

Radioactive contamination of enriched $^{106,116}\text{CdWO}_4$ crystal scintillators

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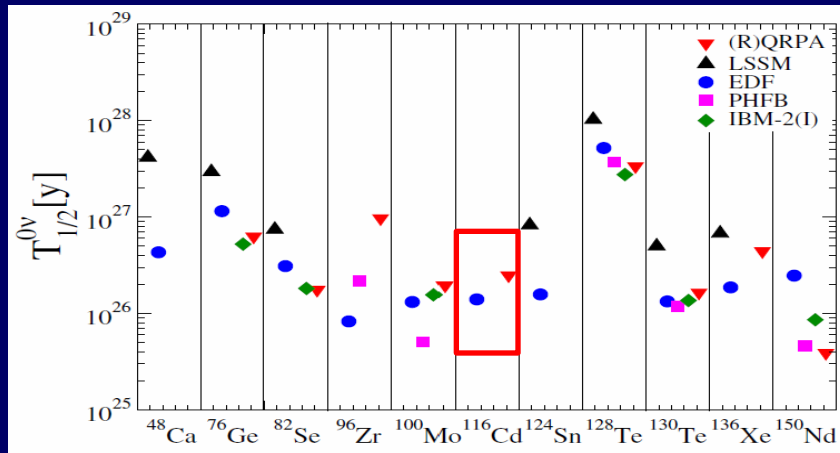
- Motivation
- Development of $^{106,116}\text{CdWO}_4$ crystal scintillators
- Experiments to search for 2β decay of ^{106}Cd and ^{116}Cd
- Radioactive contamination $^{106,116}\text{CdWO}_4$ scintillators
- Segregation of thorium, radium and potassium
- Conclusions

Motivation

^{116}Cd

One of the most promising isotopes to search for $0\nu 2\beta$ decay

- $Q_{2\beta} = 2813$ keV, $\delta = 7.5\%$
- promising theoretical calculation
- the possibility of isotopic enrichment in large amount

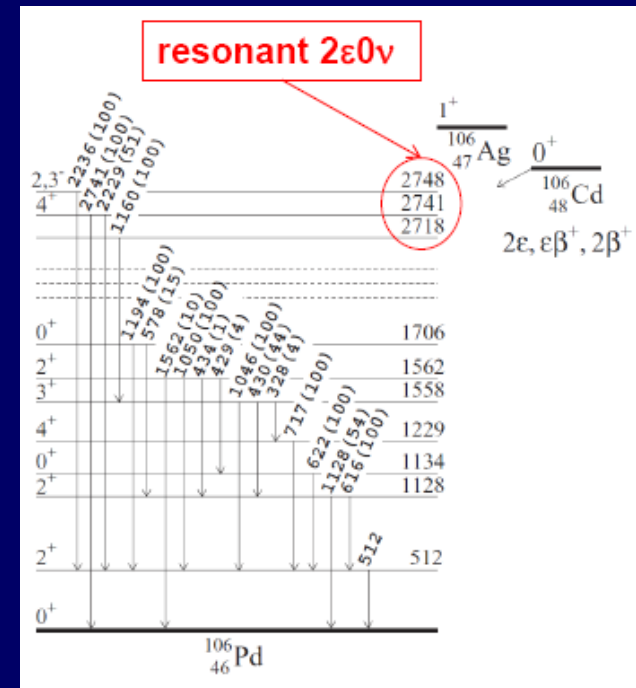


- good scintillation properties
- low levels of internal contamination
- particle discrimination ability to reduce background
- well established production

^{106}Cd

The one among six $2\beta^+$ isotopes (2ε , $\varepsilon\beta^+$, $2\beta^+$)

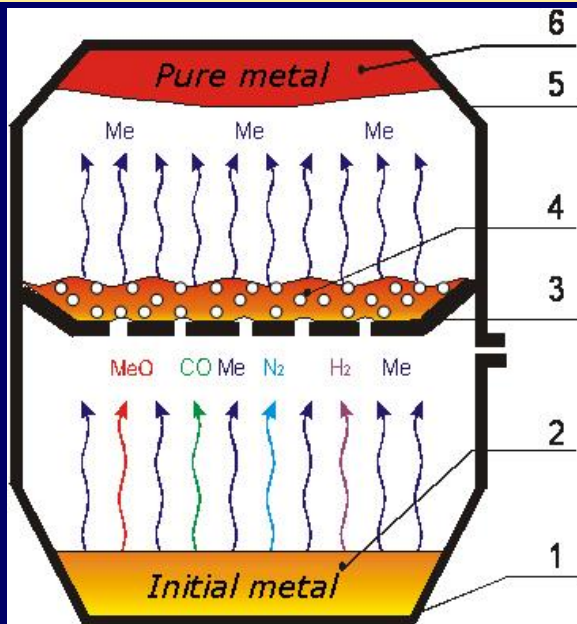
- $Q_{2\beta} = 2775$ keV, $\delta = 1.3\%$
- possibility to distinguish mass and RCH mechanism
- resonant $2\varepsilon 0\nu$ decay on the excited levels of ^{106}Pd



Production of $^{106,116}\text{CdWO}_4$

Purification of ^{106}Cd and ^{116}Cd metal samples

Distillation through getter filters



1 – crucible; 2 – initial metal; 3 – plate with holes; 4 – getter; 5 – condenser; 6 – purified metal

Concentration of impurities in ^{106}Cd (ppm)

Element	Before	After
K	11	0.04
Ni	0.6	< 0.2
Cu	5	0.5
Fe	1.3	0.4
Mg	12	<0.05
Mn	0.1	0.1
Cr	9	<0.1
Pb	270	<0.3

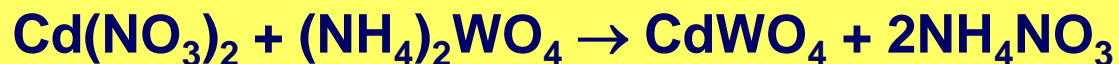
result →

R.Bernabey *et al.*, Metallofiz. Nov. Tekhn. 30 (2008) 477
G.P.Kovtun *et al.*, Functional Materials 18 (2011) 121

Production of $^{106,116}\text{CdWO}_4$

Synthesis of $^{106,116}\text{CdWO}_4$ compounds

After dissolving the metallic cadmium in nitric acid, the purification was realized by co-precipitation on a collector. Solutions of cadmium nitrate and ammonium para-tungstate were mixed and then heated to precipitate cadmium tungstate:



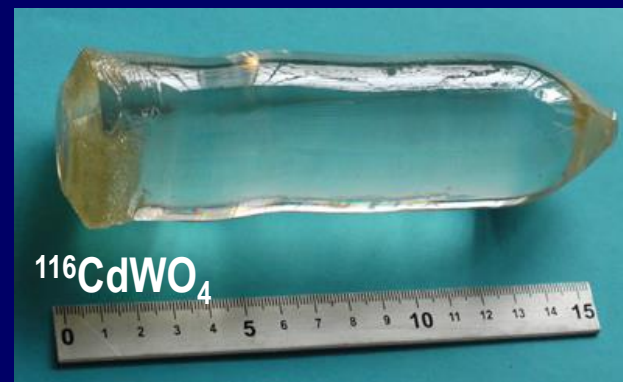
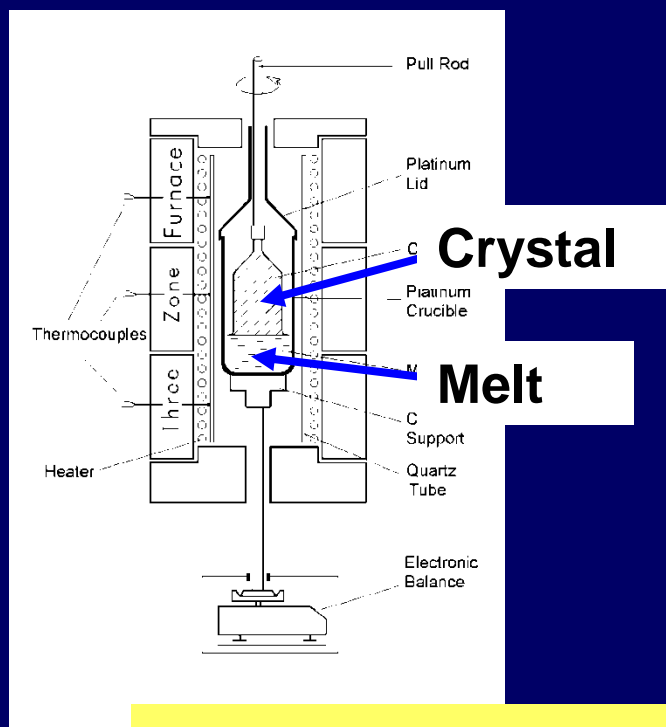
- All the operations were carried out by using quartz or polypropylene lab-ware, materials with low level of radioactive contaminations
- Reagents of high purity grade (concentration of any metal less than 0.01 ppm)
- Water, acids and ammonia were additionally distilled by laminar evaporation in quartz installation
- Additional recrystallization was performed to purify ammonium para-tungstate

P. Belli *et al.*, NIMA 615 (2010) 301

A. Barabash *et al.*, JINST 6 (2011) P08011

Production of $^{106,116}\text{CdWO}_4$

Crystal growth by Low-Thermal-Gradient Czochralski technique



	<u>standard</u>	<u>LTG-C</u>
Output	25-30%	<u>up to 90%</u>
Quality		<u>typically higher</u>
Radiopurity		<u>expected better</u>
Losses of powder	2-3%	<u><0.3%</u>

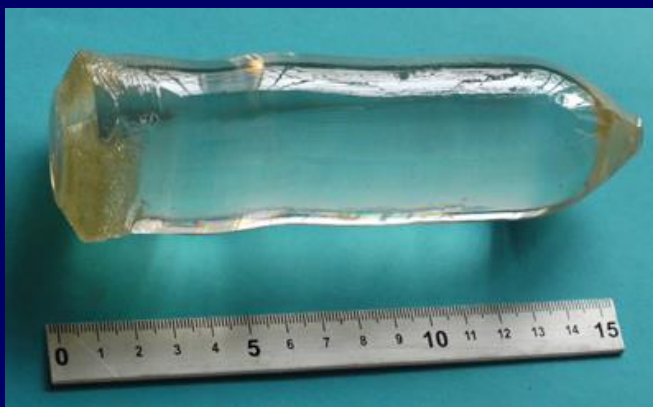
[1] A.A. Pavlyuk *et al.*, Proc. APSAM-92, April 26–29, Shanghai, China (1992)

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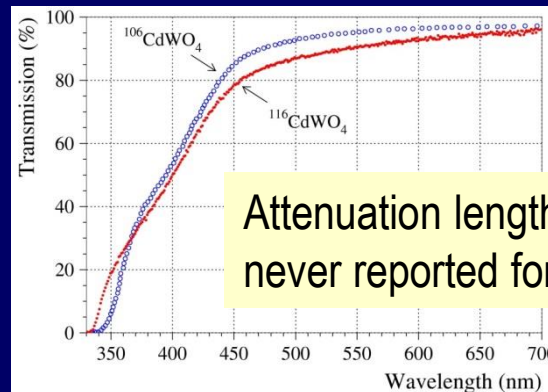
Optical properties and energy resolution



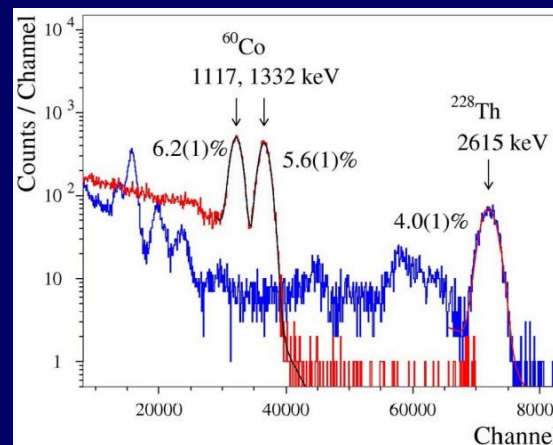
$^{106}\text{CdWO}_4$ 231 g (87%) [1]



$^{116}\text{CdWO}_4$ 1868 g (87%) [2]



Attenuation length \approx 50-60 cm
never reported for CdWO_4



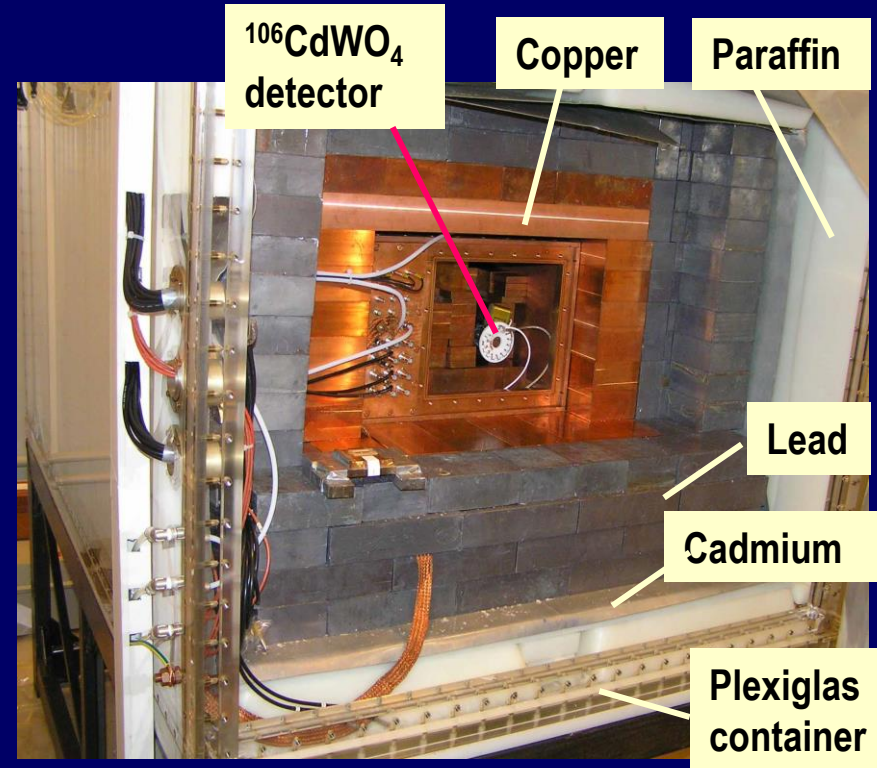
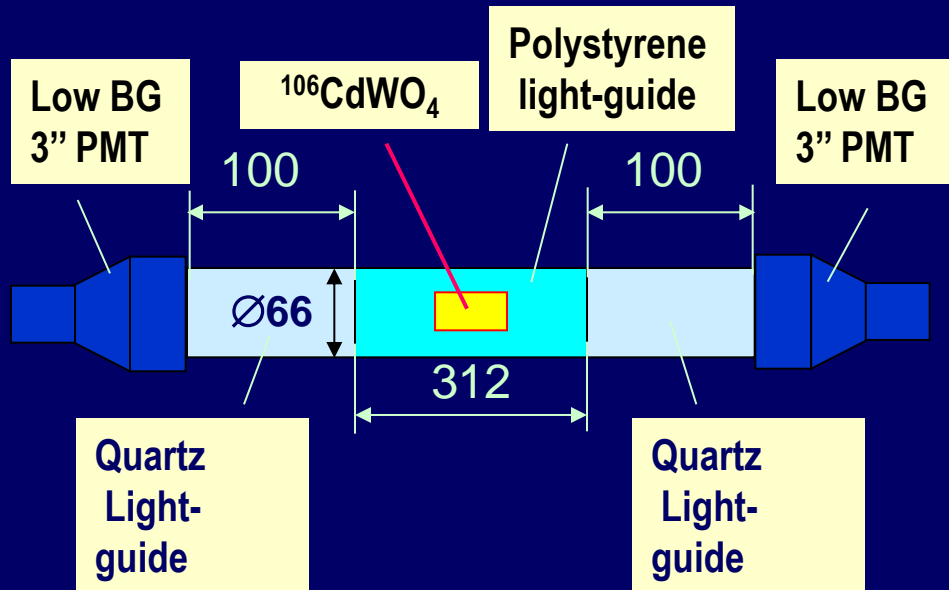
Excellent optical and scintillation properties thanks to special R&D to purify raw materials and Low-thermal-gradient Czochralski technique to grow the crystal

[1] P. Belli *et al.*, NIMA 615 (2010) 301

[2] A.S. Barabash *et al.*, JINST 6 (2011) P08011

The total losses of $^{106}\text{Cd} \approx 2\%$

Low background $^{106}\text{CdWO}_4$ detector in DAMA R&D at LNGS

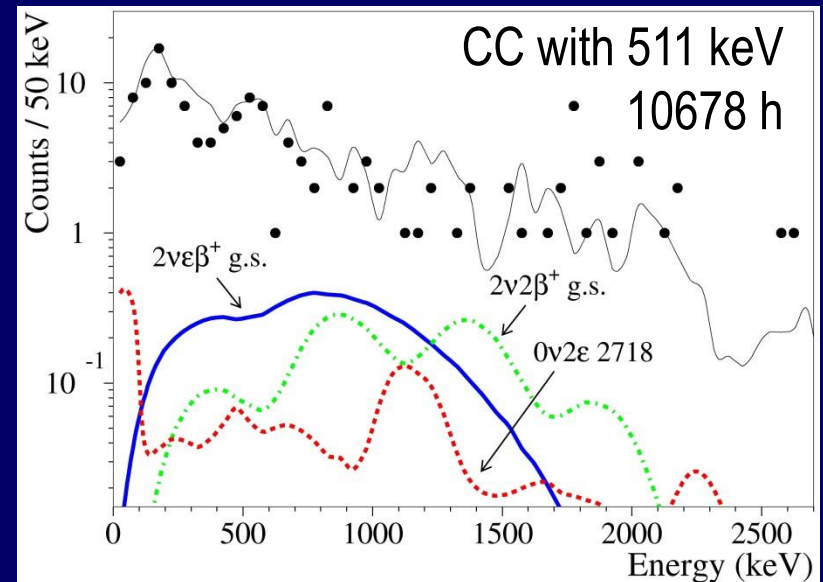


P. Belli *et al.*, PRC 85 (2012) 044610

$^{106}\text{CdWO}_4$ in the GeMulti setup with 4 HPGe

4 HPGe, $\sim 225 \text{ cm}^3$ each, in one cryostat

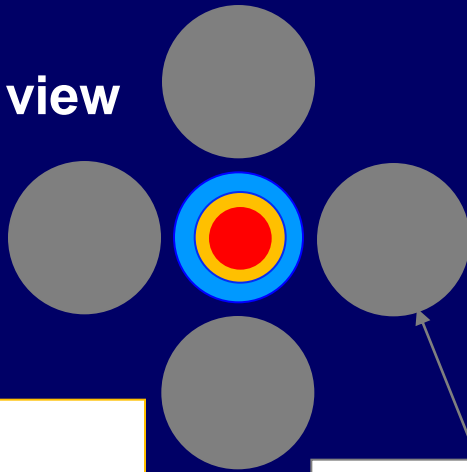
$^{106}\text{CdWO}_4$ in coincidence / anticoincidence with HPGe



Sensitivity to 2ε , $\varepsilon\beta^+$ and $2\beta^+$ decay of ^{106}Cd :

$$T_{1/2} \sim 10^{19} - 10^{21} \text{ yr}$$

bottom view



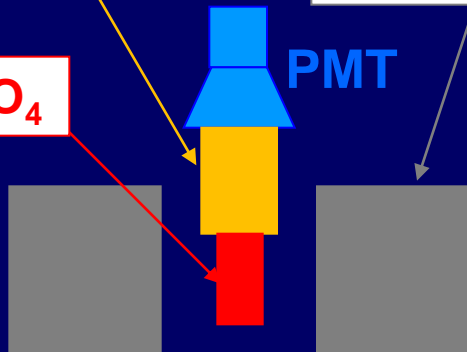
PbWO_4
(archeological lead)

HPGe 225 cm^3

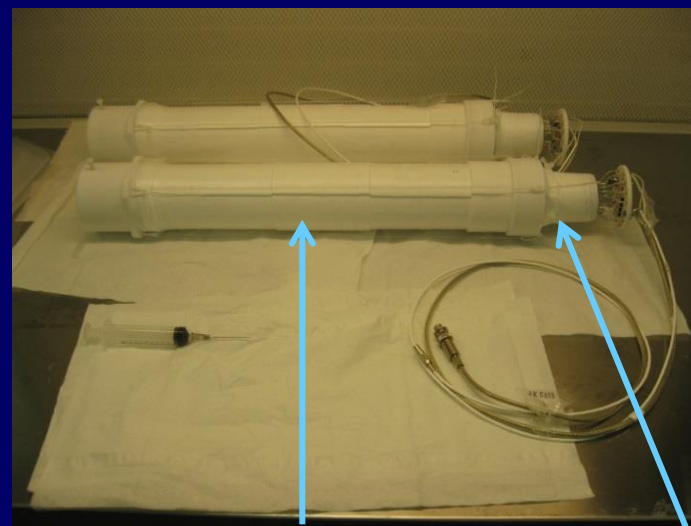
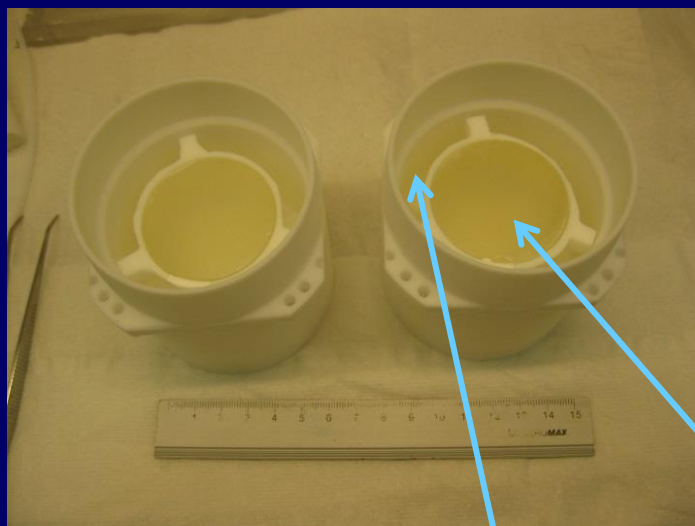
$^{106}\text{CdWO}_4$

PMT

side view



$^{116}\text{CdWO}_4$ detector

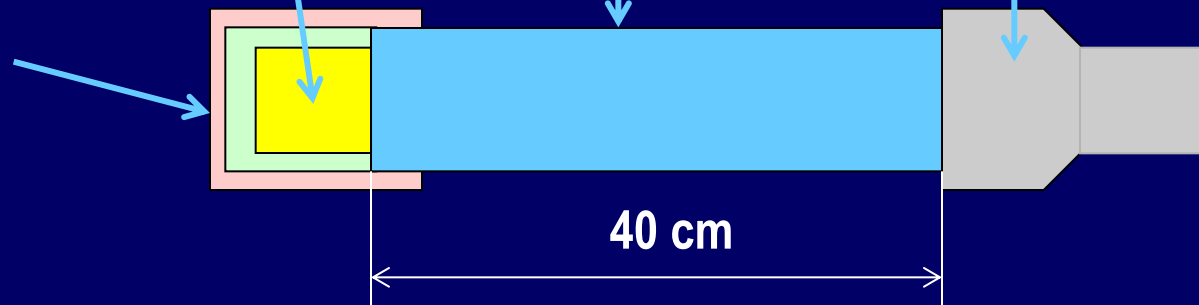


$^{116}\text{CdWO}_4$

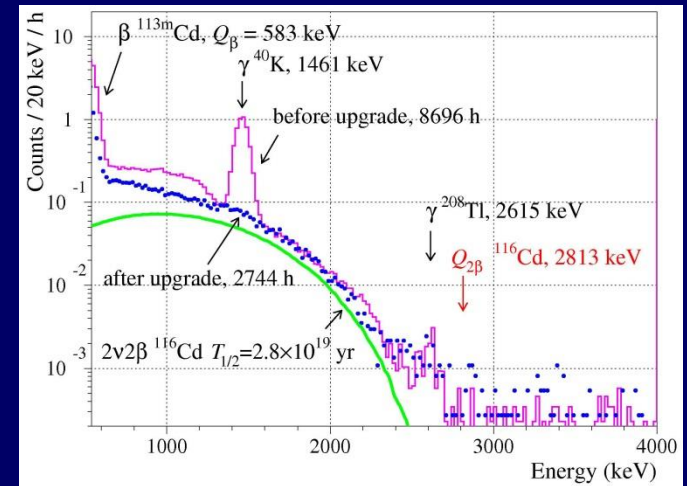
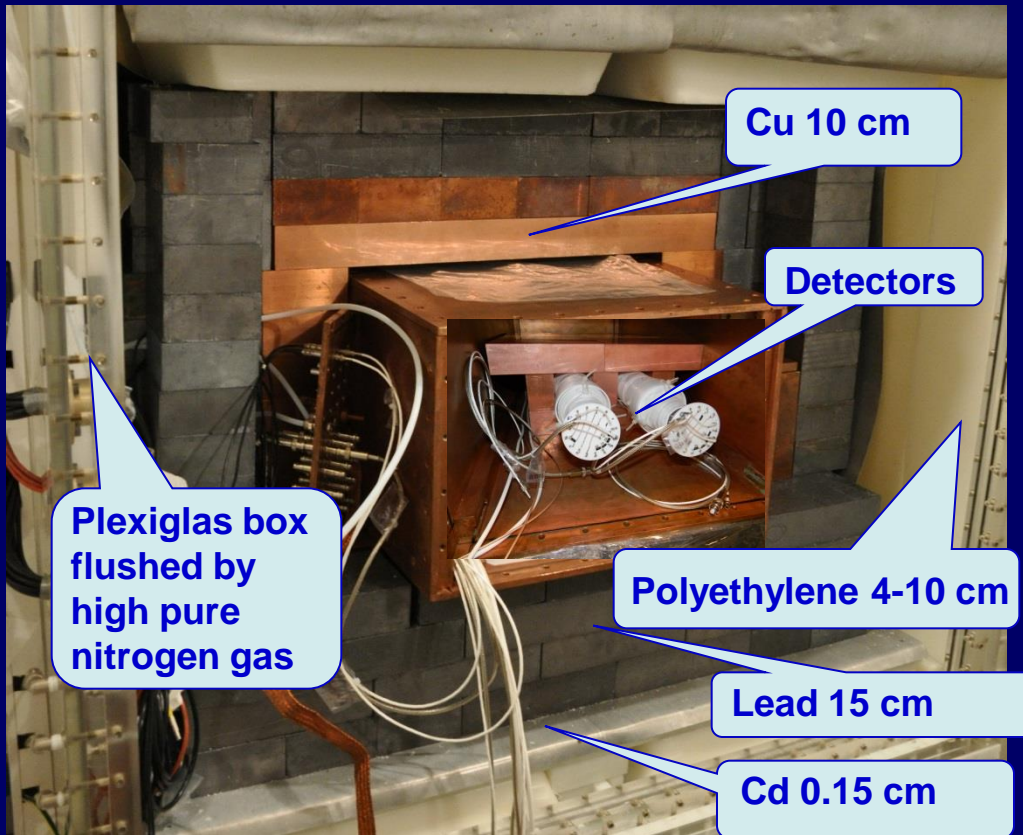
HP quartz
light-guide

PMT

Borexino liquid scintillator in
Teflon container



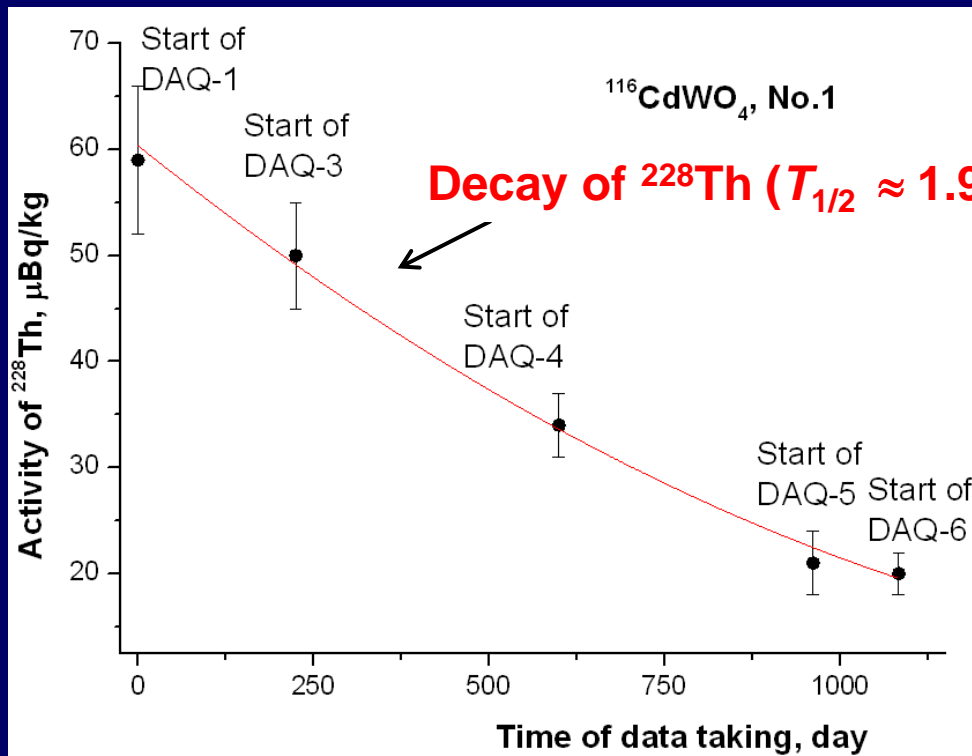
$^{116}\text{CdWO}_4$ set-up (DAMA R&D at LNGS)



^{228}Th and ^{226}Ra by time-amplitude analysis

D.V. Poda et al., EPJ WC 65 (2014) 01005.

Decay of ^{228}Th in $^{116}\text{CdWO}_4$



Activity of ^{228}Th (in $\mu\text{Bq/kg}$)
(reference date: May 2014)

No.1	20(2)
No.2	39(3)

- Contamination is mainly by thorium
- Radium is much lower

Radiopurity of $^{106,116}\text{CdWO}_4$ and CdWO_4

(mBq/kg) Ref data April 2013

Nuclide	$^{106}\text{CdWO}_4$ [1]	$^{116}\text{CdWO}_4$ [2]	CdWO_4 [3,4]
^{40}K	<1.4	<1	< (1.7 – 5)
$^{110\text{m}}\text{Ag}$	<0.06	= 0.12(4)	–
^{113}Cd	= 182(1)	= 100(10)	= 558(4)
$^{113\text{m}}\text{Cd}$	= 116 000(4000)	= 460(20)	< 3.4 – 150
^{232}Th	<0.07	<0.08	< 0.03
^{228}Th	= 0.042(4)	= 0.060(6)	< (0.003 – 0.014)
^{238}U	<0.6	<0.5	<1.3
^{226}Ra	= 0.012(3)	<0.005	< (0.007 – 0.02)
^{210}Po	<0.2	<0.5	< 0.06
Total α	= 2.1(2)	= 1.9(2) – 2.7(3)	= 0.26(4)

[1] P. Belli *et al.*, PRC 85 (2012) 044610

[3] F.A. Danevich *et al.*, Z. Phys. A 355 (1996) 433

[2] A. Barabash *et al.*, JINST 6 (2011) P08011

[4] P. Belli *et al.*, Phys. Rev. C 76 (2007) 064603

Possibility to improve the radiopurity of $^{116}\text{CdWO}_4$ by recrystallization

Activity of ^{228}Th :

10(2)

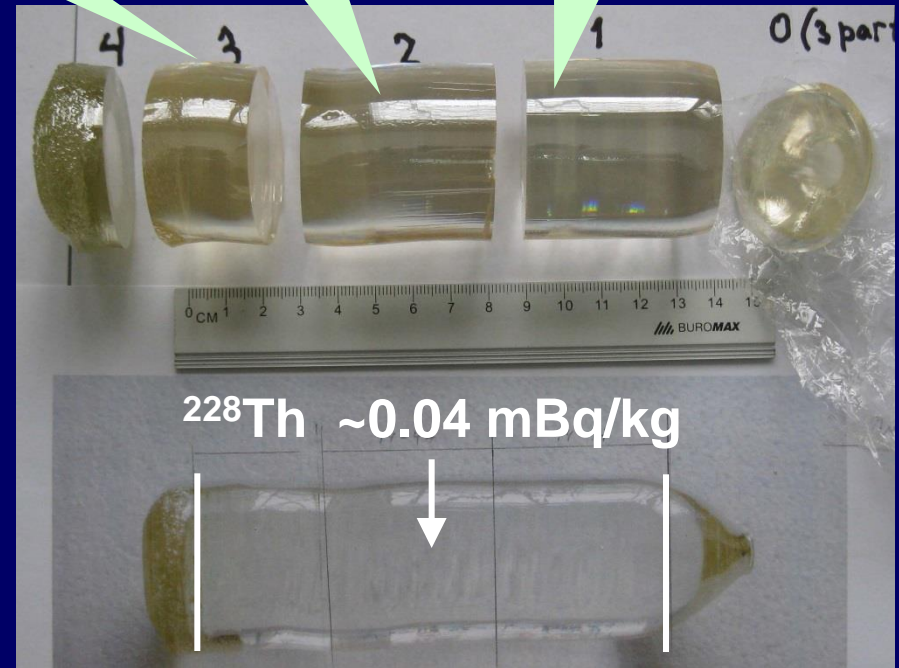
0.09(1)

0.04(1)

0.02(1)

May 2014

rest of the melt after the
crystal growth 279 g



Nuclide	Crystal	Rest of melt
^{40}K	<1	27(11)
^{226}Ra	<0.005	64(4)
^{228}Th	0.02 – 0.09	10(2)

^{228}Th in charge ~ 1.4 mBq/kg

We expect to reduce K, Th, U and Ra contamination by recrystallization

Thorium expected to be reduced by a factor $\sim 35 \Rightarrow 1 \mu\text{Bq/kg}$

conclusions

- **High quality CdWO_4 crystal scintillators were developed from enriched ^{106}Cd and ^{116}Cd with output 87%, irrecoverable losses 2%**
- **Radioactive contamination of the $^{106,116}\text{CdWO}_4$ crystals is on the level of 0.05 mBq/kg ^{228}Th , $\sim < 0.01\text{mBq/kg}$ of ^{226}Ra**
- **Strong segregation of thorium is observed in CdWO_4 crystals, substantial reduction of Th (~ 35 times) may be achieved by recrystallization**