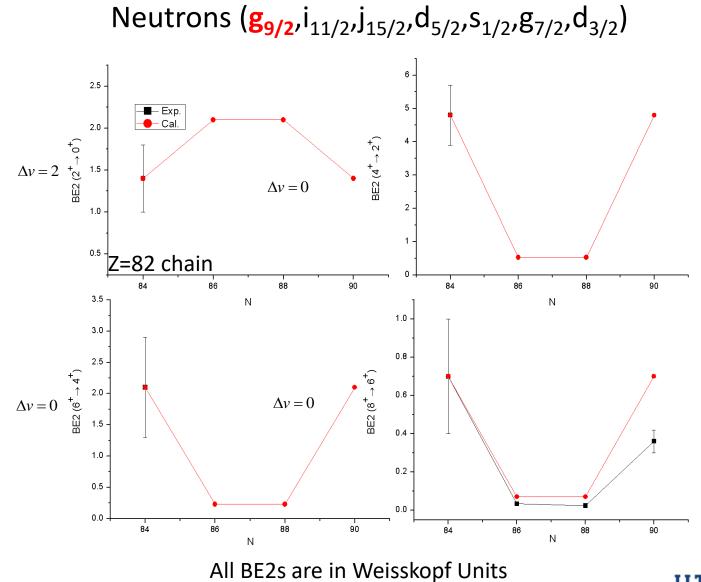




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B(E2)s in the even-even Z=82, 8+ isomers

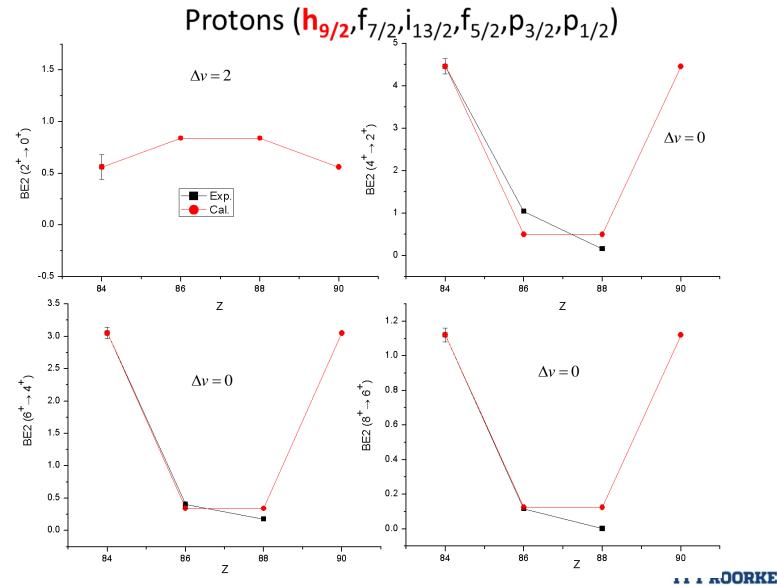




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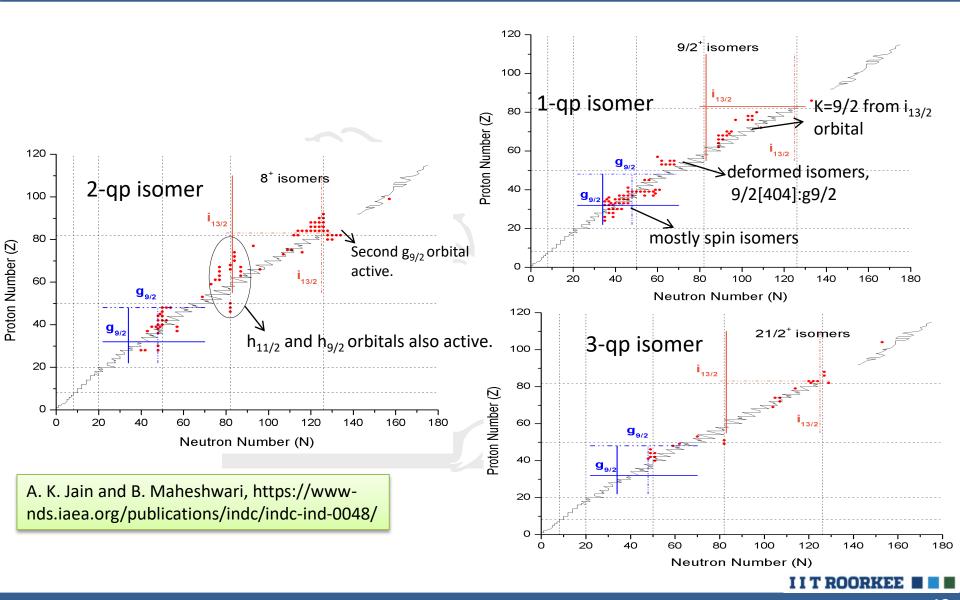
B(E2)s in the even-even N=126 chain, 8+ isomers





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g_{9/2}: 9/2⁺, 8⁺ and 21/2⁺ isomers





B(E2) values from Seniority scheme



$$B(EL) = \frac{1}{2J_{i}+1} \left| \left(J_{f} \left\| \sum_{i} r_{i}^{L} Y^{(L)}(\theta_{i},\phi_{i} \right\| J_{i} \right) \right|^{2}$$

In single-j case,
For L even

$$\left\langle u = \frac{1}{2}(2j+1) \right\rangle$$

seniority

$$\left\langle j^{n} v l J_{f} \left\| \sum_{i} r_{i}^{L} Y^{(L)}(\theta_{i},\phi_{i}) \right\| j^{n} v l \cdot J_{i} \right\rangle = \left(\frac{\Omega - n}{\Omega - v} \right) \left\langle j^{v} v l J_{f} \left\| \sum_{i} r_{i}^{L} Y^{(L)}(\theta_{i},\phi_{i}) \right\| j^{v} v l \cdot J_{i} \right\rangle$$

$$\left\langle j^{n} v l J_{f} \left\| \sum_{i} r_{i}^{L} Y^{(L)}(\theta_{i},\phi_{i}) \right\| j^{n}, v \mp 2, l' J_{i} \right\rangle = \sqrt{\frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}} \left\langle j^{v} v l J_{f} \left\| \sum_{i} r_{i}^{L} Y^{(L)}(\theta_{i},\phi_{i}) \right\| j^{v}, v \mp 2, l' J_{i} \right\rangle$$

It is easy to generalize these results for multi-j case by defining,

$$\begin{split} \widetilde{j} &= j \otimes j' \dots \quad \Omega = \frac{1}{2} \sum_{j} (2j+1) \quad n = \sum_{j} n_{j} \\ B(E2) \propto \left(\frac{\Omega - n}{\Omega - v} \right)^{2}, \Delta v = 0 \\ B(E2) \propto \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \end{split}$$

$$B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v + 2)(2\Omega + 2 - n - v)}{2(2\Omega + 2 - 2v)}, \Delta v = 2 \\ B(E2) \approx \frac{(n - v$$

B. Maheshwari, A. K. Jain, Phys. Lett. B753, 122 (2016)

Selections rules (Extension to multi-j shell)

Single-j shell

- What seniority says.
 - Odd L -> quasi-spin scalar -> Δv=0
 - Even L -> quasi-spin vector -> Δv=0, 2
- What EM decay says.
 - Odd L -> only magnetic transitions.
 - Even L -> only electric transitions.

Multi-j shell

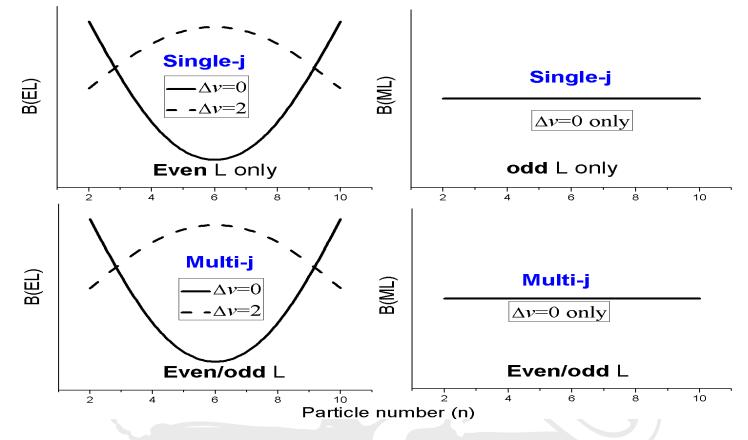
- What seniority says.
 - Odd/Even L -> Magnetic transitions -> quasi-spin scalar -> Δv=0
 - Odd/Even L -> Electric transitions -> quasi-spin vector -> Δv=0, 2
- What EM decay says.
 - Odd/Even L -> both electric/ magnetic transitions.

Odd L, Magnetic transitions, $\Delta v=0$ Even L, Electric Transitions, $\Delta v=0$, 2



Comparison of single-j and multi-j shell





B. Maheshwari and A. K. Jain, Phys. Lett. B **753**, 122 (2016).

B. Maheshwari, A. K. Jain and B. Singh, Nucl. Phys. A 952, 62 (2016).

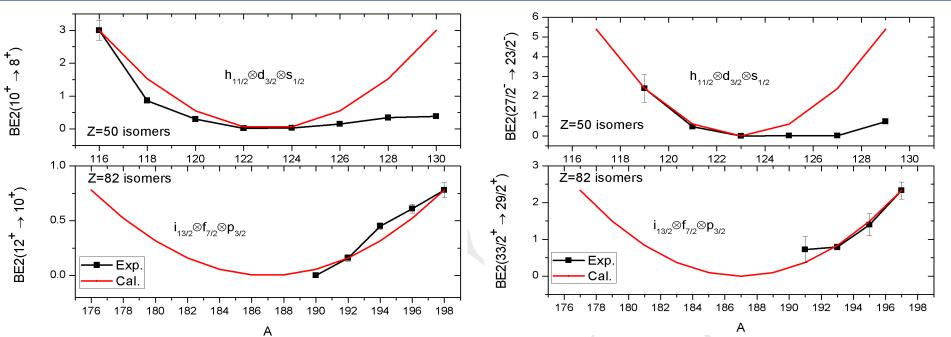
A.K. Jain and B. Maheshwari, Nuclear Physics Review 34, 73 (2017).

A.K. Jain and B. Maheshwari, Physica Scripta 92, 074004 (2017).

B. Maheshwari, Swati and A. K. Jain, Pramana-Journal of Physics 89, 75 (2017).

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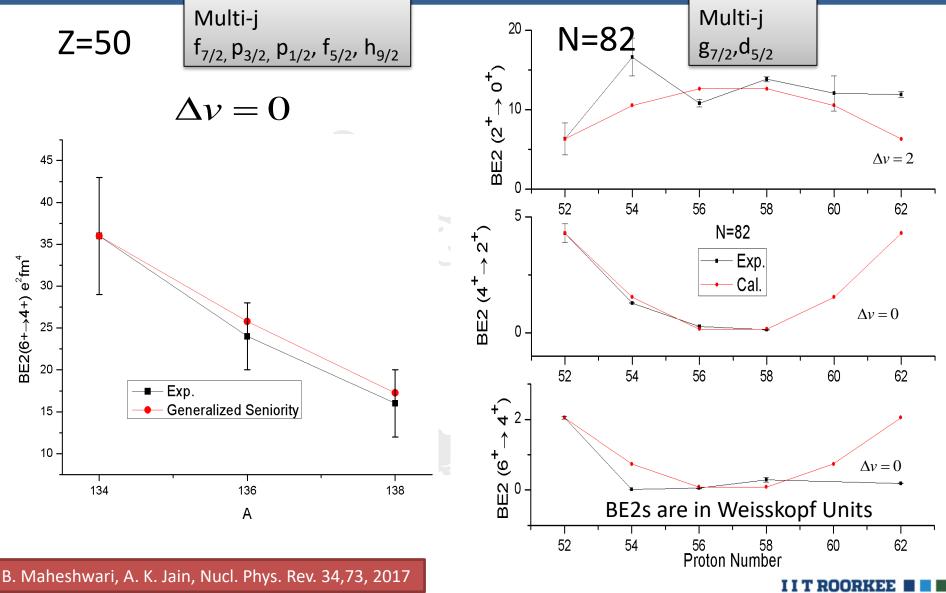
BE2s of High-spin isomers in Sn and Pb isotopes



- This concludes that the 12+, 10+, 33/2+ and 27/2+ isomers, arising mainly from the intruder orbitals, also require the configuration mixing to explain the measured data.
- The right hand sides of Z=50 isomers which do not follow this generalized seniority scheme, may further need the inclusion of the non-degenerate multi-j orbits.

B(E2)s from generalized seniority in Z=50 & N=82 chains

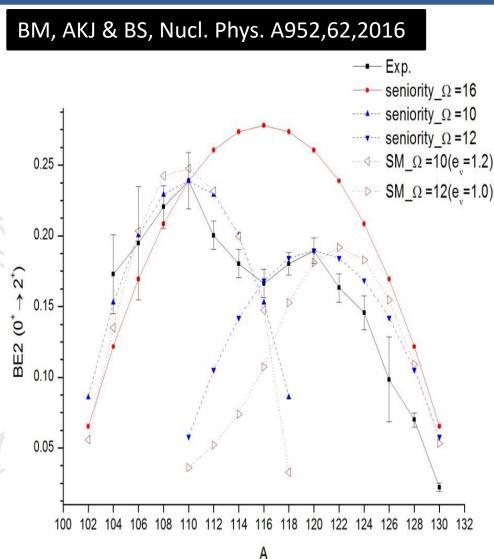




The first excited 2+ states in Sn isotopes

- Expected: a single parabolic trend coming from the generalized seniority.
- Reality: A minima in the middle of the valence space in recent measurements.
- Reason: The different sets of orbits involved before and after midshell, as also discussed by Morales et al.

I. O. Morales, P. Van Isacker and I. Talmi, Phys. Lett. B **703**, 606-608 (2011).

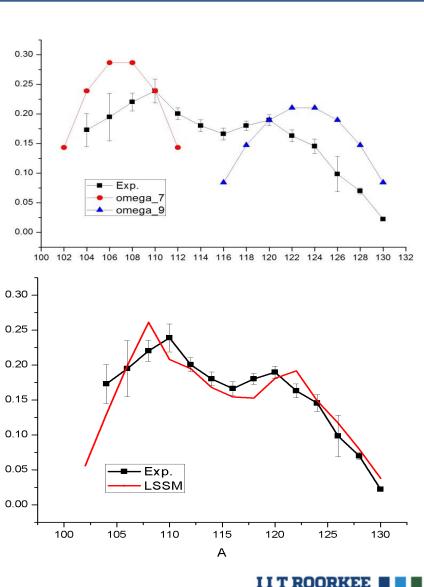




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Seniority inspired LSSM

- Failure: if we use omega 7 and 9, which comes from the g_{7/2}, d_{5/2} and h_{11/2}, d_{3/2}, s_{1/2} configuration mixing sets.
- Success: Seniority inspired and truncated LSSM explains the measured data quite well, particularly around the middle, where dimensions become quite large.
- ^{102-108,124-130}Sn have been treated in open space.

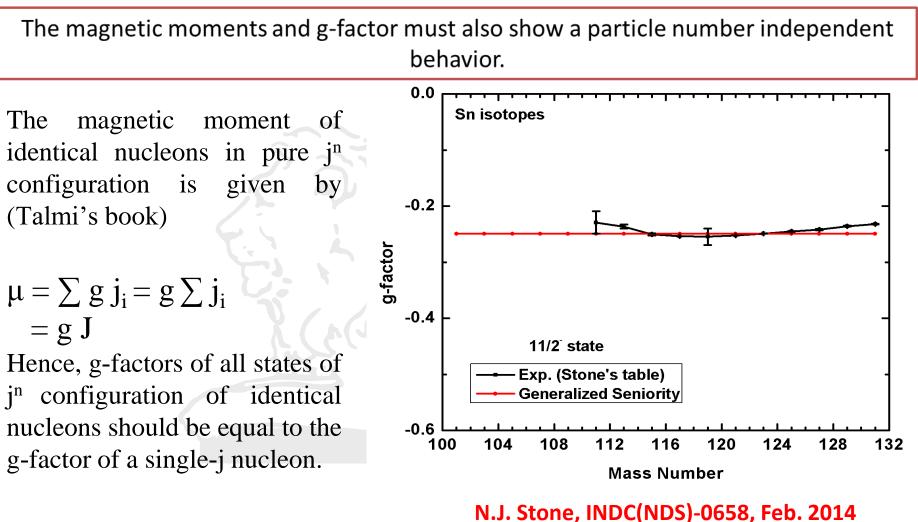


BE2 $(0^+ \rightarrow 2^+)$



g-factor





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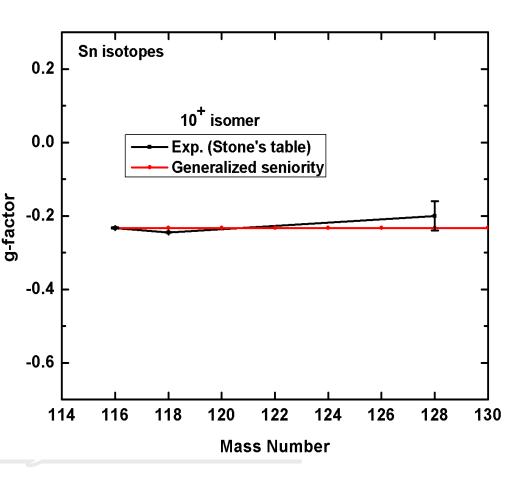
g-factor trend for 10⁺ state in Sn isotopes

The magnetic moment of identical nucleons in the mixed configuration \tilde{j}^n is given by

$$\mu = g \sum_{i=1}^{n} \widetilde{j} = gJ$$

Hence, g-factors of all 10^+ states of configuration \tilde{j}^n of identical nucleons should be equal to the g-factor of a single nucleon.

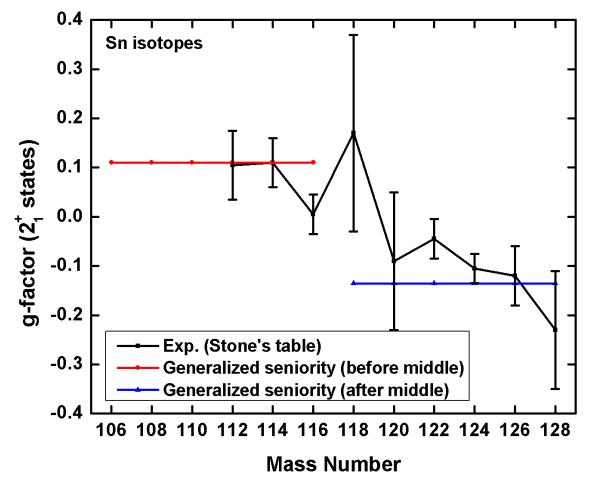
However, the data on $27/2^{-1}$ isomer is not available till date. We expect and predict the g-factor of this state to be of the same order as for 10^{+} state, since both follow same configuration mixing as shown in B(E2) results.



Points out the need of experiments in this direction!



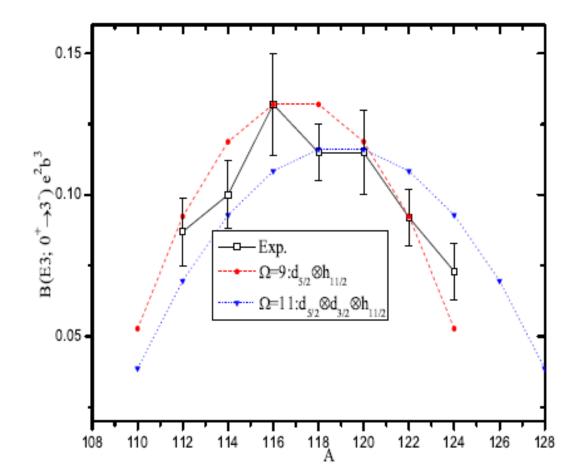




Two parabolas in B(E2) curve highlights the difference in the configuration before and after the middle for the generation of the 2+ states in Sn isotopes.

Therefore, g-factor before and after the middle are expected to be different.





d5/2,h11/2 together explain the data very well.

Suggests an octupole character for the 3- states.

This interpretation is supported by the shell model calculations.

Pramana-J.Phys. 89, 75 (2017)

Summary



- We have presented a view of the seniority nuclear isomers in various semi-magic chains of isotopes.
- Their various properties and unique features have been studied.
- Examples from Z=50, Z=82, N=126 chains have been presented.
- Both seniority conserving and seniority changing transitions can be understood very well.
- Many predictions are possible based on this scheme.

Thank You